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A COMPREHENSIVE SPECTRAL ROTATIONAL ANALYSIS OF THE INTERSTELLAR METHYL ISOCYANATE, CH₃NCO

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Abstract

Methyl isocyanate (CH₃NCO) is a recently identified interstellar molecule giving rise to many detected lines. Interestingly, its delayed identification was not due to weak lines but due to a very complex rotational spectrum. To date, the only published laboratory transitions for this molecule are those between rotational energy levels with $K \leq 3$. In the present work, Stark modulation spectroscopy was used to record the room temperature rotational spectrum of CH₃NCO in the spectral region from 32 to 90 GHz. Observation of characteristic Stark effects, measured at specific low-voltage modulation conditions, and ¹⁴N nuclear quadrupole hyperfine structure allowed unambiguous assignment of rotational transitions up to K = 10. These newly assigned transitions were subsequently followed up to 364 GHz with the aid of Loomis-Wood- type displays. Since there are no reports on astrophysical detection of ¹³C isotopic species, first laboratory measurements between 50 and 300 GHz have been also performed for CH₃N¹³CO and ¹³CH₃NCO isotopologs. A comprehensive spectral analysis undertaken in this work made possible to extend the knowledge of the rotational spectrum of CH₃NCO to more than 2500 new transitions. Furthermore, more than 1200 lines were identified and analyzed for each of the isotopologs. The extensive line lists and sets of molecular parameters reported in this work provide the basis for further astrophysical searches of CH₃NCO.

Keywords: catalogs — ISM: molecules — molecular data — techniques: spectroscopic

1. INTRODUCTION

The reported detection of methyl isocyanate (CH₃NCO) on the frozen surface of comet 67P/Churyumov-Gerasimenko (Goesmann et al. 2015) provided the impulse for its detection in the interstellar medium (ISM). Even though that cometary identification is nowadays called into question (Altwegg et al. 2017), CH₃NCO is now considered to be a well established interstellar molecule with the first reported detections toward Sgr B2(N) (Halfen et al. 2015) and Orion KL (Cernicharo et al. 2016) molecular clouds. Later on, Belloche et al. (2017) found some additional transitions in hot molecular core Sgr B2(N2). Recently, CH₃NCO has also been observed in the low-mass solar-type protostar IRAS 16293-2422 (Ligterink et al. 2017; Martin-Domenech et al. 2017). Relying on these detections, several laboratory experiments have been undertaken in order to unveil the astrochemical origin (Ligterink et al. 2017, 2018) and survival probability of CH₃NCO in interstellar and cometary ices (Maté et al. 2018). Finally, CH₃NCO has also been incorporated into astrochemical models (Belloche et al. 2017; Martin-Domenech et al. 2017; Majumdar et al. 2018; Quénard et al. 2018).

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Inspection of the current lists of detected transitions reveals that the most significant number of CH₃NCO lines has been found in the millimeter wave survey of Orion KL (Cernicharo et al. 2016). However, all of those 399 lines belong to $K \le 3$ transitions, simply because confident assignment of K > 3 transitions was not available at the time. Furthermore, astrophysical detection of ¹³C isotopic species could not be attempted, also for lack of suitable laboratory data. The identification of ¹³C isotopologs of detected molecules is of special importance in astrophysics since it provides information on abundance ratio of ¹²C over ¹³C isotopes in the ISM. The abundance of CH₃NCO in Orion KL has been found to be only a factor of ten below those of HNCO and CH₃CN (Cernicharo et al. 2016) which makes this molecule a significant potential contributor to so-called U- lines. The large number of 8000 U-lines out of 16000 lines detected in the line survey of Orion KL (Tercero et al. 2010), was reduced on successive investigations, yet CH₃NCO accounted for the strongest outstanding U-lines at the time of its detection (Cernicharo et al. 2016). Hence, K > 3 transitions of the parent species and transitions of the ¹³C isotopic species are also expected to be good candidates for future detections. The fact that there were no attempts to detect these lines is due to the lack of suitable laboratory data arising largely from the difficulties in dealing with the rotational spectrum of the CH₃NCO molecule.

The pioneer microwave measurements performed by Curl et al. (1963); Lett & Flygare (1967) and later by Koput (1984, 1986, 1988) have demonstrated that CH₃NCO displays a spectrum with practically irregular distribution of *a*-type *R*-branch transitions of comparable intensities. This spectral complexity has been attributed to the presence of two large amplitude motions in the molecule: the internal rotation of the methyl group with a very low barrier ($V_3 = 21 \text{ cm}^{-1}$) and a low- frequency CNC bending motion (v_b). A detailed analysis carried out by Koput (1986) made it possible to unambiguously assign the rotational transitions in numerous low-lying internal rotation states (described by the quantum number *m*) in the ground state ($v_b = 0$) and two lowest excited states of the CNC bending mode ($v_b = 1, 2$). The quantum number assignment was based on characteristic line patterns observed in the Stark modulation spectrum. That work served as an excellent starting point for the sequence of laboratory studies (Kisiel et al. 2010, 2015; Cernicharo et al. 2016) in which the original assignment made by Koput ($J'' \le 3, K \le 3$) was extended up to 364 GHz (J'' = 41). This resulted in a list of 1269 confidently assigned laboratory lines and detection of CH₃NCO towards Orion KL (Cernicharo et al. 2016).

The present work has been conducted in two different directions: on the one hand, to extend the knowledge of the laboratory rotational spectrum of parent CH_3NCO , and on the other hand, to provide first spectroscopic information on the $CH_3N^{13}CO$ and $^{13}CH_3NCO$ isotopic species. A combination of microwave Stark-modulation and frequency-modulation millimeter wave spectroscopy was used for this purpose. The observation of characteristic Stark effects measured at specific low-voltage conditions, of specific nuclear quadrupole hyperfine structure and the use of a graphical Loomis-Wood-type plot approach was fundamental for assignment of rotational transitions. An extensive set of new laboratory lines for the parent species and the two ^{13}C isotopic species are provided in this work. This should facilitate further identifications of CH_3NCO in space.

2. EXPERIMENTAL DETAILS

2.1. Chemical synthesis

The parent species of CH_3NCO has been prepared as previously reported (Maté et al. 2017). The syntheses of Han et al. (1989) have been modified to prepare $CH_3N^{13}CO$ and $^{13}CH_3NCO$.

CH₃N¹³CO: Sodium azide (1.63 g, 25 mmol, 2 equiv.) and dry diethylene glycol dibutyl ether (15 ml) were introduced into a three-necked flask equipped with a nitrogen inlet, a reflux condenser, a magnetic stirring bar and a dropping funnel. The suspension was heated at 80 – 95 °C and acetyl-I-¹³C chloride (1g, 13 mmol, 1 equiv.) was added dropwise. The suspension was stirred at this temperature for 1 h. After cooling to room temperature, the flask was connected to a vacuum line (0.1 mbar) equipped with two traps. The first one was immersed in a bath cooled at -30° C to remove high boiling compounds. The low boiling CH₃N¹³CO was selectively condensed in the second trap immersed in a liquid nitrogen bath. Yield : 392 mg, 52%. ¹H NMR (CDCl₃, 400 MHz) δ 3.02 (d, ³J_{CH} = 4.9 Hz, CH₃). ¹³C NMR (CDCl₃, 100 MHz) δ 28.1 (d, ²J_{CC} = 5.1 Hz, CH₃), 121.6 (s brd, NCO).

¹³CH₃NCO: Silver cyanate (2g, 13.3 mmol), dry diethyleneglycol dibutyl ether (3 mL) and iodomethane-¹³C (1g, 7.0 mmol) were mixed together in a cell equipped with a magnetic stirring bar and a stopcock. The cell was immersed in a liquid nitrogen bath and degassed. The mixture was heated under stirring at 90°C for 20 h. The cell was then connected to a vacuum line (0.1 mbar) equipped with two traps. Following the same procedure as mentioned above, low boiling ¹³CH₃NCO was selectively condensed in the second trap immersed in a liquid nitrogen bath. Yield: 243 mg (60%). ¹H NMR (CDCl₃, 400 MHz) δ 3.01 (d, ¹J_{CH} = 143.1 Hz, CH₃). ¹³C NMR (CDCl₃, 100 MHz) δ 28.1 (d, ¹J_{CH} = 143.1 Hz, ¹³CH₃), 121.4 (s brd, NCO).

CH₃NCO

2.2. Rotational spectra

A Stark modulation spectrometer, whose design has been described previously (Kolesniková et al. 2017), was used to acquire the Stark spectra of the parent and ¹³C isotopic species between 32 and 90 GHz. To cover this region, an Agilent synthesizer (≤ 20 GHz) was used in combination with an additional frequency doubler, tripler or quadrupler as a frequency source. The absorption lines were modulated with a 33.3 kHz zero-based square-wave voltage and phase sensitive detection by means of a lock-in amplifier was employed. Modulation voltages of 10 and 150 V, corresponding to electric field strengths of 21 and 319 V cm⁻¹, were applied.

Frequency modulation spectra of the ¹³C isotopic species above 50 GHz were recorded using the millimeter and submillimeter wave spectrometer in Valladolid, which has been described in Daly et al. (2014). Agilent synthesizer and a set of amplifiermultiplier chains WR15.0, WR10.0, WR6.5, and WR9.0 (VDI, Inc) in combination with additional frequency doubler and tripler (WR4.3, WR2.8, VDI, Inc.) were used to reach frequencies up to 300 GHz. The synthesizer frequency was modulated at f = 10.2 kHz with modulation depth ranging from 30 to 40 kHz. A detection system composed of either Schottky diodes or broadband quasioptical detector (VDI, Inc) was completed by a 2*f* demodulation using a phase sensitive lock-in amplifier. All spectra were recorded in 1 GHz sections in both directions (a single acquisition cycle) and averaged. The archival data in the 117-364 GHz frequency region (Cernicharo et al. 2016) recorded with the Fast Scanning Submillimeter Spectroscopic Technique (FASSST) (Petkie et al. 1997; Medvedev et al. 2004) has also been used for the parent species, and for some highest frequency lines of the ¹³C species visible in natural abundance. The spectra from the various spectrometers were combined into a single spectrum and further processed using the AABS package (Kisiel et al. 2005, 2012). The uncertainty of center frequency measurement on the recorded second derivative line shapes was estimated to be better than 50 kHz. In the experiments, the samples of CH₃NCO were kept at room temperature, typically at a pressure of 20 μ bar (15 mTorr).

3. ANALYSIS AND RESULTS

3.1. Parent species

Considerable progress in the assignment of higher K transitions was made upon measurement of the Stark spectrum of CH₃NCO at the low modulation voltage of 10 V. We were able to build upon the classification of Stark patterns made by Koput (1984). The frequency spread of the expected Stark pattern for this near-prolate molecule is expected to be directly proportional to K. At low modulation voltage, K = 0 transitions disappear, while the target K > 3 transitions should give rise to Stark patterns with the broadest frequency spans and in the form of equidistant combs of resolved Stark components. Under the phase detection used in the Stark-modulation method these have the form of negative patterns of Stark lines on both sides of the field free line, which is displayed in the positive direction. A portion of the spectrum in the region of the $J = 5 \leftarrow 4$ rotational transition is shown in Figure 1. This part of the spectrum is removed from the clutter $J = 5 \leftarrow 4$ lines for different internal rotation m states and different CCN bending mode v_b states centered on 43600 MHz. Inspection of isolated, already assigned lines in this spectrum showed relative invariance of the Stark patterns on m and v_b , as visible in Figure 1. Comparison of the visible line patterns allowed confident identification of the line at 42731 MHz in Figure 1 as corresponding to K = 4.

