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1 Achieving Ultrahigh Piezoelectricity in Organic-  
2 Inorganic Vacancy-Ordered Halide Double  
3 Perovskites for Mechanical Energy Harvesting

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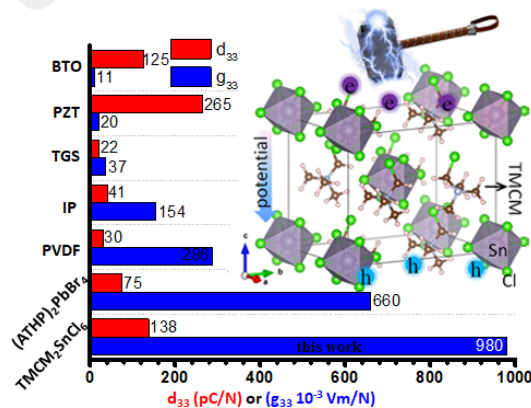
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## 1 ABSTRACT

2 Piezoelectric charge coefficient ( $d_{33}$ ) and piezoelectric voltage coefficient ( $g_{33}$ ) are the two most  
3 critical parameters that define output performance of piezoelectric nanogenerators (PNGs). Herein,  
4 we propose a vacancy-ordered double perovskite of  $\text{TMCM}_2\text{SnCl}_6$  (where TMCM is  
5 trimethylchloromethyl ammonium) with a large  $d_{33}$  of 137 pC/N and  $g_{33}$  of  $980 \times 10^{-3} \text{ V} \cdot \text{m/N}$ . The  
6 piezoelectric coefficients are considered from the halogen-bonding mediated synergistic  
7 movements of atomic displacement in inorganic  $[\text{SnCl}_6]^{2-}$  octahedrons, as well as the molecular  
8 rotation of organic  $\text{TMCM}^+$ , which is revealed by a combined density functional theory (DFT) and  
9 experimental study. The  $\text{TMCM}_2\text{SnCl}_6$  possesses a high saturated polarization ( $P_s$ ) of  $8.7 \mu\text{C}/\text{cm}^2$ ,  
10 a high Curie temperature ( $T_c$ ) of 365 K, and a low coercive field ( $E_c$ ) of 0.6 kV/cm. The output  
11 voltage ( $V_{oc}$ ) and current ( $I_{sc}$ ) of the PNGs are 81 V and  $2 \mu\text{A}$  at an applied mechanical excitation  
12 of (4.9 N, 40 Hz). We hope this work will provide guidance in energy harvesting by innovatively  
13 designing highly piezoelectric perovskites for the PNGs.

## 14 TOC GRAPHICS



15

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1 Piezoelectric nanogenerators (PNGs) have been emerging as a promising power source for self-  
2 powered electronics owing to their direct power conversion from mechanical to electrical energy.<sup>1-</sup>  
3 <sup>5</sup> To maximize the output power of PNGs, both the  $d_{33}$  and  $g_{33}$  of the piezoelectric host are  
4 important, which determines the output current ( $I_{sc}=(d_{33}\times\Delta F)/\Delta t$ , where  $\Delta F$  is the applied force  
5 and  $\Delta t$  is the time) and voltage ( $V_{oc}=g_{33}\times\Delta P\times L$ , where  $\Delta P$  is the applied pressure and  $L$  is the  
6 original film thickness), respectively.<sup>6-9</sup> In the past decade, a wide range of piezoelectric materials,  
7 targeting high  $g_{33}$  or  $d_{33}$ , have been synthesized for the efficient PNGs. For example, the organic  
8 polyvinylidene fluoride (PVDF) possesses a high  $g_{33}$  ( $\sim 286 \times 10^{-3} \text{ V}\cdot\text{m/N}$ ), leading to a high output  
9 piezoelectric voltage. Unfortunately, the resultant current is limited due to its relatively-low  $d_{33}$   
10 ( $\sim 30 \text{ pC/N}$ ).<sup>10-13</sup> Conversely, inorganic perovskite oxide ceramics, including  $\text{PbZr}_x\text{Ti}_{1-x}\text{O}_3$  (PZT)  
11 and  $\text{BaTiO}_3$  (BTO), exhibit a high  $d_{33}$  ( $>100 \text{ pC/N}$ ) but their  $g_{33}$  is rather low ( $\sim 20 \times 10^{-3} \text{ V}\cdot\text{m/N}$ ).<sup>14-</sup>  
12 <sup>18</sup> Considering the relation between  $d_{33}$  and  $g_{33}$  ( $g_{33}=d_{33}/(\epsilon_0\times\epsilon_r)$ ), where  $\epsilon_r$  is material relative  
13 permittivity, it is challenging to improve  $d_{33}$  while maintaining a high  $g_{33}$  because the increase of  
14  $d_{33}$  is usually associated with an even larger increase of  $\epsilon_r$ .<sup>19</sup>

