

Effects of cadmium, inorganic mercury and methyl-mercury on the physiology and metabolomic profiles of shoots of the macrophyte Elodea nuttallii

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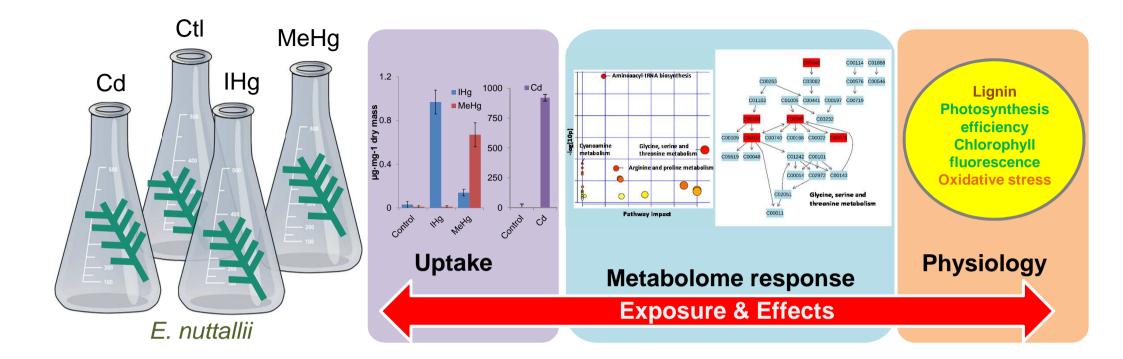
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	Journal Pre-proof
1	Effects of cadmium, inorganic mercury and methyl-mercury on the physiology and metabolomic
2	profiles of shoots of the macrophyte Elodea nuttallii
3	
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13 ABSTRACT

Macrophytes are known to bioaccumulate metals, but a thorough understanding of tolerance strategies and molecular impact of metals in aquatic plants is still lacking. The present study aimed to compare Hg and Cd effects in a representative macrophyte, *Elodea nuttallii* using physiological endpoints and metabolite profiles in shoots and cytosol.

18 Exposure 24h to methyl-Hg (23 ng·L⁻¹), inorganic Hg (70 ng·L⁻¹) and Cd (281 μ g·L⁻¹) did not affect 19 photosynthesis, or antioxidant enzymes despite the significant accumulation of metals, confirming a 20 sublethal stress level. In shoots, Cd resulted in a higher level of regulation of metabolites than MeHg, 21 while MeHg resulted in the largest number of regulated metabolites and IHg treatment regulated no 22 metabolites significantly. In cytosol, Cd regulated more metabolites than IHg and only arginine, 23 histidine and mannose were reduced by MeHg exposure. Kyoto Encyclopedia of Genes and Genomes 24 (KEGG) pathway analysis of data suggested that exposure to MeHg resulted in biochemical changes 25 including aminoacyl-tRNA biosynthesis, glycine, serine and threonine metabolism, nitrogen 26 metabolism, arginine and proline metabolism, cyanoamino acid metabolism, while the treatment of 27 Cd stress caused significant variations in aminoacyl-tRNA biosynthesis and branched-chain amino 28 acids pathways. Data supports an impact of MeHg on N homeostasis, while Cd resulted in an osmotic 29 stress-like pattern and IHg had a low impact. Marked differences in the responses to MeHg and IHg 30 exposure were evidenced, supporting different molecular toxicity pathways and main impact of 31 MeHg on non-soluble compartment, while main impact of IHg was on soluble compartment. 32 Metabolomics was used for the first time in this species and proved to be very useful to confirm and 33 complement recent knowledge gained by transcriptomics and proteomics, highlighting the high 34 interest of multi-omics approaches to identify early impact of environmental pollution.

35

36 **Keywords:** amino acids, cytosol, macrophyte, shoots, sugars.

37

38 Capsule: Metabolomics reveal different cellular toxicity responses to sublethal Cd, MeHg and IHg

39 treatment in Elodea nuttallii

41 **1. INTRODUCTION**

42 Anthropogenic activities result in environmental contamination by mercury (Hg) and cadmium (Cd) (Loizeau et al., 2013; Azimi and Rocher, 2016). As a result, biota is affected by the toxicity of these 43 44 toxic metals, which can also affect human populations via contaminated food (Croteau et al., 2005; 45 Bravo et al., 2010; Niane et al., 2014). Metals can enter the food chain through primary producers, 46 including macrophytes, which represent a key group within communities of shallow water, in terms of biomass, primary production, food source and shelter for fauna. Importantly, macrophytes are 47 48 also central to oxygen, pollutants and nutriments biogeochemical cycles (Wetzel, 2001; Chambers et 49 al., 2008; Bonanno and Lo Giudice, 2010; Noges et al., 2010). High levels of Cd, inorganic Hg (IHg) 50 and methyl-Hg (MeHg) can be observed in aquatic ecosystems colonized by plants (Samecka-Cymerman and Kempers, 2003; Castro et al., 2009; Regier et al., 2013b). Therefore, bioaccumulation 51 52 of metals in macrophytes represents a significant and current environmental concern (Beauvais-Fluck et al., 2016; Beauvais-Fluck et al., 2017). 53

54 Due to their ability to accumulate and tolerate high concentrations of metals (Regier et al., 2013b; Cosio et al., 2014), aquatic plants are often seen as good candidates for phytoremediation, 55 biomonitoring and ecotoxicology investigations (Valega et al., 2009; Bonanno and Lo Giudice, 2010; 56 57 Cosio et al., 2014). However, physiological mechanisms associated with metal accumulation and prevention of their toxicity are still poorly understood in those plants, preventing the full 58 59 optimization of their use for environmental applications. Increasing the knowledge on the biology and physiology of these plants, and more particularly the impact of metal accumulation on their 60 metabolism is therefore a research priority (Cosio et al., 2014). 61

62 Elodea nuttallii is a macrophyte widely distributed in temperate regions, and it is well-known for its 63 accumulative capacity of metals (Larras et al., 2013; Regier et al., 2013b). Several recent studies used transcriptomics and proteomics approaches to describe cellular toxicity of Cd and Hg in E. nuttallii, 64 65 with the aim of using this plant for active biomonitoring. Data revealed rapid molecular changes in metal-exposed plants (Larras et al., 2013; Regier et al., 2013a; Regier et al., 2013b; Regier et al., 66 2016; Beauvais-Fluck et al., 2018a; Beauvais-Fluck et al., 2018b). These studies, spanning a wide 67 68 range of concentrations, globally suggest toxic metals to impact genes and proteins involved in cell 69 structure (lignin content), energy metabolism (sugar and photosynthesis), amino acid metabolism, 70 and to trigger anti-oxidative stress responses (Larras et al., 2013; Regier et al., 2013a; Regier et al., 71 2013b; Regier et al., 2016; Beauvais-Fluck et al., 2018b). Besides, several differences were observed between IHg and MeHg responses. At the transcriptome level, MeHg deregulated more genes than 72 73 IHg at similar intracellular concentrations (Beauvais-Fluck et al., 2018b). At the proteome level, 200 $ng \cdot L^{-1}$ IHg had a significant and similar effect than 500 $\mu g \cdot L^{-1}$ Cd exposure, while 30 $ng \cdot L^{-1}$ MeHg was 74

similar to controls, showing non-significant changes in proteome (Larras et al., 2013). Data thus
suggest different cellular toxicity pathways for IHg and MeHg. However, for both transcriptomic and
proteomic, many regulated contigs or peptides had an unknown function indicating considerable
potential for new discoveries in the biology of toxic metals.

79 Views obtained from transcriptomics, proteomics and physiological studies depict distinct levels of 80 regulation that exist in the organism. Transcriptomic informs on the level of expression of genes, but 81 cellular functions of a particular gene are carried out by its protein. Regulation of proteins in turn 82 results in alterations in levels of small metabolites involved in biochemical reactions in the functional 83 organism. Gene regulation might happen to maintain protein levels, whereas changes in the abundance of proteins can be related to posttranslational modifications of proteins. Eventually, 84 proteins carry out a myriad of functions within the cell, including the metabolism of small molecules 85 86 such as amino acids and sugars, known to have functional roles in plant abiotic stress tolerance and signaling (Yancey, 2005). Physiological adjustments and degradation of pollutants is thus ultimately 87 88 sustained by the metabolic network, whose production of energy, and primary and secondary 89 metabolites, are central for physiological repair and remodeling of cells. In this context, 90 metabolomic enables a large physiological representation of the effect of pollutants by incorporating 91 metabolic fluxes, enzymatic kinetics, and the resulting action of the stressors at organismal level. 92 Metabolic reconfigurations in plants exposed to pollutants, and more generally to a range of abiotic 93 stressors, result in the up-regulation and down-regulation of specific metabolic functions to sustain 94 biological functions (sustainers) and control/modulate metabolite fluxes (modulators) (Topfer et al., 95 2014; Tohge et al., 2015).

