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# 1 Biodegradation of Diclofenac by two green microalgae: Picocystis sp. and

# 2 Graesiella sp.

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#### 20 Abstract

The aim of the present study was to provide an integrated view of algal removal of diclofenac (DCF). Two isolated microalgal strains *Picocystis* sp. and *Graesiella* sp. were cultivated under different DCF concentrations and their growth, photosynthetic activity and diclofenac removal efficiency were monitored.

Results showed that DCF had slight inhibitory effects on the microalgal growth which did not 25 exceed 21% for *Picocystis* and 36% for *Graesiella* after 5 days. Both species showed different 26 patterns in terms of removal efficiency. In presence of Picocystis sp., the amounts of removed 27 DCF were up to 73%, 43% and 25% of 25, 50 and 100 mg L<sup>-1</sup> respectively; whereas only 28 29 52%, 28% and 24% were removed in the presence of Graesiella at same DCF tested 30 concentrations. DCF removal was insured mainly by biodegradation. To better reveal the mechanism involved. 31 metabolites analyses were performed. Two DCF biodegradation/biotransformation products were detected in presence of Picocystis. 32

This study indicated that *Picocystis* performed a satisfactory growth capacity and DCF removal efficiency and thus could be used for treatment of DCF contaminated aqueous systems.

#### 36 Keywords

37 Microalgae; Extremophiles; Diclofenac; Removal; Biodegradation; Biotransformation.

#### 38 **1. Introduction**

The presence of emerging contaminants as pharmaceutical and personal care products (PPCPs) in the aquatic environment continues to give rise to concern due to their environmental risk and toxicological properties. Indeed, these chemical compounds can cause multiple changes in the physiological state of organisms, and their occurrence in the environment may affect
non-target species (Valavanidis et al., 2014).

Among the most used PPCPs, Diclofenac (DCF) is a non-steroidal anti-inflammatory drug 44 45 (NSAID) widely prescribed as an antipyretic analgesic. It is often found as a persistent toxic waste and one of the most widely available drugs in the world. Recent studies based on the 46 Intercontinental Marketing Services (IMS) health data (which serves 82% of the global 47 population) from 86 countries estimated that about  $1443 \pm 58$  tons of DCF are consumed 48 globally on an annual basis (Acuña et al., 2015). However, this is only an indication of the 49 DCF consumption for human health-related applications and does not include DCF's 50 veterinary uses. 51

A large part of the consumed DCF is excreted in urine and feces in original form so entering 52 municipal wastewater. DCF is ineffectively removed by conventional wastewater treatment 53 54 plants (WWTPs) (Langenhoff et al., 2013; Sophia and Lima, 2018). Thus, it can be discharged into the environment with treated wastewater effluent, recycled water, and 55 56 wastewater plant sludge. Actually, DCF is detected in several aquatic environments all over the world at concentration ranges from few hundreds to thousands of ng/L (Lonappan et al., 57 2016). It may be transported through food chains (Cuellar-Bermudez et al., 2017) causing 58 toxic adverse effects on many aquatics organisms, even at environmentally low 59 concentrations (Xu et al., 2019). 60

Therefore, it is mandatory to investigate alternative treatments for DCF removal from wastewaters. Among them, those based on the use of microalgae are emerging as a sustainable and economical solution (**Escapa et al., 2017**). Recent studies reported that many microalgae species could remove several pharmaceutical contaminants, including DCF (**Xiong et al., 2018**). Therefore, using microalgae to remove DCF from wastewater could be prospective.

To date, only very few studies have been carried out on the assessment of DCF removal using microalgae (**Escapa et al., 2017; Villar-Navarro et al., 2019**). Furthermore, the ability of microalgae to accumulate DCF as well as its biodegradation and/or biotransformation products were not yet investigated. Thus, information regarding DCF effects, removal and biotransformation products in the presence of microalgae are required.

Among microalgae, a particular interest is given to the use of extremophilic species in bioremediation systems (**Varshney et al., 2014**). Such organisms are assumed to have specific qualities allowing them to tolerate high concentration of several pollutants and further could be more efficient in the fast contaminant removal from wastewaters (**Peeples, 2014**).