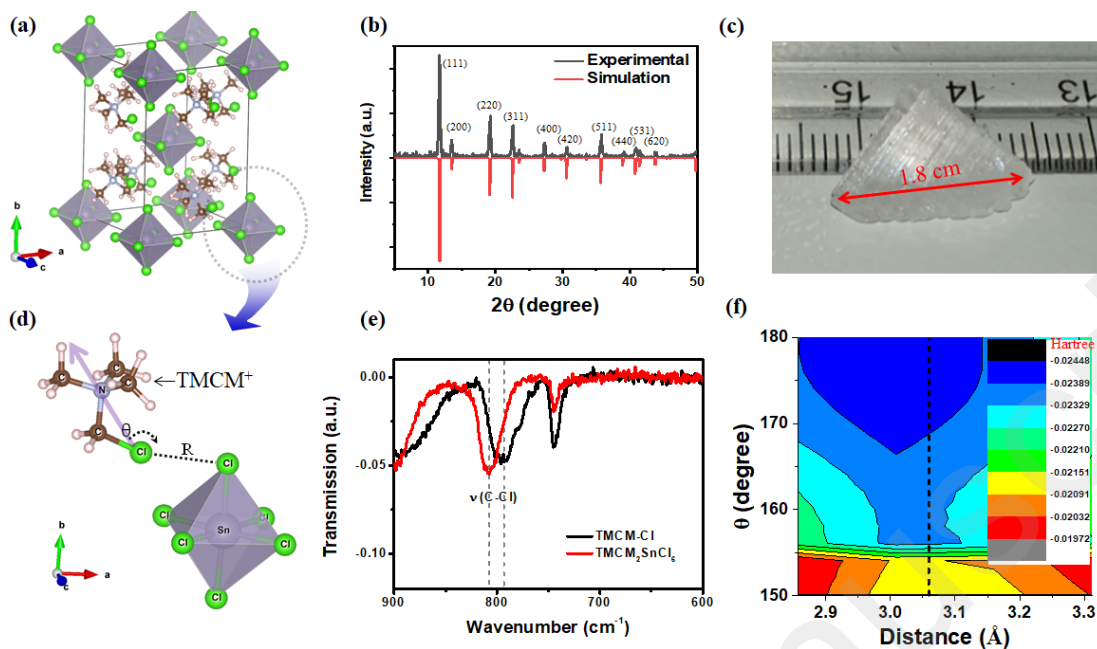
15 Organic-inorganic hybrid perovskites (OIHPs) combine the merits of both organic molecules  
16 and crystalline inorganic solids at a molecular level. The OIHPs were firstly introduced to solar  
17 cells (SCs) due to their excellent optical properties in 2009 and then adopted in piezoelectric  
18 research.<sup>20-25</sup> In 2015, the  $\text{MAPbI}_3$  was firstly employed in PNGs which yielded an output voltage  
19 and current density of 2.7 V and  $140 \text{ nA/cm}^2$  at a pressure of 0.5 MPa.<sup>26</sup> Subsequently, a series of  
20 perovskites, including  $\text{MASnBr}_3$ ,  $\text{FAPbBr}_2\text{I}$  and  $\text{Cl/Br-MAPbI}_3$ , have been developed for efficient  
21 PNGs via the piezoelectric composite films.<sup>27-31</sup> However, the aforementioned OIHPs evolved  
22 from the shadows of the SCs possess limited piezoelectric coefficients. Therefore, extensive efforts  
23 have been devoted to searching for highly piezoelectric OIHPs. Remarkably, one-dimensional

1 ABX<sub>3</sub>-type OIHPs with high d<sub>33</sub>, including the TMCM-MnCl<sub>3</sub> and A-site mixed  
2 (TMFM)<sub>x</sub>(TMCM)<sub>1-x</sub>CdCl<sub>3</sub>, were synthesized in 2017 and 2019, respectively.<sup>32, 33</sup> Two-  
3 dimensional (ATHP)<sub>2</sub>PbBr<sub>4</sub> OIHPs with large d<sub>33</sub> of 75 pC/N and giant g<sub>33</sub> of 660 ×10<sup>-3</sup> V•m/N  
4 were reported in 2019.<sup>34</sup> In 2020, the OIHPs were further developed and successfully employed in  
5 energy harvesting applications, including (FMTMA)PbCl<sub>2</sub>I, HETMA-CdCl<sub>3</sub>, (TMFM)FeBr<sub>4</sub> and  
6 [Ph<sub>3</sub>MeP]<sub>4</sub>[Ni(NCS)<sub>6</sub>].<sup>35-38</sup> Despite their large piezoelectric coefficients, the newly-developed  
7 OIHPs may suffer substantial difficulties in mass production and implementation. For example,  
8 the TMCM-MnCl<sub>3</sub> has intrinsic chemical instability as the bivalent Mn is prone to be oxidized to  
9 tetravalent, while the Cd (or Pb) in (TMFM)<sub>x</sub>(TMCM)<sub>1-x</sub>CdCl<sub>3</sub> (or (ATHP)<sub>2</sub>PbBr<sub>4</sub>) is not  
10 environmental-friendly. Unlike the ABX<sub>3</sub>-type OIHPs, the A<sub>2</sub>BX<sub>6</sub> OIHPs (such as K<sub>2</sub>PtCl<sub>6</sub>) are  
11 potentially more chemically stable because their B-site is already in the tetravalent state.<sup>39-41</sup>

12 Herein, we proposed an alternative crystal model, which was vacancy-ordered double OIHPs,  
13 for novel piezoelectric material design. The chemical formula was confirmed as TMCM<sub>2</sub>SnCl<sub>6</sub> via  
14 powder X-ray diffraction (XRD) and X-ray photoelectron spectroscopy (XPS). The large d<sub>33</sub> of  
15 137 pC/N and giant g<sub>33</sub> of 980 ×10<sup>-3</sup> V•m/N were experimentally measured for the prepared  
16 TMCM<sub>2</sub>SnCl<sub>6</sub> crystal, which was further confirmed by the DFT study and the outputs of PNGs.  
17 The PNGs were fabricated via sandwiching the TMCM<sub>2</sub>SnCl<sub>6</sub>@PDMS composite film into two  
18 copper electrodes, where PDMS represents Poly(dimethylsiloxane). A self-powered wireless  
19 communication node (SWCN) was successfully driven by the PNG. The detailed material  
20 synthesis and device fabrication processes are shown in the supporting information.