96 In this context, metabolomics give the opportunity to confirm or reveal effects of toxicants in a way 97 that cannot be fully exploited by other molecular approaches (e.g. proteomics and transcriptomics). 98 In the present study, we analyzed the metabolic profiles of cytosol and shoots in *E. nuttallii* exposed 99 to Cd, IHg or MeHg, and their control counterparts, to obtain a comprehensive phenotyping of the 100 effects of sublethal concentrations of these toxicants. Metal uptake and physiological endpoints 101 were also measured. We hypothesized that amino acids and sugars should be associated with the 102 high tolerance of E. nuttallii to toxic metals, and thus, should be measured in higher amounts in 103 metal-exposed plants.

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105 2. EXPERIMENTAL SECTION

106 2.1. Lab-ware and reagents

107 All laboratory material was soaked in 10% v/v HNO₃ (*pro analysis*, Merck, Nyon, Switzerland) 108 followed by two 10% v/v HCl (*pro analysis*, Merck, Nyon, Switzerland) acid baths for \ge 1 week,

thoroughly rinsed with ultrapure water (<18.2 MΩ, MilliQ Direct system, Merck Millipore, Darmstadt, Germany) and dried under a laminar flow hood. MeHg (CH₃HgCl) standard solution (1 g·L⁻ 111 ¹) was obtained from Alfa Aesar (Ward Hill, MA, USA), while IHg (Hg(NO₃)₂) and Cd (Cd(NO₃)₂) standard solutions (1 g·L⁻¹) were obtained from Sigma-Aldrich (Buchs, Switzerland).

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2.2. Plant culture and exposure to toxic metals

Shoots of Elodea nuttallii (Planch.) St. John were collected in Lake Geneva (Switzerland, N46° 16' 115 28.7" E6° 10' 15.7"), and a culture was established and maintained in microcosms as previously 116 described (Regier et al., 2013b). Cultures and experiments were conducted in the laboratory under 117 controlled conditions (20 °C, 16/8 h with 5.84 W m⁻² photosynthetically active radiation [PAR], 1000 118 lux, 20 ± 1 °C). We exposed (or not: control) for 24h in triplicate three 10 cm-long shoots without 119 roots in 1L bechers filled with 1.2 µm filtered Lake Geneva water spiked with 30 ng·L⁻¹ MeHg 120 (nominal concentration 30 $\text{ng}\cdot\text{L}^{-1}$, effective concentration 23 ± 7 $\text{ng}\cdot\text{L}^{-1}$ MeHg), 70 $\text{ng}\cdot\text{L}^{-1}$ IHg (nominal 121 concentration 200 ng·L⁻¹, effective concentration 70 ± 14 ng·L⁻¹ IHg) or 280 μ g·L⁻¹ Cd (nominal 122 concentration 500 μ g·L⁻¹, effective concentration 281 ± 24 μ g·L⁻¹ Cd). 123

124

125 2.3. Hg and Cd uptake

After the 24h-long exposure to toxic metals, shoots were rinsed with 1 mM EDTA for IHg or Cd treatments, and with 1 mM EDTA + 1 mM cysteine for MeHg treatments, to assess intracellular metal concentrations (*i.e.* not extractable portion) (Larras et al., 2013).

For total Hg (THg = IHg + MeHg), lyophilized shoots of *E nuttallii* were analyzed by AMA-254. For MeHg, shoots were ground, freeze-dried, and extracted by HNO₃ leaching/CH₂Cl₂ extraction. Then, they were analyzed by ethylation onto Tenax traps, followed by GC separation and cold vapor atomic fluorescence spectrometry (GC-CV-AFS), as described in Liu et al. (2012). For Cd, shoots were ground, freeze-dried and analyzed by ICP-MS (HP 4500, Agilent) after mineralization into 4 mL HNO₃ s.p. and 1 mL H₂O₂ s.p.

THg and Cd in cytosol (see cytosol extraction protocol below) were measured by cold vapor atomic fluorescence spectroscopy (CV-AFS, 2500 Tekran) and ICP-MS, respectively. MeHg in cytosol was measured by the hydride generation method with cryogenic trapping, gas chromatography and atomic fluorescence spectrometry (CT-GC-AFS) (Stoichev et al., 2004).

For all metal analysis methods, the analytical quality was assured by analyzing certified referencematerial, blanks and several samples twice.

141

142 2.4. Physiological endpoints

143 Chlorophyll content was measured by spectrophotometry (Cosio and Dunand, 2010). Chlorophyll 144 fluorescence allows determining PSII efficiency, and was measured using a Handy PEA fluorimeter 145 (Hansatech Instruments Ltd, Norfolk, UK). Plants were dark adapted for 15 min before illuminating 146 them for 1 sec while recording increase in fluorescence. The performance index PI(abs) was 147 calculated from measured fluorescence. The activities of class III peroxidases (POD) and superoxide 148 dismutases (SOD) were assessed by spectrophotometry (Cosio and Dunand, 2010; Regier et al., 149 2016). Lignin content was measured by thyoglycolic attack (Cosio and Dunand, 2010).

150

151 **2.5. Cytosol extraction**

Fresh plants were prepared as described in Regier et al. (2013b). Briefly, fresh plants were all collected at the same time, and weighed before being ground into 10 mL buffer (0.25M sucrose, 1 mM DTT, 50 mM Tris HCl pH 7.5). The resulting mixture was centrifuged at 500 *g* for 5 min at 4 °C. Supernatant, which contained cytosol, was collected and frozen until further used for analysis of metabolite concentrations. The dry mass of the whole shoots was calculated from the fresh biomass of each shoot weighed before the sap extraction based on previous observations that dry mass represents in average 18.3% of fresh mass for shoots.

159

160 **2.6. Analysis of metabolites**

For metabolomics, shoots of the plants exposed to the different experimental conditions for 24h 161 were all collected at the same time, and immediately frozen in liquid nitrogen. Shoot were kept 162 frozen at -30 °C until analysis. Three independent replicates, consisting of a pool of three shoots, or 163 164 corresponding to the pooled cytosols of three plants (see section 2.5 for the method used for cytosol 165 extraction), were run for each experimental condition. The plant samples were homogenized into 166 600 μL of ice-cold (-20 °C) methanol-chloroform solution (2:1, v:v) using a tungsten-bead beating 167 apparatus (RetschTM MM301, Retsch GmbH, Haan, Germany) at 25 agitations per second for 1.5 min. Then, a volume of 400 µl of ice-cold ultra-pure water was added to each sample, and samples 168 169 were subsequently vortexed and centrifuged at 4000 g for 10 min at 4 °C. A volume of 600 μ L of the 170 upper aqueous phase (which contained polar metabolites) was transferred to new microtubes, and 171 further used for GC-MS and LC assays.

172

173 GC-MS Analyzes

Sixty (cytosol) or 120 (shoots) μL of the upper aqueous phase were transferred to new glass vials,
and these aliquots were vacuum-dried (MiVac, Genevac Ltd., Ipswitch, England) at 32 °C for 45 min.
Then, we used the derivatization process and GC-MS settings described in Khodayari et al., (2013).

Briefly, the dried aliquots were resuspended in 30 μ l of 20 mg·L⁻¹ methoxyamine hydrochloride 177 (Sigma-Aldrich, St. Louis, MO, USA) in pyridine, incubated under automatic orbital shaking at 40 °C 178 179 for 90 min. Subsequently, 30 µl of N-methyl-N-(trimethylsilyl) trifluoroacetamide (MSTFA; Sigma) 180 was added and the derivatization was conducted at 40 °C for 45 min under agitation. Our GC-MS 181 system consisted of a Trace GC Ultra chromatograph and a Trace DSQII quadrupole mass 182 spectrometer (Thermo Fischer Scientific, see Khodayari et al., (2013) for more details). Randomized 183 sample sequences were established for sample injection, chromatograms were deconvoluted using 184 XCalibur v2.0.7. Standard samples, consisting of 62 pure reference compounds at 1, 2, 5, 10, 20, 50, 100, 200, 500, 750, 1000, and 1500 μM were run, and metabolite levels were quantified using the 185 186 quadratic calibration curves for each reference compound.

187

188 LC analyzes

Amino acid analysis was adapted from Renault et al. (2010) and (2016). Briefly, 100 µL of the 189 190 metabolites in solution in the methanol-ultra-pure water extract was vacuum-dried (MiVac, Genevac 191 Ltd., Ipswitch, England) at 32 °C for 30 min, and the pellet was re-suspended into 100 µL of ultra-192 pure water. Then, aliquots (10 µL) of the aqueous extracts were used for amino acids derivatization 193 according to the AccQ•Tag ultra derivatization kit protocol (Waters Corporation, Milford, MA). 194 Amino acids were analyzed using an Acquity UPLC®533 system (Waters Corporation, Milford, MA) by 195 injecting 1 μ L of the derivatization mix onto an Acquity UPLC[®]534 BEH C18 1.7 μ m 2.1 x 100 mm 196 column heated at 55 °C. External standards were run and used for the drawing of calibration curves, 197 further allowing the quantification of the metabolites from the samples.