Another useful feature, observed in the lowest *J* rotational spectra of CH₃NCO, is the nuclear quadrupole hyperfine structure due to the presence in the molecule of a single ¹⁴N nucleus. At the current resolution of the Stark spectra in Figure 1, this gives rise to a noticeable asymmetry in the lineshapes for K = J'' transitions (as for the K = 4 line in Figure 1). As shown in Figure 2, these doublets can be better resolved in the frequency-modulated spectra not only for K = J'' but also for K = J'' - 1 transitions. The frequency splitting in the doublet is measurable and is compared in Figure 2 with predictions based on the most recent hyperfine splitting constants for CH₃NCO (Cernicharo et al. 2016). The hyperfine patterns thus provide independent confirmation of the assignments based on the Stark patterns. The behavior of the Stark and hyperfine patterns as a function of *K* and *J''* for lines that are crucial to the higher *K* assignment is further illustrated in Figure 3.

The application of these criteria and step by step inspection of spectral regions corresponding to higher *J* transitions then allowed the identifications of K = 5 - 10 transitions. The availability of two different criteria was especially useful in the denser portions of the spectra where the characteristic Stark lobes could not be clearly distinguished due to overlaps with those of neighboring lines. Final confirmation of the *K* assignments was made simply by the absence of the corresponding lines in the spectral region for J < K. Once the K > 3 transitions were successfully assigned in the low-frequency Stark spectra for lowest-*J*, their higher-*J* counterparts could be easily located in the higher-frequency spectra (up to 364 GHz) measured in our previous work (Cernicharo et al. 2016). Graphical Loomis-Wood-type plotting techniques as built into the AABS package (Kisiel et al.



Figure 1. Upper panel: The CH₃NCO molecule and a section of the low-voltage, Stark-modulation spectrum (10 V) in the region of the $J = 5 \leftarrow 4$ rotational transition. Lower panel and inset: a detail of the spectrum showing a characteristic Stark effect and line shape asymmetry for isolated transitions. While K = 1 and K = 2 transitions are not well modulated, clear fast first-order Stark lobes (see text) can be observed for the K = 3 transitions and the increase of the Stark lobe span with K allows easy identification of the K = 4 transition. Furthermore, the line shape of the K = 4 transition is affected by the nuclear quadrupole hyperfine structure, which is in agreement with predictions in Figure 2.

2005, 2012) proved invaluable for this purpose. The *J* dependence of frequencies for a given set of *K*, *m* and v_b quantum number values is a smooth function so that such transition sequences are readily identifiable.

Koput (1986) has shown that the global fitting of the rotational spectrum of CH₃NCO to experimental accuracy is an extremely challenging task. For this reason, the experimental spectrum has been broken down in our previous work into more than 220 (K, m, v_b) sequences of correlated lines, with each sequence analyzed separately (see Cernicharo et al. 2016, for details). The same pragmatic approach is employed in this work. Each of the identified K > 3 sequences was thus fitted to a suitable linear rotor-type J(J + 1) power series expansion for the rotational energies

$$E_{\rm rot} = BJ(J+1) - D_J J^2 (J+1)^2 + H_J J^3 (J+1)^3 + L_J J^4 (J+1)^4 + P_J J^5 (J+1)^5 + P_{12} J^6 (J+1)^6 + P_{14} J^7 (J+1)^7 + P_{16} J^8 (J+1)^8 + P_{18} J^9 (J+1)^9$$
(1)

where the lengths of the centrifugal distortion expansions were established empirically. In cases of partially resolved nuclear quadrupole hyperfine structure, such as those in Figure 2, an intensity-weighed frequency average has been the measured frequency used in the fitting procedure. The determined linear rotor-type parameters are listed in Table 1 while the experimental frequencies are collected in Table 2. It has to be noted that while the assignment of the *K* quantum numbers is confident, the assignment of *m* quantum numbers is rather uncertain, although a general decrease of intensity with increasing *m* may be expected. For this reason in Table 1 we list the sequences in order of decreasing intensity for each *K* and adopt a modified naming convention in relation to that used in Cernicharo et al. (2016). For example, for K = 4 transitions, the labeling of corresponding sequences is as follows: K4a, K4b, K4c, etc. where a, b, c, are assigned in the order of decreasing relative intensity at room tem-



Figure 2. Line shapes for the higher-*K* transitions at the lowest possible values of J'' illustrating the characteristic, asymmetric doubling arising from the partially resolved nuclear quadrupole hyperfine structure. (a) Comparison of the resolution in Stark and frequency modulated (FM) spectra. (b) Illustration of the relatively constant, resolvable doubling visible in FM spectra for the J'' = K lines, which decreases rapidly for J'' = K + 1 transitions. (c) Calculated values of the splitting to compare with the experimental values indicated in the previous panel.



Figure 3. A diagram of Stark line shapes for some K = J'', K = J'' - 1, and K = J'' - 2 transitions showing the increasing frequency span of the Stark lobe pattern on increasing *K* for a given value of *J''*, while for a given *K* the width of Stark pattern and the hyperfine splitting decrease rapidly with increasing *J''*.



Figure 4. A typical Stark pattern (10 V modulation voltage) observed for the m = 0, $K_a = 2$ asymmetry doublet in the ground state $v_b = 0$ and two successive excitations of the CNC bending mode $v_b = 1$ and $v_b = 2$. The irregular frequency progression of the three marked doublets is indicative of the anharmonicity/perturbations in the CNC bending motion.

perature (a strongest, b second strongest, etc.). Many sequences for K = 4 and 5 consist of degenerate lower-*J* parts, separating into two distinct series at high-*J*. In such cases L and U letters are added (e.g. K4aL, K4aU) in order to distinguish between the components separating to lower and upper frequency, respectively. In some cases a line sequence displayed a *J* dependence of transition frequencies which was smooth but sufficiently non-linear to fall outside the tractability as a whole with Eq. 1. This necessitated splitting the sequence into two parts (with prefixes V and X), such as for the strongest K = 4 sequence in Table 1 reproduced by parts V129 and X129.

Apart from the K > 3 transitions, a comprehensive investigation of the Stark spectra made it possible to assign also some $K \le 3$ transitions in the $v_b = 0$ state of the parent isotopic species, such as K = -2, m = 1 or K = -1, m = 3L, that were missing in our preceding paper and that are expected to result in identifiable spectral features in the interstellar line surveys. Attempts were made to locate also K = -2, m = -2; K = -2, m = 3L; K = -2, m = 3U sequences of transitions, however, they seem to be strongly perturbed. For completeness, $K \leq 3$ transition sequences in the higher energy m = 4, -5, and 6 internal rotor states of $v_b = 0$, assigned by Koput (1986), were also extended into the millimeter wave region. Ultimately, several $K \le 3$ sequences of transitions could also be identified in $v_b = 1$ and $v_b = 2$ excited vibrational states as shown in Figure 4. For all these K < 3 transitions we follow the notation from Cernicharo et al. (2016). In the case of the m = 0 state for which asymmetric rotor notation K_a and K_c would normally be more appropriate, we use the notation K = nL, nU to distinguish between lower ($K_a + K_c = J + 1$) and upper $(K_a + K_c = J)$ frequency sequences for the same value of $K_a = n$. Analogously, we use the notation m = 3L and 3U to distinguish between the two nearly degenerate m = 3 substates giving rise to lower and upper frequency transition sequences, respectively. Although in the foreseeable future astrophysical detection of lines corresponding to these higher vibrational energy states appears more challenging, it cannot be completely discounted as shown recently by detection of several lines of CH₃NCO in $v_b = 1$ toward Sgr B2(N2) (Belloche et al. 2017). The sets of parameters of the linear rotor-type, J(J + 1) power series for all these additionally assigned $K \leq 3$ sequences are given in Table 3 and the experimental frequencies are gathered in Table 4. Finally, JPL catalog line list for these sequences is provided in Table 5.

A useful overview of the features hindering the solution of the spectroscopic problem for CH₃NCO is provided in Figure 5. In this reduced frequency representation based on Eq. 1 the sequences are expected to converge at low *J* to the effective values of *B* as tabulated in Tables 1 and 3. Only moderate (and uniform) curvature in the vertical direction is expected. A useful reference is provided by the quasi-linear NCNCS molecule (see Figure 8 of Winnewisser et al. (2010)). For NCNCS the behaviour appears to be less perturbed than that visible in Figure 5 but in that case reproduction of the rotational spectrum to within experimental accuracy has not yet been achieved. In the present case of CH₃NCO the situation is considerably more complex since the most important, lowest *K*, sequences in the lowest internal rotation substates m = 0 and 1 appear to show the most unusual behavior. The m = 0, K = 3 sequence is already significantly shifted to low frequency relative to limited fits made with the standard asymmetric rotor model (as reproduced in the discussion of the ¹³C species further below). Furthermore, the most intense K=4 sequence, and thus the first candidate to be assigned to m=0, is displaced even more to low frequency and is highly nonlinear. It is only at higher *K* where the sequences are closer to the positions expected from standard asymmetric rotor theory, but their ordering is by no means regular. The behaviour of the m = 1 sequences as assigned by Koput (1986), in particular of that for



Figure 5. Illustration of the complexity underlying the rotational spectrum of CH_3NCO as revealed by a reduced frequency plot of identified transition sequences. In this representation transition series for a 'well behaved' molecule are expected to show only relatively small curvature from the vertical, such as in the right hand side of the figure. This is clearly not the case for many important series. Color coding: red - the lowest *K* ground state (m = 0) sequences, blue - the lowest *K* m = 1 sequences, black - the most intense sequences for each value of *K* for the presently assigned K > 3 series, green - the remaining presently assigned K > 3 series, grey - other series. The values of *K* for selected series are indicated.

K = 0, is also very challenging. In principle, low barrier methyl group internal rotation is treatable with the RAM36 program of Ilyushin et al. (2010). The program has been very successful in treating toluene, which is a molecule with a very similar energy distribution of internal rotor states to that in CH₃NCO, and also displaying a plethora of perturbations between m = 0 and 1 substates (Ilyushin et al. 2017). In fact, RAM36 has already been applied to CH₃NCO (Halfen et al. 2015), albeit to a very small data set, which has later been shown to contain some misassignments (Cernicharo et al. 2016). The reported predictions allowed confident assignment in Sagittarius B2 of some conservatively chosen CH₃NCO transitions (Halfen et al. 2015) but turned out to be of a very limited character when confronted with the complete laboratory data (Cernicharo et al. 2016). The currently available data set is being used to reach a more comprehensive RAM36 fit. Progress is slow, one of the reasons being that multiple competing parameters of fit are involved. The choice of numerical values of even the leading parameters defining internal rotation for this low barrier case, such as ρ , F, V_3 , D_{ab} and Q_b (see Lin & Swalen 1959) is already found to be critical. Their values can be precalculated with moderate confidence but, in order to reach spectroscopic accuracy, each such parameter also needs to be supported by an associated centrifugal type expansion, which is much more poorly understood. Furthermore, the J dependence of the observed line sequences also lacks the sharp resonances that allow precise identification of the positions of interacting vibration-rotation energy levels (as has been the case for toluene, Ilyushin et al. (2017)). For the time being we prefer to make the extensive laboratory data available for applications and to cite Curl et al. (1963), who on being confronted with the CH₃NCO spectrum stated: There seems to be little value in reporting values of the constants obtained or details of the fits. Nevertheless, the analysis in terms of transition series of the complete rotational spectrum of CH₃NCO up to 364 GHz allows approximate estimation of most of the rovibrational energy levels within 400 cm⁻¹ or so from J = 0 of the ground state with E_{vr} (cm⁻¹) = 0.14462 $J(J + 1) + 4.138K^2 + E_{vib}(m, v_b)$ where the two numerical constants are, respectively, (B + C)/2 and A - (B + C)/2 (from Cernicharo et al. (2016)) and $E_{vib}(m, v_b)$ are the vibrational substate energies from Koput (1986).