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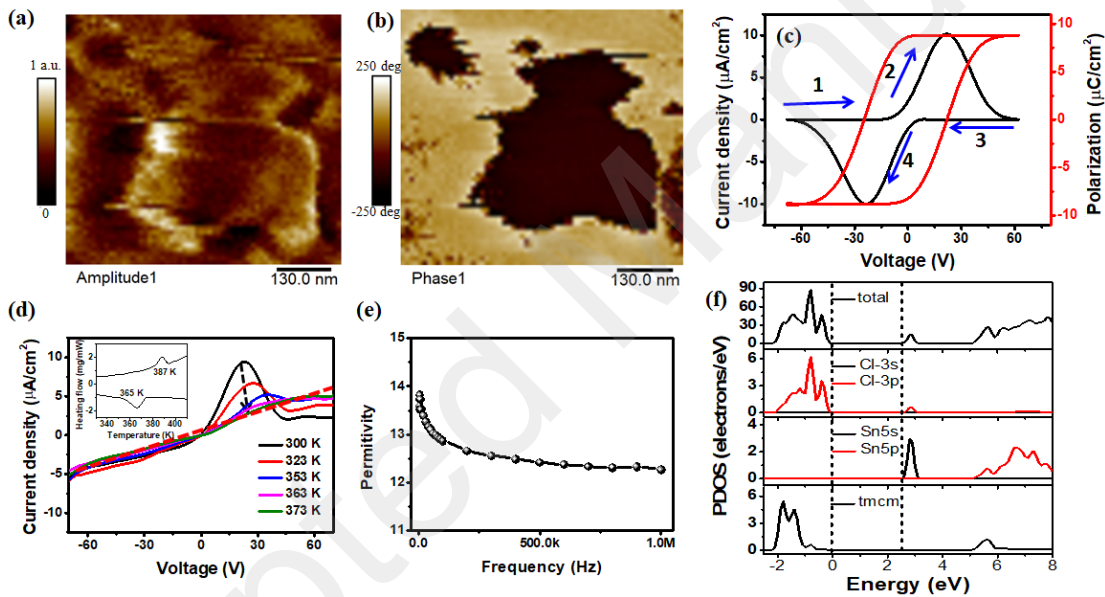
1  
 2 **Figure 1. Crystal structure of  $\text{TMCM}_2\text{SnCl}_6$  double perovskites:** (a) structural model; (b)  
 3 calculated and experimental XRD patterns; (c) digital photograph of single crystal; (d) halogen  
 4 bonding interaction. The arrow from Cl to N indicates the dipole direction of  $\text{TMCM}^+$ ; (e) FTIR  
 5 spectra of pure  $\text{TMCM-Cl}$  molecules and  $\text{TMCM}_2\text{SnCl}_6$ ; (f) calculated energy mapping of halogen  
 6 bond for the simplified model in Figure 1d.

7  
 8  $\text{TMCM}_2\text{SnCl}_6$  is a vacancy-ordered double perovskite because its 50%  $[\text{SnCl}_6]^{2-}$  octahedrons  
 9 are replaced by periodic vacancies, in which the asymmetric  $\text{TMCM}^+$  takes the A sites (**Figure**  
 10 **1a**). This is different from the reported vacancy-ordered double perovskites with spherical  
 11 symmetric cations like trimethylmethyl ammonium (TMA) or monovalent alkali metals (Na, K,  
 12 Cs...) taking the A sites.<sup>42-44</sup> **Figure 1b** shows the XRD measured from the  $\text{TMCM}_2\text{SnCl}_6$  single  
 13 crystals (black curve) in comparison to the simulated XRD (red curve) from the projected structure  
 14 in Figure 1a. The single crystal of  $\text{TMCM}_2\text{SnCl}_6$  with a length of 1.8 cm is shown in **Figure 1c**.

1 The agreement of XRD peaks confirms the crystal structure and its high phase purity. The  
2  $\text{TMCM}_2\text{SnCl}_6$  was aged to 6 months in an ambient atmosphere and the XRD was measured in  
3 **Figure S1a**. The reflection peaks of  $\text{TMCM}_2\text{SnCl}_6$  almost remained unchanged and no new  
4 reflection peaks appeared. These indicate the  $\text{TMCM}_2\text{SnCl}_6$  crystals have not degraded into other  
5 chemicals and exhibit good stability. The  $\text{TMCM}_2\text{SnCl}_6$  also exhibits good thermal stability, as  
6 indicated by a negligible change in the measured permittivity at 340K for one hour (**Figure S1d**).  
7 The chemical states of  $(\text{TMCM})_2\text{SnCl}_6$  were measured via XPS in **Figure S2**. The peaks at binding  
8 energy of 495.8 eV and 487.3 eV are assigned to  $\text{Sn}^{4+}$  and corresponding to its  $3d_{3/2}$  and  $3d_{5/2}$   
9 orbitals, respectively. The peaks at 200.6 eV and 198.8 eV are assigned to  $3p_{1/2}$  and  $3p_{3/2}$  orbitals  
10 of  $\text{Cl}^-$ . This is well consistent to the chemical states in  $[\text{SnCl}_6]^{2-}$  octahedrons, where Sn is in  
11 tetravalent state and Cl in minus state.