The aim of the present work is to evaluate and compare, under laboratory culture conditions, the effects and the removal efficiency of DCF by two extremophilic microalgae strains *Picocystis* sp. and *Graesiella* sp. isolated from two polluted ponds in Tunisia for the purpose of selecting resistant species for their bioremediation use. To fulfill those purposes, the effect of DCF on growth and photosynthesis of both species was assessed together with their abilities to remove and accumulate DCF. Then, the resulting biotransformation and biodegradation products were addressed.

#### 83 **2.** Materials and methods

84

# 2.1 Chemicals and reagents

Diclofenac sodium salt (≥98%) was purchased from Sigma Aldrich. Ultra-pure water (UPW)
was delivered by Elga Pure Lab System (resistivity 18.2 MΩ.cm, COT <50 µg C/L).</li>
Acetonitrile (ACN) with 0.1% formic acid (FA) was purchased from JT Baker (LC-MS grade)
and used in association with UPW with 0.1% formic acid.

Salts and reagents for medium preparation suppliers are specified in Supplementary material 1.All chemicals used were of analytical grade.

91

#### 2.2 Algal strains and culture conditions

Two extremophilic chlorophyta species were investigated in this study: an alkaliphilic 92 93 Picocystis sp. CINS 23 and a thermophilic Graesiella sp. CINS 60. The strains have been deposited in the National Institute of Marine Sciences and Technologies Collection. 94 Picocystis sp. was isolated from a household sewage "Essed valley" located in Center East 95 of Tunisia (35° 59′ 23″ N, 10° 30′ 10″ E) and *Graesiella* sp. from a hot water catchment 96 basin emerging from "Ain Echfa," a hot spring located in northern Tunisia (36°49' N, 10°34' 97 E). Preliminary laboratory experiments were conducted to define the optimal autotrophic 98 growth conditions: culture media, temperature and light intensity (Ben Ali et al., 2017; Ben 99 Ouada et al., 2018; Mezhoud et al., 2014). The strains were cultivated separately in batch 100 101 culture under sterile conditions: Graesiella in Bold's Basal Medium (BBM) (Bischoff and Bold, 1963) modified according to the Elser concept for freshwater microalgae with C:N:P 102 103 ratio equal to 166:20:1 (Elser et al., 2000), and Picocystis in Zarrouk medium (Zarrouk, 104 1966) (Supplementary material 1). The initial pH was adjusted to 6.8 and 8.4 for BBM and Zarrouk media, respectively. Cultures were maintained in optimal growth conditions at 105 temperature of 30°C and light intensity of 75 µmol m<sup>-2</sup> s<sup>-1</sup> under 8/16-h illumination cycle as 106 107 established by preliminary laboratory experiments. Experimental cultures were conducted at these optimal growth conditions for all the designed experiences. 108

Exponentially growing cultures from *Picocystis* and *Graesiella* were separately inoculated in 500 mL Erlenmeyer flasks containing respectively 300 mL of Zarrouk and BBM media, with an initial cell density of  $10^6$  cells mL<sup>-1</sup>. Cultures were exposed separately for 5 days to DCF at 0 (control), 25, 50, 100 and 200 mg L<sup>-1</sup> initial concentrations. For each experimental DCF concentration and separately for each tested strain, a series of six Erlenmeyer flask cultures were used in triplicate. Daily, 3 ml of each culture served for the determination of the optical density and the photosynthetic activity. The determination of the residual, adsorbed, 116 accumulated, and biodegraded DCF amounts of 25, 50 and 100 mg L<sup>-1</sup> initial tested 117 concentrations, was carried out at the end of the experiment (day 5). Blank controls (without 118 algae cells) served also for the determination of the DCF abiotic removal (Supplementary 119 material 2).

120 The stock solution of diclofenac with the concentration of 1 g  $L^{-1}$  was prepared by dissolving 121 an appropriate amount of the compound in distilled water and it was stored below the 122 temperature of 5 °C. The testing solutions were diluted to the desired concentrations during 123 the experiments.

## 124 **2.2** Growth inhibition test

Growth was followed spectrophotometrically in each culture condition by measuring the daily
changes in OD<sub>680 nm</sub> during five days of DCF exposure.

Growth curves of microalgae population were obtained by plotting OD<sub>680 nm</sub> values and incubation time. The areas under each growth curve were evaluated using Origin 8.5 software (OriginLab, Northampton, USA). Inhibition of algal activity was expressed as the percentage inhibition (PI%) defined according to **De Orte et al. (2013)** by the following equation:

131 
$$PI\% = \left[\frac{Control area - Treated area}{Control area}\right] \times 100$$
 (Eq. A1)

Where, Control area and Treated area are the integrated areas under the growth curve ofcontrol and treated culture, respectively.