198

199 **2.7. Data Analysis**

200 Principal component analysis (PCA) was performed to identify patterns in response of variables and 201 individuals, and compare the responses of the plants exposed to the three metal treatments, and 202 the control. PCA were conducted with the packages FactoMineR (http://factominer.free.fr/) and 203 ade4 (http://pbil.univ-lyon1.fr/ade4) in the R software (http://www.r-project.org). Normality and 204 homoscedasticity were tested, and a student's t-test was used in Excel (Microsoft, Redmond, WA, 205 USA) to determine if two sets of data were significantly different. For the metabolic pathway analysis, the name of compounds that exhibited significant variations were analyzed in 206 207 MetaboAnalyst (http://www.metaboanalyst.ca). The library of metabolic pathways was assembled 208 from the KEGG database, and the significance of the pathway name was assessed with the Holm-209 Bonferroni method.

211 **3. RESULTS**

3.1. Bioaccumulation of metals in the plants and effects on physiological endpoints

The 24h-exposure of *E. nuttallii* to Cd, IHg and MeHg resulted in a significant bioaccumulation of metals in shoots (Table 1). Yet, this accumulation had no significant impact on measured physiological endpoints: lignin and chlorophyll contents, photosynthesis, POD and SOD activities, whose values were non-significantly different from control plants (Table 1).

217

218 **3.2. Metabolites in shoots**

219 We investigated the impact of short sublethal exposures to Cd, IHg and MeHg on the metabolic 220 phenotypes of shoots of E. nuttallii (Table S1). In whole shoots, Cd treatment increased the 221 concentrations of 11 amino acids, two sugars, one polyol (adonitol) and one organic acid (pipecolic 222 acid), and reduced the amounts of two amino acids (aspartatic acid, methylcysteine), relative to control plants. Most strongly up-regulated amino acids by Cd included leucine, isoleucine and 223 224 tryptophan showing a 16× increase, and tyrosine as well as valine showing 11× and 10× increase, 225 respectively, after 24h of exposure to metals. Aspartatic acid and methylcysteine were reduced 226 0.05× and 0.3×, respectively by the 24h exposure to Cd. Similarly, MeHg exposure increased the 227 amounts of ammonium and of 13 amino acids, and reduced the concentration of one polyol (arabitol 228 0.5×) (Table 2). Most strongly up-regulated amino acids included BABA (4×), tryptophan and 229 methionine (2×). Exposure to IHg had no significant impact on concentrations of the measured 230 metabolites in shoots, as compared with control plants.

231 Principal component analysis (PCA) was performed to compare metabolic profiles among 232 treatments: 52% of the total variance was explained on axes 1 and 2 (Figure 1A and S1). In this PCA, 233 Cd-exposed shoots showed the most distinct pattern, while centroids of IHg and MeHg-treated 234 shoots were relatively similar to controls. Axis 1 was mainly constructed by the variations of 235 adonitol, maltose, isoleucine, valine, tryptophan, tyrosine and leucine. Axis 2 was mainly supported by the variations of galactonolactone, glycerol, phosphoric acid, galactose, asparagine, glucose-6-236 237 phosphate and glyceric acid. In shoots, the metabolic pathway analysis performed thanks to KEGG 238 databases revealed two biological metabolic pathways regulated by Cd: aminoacyl-tRNA 239 biosynthesis, and valine, leucine and isoleucine biosynthesis. IHg had no significant impact at the level of pathways, while MeHg significantly impacted five metabolic pathways: aminoacyl-tRNA 240 241 biosynthesis, glycine, serine and threonine metabolism, nitrogen metabolism, arginine and proline 242 metabolism, cyanoamino acid metabolism.

243

3.3. Metabolites in the cytosol

245 Because amino acid and sugars transfer happens at the cellular level, we also investigated the 246 impact of short sublethal exposures to Cd, IHg and MeHg on the metabolic phenotypes in the cytosol 247 of E. nuttallii. Many amino acids are synthesized in the chloroplast and transported into the cytosol 248 for protein and secondary metabolite synthesis, or transported and stored in the vacuole. Similarly, 249 the transport of sugars and distribution of metabolites between compartments is coordinated in 250 plants to drive growth and development. In the cytosol of plants, treatments significantly modified 251 the concentration of several metabolites (Table 3): amounts of ammonium $(1.7\times)$, glycine $(2.2\times)$, 252 hydroxyproline $(1.6\times)$ and phosphoric acid $(2.2\times)$ were increased by exposure to Cd. Concentrations 253 of alpha-alanine (0.6×), aspartic acid (0.5×), galactose (0.4×), and mannose (0.3×) were reduced in 254 Cd-exposed plants. Ammonium concentration rose by 1.6× in plants exposed to IHg, while this 255 exposure reduced the amount of arginine $(0.2\times)$. In MeHg-exposed plants, changes of the 256 concentration of individual amino acids, relative to controls, were significantly reduced for histidine (completely depleted) and arginine (0.2×). Regarding sugars, mannose was also reduced by IHg 257 258 $(0.6\times)$ and MeHg $(0.3\times)$, and galactose by MeHg $(0.3\times)$. No significant variation was reported at the 259 level of the metabolic pathways in the cytosol of the treated plants, as compared with control 260 plants.

261 Principal component analysis of metabolites in cytosol explained 60% of the variance of samples on 262 the first and second axes (Figure 1B and S2). Plants exposed to Cd showed the most distinct 263 metabolic phenotypes relative to control plants on both axes 1 and 2. While MeHg-treated plants 264 were separated from controls on axis 2, the centroid of IHg-treated plants was close to the one of their control counterparts. The first axis of the PCA was mainly supported by the variations of 265 266 tryptophan, valine, proline, glutamic acid, phenylalanine, GABA and threonine. The second axis was 267 constructed by the variations of glucose, galactose, mannose, alpha-alanine, inositol, aspartic acid 268 and fructose.

269

4. DISCUSSION

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4.1. Bioaccumulation and physiological endpoints

Both bioaccumulation and physiological endpoint analyses confirmed that metal concentrations used here resulted in their significant bioaccumulation, with sublethal effects for *E. nuttallii*, as earlier reported in other studies (Larras et al., 2013; Regier et al., 2013b; Beauvais-Fluck et al., 2018b). Chosen concentrations of metal in our work are representing a compromise between high concentrations resulting in unambiguous metal accumulation pattern in macrophytes, and concentrations with environmental relevance. Earlier investigations using the same concentrations of Cd (281 μ g·L⁻¹) and IHg (70 ng·L⁻¹) for 7d revealed that root growth was reduced, while lignification

279 of stems was increased in E. nuttallii (Larras et al., 2013). Nevertheless, these treatments had no effects on chlorophyll content and shoot growth (Larras et al., 2013; Regier et al., 2013b). 280 Conversely, exposure to MeHg (23 $ng L^{-1}$) for 7d increased root growth, suggesting a hormesis effect 281 282 that is seen as an overcompensation of a moderate stress (Larras et al., 2013). Similarly in another study, exposure for 2h to 10 ng·L⁻¹ or 10 μ g·L⁻¹ MeHg had no significant effect on chlorophyll content, 283 POD and SOD activities in E. nuttallii (Beauvais-Fluck et al., 2018b). In sum, consistent with the 284 literature and based on the absence of effects of the physiological endpoints, our experimental 285 286 conditions did not induce an acute stress in E. nuttallii shoots. Eventually, these data confirmed that 287 E. nuttallii is highly tolerant to metals.

288

289 **4.2. Effects of metals on the metabolic phenotypes of macrophytes**

290 We were interested in obtaining a general picture of effects of toxic metals on the physiology of the plants by comparing their metabolic signatures. Our metabolic profiling revealed that all treatments 291 292 impacted metabolite amounts in shoots, cytosol or both, as compared with control plants. The Cd-293 treated plants were characterized by the most distinct metabolic phenotypes according to PCA, in 294 both shoots and cytosol. This finding highlights a higher impact of this metal on the amount of each 295 metabolite, as also evidenced by a higher level of regulation (up to 16×) than the one measured in 296 IHg and MeHg treatments. Even if this distinction may partially result from the higher Cd 297 concentration used as compared with IHg and MeHg, we must re-emphasize that we did not find 298 differences among the three treatments and the control for the other physiological endpoints. 299 Moreover, metabolic pathway analysis suggested that MeHg impacted more metabolic pathways in 300 shoots than did Cd in our experimental conditions. A previous proteomic study with identical 301 experimental conditions suggested similar responses, and stress-induced level of IHg and Cd in E. 302 nuttallii, and an absence of response for MeHg (Larras et al., 2013). Conversely, transcriptomic 303 analyses suggested similar stress levels for Cd and IHg and a higher molecular impact of MeHg 304 (Regier et al., 2013a; Beauvais-Fluck et al., 2018b). When examining responses of the two considered 305 compartments (shoots and cytosol), we observed that Cd and MeHg both significantly impacted 306 several pathways in shoots, while IHg had no significant impact on metabolites pathways. For 307 cytosol, Cd had a higher impact than IHg, while MeHg had lower impact than IHg. Both cytosolic compartment analysis and proteomics target soluble molecules, suggesting that IHg might impact 308 309 mainly soluble compartments, while MeHg impacts mainly the non-soluble compartments. However, 310 these differences reflect different molecular toxicity pathways of the three metals, and highlight the 311 interest of coupling analysis at different levels of organization. Indeed, the present study brings a 312 different pattern as compared with those previously observed by transcriptomics. For instance,

313 MeHg was earlier reported as a stronger dysregulator than IHg, by affecting a much higher number 314 of genes in E. nuttallii (Beauvais-Fluck et al., 2018b). In our study, metabolomics confirmed that 315 MeHg has a stronger molecular effect than IHg in shoots, but not in cytosol. In addition, Cd, IHg, and 316 MeHg regulated genes involved in biosynthesis of sulfur-containing amino acids such as methionine 317 and cysteine (Regier et al., 2013a; Beauvais-Fluck et al., 2018a; Beauvais-Fluck et al., 2018b). 318 Methionine was increased 2× by MeHg in shoots supporting its role in MeHg response, while 319 cysteine was not altered after metal exposure in shoots or cytosol, suggesting that regulation at the 320 gene and protein level was mostly happening to maintain the level of this metabolite.