12 A halogen bond has been considered as an efficient driving force to achieve the well-defined  
13 structures during crystal growth for piezoelectric properties.<sup>45, 46</sup> In  $\text{TMCM}_2\text{SnCl}_6$ , there is a net  
14 attractive interaction between an electrophilic region associated with the Cl atom in  $\text{TMCM}^+$  and  
15 the nucleophilic  $\text{Cl}^-$  in the  $[\text{SnCl}_6]^{2-}$  octahedron (**Figure 1d**). The attractive nature of the halogen  
16 bond is determined by the bond distance (R, in the range of 2-6 Å) and the bond angle ( $\theta$ , in the  
17 range of 150-180°), especially when R is shorter than the sum of Van der Waals radii (~ 3.5 Å).  
18 The halogen bond in the  $\text{TMCM}_2\text{SnCl}_6$  was reflected by Fourier transform infrared spectroscopy  
19 (FTIR) in **Figure 1e**. The absorption peak at  $793\text{ cm}^{-1}$  (black line) is attributed to the stretching  
20 vibration of Cl-C bond of  $\text{TMCM}^+$ . For the  $\text{TMCM}_2\text{SnCl}_6$ , the absorption peak has a red-shift of  
21  $14\text{ cm}^{-1}$  to  $807\text{ cm}^{-1}$  (red line). The red shift should be attributed to the increased force constant  
22 when the electrophilic Cl in  $\text{TMCM}^+$  is attracted by the nucleophilic Cl in  $[\text{SnCl}_6]^{2-}$  octahedron  
23 according to harmonic motion for diatomic model.<sup>47, 48</sup> The halogen bond energy was calculated

1 by using DFT as a function of the bond distance (R) and the bond angle ( $\theta$ ) in **Figure 1f and**  
 2 **Figure S3**. With the decreasing of R, the calculated system energy of  $\text{TMCM}^+ \cdots [\text{SnCl}_6]^{2-}$   
 3 decreases as the increased Coulomb attraction, and followed by a quick increase afterwards due to  
 4 the dominating nuclear repulsion. The halogen bond energy is evaluated from the absolute value  
 5 of valley energy. The lowest energy of halogen bond is observed to be 0.021 Hartree at a  $\text{Cl} \cdots \text{Cl}$   
 6 bond distance of 3.06 Å for a certain  $\text{C}-\text{Cl} \cdots \text{Cl}$  bond angle ( $\theta = 150^\circ$ ). The bond energy increases  
 7 from 0.021 to 0.024 Hartree (dash line) as  $\theta$  increases from  $150^\circ$  to  $180^\circ$ .



8

9 **Figure 2. Piezoelectric and ferroelectric properties of  $\text{TMCM}_2\text{SnCl}_6$  double perovskite:**

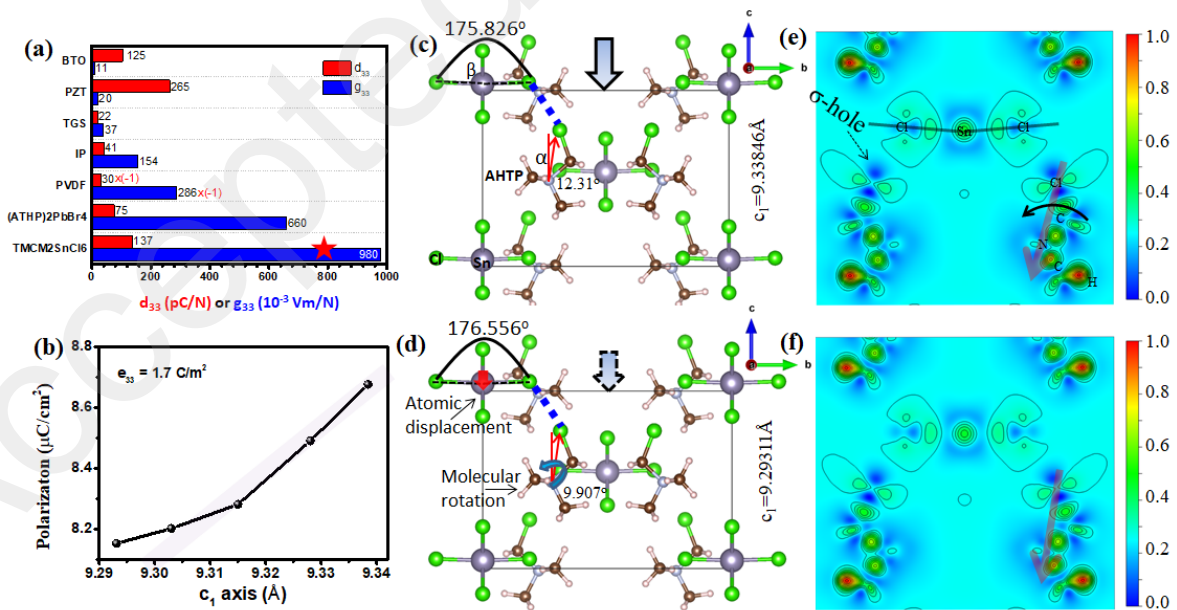
10 vertical PFM amplitude (a) and phase (b) images; (c) current density-voltage (J-V) curve and the  
 11 integrated polarization-voltage (P-V) hysteresis loop of the  $\text{TMCM}_2\text{SnCl}_6$  film at room  
 12 temperature ( $25^\circ\text{C}$ ) by using the double-wave method; (d) temperature-dependent J-V curves.  
 13 Voltage was swept from -70 to 70 V. The inset is the differential scanning calorimetry (DSC) curves



1 of  $\text{TMCM}_2\text{SnCl}_6$  crystals with the scanning rate of 10 K/min; (c) frequency-dependent permittivity  
2 tested from 0.01 to 1 MHz; (f) calculated density of states of  $\text{TMCM}_2\text{SnCl}_6$ .