3.2. ^{13}C isotopic species

Three different spectroscopic techniques were used for the assignment of rotational transitions of CH₃N¹³CO and ¹³CH₃NCO isotopologs. In a first step, their rotational signatures were searched for in natural abundance in the broadband chirped-pulse Fourier transform microwave spectrum (6 – 18 and 25 – 26 GHz) acquired in our previous study (Cernicharo et al. 2016). This made possible the assignment of $K_a = 0$, m = 0, $v_b = 0$ lines solely on the basis of predicted frequency shifts induced by isotopic substitution. In a next step, Stark modulation spectroscopy on isotopically enriched samples was used to discriminate between transitions with different values of the *K* quantum number. The assignments of *m* and v_b quantum numbers were subsequently

transferred from the parent species on the basis of similarities of spectral patterns for corresponding J regions (see Figure 6). Finally, the assigned sequences were extended into the millimeter wave range. The carbonyl carbon atom is very close to the center of mass of CH₃NCO, so that the $J = 6 \leftarrow 5$ spectrum of CH₃N¹³CO is shifted to low frequency by only *ca* 170 MHz. The effect on the line pattern in Figure 6 is seen to be minimal so that quantum number assignment is readily transferable from the parent species. For ¹³CH₃NCO the corresponding isotopic frequency shift is *ca* 1440 MHz, and changes in the line pattern are seen to be more significant. Nevertheless, the key lines are also assignable on the basis of Stark behaviour and relative intensity. The experimental line lists for the assigned transition sequences of CH₃N¹³CO and ¹³CH₃NCO with the lowest vibrational energies that are expected to be key in future astrophysical identifications are given in Tables 6 and 7. The labeling of the individual sequences as well as the quantum number nomenclature is kept the same as for the parent species. Similarly to the parent species, only $K_a = 0, 1, 2, m = 0, v_b = 0$ transitions could be fitted together with reasonable terms in Watson's semi-rigid asymmetric rotor Hamiltonian. Results of such fits are given in Tables 8 and 9. Table 10 then summarizes the spectroscopic parameters for each of the identified line sequences of CH₃N¹³CO and ¹³CH₃NCO obtained from the linear-rotor fits to Eq. 1. The similarities in their values with those for the parent species as compared in Table 10 corroborate the correct assignments. The JPL catalog line lists for CH₃N¹³CO and ¹³CH₃NCO that are most directly useful for astrophysical applications are reported in Tables 11 and 12, respectively.

Evaluation of transition intensities is critically dependent on the value used for the vibration-rotation partition function q_{vr} . We tested for the possible isotopic changes to q_{vr} , but found those to be small and for simplicity decided to use the recommended parent species function from Table A.4 of Cernicharo et al. (2016). Isotopic changes result from changes in rotational constants and changes in the vibrational energy distribution, both of which would be most marked for ¹³CH₃NCO. On repeating the assumptions used in evaluating the recommended partition function for the parent species we find that the 300K value of q_{vr} would be greater by 3.3% for ¹³CH₃NCO, but by less than 0.4% for CH₃N¹³CO. The distribution of internal rotation substates in this very low barrier case is rather close to the free internal rotation limit of $m^2 F$ (Lin & Swalen 1959). The internal rotation F parameter is estimated to be near 8.3 cm⁻¹ and would be lower by *ca* 0.019 cm⁻¹ in ¹³CH₃NCO and 0.001 cm⁻¹ for CH₃N¹³CO. This would result in only negligible changes in internal rotor energies, which would increase in proportion to m^2 , and be equal to 0.68 cm⁻¹ for the highest considered m=6 substate in ¹³CH₃NCO. Finally, anharmonic *ab initio* calculations show that the fundamental wavenumber of the CNC bending mode decreases by 1.4 cm⁻¹ for ¹³CH₃NCO and 0.4 cm⁻¹ for CH₃N¹³CO. All of these changes are deemed insignificant in relation to the assumptions typically made in partition function evaluation, as discussed in detail in Cernicharo et al. (2016).



Figure 6. Stark spectra of the $J = 6 \leftarrow 5$ region for the parent and the two ¹³C isotopic species of CH₃NCO measured at modulation voltage of 10 V, with frequency axes aligned in way highlighting the relatively small isotopic variation of the observed spectroscopic patterns. Comparison of the two top spectra gives evidence for the achieved high isotopic purity since no lines of the parent species are visible in the spectrum of CH₃N¹³CO.

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₁₈ (aHz)		$7(14) \times 10^{8}$	-1.34(14)	$2(17) \times 10^{8}$	-1.35(14)	1.34×10^{10}	0.373(66)	1.34×10^{10}	0.141(38)		2.12(42)	4.23(46)					-1.438(29)	-1.623(38)		-0.239(13)												28.8(11)			
P_{16} (fHz) P		.69(19)×10 ⁸ -1.1	14.6(13)	$.62(22) \times 10^8$ -1.1	14.7(13)	1.1835×10^{10}	-3.53(59)	1.1829×10^{10}	-1.36(34)		-14.7(24)	-26.7(25)					11.69(23)	13.11(28)		1.58(11)												-152.0(59)			
P_{14} (pHz)		$-1.05(10) \times 10^{8}$ 1	-70.0(58)	$-1.02(12) \times 10^{8}$ 1	-70.4(58)	45.165×10 ⁸ -	14.4 (22)	45.145×10 ⁸ -	5.6 (13)		45.0(59)	73.7(58)					-39.45(76)	-43.93(90)		-3.97(40)			0.226(11)		0.0204(40)	0.0382(42)						311.(13)		-0.284(64)	0.0314(56)
P_{12} (nHz)		$3.79(33) \times 10^7$	194.(13)	$3.66(39) \times 10^7$	195.(13)	-97.67×10^{7}	-32.9 (46)	-97.63×10^{7}	-13.2 (28)		-81.1(79)	-118.0(74)					73.7(14)	81.2(15)		5.22(78)			-0.998(64)		-0.097(25)	-0.213(27)					0.0335(21)	-286.(16)		1.03(28)	-0.160(35)
P_{J} (μ Hz)		-8.59(65) ×10 ⁶	-343.(19)	$-8.33(77) \times 10^{6}$	-345.(19)	131.94×10^{6}	46.3 (60)	131.91×10^{6}	19.2 (36)	0.0428(67)	96.7(62)	124.3(55)	-0.0216(12)	-0.0031(10)			-89.8(15)	-96.9(16)		-4.27(89)		-0.0082(15)	0.73(15)		-0.082(63)	0.210(68)					-0.110(11)	88.(11)		-1.49(50)	0.287(90)
L _J (mHz)		1.287(81) ×10 ⁶ -	404.(18)	1.255(96) ×10 ⁶ -	405.(18)	-11.571×10 ⁶	-40.4 (47)	-11.568×10 ⁶	-16.9 (30)	-0.085(20)	-84.5(30)	-96.4(25)	0.1082(55)	0.0123(47)		-0.00129(56)	81.49(96)	85.54(96)	0.00635(66)	1.08(61)	-0.00105(68)	0.0435(70)	1.02(17)	0.00323(56)	0.698(80)	0.350(89)		0.02590(70)	0.00123(45)		-0.152(25)	9.3(52)		1.09(44)	-0.24(11)
H_{J} (Hz)		$1.295(64) \times 10^{5}$	-305.(10)	$1.269(77) \times 10^{5}$	-306.(10)	6.163×10^{5}	18.5 (23)	6.152×10^{5}	5.7 (15)	0.044(22)	61.43(83)	64.27(67)	-0.1645(92)	-0.0254(79)		0.0252(21)	-57.56(37)	-58.88(34)	0.0136(25)	-2.24(25)	-0.0084(25)	-0.085(12)	1.06(11)	-0.0057(21)	0.319(55)	0.525(62)		-0.0358(26)	0.0021(17)	-0.00678(26)	0.335(25)	13.6(13)	0.00262(32)	-0.46(21)	0.091(78)
D _J (kHz)		-8.74(32) ×10 ³ -	-121.8(37)	-8.62(38) ×10 ³ -	-122.1(37)	24.017×10^{3}	-7.18 (67)	24.014×10^{3}	-11.43(46)	1.536(10)	41.59(12)	41.936(98)	2.0690(67)	1.9297(57)		2.2359(26)	-43.955(76)	-44.176(67)	1.4947(31)	6.558(55)	2.2318(32)	2.0469(83)	-2.796(35)	1.6359(25)	-0.481(20)	-0.424(22)		0.5959(33)	2.0963(21)	1.89726(79)	0.378(12)	-5.10(20)	2.06674(93)	1.884(55)	2.313(27)
B^{g} (MHz)		3830.2(86)	4200.20(73)	3834.(10)	4200.15(72)	4746.97	4232.88(10)	4746.93)	4232.152(74)	4346.5608(20)	4405.4265(92)	4405.4441(68)	4389.1854(20)	4365.1008(16)		4365.4070(12)	4306.3355(75)	4306.3197(62)	4350.4222(15)	4341.4381(58)	4399.9696(15)	4387.7577(25)	4355.8543(52)	4354.3392(12)	4356.1661(32)	4356.1725(35)		4353.4867(16)	4374.6209(10)	4355.09118(70)	4347.7439(26)	4412.075(14)	4386.55941(80)	4411.8773(63)	4388.1291(45)
σ_w^f		3.159	4.944	3.764	4.904	0.0	2.597	0.254	1.349	0.869	1.304	0.914	1.228	0.947		1.016	1.192	0.932	1.240	1.030	1.260	1.282	1.201	0.993	0.833	0.916		1.291	0.849	0.874	1.036	0.967	0.942	1.206	1.121
I Ne		0 (7 0	0 (7 0	0 (0 (0	0 (9 (7 0	0	0 2) 5		3	4	3	0	.1	1 0	5 1	0	-	0	0		3	1 0	5	1 0	7 0	1	4 3	. 33
		10	58 27	10	59 27	5	34 25	5	37 25	27 25	94 37	14 37	47 37	93 30		88 33	14 31	01 31	76 35	73 36	41 34	27 2£	36 25	69 32	79 25	72 3(00 32	34 34	92 32	07 34	92 27	44 31	12 24	11 29
lame ^b I ^c _{re}		V129:	X129: 0.7:	V130:	X130: 0.7:	V102:	X102: 0.7.	V103:	X103: 0.7.	V004: 0.7.	V141: 0.6	V140: 0.7	V051: 0.5-	V108: 0.45		V022: 0.9	V109: 0.7	V110: 0.70	V001: 0.6'	V092: 0.5'	V062: 0.4-	V049: 0.4	V122: 0.3.	V087: 0.24	V238: 0.1	V239: 0.1		V009: 1.00	V036: 0.8.	V011: 0.5	V003: 0.51	V233: 0.4	V048: 0.3-	V184: 0.3	V145: 0.3
K^a N	K = 4	K4aL		K4aU		K4bL '		K4bU		K4c	K4dL	K4dU	K4e	K4f	K = 5	K5a '	K5bL '	K5bU '	K5c '	K5d	K5e	K5f	K5g '	K5h	KSiL	K5iU '	K = 6	K6a	K6b	K6c	K6d	K6e	K6f	K6g	K6h