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322

4.3 Early signs of stress responses in *Elodea nuttalli* treated with Cd

Exposure of *E. nuttalli* resulted in a significant re-patterning of the metabolic phenotype, with for 323 324 instance a significant effect on the concentrations of metabolites involved in aminoacyl-tRNA 325 biosynthesis, and on the branched-chain amino acids. Aminoacyl-tRNA biosynthesis is involved in the 326 synthesis of proteins in organisms, and are particularly central to growth (Banerjee et al., 2011; 327 Raina and Ibba, 2014). Most often, the activity of this pathway declines when plants are exposed to 328 stressful or restrictive conditions, as requirements for protein synthesis is reduced to favor the 329 activation of stress-response genes (Holcik and Sonenberg, 2005). In parallel, we also found that Cd 330 treatment impacted the concentration of branched-chain amino acids (i.e. leucine, isoleucine and 331 valine) in shoots of E. nutalli. Branched-chain amino acids share four enzymes for their biosynthesis, 332 which are coordinately regulated. Plants that accumulate and tolerate high Cd amounts, such as Solanum nigrum, have been demonstrated to accumulate isoleucine and valine in roots (Xu et al., 333 334 2012). Branched amino acids derive from the shikimate pathway, and are required for protein 335 synthesis and production of aromatic secondary metabolites, e.g. flavonoids, lignin cell wall components, and anthocyanins, which are important for cell wall extensibility and non-enzymatic 336 337 anti-oxidative response both involved in metal tolerance in plants (Tzin and Galili, 2010; Zemanova et al., 2017). Branched-chain amino acids also serve as precursors for the production of secondary 338 339 metabolites (e.g. alkaloids, glycosides) involved in responses to biotic and abiotic stress (Rizhsky et 340 al., 2004). These secondary metabolites notably play a role during osmotic stress (Zemanova et al., 341 2017). Our metabolite profiling thus suggests that Cd exposure results in an osmotic stress to E. 342 nuttallii, as depicted by the 10× to 16× increase of leucine, isoleucine and valine that typically 343 accumulate in osmotically challenged plants. This hypothesis is further supported by the significant 344 accumulation of proline, a well-known biomarker of water stress in plants. Indeed, proline can act as an osmolyte, a metal chelator, an antioxidative defense molecule and a signaling molecule (Hayat et 345 346 al., 2012).

347 The probable osmotic stress induced by Cd-exposure likely increased the amount of oxidative stress 348 experienced by the plants. Oxidative stress has been shown to alter the activity of the Krebs cycle 349 (Obata and Fernie, 2012), with potential side effects on the pathway connected to the TCA, and thus 350 on the levels of TCA-cycle-derived compounds. This includes aspartic acid, whose levels decreased in 351 shoots (and to a lesser extent in the cytosol) of Cd-treated plants. Besides, we also measured 352 decreased amounts of mannose and galactose in shoots of *E. nuttalli*. These two sugars are involved 353 in the Smirnoff-Wheeler pathway, and their reduction suggests that they were used for the 354 biosynthesis of ascorbic acid. Ascorbic acid is widely recognized as an important ROS scavenger in 355 plants, and earlier studies reported that accumulation or enhanced production of ascorbic acid 356 boosted the ability of plants to cope with salt or drought stress, by limiting the peroxidation of 357 membrane lipids and reducing losses of the chlorophyll content (Venkatesh and Park, 2014). 358 Increased ROS production in Cd-treated plant is very likely, as this is a reported toxic effect of metal exposure at higher concentrations (Kieffer et al., 2009; Xu et al., 2012; Larras et al., 2013). 359

Finally, the accumulation of leucine, isoleucine and valine may serve to promote stress-induced protein synthesis, and may act as signaling molecules to regulate gene expression (Joshi et al., 2010). The breakdown products of these three amino acids, *e.g.* acetyl-CoA, propionyl-CoA, and acetoacetate, are potential energy sources for plants (Joshi et al., 2010). The global increase of numerous amino acids likely depicts increased degradation and synthesis of proteins (Vital et al., 2017). Similar observation was reported in crop cultivars, and shown to be linked to impact of osmotic stress on the proteasome (Vital et al., 2017).

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368 369

4.4. MeHg affects the nitrogen metabolism and triggers chemical defenses against abiotic stress

In macrophytes exposed 24h to MeHg, the amount of several amino acids, including aspartic acid, glutamine, glycine, phenylalanine, proline, threonine and tryptophan were altered in shoots. This finding is congruent with previous transcriptomic studies conducted on *E. nuttallii* exposed to MeHg, which showed that genes involved in synthesis of proteins represented a high proportion of regulated genes (Beauvais-Fluck et al., 2018a). Altogether, data suggest a rapid adjustment of the plants to MeHg exposure. Several of these amino acids are expected to have a role in chemical defenses against abiotic stresses, notably glycine, threonine and proline (as discussed above).

Our metabolomic data, and more particularly the significant rise of glutamine and ammonium amounts, further support that MeHg affects nitrogen (N) metabolism and homeostasis in the plant. The increase of the amount of aspartatic acid and glutamine in MeHg-treated plants, and the concomitant reduction of ammonium amount in cytosol, suggest that N transport is altered. In

381 plants, the main pathway of ammonia assimilation is the GS/GOGAT cycle, where glutamine 382 synthetase / glutamate synthase cooperate (Masclaux-Daubresse et al., 2006). Plants may also 383 utilize the reaction catalyzed by glutamate dehydrogenase for ammonia assimilation, especially under environmental stress conditions (Masclaux-Daubresse et al., 2006). Consequently, in plants, 384 385 aspartic acid, asparagine, glutamic acid and glutamine are the main amino acids transporter of N 386 (Miesak and Coruzzi, 2002). Here, the increase in ammonia and glutamine (high N/C ratio) suggest 387 that nitrate assimilation is affected, and both ammonia and amino acids are released by protein 388 degradation and hydrolysis. Similarly, previous transcriptomic study in E. nuttallii exposed to IHg, MeHg or Cd highlighted a down-regulation of genes coding for nitrate and ammonium transport (e.g. 389 390 high affinity nitrate transporters) suggesting a global impact of tested metals on nutrition (Regier et 391 al., 2013a; Beauvais-Fluck et al., 2018a).

392 Phenylalanine, threonine, and tryptophan are aromatic amino acids coupled with the phenylpropanoid and the shikimate pathways. Phenylpropanoids are a diverse group of compounds 393 394 derived from the carbon skeleton of phenylalanine. These compounds are involved in plant defense 395 and generate cell wall components (e.g. lignin) and antioxidant metabolites (e.g. anthocyanin) 396 (Fraser and Chapple, 2011). In previous study, a significant increase of anthocyanin, acting as non-397 enzymatic antioxidants (Beauvais-Fluck et al., 2018a; Beauvais-Fluck et al., 2018b), was measured in *E. nuttallii* exposed 2h to 10 ng·L⁻¹ MeHg and higher concentrations, but not after IHg treatments 398 (Beauvais-Fluck et al., 2018b). The increase amount of phenylalanine, threonine, and tryptophan 399 400 suggests enhanced antioxidant activity in MeHg-treated plants.