3 The  $d_{33}$  of  $\text{TMCM}_2\text{SnCl}_6$  perovskites was measured to be 137 pC/N by using a quasi-static  
4 method in **Figure S4**. The ferroelectric domains were confirmed by a clear contrast in amplitude  
5 and phase images of vertical piezoelectric force microscopy (V-PFM) in **Figure 2a-b**. There was  
6 no crosstalk between the topography (**Figure S5b**) and the V-PFM phase. Meanwhile, the lateral  
7 PFM (L-PFM) was recorded in **Figure S5c-d** where the signal strength was found relatively weak,  
8 indicating that the spontaneous polarization was almost along the out-of-plane direction.<sup>33</sup> The P-  
9 V hysteresis loop was measured for  $\text{TMCM}_2\text{SnCl}_6$  films with the thickness of 400  $\mu\text{m}$  by using  
10 the double-wave method (DWM) in **Figure 2c**.<sup>49, 50</sup> The arrows in the J-V curve indicate the  
11 voltage sweep directions (-60 V  $\rightarrow$  0 V  $\rightarrow$  60 V  $\rightarrow$  0 V  $\rightarrow$  -60 V). The J-V curve exhibits two  
12 apparent current peaks at the ferroelectric switching voltage of  $\pm 23$  V. The  $E_c$  of  $\text{TMCM}_2\text{SnCl}_6$ ,  
13 expressed by the ratio of switching voltage to the film thickness, is 0.6 kV/cm. This value is  
14 comparable to that of conventional piezo-ceramics such as bulk BTO single crystals (1 kV/cm).<sup>51</sup>  
15 The  $P_s$  from the P-V hysteresis loop is up to 8.7  $\mu\text{C}/\text{cm}^2$ , close to the calculated value of  $8.4 \pm 0.3$   
16  $\mu\text{C}/\text{cm}^2$ . When the temperature is higher than  $T_c$ , the ferroelectric material will undergo a phase  
17 transition and lose its spontaneous polarization. The phase transition of  $\text{TMCM}_2\text{SnCl}_6$  was  
18 considered from orthorhombic to cubic, as indicated by some disappearing diffraction peaks at  
19 around 26.3°, 33.5° and 36.5° in temperature-dependent XRD (**Figure S1b**). The  $T_c$  of  
20  $\text{TMCM}_2\text{SnCl}_6$  was studied via monitoring the change of peak intensity at 23 V in the temperature-  
21 dependent J-V curves. As indicated in **Figure 2d**, the current peaks disappear when the  
22 temperature rises to 353-363 K. This is consistent to a pair of thermal anomalies at 365/387 K in  
23 the differential scanning calorimetry (DSC) curves (the inset of Figure 2d) and an increase at 350-

1 380 K in the temperature dependent permittivity (**Figure S1c**). The  $\text{TMCM}_2\text{SnCl}_6$  possesses a  $T_c$   
 2 above room temperature, making it suitable for PNG applications. The  $\epsilon_r$  of perovskite oxide  
 3 ceramics, such as PZT, is observed as thousands, which stems from the hopping polaron within  
 4 semiconducting grains and interfacial polarization at the insulating grain boundary.<sup>52</sup> The  
 5  $\text{TMCM}_2\text{SnCl}_6$  exhibits a relative low  $\epsilon_r$  of 14-12 with the frequency from 0.01 to 1 MHz in **Figure**  
 6 **2e**. This should be attributed to the insulating properties of  $\text{TMCM}_2\text{SnCl}_6$ : where the calculated  
 7 bandgap of above 3.2 eV and a narrow conduction band width are defined by the 5s orbits of Sn  
 8 and the 3p orbits of Cl, as indicated in **Figure 2f** and **Figure S6**. According to  $g_{33} = d_{33} / (\epsilon_0 \times \epsilon_r)$ ,  
 9 where  $\epsilon_0$  is the permittivity of free space and  $\epsilon_r = 16$ , the calculated  $g_{33}$  is  $980 \times 10^{-3} \text{ V} \cdot \text{m/N}$ . Notably,  
 10 a higher  $\epsilon_r$  was chosen to match the following PNG working frequency at 40 Hz. In comparison to  
 11 traditional metal oxides and other molecular materials listed in **Figure 3a** (or **Table S1**), the  
 12  $\text{TMCM}_2\text{SnCl}_6$  perovskites are more favorable for the PNG applications due to the both large  $d_{33}$   
 13 and giant  $g_{33}$ .



14

15 **Figure 3. Origin of piezoelectricity in the  $\text{TMCM}_2\text{SnCl}_6$  double perovskite:** (a) the  $d_{33}$  and  $g_{33}$