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#### 2.3 Photosynthetic parameters

Chlorophyll fluorescence for each culture condition was measured every day using the
Aquapen-CAP-C 100 fluorometer (Photon Systems Instruments, Brno, Czech Republic). A 3
mL sample was collected daily from each experimental culture and diluted with fresh medium

to a final optical density at 680 nm of 0.3. Before the measurement, each micro-algal sample 138 was dark-adapted for 15 min at room temperature to allow the complete re-oxidation of PSII 139 reaction centers. The chlorophyll fluorescence transients (OJIP) were induced by light pulses 140 at a fixed excitation wavelength of 650 nm and intensity of 3000  $\mu mol$  photons  $m^{\text{-2}} \ \text{s}^{\text{-1}}$  and 141 recorded for up to 1 s on a logarithmic time scale. The OJIP-test parameters were determined 142 from transient analysis according to Strasser et al. (2000) and the maximal PSII 143 photochemical efficiency (Fv/Fm) was derived, wherein Fv = Fm-Fo is the variable 144 fluorescence, and Fo and Fm are minimal and maximal fluorescence yields in dark-adapted 145 state, respectively. 146

147 The non-photochemical quenching (NPQ) was measured according to the manufacturer's 148 instructions (protocol NPQ1, Photon Systems Instrument). The intensities of actinic light (450 149 nm) and pulse-saturating light were 300 and 3000 µmol photons m<sup>-2</sup> s<sup>-1</sup>, respectively. The 150 NPQ was calculated as  $((F_m - F'_m)/F'_m)$  (**Campbell et al., 1998**), wherein F'm is the 151 maximal fluorescence in a light-adapted state.

# 152 **2.4** Determination of DCF removal and degradation intermediates

153 After 5 days of DCF exposure, 150 mL of each algae culture were centrifuged at 10.000×g for 10 minutes. The supernatant was used for the determination of the residual DCF in the 154 medium. The cell pellets were washed three times with de-ionized water and then centrifuged. 155 The supernatant was used for the determination of DCF adsorbed on the microalgae surface. 156 The cell pellets were mixed with 2 mL of dichloromethane-methanol (1:2 v/v) and digested 157 by sonication for 45 min (Ji et al., 2014). After centrifugation for 10 min at 4500×g, the 158 resulting supernatant was conserved to determine the accumulated DCF within the microalgae 159 cells. 160

161 The same extraction procedure was applied for media without algae cells at all tested DCF 162 concentrations exposed to the same culture conditions during five days to determine the 163 abiotic removal of DCF.

164 Total DCF removal percentage was calculated according to the following equation:

165 DCF removal (%) =  $\frac{\text{(initial DCF concentration-final DCF concentration)}}{\text{initial DCF concentration}} \times 100$  (Eq. A2)

Biotransformation/biodegradation percentage (Bp) of DCF by microalgae was calculatedaccording to the following equation (Li et al., 2009):

168 
$$Bp(\%) = (CI - Ar - Aa - Ad - Ac) \times 100/CI$$
 (Eq. A3)

Where CI is the initial amount of DCF, Ar is the residual amount in the medium, Aa is the amount of abiotic removal, Ad is the amount adsorbed to the algal cells and Ac is the amount accumulated on algal cells. Specific removal of DCF (RS) was calculated according to the following equation:

173 RS (mg g<sup>-1</sup>) = 
$$\frac{(Ad+Ac+Bp)}{dw}$$
 (Eq. A4)

Where Ad is the DCF amount adsorbed to the algal cells, Ac is the amount accumulated on algal cells, Bp is the biodegraded amount and dw the dry weight of microalgae cells (g  $L^{-1}$ ).