401

402

4.5. Toxic metals had little effects on the concentration of sugars

403 A previous study hypothesized that the accumulation of fructose, galactose, sucrose and glucose in 404 poplar (Populus tremula L.) exposed to Cd might act as osmoprotectants playing critical roles in 405 osmotic adjustments, and protection of cell membranes against toxic trace metals (Kieffer et al., 406 2009). Here, exposures decreased mannose and galactose amounts in cytosol. Both sugars are 407 involved in the ascorbic acid biosynthetic pathway, a major antioxidant in higher plants, and their 408 decrease is thus in line with expected cellular toxicity of metals (Ma et al., 2014). Other studies 409 suggested cellular adjustments to control resources distribution within cells: reduced sugar utilization in the presence of Cd and consequently photoassimilates partial storage as sugars (Moya 410 411 et al., 1995; Bailey et al., 2003). In the aquatic plant Wolffia arrhizal, sugar and protein contents 412 were negatively correlated with toxic metal concentrations (Piotrowska et al., 2010; Zhang et al., 413 2017). In the same line, sugars accumulated in correlation with a decline of net photosynthetic rates 414 in rice exposed to Cd (Moya et al., 1995). Similarly, amino acid metabolism often is expected to be

415 closely related to abiotic stress tolerance (Zhang et al., 2017). However, in our study few sugars 416 varied between treated and control plants, both in cytosol and shoots. Nevertheless, we observed 417 an increase of sugars in shoots, while their content concomitantly decreased in cytosol. The 418 transport of sugars and distribution between storage, respiration and biosynthesis in sink tissues are 419 coordinated activities in plants (Rosa et al., 2009a; Lemoine et al., 2013). Stress-related increase in 420 sugar delivery to sinks is important for growth, cell turgor and water potential maintenance 421 (Lemoine et al., 2013). Besides, soluble sugars have several roles in cells, including as metabolic 422 resources, structural constituents of cells, but also in stress signaling (Rosa et al., 2009b). 423 Transporters are required for efficient movement of sugars that are polar solutes across membranes 424 (Patrick et al., 2013). It is likely that multiple sugar transporters operate for fine-tuning of sugar 425 fluxes for homeostasis and interaction with other proteins for sugar sensing and signaling, including 426 efficient unloading into cell wall spaces, uptake of sucrose or transport via the apoplasm. However, metal exposure concentrations used here might be too low to impact significantly sugar metabolic 427 428 pathways. Metabolic data thus support that previously observed regulation of genes and proteins 429 involved in sugar metabolism was mostly happening to maintain the level of those metabolites 430 (Regier et al., 2013a; Beauvais-Fluck et al., 2018a; Beauvais-Fluck et al., 2018b).

431

432 **5. CONCLUSION**

433 Metabolomic analysis showed that exposure of *E. nuttallii* to Cd, IHg and MeHg resulted in different 434 metabolomes in each compartment (shoot and cytosol) with some level of coordinated regulation for metabolites between shoots and cytosol. Metabolomic confirmed previous hypothesis of distinct 435 436 toxicity pathway for IHg and MeHg based on transcriptomics, proteomics and physiology (Larras et 437 al., 2013; Regier et al., 2013a; Beauvais-Fluck et al., 2018a). In contrast to previous proteomics and 438 transcriptomics data, Cd showed a higher impact on individual metabolites, while MeHg impacted 439 more metabolic pathways in shoots, but IHg showed a higher impact in cytosol than MeHg. New 440 omics methods are fundamentally transforming biology approaches, but the fact that many genes 441 and proteins identified by transcriptomics and proteomics have an unknown function results in 442 considerable uncertainty in the analysis of responses. Metabolomics give the opportunity to confirm 443 and reveal effects of toxicants in a complementary way to other omics methods, as it identified early 444 responses that were effective in revealing metal uptake and impact as well as were congruent with 445 expected adverse outcome pathways.

446

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Table 1. Lignin content, chloropyll content, performance index of PSII [PI(abs)], class III peroxidase

[POD] and superoxide dismutase [SOD] activities and bioaccumulation in shoots of Elodea nuttallii

exposed to 280 μ g·L⁻¹ Cd, 70 ng·L⁻¹ IHg, or 30 ng·L⁻¹ MeHg or not (control). For each treatment, three

replicates were analyzed. Data are presented as mean ± standard deviation (SD; n= 3). Bold font = p-

value <0.05 relative to control plants.

	Shoots					В	Bioaccumulation		
	Lignin	Chlorophyll	PI	POD	SOD	IHg	MeHg	Cd	
	content		(abs)						
	(µg∙mg⁻¹	fresh mass)		(nkat⋅mg ⁻¹ ((units∙µg⁻¹	()	(µg⋅g ⁻¹ dry mass)		
				protein)	protein)				
Control	7.2 ± 1.3	0.90 ± 0.27	0.79 ± 0.02	18.4 ± 1.63	0.93 ± 0.08	0.03 ± 0.03	0.01 ± 0.01	0.82 ± 0.12	
Cd	6.4 ± 1.4	0.78 ± 0.30	0.80 ± 0.01	28.5 ± 5.6	1.04 ± 0.06	NA	NA	921 ± 28	
IHg	5.6 ± 1.3	0.75 ± 0.16	0.79 ± 0.02	17.0 ± 1.4	1.08 ± 0.05	0.97 ± 0.11	0.01 ±0.01	NA	
MeHg	5.6 ± 2.2	0.88 ± 0.23	0.80 ± 0.01	18.3 ± 6.03	1.01 ± 0.08	0.14 ± 0.03	0.67 ± 0.11	NA	
616 N	A (not analy	zed).							
617									
618									

Table 2. Concentration (nmoles mg⁻¹ dry mass) of amino acids, organic acids, polyols and sugars

620 (n=3; mean ± SD) whose levels significantly varied in shoots of *E. nuttallii* exposed for 24h to Cd, IHg

and MeHg (bold font = p-value <0.05), relative to control plants.