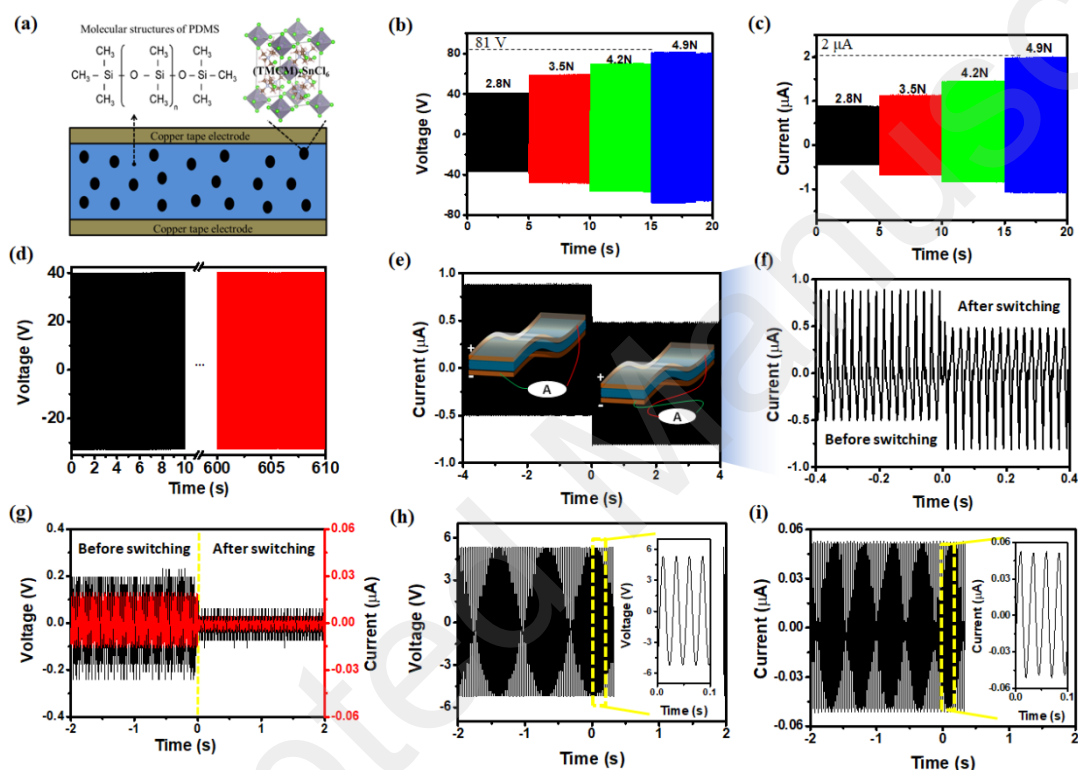
1 of TMCM<sub>2</sub>SnCl<sub>6</sub> compared with the reported inorganic and molecular materials, including BTO,  
2 PZT, Triglycine sulfate (TGS), Imidazolium perchlorate (IP), PVDF, and (ATHP)<sub>2</sub>PbBr<sub>4</sub>; (b) the  
3 calculated polarization as a function of  $c_1$  by using Berry phase method; (c) side views of the  
4 unstrained structure and (e) corresponding deformation charge density for  $c = 9.33864 \text{ \AA}$ ; (d) the  
5 strained structure and (f) deformation charge density for  $c = 9.29311 \text{ \AA}$ . The above deformation  
6 charge density is in the [001] planes including the Sn-Cl···Cl-C chain. The blue and red colors  
7 represent the electron loss and gain, respectively.

8  
9 To investigate the piezoelectric origin of TMCM<sub>2</sub>SnCl<sub>6</sub> double perovskites, the polarization  
10 ( $P_{\text{total}}$ ) was calculated by the Berry phase method as a function of  $c_1$ , where  $c_1$  is the lattice constant  
11 during the applied strain along the  $c$  axis (**Figure 3b**). The slope of the linear  $P_{\text{total}}$  vs.  $c_1$  curve  
12 yields the piezoelectric stress coefficient ( $e_{33}$ ).<sup>53, 54</sup> By fitting the  $P_{\text{total}}$  vs.  $c_1/c_0$  curve in Figure 3b,  
13  $e_{33}$  is evaluated as  $1.7 \text{ C/m}^2$ . According to the calculated elastic modulus tensor ( $c_{33} = 187.7 \text{ kBar}$ )  
14 in **Table S2**, the  $d_{33}$  is calculated as  $90 \text{ pC/N}$  by using the expression of  $d_{33} = e_{33}/c_{33}$ . The calculated  
15  $d_{33}$  is lower than the experimental value of  $137 \text{ pC/N}$  as the  $P_{\text{total}}$  contribution resulted from lateral  
16 strain was neglected.<sup>55</sup> **Figure 3c** and **3d** show the crystal structure (side view) of unstrained  
17 TMCM<sub>2</sub>SnCl<sub>6</sub> with a lattice constant  $c_1$  of  $9.33846 \text{ \AA}$  and that of strained TMCM<sub>2</sub>SnCl<sub>6</sub> with  $c_1$  of  
18  $9.29311 \text{ \AA}$  (corresponding to a strain of  $-0.4\%$ ), respectively. Under the applied strain, the angle  
19 ( $\alpha$ ) in the dipole direction of TMCM<sup>+</sup> with respect to  $c$  axis decreases from  $12.31^\circ$  to  $9.907^\circ$ .  
20 According to the equation of  $P_{\text{TMCM}} = \mu \times \cos(\alpha) / V_u$ , where  $\mu$  is the dipole moment of TMCM<sup>+</sup> ( $4.65$   
21 Debye) and  $V_u$  is the cell volume, the polarization from the TMCM<sup>+</sup> increases when  $\alpha$  decreases.  
22 The deformation charge density profiles of the unstrained and strained crystal were shown in  
23 **Figure 3e** and **3f**, respectively. The Cl atom in TMCM<sup>+</sup> is involved in the formation of the covalent