# 176 DCF Analysis

177 DCF analysis was carried out using a Waters (Aquity UPLC) liquid chromatographic system 178 coupled to a mass spectrometer detector (Quattro Premier; Micromass) equipped with an 179 electrospray ionization source. Chromatographic separation was performed on a BEH-C18 180 chromatographic column (100 mm  $\times$  2.1 mm ID; 1.7 µm). LC elution was performed with 181 100% acetonitrile as mobile phase A and an ultrapure water 9:1 acetonitrile (v/v) as mobile

phase B. The pH of the eluent was adjusted to 2 with 0.1 % formic acid. Gradient elution at a 182 flow rate of 0.4 mL min<sup>-1</sup> consisted in 0% A (100% B) from 0 to 1 min, followed by a linear 183 increase to 90% A (10% B) in 4 min and held for one minute; finally, from 6 to 7 min, mobile 184 phase A decreased back to 0%. Nitrogen was used as the collision and nebulizing gas. The 185 injection volume was 5 µL. The analysis was performed using ESI positive mode under the 186 following conditions: capillary, 23 V; collision, 29 eV; column temperature, 45°C; sample 187 temperature 5°C. The multiple-reaction monitoring (MRM) mode was applied for the DCF 188 detection and quantification. The retention time is 4.47 min and MRM parameters: parent and 189 daughter ions are 296.35 and 214.21 Da, respectively. 190

191 Complete calibration curves were performed at the beginning and at the end of the sample set, 192 the mean slope value of these curves was used for DCF quantification. The limit of detection 193 (LOD) and the limit of quantification (LOQ) were 0.180 and 0.547 mg L<sup>-1</sup>, respectively.

For the detection of DCF degradation intermediates, the full scan (FS) mode was performed, in a scan range of 60 to 500 m/z. Spectra were recorded with an average of 20-25 scans under identical experimental conditions. The reference full scan spectrum is presented in Fig. 1. The structural identity of each biodegradation product was performed with the LC-MS/MS fragmentation analysis. The analytical device was controlled by Micromass MassLynx 4.1 software.

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# 2.5 Statistical analysis

Mean and standard deviation values of the three biological replicates were calculated for each treatment and for the control. In all cases, data are given as mean value with standard deviation. Two-way ANOVA was used to analyze the effect of DCF tested concentrations and culture duration on growth and photosynthetic activities and one-way ANOVA was performed to determine the significant differences between removal efficiencies of studied

species among DCF tested concentrations, using IBM SPSS Statistic 21.0.0 software. A Pvalue < 0.05 was considered statistically significant. When significant differences were observed, means were compared using the Post Hoc test for Two-way ANOVA analysis at a level of significance of 0.05 (P<0.05). For the DCF removal efficiencies, the significance of differences between the control and treated samples was analyzed by the one-way ANOVA analysis using the multiple-range Duncan test at the same level of significance (P<0.05).

**3. Results** 

213

# 3.1 Effect of DCF on microalgae growth

The effect of DCF concentrations on Picocystis sp. and Graesiella sp. growth during 5 days of 214 215 exposure was evaluated by the growth inhibition percentage (PI%) (Table 1). Results showed 216 that the DFC has slight effect on both species. The PI of Picocystis sp. and Graesiella sp. did not reach 40 % for both species during the whole period of exposure and even at the highest 217 DCF tested concentration (200 mg L<sup>-1</sup>). The maximum PI recorded was 21% for *Picocystis* 218 and 36% for Graesiella implying that Picocystis was relatively more tolerant to DCF. 219 Furthermore, a negative PI was observed when Picocystis was exposed to 25 and 50 mg/L 220 DCF, which reflects a stimulation of the microalgae growth by 4 and 21 %, respectively. 221

# **3.2 Effect of DCF on microalgae photosynthetic activity**

The effect of DCF on the photochemical efficiency of photosystem II (Fv/Fm) and the nonphotochemical quenching (NPQ), an indicator of the excess-radiant energy dissipation to heat in PSII antenna complexes during light-adapted state, of *Picocystis* and *Graesiella* were determined during 5 days of DCF exposure at 25, 50, 100 and 200 mg L<sup>-1</sup> initial DCF concentrations. Compared to the controls, the Fv/Fm ratios of *Picocystis* treated cultures were not inhibited during the first 3 days of DCF exposure (Fig. 2). A slight decrease was recorded

only at the end of the exposure duration (4th and 5th day) in *Picocystis* cultures exposed to 229 high DCF concentrations (50 - 200 mg L<sup>-1</sup>). For Graesiella, no inhibitory effect on Fv/Fm 230 ratios was recorded during the whole experiment period regardless the DCF tested 231 concentrations. Contrariwise, an increase in the NPQ values of both species was observed 232 when exposed to high DCF tested concentrations (50 - 200 mg  $L^{-1}$ ) as compared to the 233 controls especially since the third day of DCF exposure (Fig. 3). After 5 days' exposure to 234 200 mg L<sup>-1</sup> DCF, the NPQ values of *Picocystis* and *Graesiella* treated cells were about two 235 folds of the control groups. 236