BABA 0.02 ± 0.01 0.09 ± 0.06 0.07 ± 0.05 0.08 ± 0.06					
Aspartic Acid 15.22 ± 2.79 0.72 ± 0.84 13.41 ± 2.44 21.77 ± 1.7 BABA 0.02 ± 0.01 0.09 ± 0.06 0.07 ± 0.05 0.08 ± 0.06 Glutamine 9.03 ± 2.06 6.73 ± 1.61 8.87 ± 2.13 14.61 ± 4.43 Glycine 1.20 ± 0.20 1.00 ± 1.43 0.97 ± 0.23 1.96 ± 0.30 Histidine 0.38 ± 0.12 1.36 ± 0.45 0.30 ± 0.07 0.47 ± 0.19 Hydroxyproline 0.00 ± 0.00 0.04 ± 0.04 0.01 ± 0.01 0.02 ± 0.01 Isoleucine 0.24 ± 0.17 3.73 ± 1.14 0.15 ± 0.12 0.40 ± 0.24 Leucine 0.17 ± 0.08 2.74 ± 0.70 0.12 ± 0.05 0.28 ± 0.08 Lysine 0.25 ± 0.21 1.22 ± 0.42 0.18 ± 0.18 0.41 ± 0.34 Methionine 0.03 ± 0.01 0.04 ± 0.01 0.03 ± 0.01 0.66 ± 0.01 Methylcysteine 0.10 ± 0.05 0.03 ± 0.03 0.07 ± 0.03 0.11 ± 0.04 Phenylalanine 0.38 ± 0.08 2.15 ± 0.78 0.34 ± 0.03 0.66 ± 0.15 Proline 0.39 ± 0.15 1.93 ± 0.46 0.30 ± 0.13 0.64 ± 0.23 Serine 4.49 ± 1.05 4.64 ± 0.56 4.29 ± 1.03 7.21 ± 1.02 Threonine 1.91 ± 0.37 5.68 ± 1.72 1.73 ± 0.35 3.08 ± 0.47 Tryptophan 0.10 ± 0.02 1.60 ± 0.33 0.12 ± 0.05 0.21 ± 0.03 Tyrosine 0.14 ± 0.04 1.54 ± 0.38 0.12 ± 0.03 0.24 ± 0.04 Adonitol 0.45 ± 0.09 0.75 ± 0.19 $0.59 \pm$		Control	Cd	IHg	MeHg
BABA 0.02 ± 0.01 0.09 ± 0.06 0.07 ± 0.05 0.08 ± 0.06 Glutamine 9.03 ± 2.06 6.73 ± 1.61 8.87 ± 2.13 14.61 ± 4.43 Glycine 1.20 ± 0.20 1.00 ± 1.43 0.97 ± 0.23 1.96 ± 0.30 Histidine 0.38 ± 0.12 1.36 ± 0.45 0.30 ± 0.07 0.47 ± 0.19 Hydroxyproline 0.00 ± 0.00 0.04 ± 0.04 0.01 ± 0.01 0.02 ± 0.01 Isoleucine 0.24 ± 0.17 3.73 ± 1.14 0.15 ± 0.12 0.40 ± 0.24 Leucine 0.17 ± 0.08 2.74 ± 0.70 0.12 ± 0.05 0.28 ± 0.08 Lysine 0.25 ± 0.21 1.22 ± 0.42 0.18 ± 0.18 0.41 ± 0.34 Methionine 0.03 ± 0.01 0.04 ± 0.01 0.03 ± 0.01 0.06 ± 0.01 Methylcysteine 0.10 ± 0.05 0.03 ± 0.03 0.07 ± 0.03 0.11 ± 0.04 Phenylalanine 0.38 ± 0.08 2.15 ± 0.78 0.34 ± 0.03 0.66 ± 0.15 Proline 0.39 ± 0.15 1.93 ± 0.46 0.30 ± 0.13 0.64 ± 0.23 Serine 4.49 ± 1.05 4.64 ± 0.56 4.29 ± 1.03 7.21 ± 1.02 Threonine 1.91 ± 0.37 5.68 ± 1.72 1.73 ± 0.35 3.08 ± 0.47 Tryptophan 0.10 ± 0.02 1.60 ± 0.33 0.12 ± 0.03 0.24 ± 0.03 Valine 0.52 ± 0.28 5.24 ± 1.44 0.37 ± 0.24 0.82 ± 0.41 Adonitol 0.45 ± 0.09 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.08 Arabinose 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 </td <td>Ammonium</td> <td>4.21 ±1.03</td> <td>6.12 ± 2.35</td> <td>3.90 ± 0.89</td> <td>7.72 ± 2.06</td>	Ammonium	4.21 ±1.03	6.12 ± 2.35	3.90 ± 0.89	7.72 ± 2.06
Glutamine 9.03 ± 2.06 6.73 ± 1.61 8.87 ± 2.13 14.61 ± 4.43 Glycine 1.20 ± 0.20 1.00 ± 1.43 0.97 ± 0.23 1.96 ± 0.30 Histidine 0.38 ± 0.12 1.36 ± 0.45 0.30 ± 0.07 0.47 ± 0.19 Hydroxyproline 0.00 ± 0.00 0.04 ± 0.04 0.01 ± 0.01 0.02 ± 0.01 Isoleucine 0.24 ± 0.17 3.73 ± 1.14 0.15 ± 0.12 0.40 ± 0.24 Leucine 0.17 ± 0.08 2.74 ± 0.70 0.12 ± 0.05 0.28 ± 0.08 Lysine 0.25 ± 0.21 1.22 ± 0.42 0.18 ± 0.18 0.41 ± 0.34 Methionine 0.03 ± 0.01 0.04 ± 0.01 0.03 ± 0.01 0.06 ± 0.01 Methylcysteine 0.10 ± 0.05 0.03 ± 0.03 0.07 ± 0.03 0.11 ± 0.04 Phenylalanine 0.38 ± 0.08 2.15 ± 0.78 0.34 ± 0.03 0.66 ± 0.15 Proline 0.39 ± 0.15 1.93 ± 0.46 0.30 ± 0.13 0.64 ± 0.23 Serine 4.49 ± 1.05 4.64 ± 0.56 4.29 ± 1.03 7.21 ± 1.02 Threonine 1.91 ± 0.37 5.68 ± 1.72 1.73 ± 0.35 3.08 ± 0.47 Tryptophan 0.10 ± 0.02 1.60 ± 0.33 0.12 ± 0.03 0.24 ± 0.34 Value 0.52 ± 0.28 5.24 ± 1.44 0.37 ± 0.24 0.82 ± 0.41 Adonitol 0.45 ± 0.09 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.07 Arabinose 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.07 Pipecolic Acid 0.65 ± 0.20 0.95 ± 0.19 $0.59 $	Aspartic Acid	15.22 ± 2.79	0.72 ± 0.84	13.41 ± 2.44	21.77 ± 1.77
Glycine 1.20 ± 0.20 1.00 ± 1.43 0.97 ± 0.23 1.96 ± 0.30 Histidine 0.38 ± 0.12 1.36 ± 0.45 0.30 ± 0.07 0.47 ± 0.19 Hydroxyproline 0.00 ± 0.00 0.04 ± 0.04 0.01 ± 0.01 0.02 ± 0.01 Isoleucine 0.24 ± 0.17 3.73 ± 1.14 0.15 ± 0.12 0.40 ± 0.24 Leucine 0.17 ± 0.08 2.74 ± 0.70 0.12 ± 0.05 0.28 ± 0.08 Lysine 0.25 ± 0.21 1.22 ± 0.42 0.18 ± 0.18 0.41 ± 0.34 Methionine 0.03 ± 0.01 0.04 ± 0.01 0.03 ± 0.01 0.06 ± 0.01 Methylcysteine 0.10 ± 0.05 0.03 ± 0.03 0.07 ± 0.03 0.11 ± 0.04 Phenylalanine 0.38 ± 0.08 2.15 ± 0.78 0.34 ± 0.03 0.66 ± 0.15 Proline 0.39 ± 0.15 1.93 ± 0.46 0.30 ± 0.13 0.64 ± 0.23 Serine 4.49 ± 1.05 4.64 ± 0.56 4.29 ± 1.03 7.21 ± 1.02 Threonine 1.91 ± 0.37 5.68 ± 1.72 1.73 ± 0.35 3.08 ± 0.47 Tryptophan 0.10 ± 0.02 1.60 ± 0.33 0.12 ± 0.05 0.21 ± 0.03 Valine 0.52 ± 0.28 5.24 ± 1.44 0.37 ± 0.24 0.82 ± 0.41 Adonitol 0.45 ± 0.09 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.07 Arabinose 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.08 Pipecolic Acid 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.08	BABA	0.02 ± 0.01	0.09 ± 0.06	0.07 ± 0.05	0.08 ± 0.06
Histidine 0.38 ± 0.12 1.36 ± 0.45 0.30 ± 0.07 0.47 ± 0.19 Hydroxyproline 0.00 ± 0.00 0.04 ± 0.04 0.01 ± 0.01 0.02 ± 0.01 Isoleucine 0.24 ± 0.17 3.73 ± 1.14 0.15 ± 0.12 0.40 ± 0.24 Leucine 0.17 ± 0.08 2.74 ± 0.70 0.12 ± 0.05 0.28 ± 0.08 Lysine 0.25 ± 0.21 1.22 ± 0.42 0.18 ± 0.18 0.41 ± 0.34 Methionine 0.03 ± 0.01 0.04 ± 0.01 0.03 ± 0.01 0.06 ± 0.01 Methylcysteine 0.10 ± 0.05 0.03 ± 0.03 0.07 ± 0.03 0.11 ± 0.04 Phenylalanine 0.38 ± 0.08 2.15 ± 0.78 0.34 ± 0.03 0.66 ± 0.15 Proline 0.39 ± 0.15 1.93 ± 0.46 0.30 ± 0.13 0.64 ± 0.23 Serine 4.49 ± 1.05 4.64 ± 0.56 4.29 ± 1.03 7.21 ± 1.02 Threonine 1.91 ± 0.37 5.68 ± 1.72 1.73 ± 0.35 3.08 ± 0.47 Tryptophan 0.10 ± 0.02 1.60 ± 0.33 0.12 ± 0.03 0.24 ± 0.03 Valine 0.52 ± 0.28 5.24 ± 1.44 0.37 ± 0.24 0.82 ± 0.41 Adonitol 0.45 ± 0.09 0.75 ± 0.19 0.59 ± 0.09 0.47 ± 0.07 Arabinose 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.03 Pipecolic Acid 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.03	Glutamine	9.03 ± 2.06	6.73 ± 1.61	8.87 ± 2.13	14.61 ± 4.43
Hydroxyproline 0.00 ± 0.00 0.04 ± 0.04 0.01 ± 0.01 0.02 ± 0.01 Isoleucine 0.24 ± 0.17 3.73 ± 1.14 0.15 ± 0.12 0.40 ± 0.24 Leucine 0.17 ± 0.08 2.74 ± 0.70 0.12 ± 0.05 0.28 ± 0.08 Lysine 0.25 ± 0.21 1.22 ± 0.42 0.18 ± 0.18 0.41 ± 0.34 Methionine 0.03 ± 0.01 0.04 ± 0.01 0.03 ± 0.01 0.06 ± 0.01 Methylcysteine 0.10 ± 0.05 0.03 ± 0.03 0.07 ± 0.03 0.11 ± 0.04 Phenylalanine 0.38 ± 0.08 2.15 ± 0.78 0.34 ± 0.03 0.66 ± 0.15 Proline 0.39 ± 0.15 1.93 ± 0.46 0.30 ± 0.13 0.64 ± 0.23 Serine 4.49 ± 1.05 4.64 ± 0.56 4.29 ± 1.03 7.21 ± 1.02 Threonine 1.91 ± 0.37 5.68 ± 1.72 1.73 ± 0.35 3.08 ± 0.47 Tryptophan 0.10 ± 0.02 1.60 ± 0.33 0.12 ± 0.03 0.24 ± 0.03 Valine 0.52 ± 0.28 5.24 ± 1.44 0.37 ± 0.24 0.82 ± 0.41 Adonitol 0.45 ± 0.09 1.07 ± 0.31 0.46 ± 0.07 0.44 ± 0.08 Arabinose 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.03 Pipecolic Acid 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.08	Glycine	1.20 ± 0.20	1.00 ± 1.43	0.97 ± 0.23	1.96 ± 0.30
Isoleucine 0.24 ± 0.17 3.73 ± 1.14 0.15 ± 0.12 0.40 ± 0.24 Leucine 0.17 ± 0.08 2.74 ± 0.70 0.12 ± 0.05 0.28 ± 0.08 Lysine 0.25 ± 0.21 1.22 ± 0.42 0.18 ± 0.18 0.41 ± 0.34 Methionine 0.03 ± 0.01 0.04 ± 0.01 0.03 ± 0.01 0.06 ± 0.01 Methylcysteine 0.10 ± 0.05 0.03 ± 0.03 0.07 ± 0.03 0.11 ± 0.04 Phenylalanine 0.38 ± 0.08 2.15 ± 0.78 0.34 ± 0.03 0.66 ± 0.15 Proline 0.39 ± 0.15 1.93 ± 0.46 0.30 ± 0.13 0.64 ± 0.23 Serine 4.49 ± 1.05 4.64 ± 0.56 4.29 ± 1.03 7.21 ± 1.02 Threonine 1.91 ± 0.37 5.68 ± 1.72 1.73 ± 0.35 3.08 ± 0.47 Tryptophan 0.10 ± 0.02 1.60 ± 0.33 0.12 ± 0.03 0.24 ± 0.03 Valine 0.52 ± 0.28 5.24 ± 1.44 0.37 ± 0.24 0.82 ± 0.41 Adonitol 0.45 ± 0.09 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.07 Arabinose 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.03 Pipecolic Acid 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.03	Histidine	0.38 ± 0.12	$\textbf{1.36} \pm \textbf{0.45}$	0.30 ± 0.07	0.47 ± 0.19