1 bond. In addition, there is a region of lower electron density ( $\sigma$ -hole) where the potential is positive,  
2 which generates a cap of depleted electron density on the elongation of the covalent bond. Thus, a  
3 halogen-bonding interaction will be formed between the Cl of  $\text{TMCM}^+$  and the electron-rich Cl  
4 atoms in  $[\text{SnCl}_6]^{2-}$ . Therefore, the rotation of  $\text{TMCM}^+$  can trigger the atomic displacement in  
5  $[\text{SnCl}_6]^{2-}$  octahedrons simultaneously, indicated by the increased angle ( $\beta$ ) of Cl-Sn-Cl in  
6  $[\text{SnCl}_6]^{2-}$  octahedrons from  $175.6^\circ$  to  $176.5^\circ$ . The polarization from  $[\text{SnCl}_6]^{2-}$  is calculated by the  
7 equation of  $P_{\text{Sn}} = P_{\text{total}} - P_{\text{TMCM}}$ . Via fitting the decomposed polarization  $P_{\text{TMCM}}$  and  $P_{\text{Sn}}$  in **Figure S7**,  
8 the  $e_{33}$  of  $[\text{SnCl}_6]^{2-}$  and  $\text{TMCM}^+$  is  $1.5 \text{ C/m}^2$  and  $0.2 \text{ C/m}^2$ , respectively. It is clear that the  $[\text{SnCl}_6]^{2-}$   
9 contributes nearly all of the piezoelectricity. After replacing the asymmetric  $\text{TMCM}^+$  by spherical  
10 symmetric  $\text{TMA}^+$ , the  $[\text{SnCl}_6]^{2-}$  in  $\text{TMA}_2\text{SnCl}_6$  is not distorted (**Figure S8**). The polarization from  
11  $\text{TMA}_2\text{SnCl}_6$  is 0 and not changed under an applied strain, indicated the loss of piezoelectricity.  
12 Unlike the traditional perovskite ceramics where the piezoelectricity is primarily arising from the  
13 atomic displacement (**Figure S9**), (i) the organic  $\text{TMCM}^+$  in  $\text{TMCM}_2\text{SnCl}_6$  can break its crystal  
14 symmetry and (ii) the piezoelectricity in  $\text{TMCM}_2\text{SnCl}_6$  is from the halogen bond-mediated  
15 synergistic movement of atomic displacement in inorganic  $[\text{SnCl}_6]^{2-}$  octahedrons and the  
16 molecular rotation of  $\text{TMCM}^+$ .

17 The piezoelectricity in the  $\text{TMCM}_2\text{SnCl}_6$  was further demonstrated by the outputs of fabricated  
18 PNGs. The precursor of piezoelectric composite films was prepared via dispersing 10 wt. %  
19  $\text{TMCM}_2\text{SnCl}_6$  nanoparticles (NPs) uniformly in the PDMS by mechanical stirring (**Figure S10**).  
20 The size of the NPs was in the range of 200-1000 nm, as shown in the SEM and elemental mapping  
21 (Figure S11b-c). The films were grown via drop-casting the above precursor solution on a glass  
22 substrate (Figure S11a). After the drop-casting, NPs with a larger size tended to precipitate to the  
23 bottom of central area rather than diffuse to the marginal area (**Figure S11a, d-f**). The fraction of

1 TMCM<sub>2</sub>SnCl<sub>6</sub> NPs in the composite film for the PNG integration was 18 wt. % measured in the  
 2 dotted circle in Figure S11d by the SEM-EDS. The film thickness at the central area was 300 μm.  
 3 Finally, the TMCM<sub>2</sub>SnCl<sub>6</sub>@PDMS composite film was sandwiched between two copper  
 4 electrodes for PNG fabrication (**Figure 4a** and **Figure S12**) (see the experimental details in the  
 5 supporting information).



6  
 7 **Figure 4. The piezoelectric output of single PNG:** (a) device structure of PNGs via inserting  
 8 TMCM<sub>2</sub>SnCl<sub>6</sub>@PDMS composite film between copper electrodes; (b)  $V_{oc}$  and (c)  $I_{sc}$  of PNGs with  
 9 the force from 2.8 to 4.9 N (40 Hz); (d) the device reliability tested for 10 mins; (e)  $I_{sc}$  and (f)  
 10 enlarged view of the PNGs under forward and reverse electrical connections. The inset is schematic  
 11 diagrams of the PNGs under forward and reverse connections; (g) polarity switching test of  $V_{oc}$   
 12 (black curve) and  $I_{sc}$  (red curve) of pure PDMS based devices; (h)  $V_{oc}$  and (k)  $I_{sc}$  of PNGs based  
 13 on TMCM<sub>2</sub>SnCl<sub>6</sub>@PDMS film (without poling). For the Figures 4d-i, the applied mechanical

1 excitation is (2.8 N, 40 Hz).