# **3.3 DCF removal from the culture medium**

The DCF removal by *Picocystis* and *Graesiella* was investigated at 25, 50 and 100 mg L<sup>-1</sup> DCF after 5 days of treatment (Table 2). Blank controls (without algae cells) were also investigated. Both species showed different patterns in terms of removal efficiency. In presence of *Picocystis* sp., the amounts of removed DCF were up to 73%, 42% and 25% of 25, 50 and 100 mg L<sup>-1</sup>, respectively; whereas only 52%, 28% and 24% were removed in the presence of *Graesiella* at the same DCF concentrations.

DCF removal in blank controls did not exceed 8% regardless of initial concentrations indicating that abiotic losses were negligible, and the observed DCF decrease was mainly due to the removal by microalgae.

The quantity of DCF adsorbed and accumulated within the microalgae cells did not exceed also 0.5% in all tested concentrations for both studied species (Table 2) that implies that the DCF removal by microalgae was insured mainly by biotransformation/biodegradation.

As illustrated in table 2, the DCF Biotransformation/biodegradation percentage (Bp) in the presence of *Picocystis* were about 69%, 36% and 21% of 25, 50 and 100 mg  $L^{-1}$  DCF, respectively. For *Graesiella*, the Bp were 44%, 21% and 18% of DCF at the same tested concentrations.

The specific removal (RS) of DCF by *Picocystis* was between 12.39 and 18.96 mg g<sup>-1</sup> for 25-100 mg L<sup>-1</sup> initial DCF concentrations. In *Graesiella*, the maximal RS was 12.23 mg g<sup>-1</sup> observed for 100 mg L<sup>-1</sup> initial DCF concentration.

#### 257

## 3.4 Characterization of biodegradation products

DCF degradation products were separated by HPLC and characterized by on-line electrospray 258 ionization-ion trap mass spectrometry (ESI-MS). In the presence of Picocystis, the 259 comparison of chromatograms obtained (c) with and (a) without microalgae (Fig. 4) showed 260 the appearance of two additional peaks at retention times of 1.21 and 3.82 min, respectively. 261 The first one, labeled DP1 in Fig. 5a, yielded a molecular ion at m/z 312 Da and two 262 additional ion fragments at m/z 266 and 230. The MS2 fragmentation spectrum of m/z 312 263 264 revealed a fragment ion at m/z 266 corresponding to the loss of one CO<sub>2</sub> molecule, and the MS3 fragmentation spectrum of m/z 266 showed a fragment ion at m/z 230 corresponding to 265 the DCF dechlorination. The difference in mass between the molecular ion of the product DP1 266 267 (312 Da) and DCF (296 Da) was 16 Da, which might be related to the hydroxylation of the compound. The addition of an hydroxyl to the molecule supported the lower retention time 268 observed for the compound by increasing the molecule's polarity (Rigobello et al., 2013). 269 270 Although information provided by LC-MS instruments is not enough to establish the precise 271 position of the OH groups in the molecule, knowledge of the reactivity of the different species and the cited literature allowed us to propose probable structure as seen in Fig. 5a. 272

The second peak, labeled DP2 in Fig. 5b, showed a molecular ion at m/z 152 Da and two additional ion fragments at m/z 134 and m/z 106. The MS2 fragmentation of m/z 152 revealed a fragment ion at m/z 134 corresponding to the dehydration by an intramolecular cyclization,

and the MS3 fragmentation spectrum of m/z 134 showed a fragment ion at m/z 106 corresponding to the loss of one CO molecule. The difference of 144 Da between diclofenac and DP2 is consistent with a loss of the dechlorinated benzene ring as proposed in Fig. 5b.

In the presence of *Graesiella*, no difference was recorded between the EIS-MSchromatograms obtained with and without microalgae.