Leucine 0.17 ± 0.08 2.74 ± 0.70 0.12 ± 0.05 0.28 ± 0.08 Lysine 0.25 ± 0.21 1.22 ± 0.42 0.18 ± 0.18 0.41 ± 0.34 Methionine 0.03 ± 0.01 0.04 ± 0.01 0.03 ± 0.01 0.06 ± 0.01 Methylcysteine 0.10 ± 0.05 0.03 ± 0.03 0.07 ± 0.03 0.11 ± 0.04 Phenylalanine 0.38 ± 0.08 2.15 ± 0.78 0.34 ± 0.03 0.66 ± 0.15 Proline 0.39 ± 0.15 1.93 ± 0.46 0.30 ± 0.13 0.64 ± 0.23 Serine 4.49 ± 1.05 4.64 ± 0.56 4.29 ± 1.03 7.21 ± 1.02 Threonine 1.91 ± 0.37 5.68 ± 1.72 1.73 ± 0.35 3.08 ± 0.47 Tryptophan 0.10 ± 0.02 1.60 ± 0.33 0.12 ± 0.05 0.21 ± 0.03 Tyrosine 0.14 ± 0.04 1.54 ± 0.38 0.12 ± 0.03 0.24 ± 0.03 Valine 0.52 ± 0.28 5.24 ± 1.44 0.37 ± 0.24 0.82 ± 0.41 Adonitol 0.45 ± 0.09 0.23 ± 0.04 0.18 ± 0.04 0.15 ± 0.03 Pipecolic Acid 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.08	Hydroxyproline	0.00 ± 0.00	0.04 ± 0.04	0.01 ± 0.01	$\textbf{0.02} \pm \textbf{0.01}$
Lysine 0.25 ± 0.21 1.22 ± 0.42 0.18 ± 0.18 0.41 ± 0.34 Methionine 0.03 ± 0.01 0.04 ± 0.01 0.03 ± 0.01 0.06 ± 0.01 Methylcysteine 0.10 ± 0.05 0.03 ± 0.03 0.07 ± 0.03 0.11 ± 0.04 Phenylalanine 0.38 ± 0.08 2.15 ± 0.78 0.34 ± 0.03 0.66 ± 0.15 Proline 0.39 ± 0.15 1.93 ± 0.46 0.30 ± 0.13 0.64 ± 0.23 Serine 4.49 ± 1.05 4.64 ± 0.56 4.29 ± 1.03 7.21 ± 1.02 Threonine 1.91 ± 0.37 5.68 ± 1.72 1.73 ± 0.35 3.08 ± 0.47 Tryptophan 0.10 ± 0.02 1.60 ± 0.33 0.12 ± 0.05 0.21 ± 0.03 Tyrosine 0.14 ± 0.04 1.54 ± 0.38 0.12 ± 0.03 0.24 ± 0.03 Valine 0.52 ± 0.28 5.24 ± 1.44 0.37 ± 0.24 0.82 ± 0.41 Adonitol 0.45 ± 0.09 0.75 ± 0.19 0.59 ± 0.09 0.47 ± 0.08 Arabinose 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.08 Pipecolic Acid 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.08	Isoleucine	0.24 ± 0.17	3.73 ± 1.14	0.15 ± 0.12	0.40 ± 0.24
Methionine 0.03 ± 0.01 0.04 ± 0.01 0.03 ± 0.01 0.06 ± 0.01 Methylcysteine 0.10 ± 0.05 0.03 ± 0.03 0.07 ± 0.03 0.11 ± 0.04 Phenylalanine 0.38 ± 0.08 2.15 ± 0.78 0.34 ± 0.03 0.66 ± 0.15 Proline 0.39 ± 0.15 1.93 ± 0.46 0.30 ± 0.13 0.64 ± 0.23 Serine 4.49 ± 1.05 4.64 ± 0.56 4.29 ± 1.03 7.21 ± 1.02 Threonine 1.91 ± 0.37 5.68 ± 1.72 1.73 ± 0.35 3.08 ± 0.47 Tryptophan 0.10 ± 0.02 1.60 ± 0.33 0.12 ± 0.05 0.21 ± 0.03 Tyrosine 0.14 ± 0.04 1.54 ± 0.38 0.12 ± 0.03 0.24 ± 0.03 Valine 0.45 ± 0.09 0.47 ± 0.31 0.46 ± 0.07 0.44 ± 0.08 Adonitol 0.45 ± 0.09 0.23 ± 0.04 0.18 ± 0.04 0.15 ± 0.03 Pipecolic Acid 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.08	Leucine	0.17 ± 0.08	2.74 ± 0.70	0.12 ± 0.05	0.28 ± 0.08
Methylcysteine 0.10 ± 0.05 0.03 ± 0.03 0.07 ± 0.03 0.11 ± 0.04 Phenylalanine 0.38 ± 0.08 2.15 ± 0.78 0.34 ± 0.03 0.66 ± 0.15 Proline 0.39 ± 0.15 1.93 ± 0.46 0.30 ± 0.13 0.64 ± 0.23 Serine 4.49 ± 1.05 4.64 ± 0.56 4.29 ± 1.03 7.21 ± 1.02 Threonine 1.91 ± 0.37 5.68 ± 1.72 1.73 ± 0.35 3.08 ± 0.47 Tryptophan 0.10 ± 0.02 1.60 ± 0.33 0.12 ± 0.05 0.21 ± 0.03 Tyrosine 0.14 ± 0.04 1.54 ± 0.38 0.12 ± 0.03 0.24 ± 0.03 Valine 0.52 ± 0.28 5.24 ± 1.44 0.37 ± 0.24 0.82 ± 0.41 Adonitol 0.45 ± 0.09 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.07 Arabinose 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.03 Pipecolic Acid 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.08	Lysine	0.25 ± 0.21	1.22 ± 0.42	0.18 ± 0.18	0.41 ± 0.34
Phenylalanine 0.38 ± 0.08 2.15 ± 0.78 0.34 ± 0.03 0.66 ± 0.15 Proline 0.39 ± 0.15 1.93 ± 0.46 0.30 ± 0.13 0.64 ± 0.23 Serine 4.49 ± 1.05 4.64 ± 0.56 4.29 ± 1.03 7.21 ± 1.02 Threonine 1.91 ± 0.37 5.68 ± 1.72 1.73 ± 0.35 3.08 ± 0.47 Tryptophan 0.10 ± 0.02 1.60 ± 0.33 0.12 ± 0.05 0.21 ± 0.03 Tyrosine 0.14 ± 0.04 1.54 ± 0.38 0.12 ± 0.03 0.24 ± 0.03 Valine 0.52 ± 0.28 5.24 ± 1.44 0.37 ± 0.24 0.82 ± 0.41 Adonitol 0.45 ± 0.09 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.08 Arabinose 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.03 Pipecolic Acid 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.08	Methionine	0.03 ± 0.01	0.04 ± 0.01	0.03 ± 0.01	0.06 ± 0.01
Proline 0.39 ± 0.15 1.93 ± 0.46 0.30 ± 0.13 0.64 ± 0.23 Serine 4.49 ± 1.05 4.64 ± 0.56 4.29 ± 1.03 7.21 ± 1.02 Threonine 1.91 ± 0.37 5.68 ± 1.72 1.73 ± 0.35 3.08 ± 0.47 Tryptophan 0.10 ± 0.02 1.60 ± 0.33 0.12 ± 0.05 0.21 ± 0.03 Tyrosine 0.14 ± 0.04 1.54 ± 0.38 0.12 ± 0.03 0.24 ± 0.03 Valine 0.52 ± 0.28 5.24 ± 1.44 0.37 ± 0.24 0.82 ± 0.41 Adonitol 0.45 ± 0.09 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.07 Arabinose 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.03 Pipecolic Acid 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.03	Methylcysteine	0.10 ± 0.05	0.03 ± 0.03	0.07 ± 0.03	0.11 ± 0.04
Serine 4.49 ± 1.05 4.64 ± 0.56 4.29 ± 1.03 7.21 ± 1.02 Threonine 1.91 ± 0.37 5.68 ± 1.72 1.73 ± 0.35 3.08 ± 0.47 Tryptophan 0.10 ± 0.02 1.60 ± 0.33 0.12 ± 0.05 0.21 ± 0.03 Tyrosine 0.14 ± 0.04 1.54 ± 0.38 0.12 ± 0.03 0.24 ± 0.03 Valine 0.52 ± 0.28 5.24 ± 1.44 0.37 ± 0.24 0.82 ± 0.41 Adonitol 0.45 ± 0.09 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.07 Arabinose 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.03 Pipecolic Acid 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.08	Phenylalanine	0.38 ± 0.08	2.15 ± 0.78	0.34 ± 0.03	0.66 ± 0.15
Threonine 1.91 ± 0.37 5.68 ± 1.72 1.73 ± 0.35 3.08 ± 0.47 Tryptophan 0.10 ± 0.02 1.60 ± 0.33 0.12 ± 0.05 0.21 ± 0.03 Tyrosine 0.14 ± 0.04 1.54 ± 0.38 0.12 ± 0.03 0.24 ± 0.03 Valine 0.52 ± 0.28 5.24 ± 1.44 0.37 ± 0.24 0.82 ± 0.41 Adonitol 0.45 ± 0.09 1.07 ± 0.31 0.46 ± 0.07 0.44 ± 0.08 Arabinose 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.07 Arabitol 0.28 ± 0.04 0.23 ± 0.04 0.18 ± 0.04 0.15 ± 0.03 Pipecolic Acid 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.08	Proline	0.39 ± 0.15	1.93 ± 0.46	0.30 ± 0.13	0.64 ± 0.23
Tryptophan Tyrosine 0.10 ± 0.02 0.14 ± 0.04 0.52 ± 0.28 1.60 ± 0.33 1.54 ± 0.38 0.12 ± 0.03 0.12 ± 0.03 0.12 ± 0.03 0.12 ± 0.03 0.24 ± 0.03 0.82 ± 0.41 Adonitol Arabinose 0.45 ± 0.09 0.65 ± 0.20 1.07 ± 0.31 0.95 ± 0.19 0.46 ± 0.07 0.59 ± 0.09 0.18 ± 0.04 0.44 ± 0.08 0.47 ± 0.07 0.18 ± 0.04 Arabitol Pipecolic Acid 0.28 ± 0.04 0.65 ± 0.20 0.95 ± 0.19 0.95 ± 0.19 0.15 ± 0.03 0.59 ± 0.09	Serine	4.49 ± 1.05	4.64 ± 0.56	4.29 ± 1.03	7.21 ± 1.02
Tyrosine Valine 0.14 ± 0.04 0.52 ± 0.28 1.54 ± 0.38 5.24 ± 1.44 0.12 ± 0.03 0.37 ± 0.24 0.24 ± 0.03 0.82 ± 0.41 Adonitol Arabinose 0.45 ± 0.09 0.65 ± 0.20 1.07 ± 0.31 0.95 ± 0.19 0.46 ± 0.07 0.59 ± 0.09 0.44 ± 0.08 0.47 ± 0.07 Arabitol Pipecolic Acid 0.28 ± 0.04 0.65 ± 0.20 0.23 ± 0.04 0.95 ± 0.19 0.15 ± 0.03 0.59 ± 0.09	Threonine	1.91 ± 0.37	5.68 ± 1.72	1.73 ± 0.35	3.08 ± 0.47
Yaline 0.52 ± 0.28 5.24 ± 1.44 0.37 ± 0.24 0.82 ± 0.41 Adonitol 0.45 ± 0.09 1.07 ± 0.31 0.46 ± 0.07 0.44 ± 0.08 Arabinose 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.07 Arabitol 0.28 ± 0.04 0.23 ± 0.04 0.18 ± 0.04 0.15 ± 0.03 Pipecolic Acid 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.08	Tryptophan	0.10 ± 0.02	1.60 ± 0.33	0.12 ± 0.05	0.21 ± 0.03
Adonitol 0.45 ± 0.09 1.07 ± 0.31 0.46 ± 0.07 0.44 ± 0.08 Arabinose 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.07 Arabitol 0.28 ± 0.04 0.23 ± 0.04 0.18 ± 0.04 0.15 ± 0.03 Pipecolic Acid 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.08	Tyrosine	0.14 ± 0.04	1.54 ± 0.38	0.12 ± 0.03	0.24 ± 0.03
Arabinose 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.07 Arabitol 0.28 ± 0.04 0.23 ± 0.04 0.18 ± 0.04 0.15 ± 0.03 Pipecolic Acid 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.08	Valine	0.52 ± 0.28	5.24 ± 1.44	0.37 ± 0.24	0.82 ± 0.41
Arabinose 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.07 Arabitol 0.28 ± 0.04 0.23 ± 0.04 0.18 ± 0.04 0.15 ± 0.03 Pipecolic Acid 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.08					
Arabitol 0.28 ± 0.04 0.23 ± 0.04 0.18 ± 0.04 0.15 ± 0.03 Pipecolic Acid 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.08	Adonitol	0.45 ± 0.09	1.07 ± 0.31	0.46 ± 0.07	0.44 ± 0.08
Pipecolic Acid 0.65 ± 0.20 0.95 ± 0.19 0.59 ± 0.09 0.47 ± 0.08	Arabinose	0.65 ± 0.20	0.95 ± 0.19	0.59 ± 0.09	0.47 ± 0.07
	Arabitol	0.28 ± 0.04	0.23 ± 0.04	0.18 ± 0.04	0.15 ± 0.03
	Pipecolic Acid	0.65 ± 0.20	0.95 ± 0.19	0.59 ± 0.09	0.47 ± 0.08
	•	2.81 ± 1.02	5.43 ± 1.39	3.02 ± 0.74	3.46 ± 1.19