Table 1 continued on next page

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								TAIMPT	commund)					
K^{a}	Name ^b	Ic .	N ^d N	Γ_{rej}^{e} σ	^f B ^g (MHz	<u></u>	D _J (kHz)	H _J (Hz)	L _J (mHz)	<i>P_J</i> (μHz)	<i>P</i> ₁₂ (nHz)	<i>P</i> ₁₄ (pHz)	P ₁₆ (fHz)	P ₁₈ (aHz)
K6j K6k	V064: C V078: 0).146	32 30	2 0.8 1 1.0	05 4401.17868(09 4426.54770((55) (94)	2.13354(74) 2.2099(10)	0.00560(24) 0.00665(34)						
K = 7							х. 7							
К7а	V013: C	.892	31	3 1.1	08 4357.7248(1-	(4)	1.6500(30)	-0.0152(23)	0.00193(62)					
К7b	V123: C	.388	30	2 1.2	80 4364.1594(1)	(0	1.8941(12)	-0.00551(40)						
K7c	V084: C	.342	27	4 0.8	39 4360.3493(1	2)	0.8838(24)	-0.1255(19)	0.01255(49)					
К7d	V063: C	0.278	31	1 0.8	75 4400.6256(2	(8)	2.318(13)	0.153(27)	-0.008(28)	-0.061(13)	0.0192(26)			
К7e	V060: C).245	30	1 1.1	40 4398.34077((86)	2.1058(11)	0.00548(36)						
К7f	V077: C	0.229	32	0 0.7	34 4425.4453(1	(0	2.1866(20)	0.0030(15)	0.00108(41)					
$\frac{K7g}{K = 8}$	V069: (0.166	33	1 1.1	14 4411.64576((68)	2.1514(10)	0.00692(33)						
K8a	V021: C	.682	26	2 0.9	60 4364.1271(1.	5)	1.9735(30)	-0.0069(23)	-0.00255(61)					
K8b	V059: C	.491	31	2 1.2	42 4397.4177(1	7)	2.1423(34)	0.0089(27)	-0.00519(71)					
K8c	V174: C	.380	31	1 1.3	03 4389.7174(4	5)	1.802(19)	0.453(38)	-0.382(36)	0.417(17)	-0.0919(30)			
K8d	V194: C).296	18	1 1.1	54 4361.0443(3	3)	0.6630(93)	0.029(11)	0.0788(68)	-0.0091(14)				
K = 9														
K9a	V165: C	.495	28	3 1.0	97 4373.1956(4	(4	-1.075(18)	0.213(35)	0.059(33)	0.085(15)	-0.0336(27)			
K9b	V068: C	.375	29	2 0.9	17 4409.55603(80)	2.13686(88)	0.00221(29)						
K9c	V152: C	0.261	25	1 0.8	(29 4377.4805(1	2)	1.6836(26)	-0.0156(20)	0.00197(53)					
K9d	V035: C).282	27	2 1.0	73 4373.4274(1	(9)	2.0711(32)	0.0191(25)	0.00245(65)					
K9e	V034: C	.146	25	3 1.3	(31 4372.0337(1	2)	2.0403(14)	0.00960(48)						
K9f	V200: C	0.105	26	3 1.2	62 4396.09603((09)	2.01705(26)							
K = 1(
K10a	V027: 6	393	27	3 1.1	96 4367.2143(1	(6	1.4495(38)	0.0095(30)	0.00302(79)					
K10b	V074: C	.241	27	2 1.0	62 4422.1124(1	6	2.1495(11)	0.00574(38)						
K10c	V173: C	0.188	29	0 1.4	:25 4392.41226((06)	1.98881(28)							
IL _p	te notation	n of in	dividu	al K se	squences as descri	ibed in t	he text followin	g Eq.1.						
$^{p \operatorname{LI}}$	te identifi	er of tł	ie line	seque	nce established in	(Cernich	1 1 2 1 2 2 1 1 1 1 2 2 1 1 1 2 2 2 2 2							
<i>с</i> А	utomatica	lly assi	igned	intensi	ty of the sequence	e relativ	e to the stronges	t sequence in the	spectrum.					
d Ll	ie number	r of lin	es in t	he line.	ar fit.									
e Tl	te number	: of coi	ıfiden	tlv assi	gned lines that we	ere perti	urbed and were	rejected from the	fit.					
4				- -	0	. ,								
U L	nitless (w	eightea	i) dev	iation c	of the effective rot	or fit.								
θ TI	e values	of the J	param	eters rc	esulting from the e	effective	linear rotor fit.							

 Table 1 (continued)

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						<i>K</i> = 4				
J'	J''	K4aL ^a	K4aU	K4bL	K4bU	K4c	K4dL	K4dU	K4e	K4f
5	4	b	b	42731.292 ^c	42731.292 ^c	43464.849 ^c	44034.686 ^c	44034.686 ^c	43890.896 ^c	43650.061 ^c
6	5	b	<i>b</i>	50964.759	50964.759	52157.325 ^c	52831.945 ^c	52831.945 ^c	52668.393 ^c	52379.568 ^c
7	6	58339.391	58339.391	59350.121	59350.121	60849.700	61624.560	61624.560	61445.688	61108.737
8	7	66984.159	66984.159	67783.970	67783.970	69541.847	70412.420	70412.420	70222.650	69837.656
9	8	75576.792	75576.792	76238.725	76238.725	78233.684	79195.280	79195.280	78999.235	78566.221
10	9	84141.319	84141.319	84705.058	84705.058	86925.110	87973.153	87973.153	87775.326	87294.305
Sequ	ence	V129+X129	V130+X130	V102+X102	V103+X103	V004	V141+X141	V140+X140	V051	V108

Table 2. Line list of K > 3 rotational transitions for CH₃NCO compiled from experimental frequencies (MHz) assigned on the basis of Stark-modulation spectroscopy and nuclear quadrupole hyperfine structure.

^a The sequence notation as specified in the text.

 b Transition for which there is no experimental value and the extrapolation from a polynomial fit is not reliable.

^c Intensity-weighted average of the hyperfine doublet.

Note—All frequencies (unless otherwise marked) are experimental measurements with estimated uncertainty of 0.05 MHz. This table is published in its entirety in the Tar Archive. A portion is shown here for guidance regarding its form and content.

	Table.	3. The	parar	neters (of the linear rote	or-type, $J(J + 1)$) power series fi	tted to the addit	ionally assigned	$K \le 3$ line seq	uences of the p	parent species.	
K^a Name ^b	I ^c	N ^d _{in} 1	ve jen	O w	B (MHz)	D _J (kHz)	H_{J} (Hz)	L _J (mHz)	$P_{J}(\mu \mathrm{Hz})$	P ₁₂ (nHz)	P ₁₄ (pHz)	P ₁₆ (fHz)	P_{18} (aHz)
$v_{\rm b} = 0, m =$	$1 E: E_{\rm vib}$	= 8.4 ci			C								
-2a V179: -2b X179:	0.579 0.789	16 22	1 0 1	.494 .016	4350.3325(51) 4344.71(97)	-78.80(10) -125.0(46)	-44.0(10) -253.(12)	-238.0(46) 309.(20)	526.(10) -260.(21)	-356.5(84) 149.(14)	··· -55.4(64)	 12.1(15)	··· -1.16(16)
$v_{\rm b} = 0, m = 0$	3U (K =	0, J = 1	h = 0	12, A1): I	$E_{\rm vib} = 79.7 {\rm cm}^{-1}$								
-1 V138:	0.663	34	5 1	577	4377.4365(70)	-10.872(77)	3.73(40)	-27.1(11)	24.8(17)	-20.7(16)	13.66(90)	-4.81(27)	0.658(34)
$v_{\rm b} = 0, m = 0$	$4 E: E_{\rm vib}$	= 140.6	cm ⁻¹										
-1 V164:	0.502	36	3	.259	4372.7579(28)	-0.421(13)	0.001(29)	0.560(30)	-0.246(14)	0.0193(27)	:	:	÷
1 V160:	0.312	29	1	.076	4372.9957(46)	1.714(48)	-1.04(23)	2.72(56)	-4.08(74)	3.13(48)	-0.97(12)	:	÷
-2 V229:	0.461	26	0	.183	4407.0035(81)	-23.20(14)	87.6(12)	-344.4(54)	243.(13)	145.(18)	-259.(13)	93.4(41)	÷
2 V158: 2' V159:	0.209 0.209	34 34	5 1 5	.213 .404	4379.0786(10) 4379.0839(17)	2.0825(11) 2.1043(38)	-0.00046(37) 0.0111(33)	··· -0.01225(93)	: :	: :	: :	: :	: :
$v_{\rm b} = 0, m =$	$-5 E: E_v$	_{ib} = 217	.5 cm ⁻	1									
-1 V045:	0.252	34	2	.854	4384.15287(41)	2.11044(22)				:	:	÷	:
2 V193:	0.413	37	2	.994	4388.1800(34)	-2.184(29)	1.09(11)	-0.28(23)	1.74(27)	-1.84(17)	0.608(61)	-0.0661(86)	:
-2 V056:	0.199	38	1 0	1961	4393.35137(72)	2.17888(83)	0.00628(28)	::	:	:	:	:	:
$v_{\rm b} = 0, m = 0$	5 A: E _{vib}	= 311.1	cm ⁻¹										
0 V175:	0.377	35	3	.248	4387.7938(21)	2.1700(70)	0.0974(97)	-0.0479(58)	0.0092(13)	:	:	:	:
-1 V136:	0.458	35	4	.089	4383.39713(85)	1.89120(97)	0.00489(33)			:	:	:	÷
1 V058:	0.314	40	0	.023	4395.72730(75)	2.19786(86)	0.00391(29)	:		:	:	:	:
-2 V147:	0.533	37	2	.054	4386.2913(12)	0.8694(25)	0.0705(20)	0.02113(54)		:	:	:	:
$v_{\rm b} = 1, m = 1$	0 : E _{vib} =	: 182.2 c	m-1										
2L V090:	0.356	36	3	.077	4342.2441(23)	3.281(11)	0.144(24)	-0.118(24)	0.062(11)	-0.0124(22)	:	÷	:
2U V088:	0.385	38	1	.054	4342.2228(12)	-2.0956(25)	-0.1958(20)	-0.02294(53)	:		:	:	:
3L V104:	0.339	32	5 0	.852	4282.6194(34)	13.690(30)	7.53(11)	2.05(21)	-4.58(22)	2.59(13)	-0.707(44)	0.0790(58)	:
3U V105:	0.325	31	9	166'	4282.6294(41)	13.818(39)	8.01(14)	1.27(27)	-3.85(28)	2.20(17)	-0.593(55)	0.0653(72)	:
$v_{\rm b} = 1, m =$	$1: E_{vib} =$	= 191.4 c	m_1										
0a V177:		21	0 0	.774	4353.3169(60)	122.30(14)	37.3(16)	429.6(88)	-1143.(25)	1306.(37)	-593.(21)	:	:
0b X177:	0.359	15	0	.703	4357.44(27)	164.3(13)	264.3(34)	-288.8(52)	202.9(47)	-83.0(22)	15.02(45)	:	÷
-1 V099:	0.379	39	2	.147	4294.0094(36)	-19.616(29)	-19.36(11)	14.63(21)	-8.80(24)	3.63(15)	-0.883(50)	0.0942(66)	:
$v_{\rm b} = 1, m =$	$-2:E_{\rm vib}$	= 222.3	cm ⁻¹										

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Table 3 continued on next page

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Table 3 (continued)	$B \text{ (MHz)} \qquad D_J \text{ (kHz)} \qquad H_J \text{ (Hz)} \qquad L_J \text{ (mHz)} \qquad P_J (\mu \text{Hz}) \qquad P_{12} \text{ (nHz)} \qquad P_{14} \text{ (pHz)} \qquad P_{16} \text{ (Hz)} \qquad P_{18} \text{ (aHz)}$	4355.3770(22) 0.539(15) 0.556(44) -1.554(66) 0.086(52) 0.190(21) -0.0384(33) 4357.25912(76) 1.43704(90) -0.00953(30)		4330.7106(23) 6.917(19) 0.415(70) -0.42(13) 0.50(15) -0.316(96) 0.102(31) -0.0134(42)	4294.05621(88) 3.05470(94) 0.07787(30)	4375.67247(93) 4.1429(10) 0.03510(37) ··· ··· ··· ··· ··· ··· ··· ··· ···	4343.4842(16) 3.1549(33) 0.0395(26) -0.00193(70) ··· ·· ·· ·· ·· ·· ·· ·· ··	4343.443(13) -1.0207(27) -0.1855(22) -0.01014(59)
I	B (MHz) D_{J} (kHz) H_{J} (F	4355.3770(22) 0.539(15) 0.556(4357.25912(76) 1.43704(90) -0.0095		4330.7106(23) 6.917(19) 0.415(4294.05621(88) 3.05470(94) 0.0778	4375.67247(93) 4.1429(10) 0.0351	4343.4842(16) 3.1549(33) 0.0395	4343.4743(13) -1.0207(27) -0.1855
	^{<i>a</i>} Name ^{<i>b</i>} f_{rel}^{c} N_{in}^{d} N_{rej}^{e} σ_{w}^{f}	0 V120: 0.345 38 2 0.838 1 V015: 0.270 32 4 0.938	$= 2, m = 0 : E_{vlb} = 357.9 \text{ cm}^{-1}$	0 V095: 0.187 41 1 0.740	L V128: 0.176 38 2 1.180	U V217: 0.156 31 2 1.139	L V091: 0.156 31 4 1.263	U V113: 0.172 33 4 1.091

^{*a*} The value of the K_a or K quantum number following the notation described in Cernicharo et al. (2016).