#### 281 **4. Discussion**

Microbial removal of pollutants relies on a trade-off between tolerance of the organisms to the 282 toxicity of the targeted compounds, and their efficiency to effectively sequester or degrade 283 284 them. To date only few studies on DCF toxicity to microalgae were reported, mostly on freshwater cyanobacteria and chlorophyta species (Supplementary material 3). The measured 285 sensitivities to diclofenac spread from 84.5% growth inhibition in the freshwater 286 cyanobacterium Cylindrospermopsis raciborskii exposed to 0.1 mg  $L^{-1}$  (Bácsi et al., 2016) to 287 50% growth inhibition (EC50) in the marine chlorophyceae Dunaliella tertiolecta 288 (DeLorenzo and Fleming, 2008) exposed to 185.7 mg L<sup>-1</sup> DCF. Comparatively, *Picocystis* 289 and *Graesiella* were characterized in the present study by a high tolerance to DCF, since an 290 exposure up to 200 mg L<sup>-1</sup> did not reach 40% inhibition for both strains. This behavior could 291 292 at least partly attributed to the origin of the model organisms, considering that extremophile algae have already reported to be tolerant to chemical contamination (Ben Ali et al., 2017; 293 Ben Ouada et al., 2018; Rehman and Shakoori, 2004). Furthermore, growth stimulation up 294 to 21% was observed in *Picocystis* exposed to 50 mg L<sup>-1</sup> DCF, as reported for other 295 chlorophyta strains exposed to 25 mg L<sup>-1</sup> DCF (Escapa et al., 2017). This behavior is 296 consistent with potential use of DCF as organic carbon source, since chlorophyta species were 297 shown able to photo-heterotrophic metabolism when culture medium was complemented by 298 organic substrates (Ratha et al., 2013; Zili et al., 2015). 299

Toxic chemicals often target the photosynthetic apparatus in microalgae (Choi et al., 2012), 300 but raw Picocystis and Graesiella PSII potentials (Fv/Fm) were not altered by DCF exposure 301 in the present study. To date, no toxicity of DCF to PSII integrity was reported for 302 microalgae, whereas higher plants were found sensitive (Kummerová et al., 2016). 303 Comparatively, the non-photochemical quenching NPQ, an indicator of the excess-radiant 304 energy dissipation during light-adapted state, increased for both species after DCF exposure. 305 The same trend was reported in river biofilms and in plant species (Kummerová et al., 2016), 306 307 and suggests an adaptive mechanism in Picocystis and Graesiella to protect PSII antenna and dissipate the excess of absorbed energy (Müller et al., 2001). 308

The high tolerance of *Picocystis* and *Graesiella* to DCF could be related to their high DCF 309 removal abilities. Among many studies about chemical remediation by microalgae, little 310 attention was paid to NSAIDs including DCF. Matamoros et al. (2015a; 2015b), Xiong et al. 311 312 (2018) and Zhou et al. (2014) demonstrated that microalgal inoculation in wastewaters enhanced pharmaceutical removal efficiencies of treatment systems up to 80%. In laboratory 313 314 monocultures, the amount of DCF removed by green microalgae were about 20 to 80 %, at DCF initial concentration  $\leq 25 \text{ mg L}^{-1}$  (Escapa et al., 2017). These values are comparable to 315 DCF removal efficiencies of both Graesiella (52%) and Picocystis (73%) at 25 mg L<sup>-1</sup> initial 316 DCF concentration. Even at higher initial concentrations, reaching 100 mg L<sup>-1</sup>, *Picocystis* and 317 Graesiella were able to remove about 25-42% and 24-28%, respectively; after 5days of 318 exposure, which demonstrated their high DCF removal abilities; even when compared to other 319 biological treatments (Langenhoff et al., 2013). 320

In the present work, a decrease of DCF initial concentrations was observed in the controls without algae suggesting abiotic degradation, most likely by photooxidation (**Michael et al.**, **2014**). However, abiotic loss of DCF did not exceed 8% at all tested concentrations, and biotic removal is the most probable process in accordance with literature data (Escapa et al.,
2017).