2 The outputs of the  $\text{TMCM}_2\text{SnCl}_6$ @PDMS based PNGs was measured by mounting the device  
3 on a shaker in a vertical compressing and releasing process (**Figure S13**). For the PNGs with 18  
4 wt. %  $\text{TMCM}_2\text{SnCl}_6$  in the composite film, the output  $V_{oc}$  and  $I_{sc}$  of 40 V and 0.9  $\mu\text{A}$  was achieved  
5 under the force of 2.8 N at a frequency of 40 Hz. The device output is reproducible as indicated  
6 by the statistical analysis of  $V_{oc}$  in **Figure S14**. With the applied force increased from 2.8 to 4.9  
7 N, the piezoelectric outputs ( $V_{oc}$  and  $I_{sc}$ ) of the PNGs increase linearly in **Figure 4b-c**. The  $V_{oc}$   
8 and  $I_{sc}$  reach 81 V and 2  $\mu\text{A}$ , respectively, at an applied force of 4.9 N. The frequency-dependent  
9  $V_{oc}$  and  $I_{sc}$  are shown in **Figure S15** as the force keeps 2.8N. Meanwhile, the device outputs were  
10 also recorded at low frequency (3 Hz) via a cylindrical hammer mounted on a linear motor and a  
11 metallic stopper. The  $V_{oc}$  and current density ( $J_{sc}$ ) were measured as 11.6 V and 1.17  $\mu\text{A}/\text{cm}^2$ ,  
12 respectively, at an applied pressure of 0.032 MPa (**Figure S16**). The concentration of  
13  $\text{TMCM}_2\text{SnCl}_6$  on the electricity generation capacity of the PNGs was optimized in **Figure S18**.  
14 The output voltage and current gradually increase to 72 V and 1.3  $\mu\text{A}$  with the increasing mass  
15 fraction of  $\text{TMCM}_2\text{SnCl}_6$  NPs up to 29 wt. %, under an applied force of 2.8 N. In comparison with  
16 traditional inorganic perovskite ceramics in **Table S3**, the  $\text{TMCM}_2\text{SnCl}_6$  were prepared at room  
17 temperature and the PNGs exhibit the competitive piezoelectric outputs. The reliability of the  
18 PNGs was explored by continually testing the device for 10 minutes and the amplitude of the  
19 output voltage shows a negligible degradation (**Figure 4d** and Video S1), confirming its excellent  
20 mechanical durability. Finally, the PNGs were used as an integrated power source to meet up the  
21 energy demand of practical electrical loads via harvesting energy from local environment, such as  
22 SWCN (see the details in **Figure S17** and Video S2).

1 To demonstrate the generated outputs were originating from the piezoelectric phenomenon, the  
2 following experiments were conducted: (i) the measured current output of the PNGs switched its  
3 polarity during the electrical connection reversal (**Figure 4e-f** and Video S3), indicating that the  
4 outputs were produced from the PNGs; (ii) For comparison, the output  $V_{oc}$  and  $I_{sc}$  of PNGs made  
5 from pure PDMS were merely 0.2 V and 0.02  $\mu$ A and lacking of the polarity switching property in  
6 **Figure 4g**. The  $V_{oc}$  is enhanced with the loading  $\text{TMCM}_2\text{SnCl}_6$  NPs in the composite film as shown  
7 in **Figure S18**. These verify the claim that - the high performance of PNGs is determined by the  
8 incorporation of  $\text{TMCM}_2\text{SnCl}_6$ ; (iii) the effect of electrical poling on the PNG outputs was  
9 investigated (see the poling procedures in the experimental section). Compared with the outputs  
10 of unpoling PNGs (**Figure 4h-k**), the PNGs after poling at 4.5 kV (Figure 4b-c) exhibited drastic  
11 increase in  $V_{oc}$  (from 5.4 to 40 V) and  $I_{sc}$  (from 0.05 to 0.9  $\mu$ A) under the same measurement  
12 conditions (2.8 N, 40 Hz). A detailed comparison of output voltage of PNGs measured under  
13 various applied poling fields is shown in **Figure S19**.

14 In summary, a novel vacancy-ordered double perovskite of  $\text{TMCM}_2\text{SnCl}_6$  with a large  $d_{33}$  of  
15 137 pC/N and giant  $g_{33}$  of  $980 \times 10^{-3} \text{ V}\cdot\text{m/N}$  was proposed for the energy harvesting applications.  
16 The  $\text{TMCM}_2\text{SnCl}_6$  perovskite possesses a high  $P_s$  of 8.7  $\mu\text{C}/\text{cm}^2$ , a high  $T_c$  of 365 K, and a low  $E_c$   
17 of 0.6 kV/cm. The ultrahigh piezoelectricity in  $\text{TMCM}_2\text{SnCl}_6$  is attributed to the halogen bonding-  
18 mediated synergistic movements of atomic displacement in the inorganic  $[\text{SnCl}_6]^{2-}$  octahedron and  
19 the dipole alignment of organic  $\text{TMCM}^+$ . The piezoelectric coefficients from the  $\text{TMCM}_2\text{SnCl}_6$   
20 was firstly experimentally measured, then verified by DFT study, and finally confirmed by the  
21 piezoelectric output of the PNGs. The piezoelectric composite films were prepared by dispersing  
22  $\text{TMCM}_2\text{SnCl}_6$  NPs into the PDMS matrix. The peak values of  $V_{oc}$  and  $I_{sc}$  of the fabricated PNGs  
23 reached up to 81 V and 2  $\mu$ A at an applied mechanical excitation of (4.9 N, 40 Hz). The SWCN

1 was successfully driven by the PNG. This work not only opens up a door for designing novel  
2 piezoelectric OIHPs with ultrahigh  $d_{33}$  and  $g_{33}$ , but also paves the way for its real applications in  
3 self-powered electronics.

#### 4 **Supporting Information**

5 The Supporting Information is available free of charge at <https://pubs.acs.org/doi/...>  
6 Experimental methods, computational details, XRD, XPS,  $d_{33}$ , and calculated band structure of  
7  $\text{TMCM}_2\text{SnCl}_6$  crystals, digital photograph, SEM images and piezoelectric potential simulation of  
8  $\text{TMCM}_2\text{SnCl}_6$ @PDMS composite films, digital photos, frequency-dependent outputs of single  
9 PNG, real application of PNG for SWCN, output performance comparison of PNGs based on  
10 PDMS matrix, table for calculated elastic modulus, and Videos (MP4) for device stability, output  
11 switching, and SWCN.

#### 12 **Notes**

13 The authors declare no competing financial interest.

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