 $b\ {\rm The}$ identifier of the line sequence established in Cernicharo et al. (2016).

 c Automatically assigned intensity of the sequence relative to the strongest sequence in the spectrum.

d The number of lines in the linear fit.

^e The number of confidently assigned lines that were perturbed and were rejected from the fit.

f Unitless (weighted) deviation of the fit.

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Table 4. Line list of rotational transitions for CH₃NCO compiled from experimental frequencies (MHz) for additional $K \le 3$ line sequences assigned on the basis of continuity from the assignment reached by Koput (1986) for $J'' \le 3$.

v_b :	= 0, m = 4	$E_{\rm vib} = 140.6 {\rm cm}$	$m^{-1} a$		
J'	$J^{\prime\prime}$	$K = -1^a$	K = 1	<i>K</i> = -2	<i>K</i> = 2
2	1	17491.620 ^b	17491.620 ^b		
3	2	26236.710 ^b	26237.850 ^b	26444.570^{b}	26274.290^{b}
4	3	34982.163	34983.529	35262.419	35032.080
5	4	43727.872	43729.167	44083.027	43789.693
6	5	52473.515	52474.398	52907.508	52547.163
7	6	61219.189 ^c	61219.500 ^c	61736.631	61304.239
8	7	69965.011	69964.211	70571.424	70060.972
Sec	quence a	V164	V160	V229	V158

^{*a*} The vibrational energies, *K* quantum number and sequence notations are the same as in Cernicharo et al. (2016).

^b Taken from Koput (1986)

^{*C*} Calculated from a J(J + 1) power series fit to the sequence (mainly due to line blending).

Note—All frequencies (unless otherwise marked) are hyperfine-free experimental measurements with estimated uncertainty of 0.05 MHz.This table is published in its entirety in the Tar Archive. A portion is shown here for guidance regarding its form and content.

Frequency	Error	Log(Int) ^a	DR ^b	$E_{\rm low}$	$g_{ m upp}$ c	TAG ^d	QNFMT ^e			QN' ^f					QN'' ^g		
(MHz)	(MHz)			(cm^{-1})				J'	K'_a	K_c'	m'	$v_{\rm b}'$	$J^{\prime\prime}$	$K_a^{\prime\prime}$	K_c''	$m^{\prime\prime}$	$v_{\rm b}^{\prime\prime}$
61299.229	0.0500	-5.1774	3	89.9688	15	01	505	7	-1	0	3	0	6	-1	0	3	0
70061.612	0.0500	-5.0059	3	92.0119	17	01	505	8	-1	0	3	0	7	-1	0	3	0
78825.997	0.0500	-4.8562	3	94.3469	19	01	505	9	-1	0	3	0	8	-1	0	3	0
87592.559	0.0500	-4.7237	3	96.9736	21	01	505	10	-1	0	3	0	9	-1	0	3	0
96361.445	0.0500	-4.6051	3	99.8922	23	01	505	11	-1	0	3	0	10	-1	0	3	0
105132.553	0.0500	-4.4982	3	103.1026	25	01	505	12	-1	0	3	0	11	-1	0	3	0

Table 5. JPL catalog line list for additional $K \le 3$ transitions of CH₃NCO.

^a Base 10 logarithm of the integrated intensity at 300 K in units of nm² MHz.

^b Degrees of freedom in the rotational partition function.

^c Upper state degeneracy.

^d Species tag or molecular identifier.

^e Format of the quantum numbers.

^f Quantum numbers for the upper state. K_a and K_c quantum numbers are defined only for the m = 0 state. For m > 0: $K_a = K$ and $K_c = 0$.

^g Quantum numbers for the lower state. K_a and K_c quantum numbers are defined only for the m = 0 state. For m > 0: $K_a = K$ and $K_c = 0$.

NOTE—The key parameters used in the evaluation of this table, $\mu = \mu_a = 2.882$ D, T = 300 K, and $Q_{vr} = 138369$, are defined in sections A.1 and A.2 of Cernicharo et al. (2016). This table is published in its entirety in the machine readable format. A portion is shown here for guidance regarding its form and content.

$v_b = 0$	0, m = 0	(ground state), E	$v_{\rm vib} = 0 \ {\rm cm}^{-1} \ a$					
J'	$J^{\prime\prime}$	$K = 0^{a}$	K = 1L	<i>K</i> = 1U	K = 2L	K = 2U	K = 3L	K = 3U
6	5	51853.337	51403.416	52340.000	51905.699	51910.851	51726.586	51726.586
7	6	60492.530	59969.439	61061.832	60555.431	60563.705	60345.859	60345.859
8	7	69130.253	68534.899	69782.881	69204.658	69217.015	68964.201	68964.201
9	8	77766.490	77099.721	78503.174	77853.205	77870.697	77581.512	77581.512
10	9	86400.869 ^b	85663.778	87222.481	86501.119	86525.020	86197.856	86197.856
11	10	95033.255 ^b	94227.186^{b}	95940.874^{b}	95148.303 ^b	95180.269 ^b	94812.773 ^b	94812.852^{b}
12	11	103663.458 ^b	102789.647 ^b	104658.087 ^b	103794.561 ^b	103836.065 ^b	103426.398 ^b	103426.526 ^b
Sequ	ence ^a	V093	V126	V131	V089	V006	V096	V097

Table 6. Line list of rotational transitions for $CH_3N^{13}CO$ compiled from experimental frequencies (MHz) for line sequences assigned on the basis of CP-FTMW, Stark-modulated, and frequency-modulated millimeter wave spectra.

^a The vibrational energies, K quantum number notation, and the labeling of the individual line sequences are the same as for the parent species in Cernicharo et al. (2016).

^b Calculated from a J(J + 1) power series fit to the sequence (interpolation due to line blending or missing data).

NOTE—All frequencies (unless otherwise marked) are hyperfine-free experimental measurements with estimated uncertainty of 0.05 MHz. This table is published in its entirety in the Tar Archive. A portion is shown here for guidance regarding its form and content.

CH₃NCO

$v_b = 0$	0, m = 0	(ground state), E	$v_{\rm vib} = 0 \ {\rm cm}^{-1} \ a$					
J'	$J^{\prime\prime}$	$K = 0^{a}$	K = 1L	K = 1U	K = 2L	K = 2U	K = 3L	<i>K</i> = 3U
6	5	50583.272	50151.340	51050.741	50634.955	50639.578	50482.694	50482.694
7	6	59010.994	58508.723	59557.732	59072.857	59080.538	58894.756	58894.756
8	7	67437.329	66865.576	68064.121	67510.342	67521.855	67306.163	67306.163
9	8	75862.078	75221.855	76569.588	75947.215	75963.479	75716.641	75716.641
10	9	84285.279 ^b	83577.376	85074.249	84383.466	84405.827	84126.212	84126.212
11	10	92706.543	91932.271 ^b	93577.977 ^b	92818.970 ^b	92848.738 ^b	92534.703 ^b	92534.775 ^b
12	11	101125.795	100286.312	102080.644	101253.657	101292.289	100942.013	100942.013
Seque	ence <i>a</i>	V093	V126	V131	V089	V006	V096	V097

Table 7. Line list of rotational transitions for ${}^{13}CH_3NCO$ compiled from experimental frequencies (MHz) for line sequences assigned on the basis of CP-FTMW, Stark-modulated, and frequency-modulated millimeter wave spectra.

 a The vibrational energies, K quantum number notation, and the labeling of the individual line sequences are the same as for the parent species in Cernicharo et al. (2016).

^b Calculated from a J(J + 1) power series fit to the sequence (interpolation due to line blending or missing data).

Note—All frequencies (unless otherwise marked) are hyperfine-free experimental measurements with estimated uncertainty of 0.05 MHz. This table is published in its entirety in the Tar Archive. A portion is shown here for guidance regarding its form and content.

Table 8. Spectroscopic constants for the ground states ($v_b = 0$, m = 0) of CH₃N¹³CO and ¹³CH₃NCO in comparison with the parent species.

Constant	Parent species ^a	CH ₃ N ¹³ CO	¹³ CH ₃ NCO
A (MHz)	128435(19)	128324.7(39)	127110(25)
B (MHz)	4414.6182(93)	4400.1847(12)	4291.1542(90)
C (MHz)	4256.7490(85)	4243.2158(11)	4140.4916(82)
Δ_J (kHz)	2.3209(11)	2.30880(17)	2.22320(57)
Δ_{JK} (kHz)	-1274.0(13)	-1273.73(12)	-1240.5(14)
δ_J (kHz)	0.4042(14)	0.40068(25)	0.37712(92)
$\delta_K (\text{kHz})$	183.1(43)	$[183.1]^{b}$	160.8(41)
Φ_J (Hz)	-0.00191(47)	$[-0.00191]^{b}$	$[-0.00191]^{b}$
Φ_{JK} (Hz)	-5.39(24)	$[-5.39]^{b}$	-4.08(33)
Φ_{KJ} (Hz)	-68478(285)	$[-68478.]^{b}$	-63001(300)
ϕ_{JK} (Hz)	-47.6(14)	$[-47.61]^{b}$	-44.94(78)
$N_{\rm lines}{}^c$	201	160	161
$\sigma_{\rm fit}{}^d$ (kHz)	94.6	88.5	99.6
$\sigma_{\mathrm{w}}{}^{e}$	1.89	1.77	1.99

^a Taken from Cernicharo et al. (2016).

^b Fixed to the parent species value.

^c The number of fitted lines.

^d Standard deviation of the fit.

^e Unitless (weighted) deviation of the fit.

Note-Numbers in parentheses are standard errors in units of the last digit.

Table 9. Results of fitting the spectroscopic constants of Table 8 for $CH_3N^{13}CO$ and $^{13}CH_3NCO$ ground states ($v_b = 0, m = 0$).