- **Table 3**. Concentration (nmoles·mg⁻¹ dry mass) of amino acids, organic acids, polyols and sugars
- (n=3; mean ± SD) whose amounts significantly varied (relative to control) in the cytosol of *E. nuttallii*
- exposed for 24h to Cd, IHg and MeHg (bold font = p-value <0.05).

	Control	Cd	ILIa	Molla
		Cd	IHg	MeHg
Alpha-alanine	0.48 ± 0.01	0.31 ± 0.05	0.50 ± 0.17	0.33 ± 0.18
Ammonium	0.52 ± 0.13	0.89 ± 0.11	0.81 ±0.07	0.62 ± 0.32
Arginine	0.16 ± 0.02	0.05 ± 0.06	0.02 ± 0.01	0.04 ± 0.04
Aspartic Acid	1.22 ± 0.12	$\textbf{0.60} \pm \textbf{0.11}$	1.23 ± 0.41	0.77 ± 0.44
Glycine	0.18 ± 0.03	$\textbf{0.40} \pm \textbf{0.03}$	0.12 ± 0.03	0.19 ± 0.15
Histidine	0.14 ± 0.01	0.58 ± 0.99	0.07 ± 0.01	0.0 ± 0.0
Hydroxyproline	3.68 ± 0.25	5.89 ± 0.84	4.42 ± 0.83	3.84 ± 2.01
Galactose	0.99 ± 0.01	0.39 ± 0.03	0.63 ± 0.21	0.35 ± 0.17
Mannose	0.08 ± 0.01	$\textbf{0.02} \pm \textbf{0.01}$	$\textbf{0.04} \pm \textbf{0.02}$	0.03 ± 0.01
Phosphoric acid	6.89 ± 1.26	15.20 ± 3.09	7.31 ± 0.82	7.26 ± 3.35

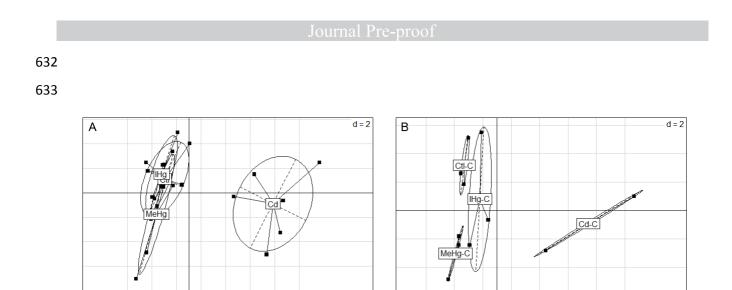


Figure 1. Principal component analysis of amino acids, organic acids, sugars and polyols measured in shoots (A) and cytosol (B) of *E. nuttallii* exposed 24h to 281 μ g·L⁻¹ Cd, 70 ng·L⁻¹ IHg and 30 ng·L⁻¹ MeHg, and their control (ctl) (n=6 and 3 ± SD, respectively). PCA were calculated and plotted with the package ade4 (http://pbil.univ-lyon1.fr/ade4) in the R software (http://www.r-project.org).

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- Cd, IHg and MeHg effects were compared in shoots and cytosol by metabolomics •
- In shoots, Cd showed higher regulation, but MeHg regulated more pathways ٠
- In cytosol, Cd had the higher impact, IHg had low effect and MeHg had very low effect
- MeHg impacts N homeostasis, and Cd results in an osmotic stress-like pattern
- Metabolomics support different molecular toxicity pathways for the three metals •

Authors declare no conflict of interest

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