СН	₃ N ¹³ C	20							¹³ C	H ₃ NC	20						
J'	K'_a	K_c'	$J^{\prime\prime}$	$K_a^{\prime\prime}$	$K_c^{\prime\prime}$	obs. a	0c. ^b	unc. c	J'	K'_a	K_c'	$J^{\prime\prime}$	$K_a^{\prime\prime}$	$K_c^{\prime\prime}$	obs. <i>a</i>	0c. ^b	unc. c
						(MHz)	(MHz)	(MHz)							(MHz)	(MHz)	(MHz)
1	0	1	0	0	0	8643.400 ^d	0.008	0.050	1	0	1	0	0	0	8431.640 ^d	0.003	0.050
2	0	2	1	0	1	17286.600^d	0.019	0.050	2	0	2	1	0	1	16863.100^d	0.015	0.050
3	0	3	2	0	2	25929.400^d	0.032	0.050	3	0	3	2	0	2	25294.240^d	0.087	0.050
6	0	6	5	0	5	51853.337	0.039	0.050	6	0	6	5	0	5	50583.272	0.076	0.050
7	0	7	6	0	6	60492.530	0.065	0.050	7	0	7	6	0	6	59010.994	0.131	0.050
8	0	8	7	0	7	69130.253	0.023	0.050	8	0	8	7	0	7	67437.329	0.118	0.050

^a Experimentally observed frequency.

^b Observed minus calculated frequency.

^c Experimental uncertainty.

^d Hyperfine-free frequency.

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Table

Species	K^a	Name ^b	$I_{\rm rel}^c$	$N^d_{ m in}$	$N_{ m rej}^e$	$\sigma_{\rm w}^f$ (MHz)	B (MHz)	D _J (kHz)	H _J (Hz)	L _J (mHz)	P _J (μHz)	P_{12} (nHz)	P_{14} (pHz)	P_{16} (fHz)	P ₁₈ (aHz)
$v_{\rm b}=0, m=0$	(gs): 1	$E_{\rm vib} = 0 c$	в_1												
CH ₃ NCO	0	V093:	0.843	41	-	0.910	4335.6996 (13)	8.4815 (43)	0.1971 (58)	0.0728 (33)	(69) 60600.0-	:	:	÷	:
CH ₃ N ¹³ CO		V093:	0.997	36	0	0.735	4321.7180 (14)	8.4036 (41)	0.1946 (52)	0.0714 (29)	-0.00887 (60)	:	:	:	:
¹³ CH ₃ NCO		V093:	0.730	37	0	1.243	4215.8348 (15)	7.8872 (29)	0.1644 (22)	0.07012 (55)	8[60600.0-]	:	:	:	:
CH ₃ NCO	IL	V126:	0.783	38	7	0.967	4297.6203 (10)	3.3744 (21)	0.1133 (16)	-0.00393 (41)	:	:	:	:	:
CH ₃ N ¹³ CO		V126:	0.898	31	0	0.659	4283.85898 (98)	3.3465 (20)	0.1100 (16)	-0.00358 (44)	:	:	÷	:	÷
¹³ CH ₃ NCO		V126:	1.000	32	2	0.862	4179.5051 (10)	3.1806 (19)	0.0960 (14)	-0.00223 (34)	:	:	:	÷	:
CH ₃ NCO	10	V131:	0.786	38	6	1.249	4376.1857 (14)	4.2752 (30)	-0.0053 (24)	0.00245 (67)	:	:	:	:	:
CH ₃ N ¹³ CO		V131:	0.948	31	0	0.730	4361.97551 (67)	4.24135 (82)	-0.00599 (29)	[0.00245]	:	:	:	:	:
¹³ CH ₃ NCO		V131:	0.943	28	-	0.770	4254.5199 (11)	4.0287 (25)	-0.0120 (22)	0.00598 (64)	:	:	:	:	÷
CH ₃ NCO	2L	V089:	0.761	38	-	1.101	4339.69609 (82)	3.28715 (94)	0.03891 (31)	:	:	:	:	:	:
CH ₃ N ¹³ CO		V089:	0.760	31	0	0.737	4325.71123 (68)	3.26463 (82)	0.03850 (30)	:	:	:	:	:	:
¹³ CH ₃ NCO		V089:	0.725	33	1	0.693	4219.79574 (53)	3.11301 (58)	0.03494 (18)	:	:	:	:	:	÷
CH ₃ NCO	2U	V006:	0.835	38	-	1.046	4339.6811 (16)	-2.8764 (56)	-0.1690 (77)	-0.0676 (47)	0.0079 (10)	÷	:	:	:
CH ₃ N ¹³ CO		V006:	0.747	30	-	0.742	4325.69904 (71)	-2.82860 (86)	-0.16306 (31)	[-0.0676]	[0.0079]	:	:	:	:
¹³ CH ₃ NCO		V006:	0.917	27	0	0.870	4219.7862 (15)	-2.5537 (40)	-0.1440 (43)	-0.0629 (15)	[0.0079]	:	:	:	÷
CH ₃ NCO	3L	.V096:	0.734	34	5	1.123	4324.6812 (18)	5.1738 (58)	-0.2812 (77)	0.1712 (44)	-0.01590 (94)	:	:	:	:
CH ₃ N ¹³ CO		:960V	0.792	29	0	0.897	4310.9131 (14)	5.0840 (30)	-0.2833 (25)	0.16765 (73)	[-0.01590]	:	:	÷	:
¹³ CH ₃ NCO		:960A	0.492	27	-	1.249	4207.1970 (20)	4.3934 (52)	-0.2735 (54)	0.1312 (18)	[-0.01590]	:	:	÷	:
CH ₃ NCO	3U	:7097	0.808	35	4	1.078	4324.6762 (18)	5.1686 (55)	-0.1769 (74)	0.1479 (43)	-0.01455 (89)	:	:	:	:
CH ₃ N ¹³ CO		:7097	0.777	29	0	0.907	4310.9126 (14)	5.0856 (31)	-0.1762 (26)	0.14395 (74)	[-0.01455]	:	:	:	:
¹³ CH ₃ NCO		:760V	0.492	24	ю	0.950	4207.1944 (17)	4.3939 (43)	-0.1719 (43)	0.1092 (14)	[-0.01455]	:	:	:	:
$v_{\rm b} = 0, m = 1$	$E: E_v$	_{ib} = 8.4 c	m_1												
CH ₃ NCO	0a	V107:	0.625	22	0	1.016	4357.5774(71)	132.34(14)	86.4(14)	226.1(67)	-665.(17)	719.(21)	-298.(11)	:	:
CH ₃ N ¹³ CO		V107:	0.935	14	0	0.603	4343.4612(83)	130.81(12)	85.56(77)	220.1(24)	-651.3(39)	709.3(23)	[-298]	:	:
¹³ CH ₃ NCO		V107:	0.796	13	0	0.362	4236.403(10)	120.50(20)	74.5(19)	194.9(96)	-549.(25)	571.(33)	-228.(17)	:	:
CH ₃ NCO	0^{p}	X107:	066.0	20	0	1.125	4348.7(10)	116.2(37)	146.2(74)	-125.4(87)	71.4(65)	-25.9(29)	5.39(74)	-0.493(81)	:
CH ₃ N ¹³ CO		X107: V107:	0.815	18	0 0	0.948	4334.99(43) 4330.37647)	116.1(14)	146.0(24)	-125.4(24)	71.5(14)	-25.91(44)	5.404(59) r < 201	[-0.493] 1.0.403]	:
CLIBINCO		v	C0C.0	<u>t</u>	0	601.1	(1+)10007+	114.4(14)	144.0(21)	(01)1.421-	(10)10.17	(+1)70.(2-	[kc·c]	[664:0-]	•
CH ₃ NCO CH ₃ N ¹³ CO	-	V101: V101:	0.755 0.917	39 31	- 0	1.000 0.572	4282.7615(32) 4269.4950(20)	-3.118(26) -3.039(13)	9.39(10) 8.933(42)	-10.67(19) -10.023(65)	6.64(21) 6.192(53)	-2.59(13) -2.434(22)	0.5668(37)	-0.0593(59) [-0.0593]	: :

Table 10. Comparison of parameters of the linear rotor-type, J(J + 1) power series for the same line sequence for the parent, CH₃N¹³CO and ¹³CH₃NCO species.

CH₃NCO

Species	K^{a}	Name ^b	Ic	$N^d_{ m in}$	N^e_{rej}	$\sigma_{\rm w}^f$ (MHz)	B (MHz)	D_{J} (kHz)	H_{J} (Hz)	L _J (mHz)	<i>P_J</i> (μHz)	P_{12} (nHz)	P_{14} (pHz)	P_{16} (fHz)	P_{18} (aHz)
¹³ CH ₃ NCO		V101:	0.742	26	0	1.171	4169.4635(56)	-1.903(49)	6.60(18)	-6.47(34)	3.62(35)	-1.43(17)	0.398(36)	[-0.0593]	:
CH ₃ NCO CH ₃ N ¹³ CO	1	V098: V098:	0.797	36 30	ŝ	0.961	4284.5720(42) 4270.9525(37)	-35.940(44) -35.408(36)	-45.38(21) -44.29(15)	49.98(57) 48.24(33)	-45.27(87) -43.37(41)	29.62(81) 28.35(28)	-12.71(44) -12.24(10)	3.086(16) 3.086(16)	-0.342(15) [-0.342]
¹³ CH ₃ NCO		V098:	0.881	26	0	0.950	4167.8312(70)	-32.204(79)	-39.92(38)	44.17(97)	-41.1(13)	27.9(11)	-12.39(47)	3.150(85)	[-0.342]
CH ₃ NCO	7	V010:	0.660	31	7	0.845	4350.3415(23)	-2.325(15)	-0.065(46)	-0.290(64)	0.284(43)	-0.109(11)	÷	:	:
CH ₃ N ¹³ CO		V010:	0.728	25	5	1.293	4336.3057(24)	-2.3166(70)	-0.0848(78)	-0.2813(29)	[0.284]	[-0.109]	:	÷	:
¹³ CH ₃ NCO		V010:	0.889	23	4	1.004	4229.9947(19)	-2.1780(61)	-0.0500(75)	-0.2959(30)	[0.284]	[-0.109]	:	:	:
CH ₃ NCO	ή	V005:	0.800	36	7	1.153	4343.3487(25)	0.096(11)	0.288(25)	-0.381(25)	0.202(12)	-0.0848(22)	:	:	:
CH ₃ N ¹³ CO		V005:	0.741	31	0	0.944	4329.3382(26)	0.074(13)	0.171(29)	-0.225(32)	0.109(17)	-0.0625(34)	:	÷	:
¹³ CH ₃ NCO		V005:	0.870	26	0	1.161	4223.2217(29)	0.190(12)	0.291(22)	-0.402(18)	0.2363(55)	[-0.0848]	:	:	:
CH ₃ NCO	б	V012:	0.642	34	б	0.973	4356.1489(11)	1.6845(24)	-0.0204(19)	0.00189(52)	:	:	:	:	:
CH ₃ N ¹³ CO		V012:	0.617	30	0	0.865	4342.10839(42)	1.67820(20)	[-0.0204]	[0.00189]	:	:	:	÷	:
¹³ CH ₃ NCO		V012:	0.726	25	1	0.891	4235.6898(10)	1.6380(15)	-0.01831(70)	[0.00189]	÷	:	:	:	:
$v_{\rm b} = 0, m = -$	-2 E:	$E_{\rm vib}=36$	6.8 cm ⁻¹												
CH ₃ NCO	0	V119:	0.774	36	1	1.001	4356.2099(29)	2.431(15)	0.837(33)	-2.676(29)	:	0.851(17)	-0.314(10)	0.0333(20)	:
CH ₃ N ¹³ CO		V119:	0.750	29	0	1.012	4342.0974(31)	2.423(12)	0.793(21)	-2.595(15)	:	0.8246(46)	-0.3063(14)	[0.0333]	:
¹³ CH ₃ NCO		V119:	0.825	25	0	0.948	4235.2124(29)	2.433(16)	0.570(38)	-2.093(34)	:	0.629(16)	-0.2422(64)	[0.0333]	:
CH ₃ NCO	-1	V008:	0.708	38	5	1.243	4353.2529(11)	1.0831(14)	:	0.00928(84)	-0.00378(31)	:	:	:	:
CH ₃ N ¹³ CO		V008:	0.664	30	-	0.520	4339.19340(41)	1.08491(35)	E	0.009162(61	[-0.00378]	:	:	÷	:
¹³ CH ₃ NCO		V008:	0.759	26	0	0.928	4232.66949(92)	1.09494(97)	:	0.00852(25)	[-0.00378]	:	:	:	:
CH ₃ NCO	-	V180:	0.807	40	0	3.750	4350.771(14)	-40.79(16)	-39.36(81)	-24.5(22)	74.5(34)	-69.5(32)	34.7(17)	-9.27(54)	1.045(69)
CH ₃ N ¹³ CO		V180:	0.743	27	-	1.061	4336.6441(84)	-40.08(10)	-37.91(53)	-26.7(14)	78.8(24)	-75.3(24)	39.0(14)	-10.93(48)	1.301(64)
¹³ CH ₃ NCO		V180:	0.901	26	0	0.872	4230.0429(99)	-35.28(13)	-27.23(83)	-41.3(27)	100.7(56)	-102.8(69)	60.6(50)	-20.1(20)	2.89(34)
CH ₃ NCO	7	V100:	0.805	39	0	1.677	4276.5950(66)	-51.195(68)	-64.24(33)	63.79(84)	-50.5(12)	29.3(11)	-11.35(59)	2.60(17)	-0.262(20)
CH ₃ N ¹³ CO		V100:	0.972	31	0	0.813	4262.8094(60)	-50.410(69)	-62.56(34)	61.94(90)	-49.7(14)	29.7(13)	-12.06(75)	2.92(23)	-0.316(30)
¹³ CH ₃ NCO		V100:	0.749	26	0	0.529	4157.4953(48)	-46.639(52)	-53.90(24)	51.23(59)	-40.60(82)	24.36(66)	-9.98(28)	2.432(50)	[-0.262]
CH ₃ NCO	ς	V024:	0.473	35	3	1.105	4366.9265(12)	1.9129(27)	-0.0098(22)	0.00672(61)		:	:	:	:
CH ₃ N ¹³ CO		V024:	0.586	28	-	0.516	4352.85172(51)	1.90448(61)	-0.00915(22)	[0.00672]		:	:	:	:
¹³ CH ₃ NCO		V024:	0.612	25	0	0.933	4246.1256(19)	1.8213(48)	-0.0204(50)	0.0153(17)	:	;	:	:	:
CH ₃ NCO	3a	V106:	0.523	21	0	0.860	4373.6038(50)	110.306(69)	93.71(41)	-32.2(12)	-11.7(16)	10.81(88)	:	:	:
CH ₃ N ¹³ CO		V106:	0.901	14	0	0.761	4359.2931(53)	108.531(50)	91.21(19)	-30.46(32)	-12.23(19)	[10.81]	:	:	:
¹³ CH ₃ NCO		V106:	0.750	15	0	0.907	4251.1541(59)	98.593(61)	78.31(23)	-22.26(38)	-14.34(22)	[10.81]		:	:
CH ₃ NCO	3b	X106:	0.884	18	0	1.049	4370.20(59)	103.8(16)	95.6(24)	-55.9(22)	20.9(11)	-4.49(32)	0.438(38)	:	:
CH ₃ N ¹³ CO		X106:	0.915	15	0	0.789	4356.21(28)	103.12(70)	94.79(92)	-55.57(67)	20.80(25)	-4.491(38)	[0.438]	:	:

Table 10 continued on next page

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Species	K^a]	Name ^b	I_{rel}^{c}	$N^d_{ m in}$	$N_{ m rej}^e$	$\sigma_{\rm w}^f$ (MHz)	B (MHz)	D _J (kHz)	H_{J} (Hz)	L _J (mHz)	P _J (µHz)	P ₁₂ (nHz)	P ₁₄ (pHz)	P_{16} (fHz)	P ₁₈ (aHz)
¹³ CH ₃ NCO		X106:	0.761	12	0	0.884	4250.21(17)	99.32(38)	90.73(40)	-53.67(21)	20.425(44)	[-4.49]	[0.438]	:	:
$w_b = w, m = 3$ $CH_3 NCO$ $CH_3 N^{13} CO$ $^{13} CH_3 NCO$		V017: V017: V017: V017:	0.562 0.562 0.637 0.625	, A2): E 38 29 23	2 1 2 0	1.223 0.708 0.655	4361.0342(19) 4346.9119(11) 4239.94822(81)	0.6664(66) 0.6646(24) 0.7203(12)	0.0255(94) 0.0166(21) 0.00654(55)	0.0843(57) 0.08616(60) [0.0843]	-0.0269(12) [-0.0269] [-0.0269]	: : :	: : :	: : :	: : :
CH ₃ NCO CH ₃ N ¹³ CO ¹³ CH ₃ NCO	7	V137: V137: V137:	0.654 0.773 0.711	37 31 26	0 0 5	1.852 1.204 0.753	4376.0833(78) 4361.8394(91) 4253.8242(60)	-11.011(83) -10.80(10) -9.695(71)	3.34(41) 3.54(55) 2.35(35)	-25.5(11) -25.5(15) -20.71(93)	21.8(17) 21.7(25) 18.5(13)	-17.3(16) -16.6(25) -15.6(11)	11.56(91) 10.7(15) 10.42(54)	-4.15(27) -3.69(50) -3.82(10)	0.576(34) 0.492(67) [0.576]
CH ₃ NCO CH ₃ N ¹³ CO ¹³ CH ₃ NCO	-	V082: V082: V082:	0.485 0.554 0.604	39 30 25	0 0 0	1.398 0.912 1.053	4362.15134(58) 4348.05955(44) 4241.26821(63)	1.68176(26) 1.67676(21) 1.64066(37)	: : :	: : :	: : :	: : :	: : :	: : :	: : :
CH ₃ NCO CH ₃ N ¹³ CO ¹³ CH ₃ NCO	0	v030: V030: V030:	0.399 0.460 0.465	35 27 23	<i>c</i> , 0, 0	1.347 1.060 1.102	4369.5669(10) 4355.46916(55) 4248.58309(71)	1.9458(12) 1.93746(26) 1.87900(41)	-0.00228(41) [-0.00228] [-0.00228]	: : :	: : :	: : :	: : :	: : :	: : :
CH ₃ NCO CH ₃ N ¹³ CO ¹³ CH ₃ NCO	с с с с	V040: V040: V040:	0.634 0.774 0.691	33 25 23	0 n n	1.276 0.635 0.787	4378.65685(60) 4364.54115(35) 4257.49302(52)	2.06406(30) 2.05499(20) 1.98668(35)	:::	: : :	: : :	: : :	: : :	: : :	: : :
$v_{\rm b} = 0, m = 3$	U (K =	0, J = 1	$1, 0 = A_2,$, A1): E	₇₅ 275	0.7 cm ⁻¹									
CH ₃ NCO CH ₃ N ¹³ CO ¹³ CH ₃ NCO	0	V018: V018: V018:	0.546 0.672 0.682	39 30 25	1 0 2	0.925 0.607 0.857	4361.4499(14) 4347.33135(97) 4240.3532(10)	0.5247(49) 0.5270(20) 0.5908(15)	0.0221(69) 0.0145(17) 0.00154(67)	0.0949(42) 0.09635(50) [0.0949]	-0.02898(93) [-0.02898] [-0.02898]	: : :	: : :	: : :	: : :
CH ₃ NCO CH ₃ N ¹³ CO ¹³ CH ₃ NCO		V020: V020: V020:	0.496 0.583 0.572	36 28 24	с г с	0.888 0.842 0.674	4363.55034(68) 4349.45678(41) 4242.61566(82)	1.72074(80) 1.71616(20) 1.6725(12)	-0.00110(27) [-0.00110] -0.00365(55)			: : :	: : :	: : :	: : :
CH ₃ NCO CH ₃ N ¹³ CO ¹³ CH ₃ NCO	2	v029: V029: V029:	0.406 0.457 0.466	34 27 24	4 0 0	1.138 0.769 0.791	4369.57115(94) 4355.47169(39) 4248.58532(51)	1.9779(10) 1.96867(19) 1.90805(29)	0.00072(36) [0.00072] [0.00072]	; : :	ŝ	:::	: : :	: : :	: : :
$v_{\rm b} = 0, m = 4$	$E: E_{\rm vib}$, = 140.4	6 cm ⁻¹												
CH ₃ NCO CH ₃ N ¹³ CO ¹³ CH ₃ NCO	0	V025: V025: V025:	0.420 0.516 0.525	37 28 24	~ 0 ~	1.039 0.658 0.864	4368.7341(11) 4354.59556(64) 4247.4371(10)	1.5662(25) 1.56131(78) 1.5348(15)	0.0039(21) 0.00303(28) -0.00030(71)	0.00218(57) [0.00218] [0.00218]	÷÷÷	E E E	:::	: : :	: : :
$v_{\rm b} = 0, m = -$ CH ₃ NCO	-5 E: E,	vib = 21. V142:	7.5 cm⁻¹ 0.300	38	-	0.973	4377.7029(11)	1.8930(23)	-0.0136(19)	0.00256(51)	÷			:	÷

 Table 10 (continued)

CH₃NCO

Table 10 continued on next page

	P_{J} (μ H
	L _J (mHz)
(continued)	H_{J} (Hz)
Table 10	D_{J} (kHz)
	B (MHz)

1Vre		rej.	(MHZ)	B (MHZ)	<i>D_J</i> (KHz)	H_{J} (Hz)	L _J (mHz)	P_{J} (μ Hz)	P_{12} (nHz)	P_{14} (pHz)	P_{16} (fHz)	P_{18} (aHz)
) 0.838 2	0 0.838 2	0.838 2	1	1363.54378(81)	1.88911(98)	-0.01291(35)	[0.00256]	:	:	:	:	:
0.958 42	0 0.958 42	0.958 42	4	56.1828(12)	1.8473(18)	-0.00969(79)	[0.00256]	÷	:	:	:	:
0.991 433	0 0.991 433	0.991 433	433.	5.3010(15)	7.8017(47)	0.2011(63)	0.0443(37)	-0.00561(76)	:	:	:	:
0.985 432	0 0.985 432	0.985 432	432	1.35444(91)	7.7351(11)	0.19638(40)	[0.0443]	[-0.00561]	:	÷	:	:
1.101 4215	0 1.101 4215	1.101 4215	4215	.5503(20)	7.2894(49)	0.1797(49)	0.0413(16)	[-0.00561]	:	:	:	:
2 1.037 4297.	2 1.037 4297.	1.037 4297.	4297.	1113(11)	3.2544(22)	0.0971(17)	-0.00294(44)	÷	:	:	:	:
0.843 4283	0 0.843 4283	0.843 4283	4283	.37801(40)	3.23616(18)	[0.0971]	[-0.00294]	:	:	:	:	:
1 1.333 4179	1 1.333 4179	1.333 4179	4179	.1157(15)	3.0893(21)	0.08971(90)	[-0.00294]	÷	:	:	:	:
3 0.969 4377	3 0.969 4377	0.969 4377	4377	7.6225(11)	4.2356(24)	0.0116(19)	0.00223(53)	÷	:	:	:	:
0 1.070 436	0 1.070 436	1.070 4363	436	3.45460(50)	4.20678(24)	[0.0116]	[0.00223]	:	:	:	:	:
1.195 425	0 1.195 425	1.195 425	425	5.9987(14)	4.0113(20)	0.01325(92)	[0.00223]	:	:	:	:	:

The value of the K_a or K quantum number following the notation described in Cernicharo et al. (2016).

 $b\ {\rm The}$ identifier of the line sequence as defined in Cernicharo et al. (2016).

 $^{\ensuremath{\mathcal{C}}}$ Intensity of the sequence relative to the strongest sequence in the spectrum.

d The number of lines in the linear fit.

 $^{\ell}$ The number of confidently assigned lines that were perturbed and were rejected from the fit.

f Unitless (weighted) deviation of the fit.

 ${\boldsymbol{g}}$ Square brackets represent the value fixed to that for the parent species.

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Table 11. JPL catalog line list for $CH_3N^{13}CO$.

Frequency	Error	Log(Int) ^a	DR ^b	$E_{\rm low}$ ^c	$g_{ m upp}{}^{d}$	TAG ^e	QNFMT ^f		(QN' g					QN″ ^h		
(MHz)	(MHz)			(cm^{-1})				J'	K'_a	K_c'	m'	$v_{\rm b}^\prime$	$J^{\prime\prime}$	$K_a^{\prime\prime}$	$K_c^{\prime\prime}$	$m^{\prime\prime}$	$v_{\rm b}^{\prime\prime}$
60492.530	0.0500	-5.0017	3	6.0737	15	01	505	7	0	7	0	0	6	0	6	0	0
69130.253	0.0500	-4.8323	3	8.0980	17	01	505	8	0	8	0	0	7	0	7	0	0
77766.490	0.0500	-4.6840	3	10.4114	19	01	505	9	0	9	0	0	8	0	8	0	0
86400.869	0.0500	-4.5525	3	13.0138	21	01	505	10	0	10	0	0	9	0	9	0	0
95033.255	0.0500	-4.4348	3	15.9051	23	01	505	11	0	11	0	0	10	0	10	0	0
103663.458	0.0500	-4.3284	3	19.0854	25	01	505	12	0	12	0	0	11	0	11	0	0

^a Base 10 logarithm of the integrated intensity at 300 K in units of nm² MHz.

^b Degrees of freedom in the rotational partition function.

^c Upper state degeneracy.

^d Energy of the lower level taken from the parent species from Cernicharo et al. (2016).

^e Species tag or molecular identifier.

^{*f*} Format of the quantum numbers.

^g Quantum numbers for the upper state. K_a and K_c quantum numbers are defined only for m = 0 state. For m > 0: $K_a = K$ and $K_c = 0$.

^h Quantum numbers for the lower state. K_a and K_c quantum numbers are defined only for m = 0 state. For m > 0: $K_a = K$ and $K_c = 0$.

Note—The key parameters used in the evaluation of this table, $\mu = \mu_a = 2.882$ D, T = 300 K, and $Q_{vr} = 138369$, were taken from the parent species. See the sections A.1 and A.2 in Cernicharo et al. (2016) for details. This table is published in its entirety in the machine readable format. A portion is shown here for guidance regarding its form and content.

4. CONCLUSIONS

A detailed rotational study of interstellar CH₃NCO is reported on the basis of Stark-modulation and frequency-modulation microwave and millimeter wave techniques. Around 2500 new transitions were assigned and measured for the parent isotopic species of CH₃NCO, extending the coverage of its rotational spectrum up to K = 10. In addition, more than 2400 $K \leq 3$ transitions were identified for the CH₃N¹³CO and ¹³CH₃NCO isotopic species. The transition frequencies reported in Tables 2, 4, 6, and 7 as well as in the JPL catalog line lists (Tables 5, 11, and 12) are almost all actual experimental frequencies. Only in exceptional cases of a frequency measurement missing for technical reasons or due to line blending the power expansion fits from Tables 1, 3, and 10 were used to interpolate between the experimentally measured lines. Although some of the expansion fits may show useful predictive behavior outside the present data region, such extrapolations would only be deemed moderately safe for the most well behaved line sequences. The criteria for this would be the shortest number of expansion terms in fits based on Eq. 1 and where the effective value of the D_J parameter is close to the 2 kHz magnitude determined in the actual joint K = 0, 1, 2 fits in Table 8. Significant deviation from these conditions indicates increased curvature of the line sequence resulting from considerable vibration-rotation coupling and in such cases extrapolation is not recommended. In summary, for the extremely challenging spectroscopic case of the CH₃NCO molecule all transitions considered in future interstellar detections should be checked against the actual experimental measured laboratory frequencies. The line lists compiled in this work provide a firm basis for further astrophysical searches of CH₃NCO and make it possible to account for a considerable number of additional U-lines in the current sensitive broadband surveys.

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Table 12. JPL catalog line list for ¹³CH₃NCO.

Frequency	Error	Log(Int) ^a	DR ^b	$E_{\rm low}$ ^c	$g_{ m upp}{}^{d}$	TAG ^e	QNFMT ^f			QN' g					QN'' ^h	1	
(MHz)	(MHz)			(cm^{-1})				J'	K'_a	K_c'	m'	$v_{\rm b}^\prime$	$J^{\prime\prime}$	$K_a^{\prime\prime}$	$K_c^{\prime\prime}$	$m^{\prime\prime}$	$v_{\rm b}^{\prime\prime}$
59010.994	0.0500	-5.0017	3	6.0737	15	01	505	7	0	7	0	0	6	0	6	0	0
67437.329	0.0500	-4.8323	3	8.0980	17	01	505	8	0	8	0	0	7	0	7	0	0
75862.078	0.0500	-4.6840	3	10.4114	19	01	505	9	0	9	0	0	8	0	8	0	0
84285.279	0.0500	-4.5525	3	13.0138	21	01	505	10	0	10	0	0	9	0	9	0	0
92706.543	0.0500	-4.4348	3	15.9051	23	01	505	11	0	11	0	0	10	0	10	0	0
101125.795	0.0500	-4.3284	3	19.0854	25	01	505	12	0	12	0	0	11	0	11	0	0

^a Base 10 logarithm of the integrated intensity at 300 K in units of nm² MHz.

^b Degrees of freedom in the rotational partition function.

^c Upper state degeneracy.

^d Energy of the lower level taken from the parent species from Cernicharo et al. (2016).

^e Species tag or molecular identifier.

f Format of the quantum numbers.

^g Quantum numbers for the upper state. K_a and K_c quantum numbers are defined only for m = 0 state. For m > 0: $K_a = K$ and $K_c = 0$.

^h Quantum numbers for the lower state. K_a and K_c quantum numbers are defined only for m = 0 state. For m > 0: $K_a = K$ and $K_c = 0$.

Nore—The key parameters used in the evaluation of this table, $\mu = \mu_a = 2.882$ D, T = 300 K, and $Q_{vr} = 138369$, were taken from the parent species. See the sections A.1 and A.2 in Cernicharo et al. (2016) for details. This table is published in its entirety in the machine readable format. A portion is shown here for guidance regarding its form and content.

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REFERENCES

Altwarg K. Balagar H. Parthaliar I. L. at al. 2017 MNDAS	Ilunchin V.V. Kisial 7. Dezazáłkowski I. Mädar H. &
Aitwegg, K., Baisigei, H., Beithener, J. J., et al. 2017, MINKAS,	ilyusiiii, v. v., Kisiei, Z., Fszczołkowski, L., Madel, H., &
469, \$130, doi: 10.1093/mnras/stx1415	Hougen, J. T. 2010, J. Mol. Spectrosc., 259, 26,
Belloche, A., Meshcheryakov, A. A., Garrod, R. T., et al. 2017,	doi: 10.1016/j.jms.2009.10.005
A&A, 601, A49, doi: 10.1051/0004-6361/201629724	Kisiel, Z., Fortman, S., Medvedev, I., et al. 2010, 65th International
Cernicharo, J., Kisiel, Z., Tercero, B., et al. 2016, A&A, 587, L4	Symposium on Molecular Spectroscopy
Curl, R. F., Rao, V. M., Sastry, K. V. L. N., & Hodgeson, J. A.	Kisiel, Z., Kolesniková, L., Alonso, J., et al. 2015, 70th
1963, J. Chem. Phys., 39, 3335, doi: 10.1063/1.1734198	International Symposium on Molecular Spectroscopy,
Daly, A., Kolesniková, L., Mata, S., & Alonso, J. 2014, J. Mol.	doi: http://hdl.handle.net/2142/79156
Spectrosc., 306, 11,	Kisiel, Z., Pszczółkowski, L., Medvedev, I. R., et al. 2005, J. Mol.
doi: http://dx.doi.org/10.1016/j.jms.2014.10.003	Spectrosc., 233, 231,
Goesmann, F., Rosenbauer, H., Bredehöft, J. H., et al. 2015,	doi: http://dx.doi.org/10.1016/j.jms.2005.07.006
Science, 349, doi: 10.1126/science.aab0689	Kisiel, Z., Pszczółkowski, L., Drouin, B. J., et al. 2012, J. Mol.
Halfen, D. T., Ilyushin, V. V., & Ziurys, L. M. 2015, ApJL, 812, L5	Spectrosc., 280, 134, doi: 10.1016/j.jms.2012.06.013
Han, DH., Pearson, P. G., & Baillie, T. A. 1989, Journal of	Kolesniková, L., Alonso, E. R., Mata, S., & Alonso, J. L. 2017,
Labelled Compounds and Radiopharmaceuticals, 27, 1371,	ApJS, 229, 26, doi: 10.3847/1538-4365/aa5d13
doi: 10.1002/jlcr.2580271204	Koput, J. 1984, J. Mol. Spectrosc., 106, 12,
Ilyushin, V. V., Alekseev, E. A., Kisiel, Z., & Pszczółkowski, L.	doi: https://doi.org/10.1016/0022-2852(84)90078-X
2017, J. Mol. Spectrosc., 339, 31,	1986, J. Mol. Spectrosc., 115, 131,
doi: 10.1016/j.jms.2017.01.005	doi: https://doi.org/10.1016/0022-2852(86)90281-X

—. 1988, J. Mol. Spectrosc., 127, 51, doi: https://doi.org/10.1016/0022-2852(88)90007-0

Lett, R. G., & Flygare, W. H. 1967, J. Chem. Phys., 47, 4730, doi: 10.1063/1.1701692

- Ligterink, N. F. W., Terwisscha van Scheltinga, J., Taquet, V., et al. 2018, MNRAS, 480, 3628, doi: 10.1093/mnras/sty2066
- Ligterink, N. F. W., Coutens, A., Kofman, V., et al. 2017, MNRAS, 469, 2219, doi: 10.1093/mnras/stx890
- Lin, C. C., & Swalen, J. D. 1959, Rev. Mod. Phys., 31, 841, doi: 10.1103/RevModPhys.31.841
- Majumdar, L., Loison, J.-C., Ruaud, M., et al. 2018, MNRAS, 473, L59, doi: 10.1093/mnrasl/slx157
- Martin-Domenech, R., Rivilla, V. M., Jimenez-Serra, I., et al. 2017, MNRAS, 469, 2230, doi: 10.1093/mnras/stx915

- Maté, B., Molpeceres, G., Tanarro, I., et al. 2018, ApJ, 861, 61
- Maté, B., Molpeceres, G., Timón, V., et al. 2017, Monthly Notices of the Royal Astronomical Society, 470, 4222, doi: 10.1093/mnras/stx1461
- Medvedev, I., Winnewisser, M., De Lucia, F., et al. 2004, J. Mol. Spectrosc., 228, 314, doi: 10.1016/j.jms.2004.06.011
- Petkie, D., Goyette, T., Bettens, R., et al. 1997, Rev. Sci. Instrum., 68, 1675, doi: 10.1063/1.1147970
- Quénard, D., Jiménez-Serra, I., Viti, S., Holdship, J., & Coutens, A. 2018, MNRAS, 474, 2796, doi: 10.1093/mnras/stx2960
- Tercero, B., Cernicharo, J., Pardo, J. R., & Goicoechea, J. R. 2010, A&A, 517, A96, doi: 10.1051/0004-6361/200913501
- Winnewisser, B. P., Winnewisser, M., Medvedev, I. R., et al. 2010, PCCP, 12, 8158, doi: 10.1039/b922023b