Biotic removal can be passive, by hydrophobic binding on microalgal cell walls or to the 326 extracellular polysaccharides (Xiong et al., 2018), and active when accumulation and 327 biotransformation/biodegradation occur within cells. In the present results, less than 1% of 328 DCF was removed by adsorption and in equal amount by intracellular accumulation. Since 329 EPS can be released into the medium (Mezhoud et al., 2014), it is possible to hypothesize 330 that free EPS contributed significantly to DCF removal. Taking into account the method used 331 in this study, the proportion of adsorbed DCF could be higher than that estimated. 332 333 Nevertheless, Bácsi et al. (2016) demonstrated in several microalgae that DCF was not permanently bound to the membranes, and that no intracellular accumulation occurred, in 334 accordance to the data presented here. As reduction of DCF concentration in the cultures 335 336 cannot be balanced by adsorbed and/or accumulated DCF in the cells, these results suggest that biodegradation processes occurred in both Picocystis and Graesiella cultures. So far, little 337 338 is known about microalgae that degrade diclofenac and the involved biodegradation pathways. 339 In this study, two degradation products of DCF were detected in Picocystis cultures, a hydroxylated compound and a mono-aromatic product. Hydroxy diclofenac has already been 340 reported as a major product of DCF metabolism in bacteria, fungi and microalgae, with a 341 toxicity reduction compare to the parent product (Domaradzka et al., 2015; Escapa et al., 342 2018). The mono-aromatic derivative results from the loss of the benzene dechlorinated ring, 343 as detected by Cooper and Song (2012) in advanced oxidation processes of DCF. 344

However, no DCF metabolites were detected in *Graesiella* cultures despite the high decrease in initial DCF concentrations observed. This can be either due to analytical issues (metabolites below detection limits, procedures not suitable for unforeseen compounds, ...) or kinetic factors, as the hypothesis of total mineralization cannot be excluded.

There is no evidence that the biodegradation is intracellular and known biodegradation 349 processes involve the production of reactive oxygen species (ROS) in the medium as 350 consequence of photosynthesis, and resulting photo-oxidation or hydroxylation of 351 contaminants (Oral and Kantar, 2019). Extracellular enzymes are also identified in 352 extracellular matrix of microalgae involved in chemicals hydrolysis (Otto et al., 2015). 353 Indeed, EPS can form a hydrated biofilm matrix acting as an external digestive system where 354 extracellular enzymes close to cells, allowing them to metabolize organic compounds 355 356 (Flemming and Wingender, 2010; Xiong et al., 2018). In this context, DCF byproducts formed could be physically trapped or included in EPS matrix and escape analysis, which 357 358 need further investigation.

#### 359 **5.** Conclusion

Based on IC50 values, *Picocystis* and *Graesiella* were, relatively, tolerant to DCF compared to other chlorophyta species as reported in the literature. The growth of both species was inhibited by less than 40% even at 200 mg L<sup>-1</sup> DCF. Interestingly, *Picocystis* and *Graesiella* exhibited high DCF removal efficiencies reaching 73% and 52% of 25 mg L<sup>-1</sup> as initial concentration, respectively.

The main proportion of DCF removal (69% and 44% of 25 mg L<sup>-1</sup> initial DCF concentration) 365 was insured by biodegradation processes. Two biodegradation products of DCF were 366 identified in the presence of *Picocystis*: hydroxy-diclofenac and a mono-aromatic derivative 367 of DCF. In Graesiella culture, no metabolites were detected despite the high DCF 368 biodegradation percentage, this may be associated either to low metabolites concentrations 369 370 below the detection limit of the analytical method used or to the total mineralization of diclofenac. Moreover, the role of EPS in removal processes may be investigated. Further 371 372 research on the biodegradation pathway and the toxicity assessment of the bioproducts still

required. The obtained results open up promising prospects for the application of the microalgae here considered, mainly *Picocystis*, in DCF bioremediation systems. The high tolerance of studied species allows their use in continuous flow culture systems particularly in the case of bioremediation of wastewaters, characterized by lower DCF concentrations. Therefore, we can concentrate or degrade the pollutant faster and with low culture volume. Although, confirmation of these results in pilot-scale at environmental conditions still needed.

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#### 385 **References**

- 386 Acuña, V., Ginebreda, A., Mor, J. R., Petrovic, M., Sabater, S., Sumpter, J., Barceló, D.,
- 2015. Balancing the health benefits and environmental risks of pharmaceuticals:
  Diclofenac as an example. Environ. Int. 85, 327–333. doi:10.1016/j.envint.2015.09.023
- Bácsi, I., B-Béres, V., Kókai, Z., Gonda, S., Novák, Z., Nagy, S.A., Vasas, G., 2016.
  Effects of non-steroidal anti-inflammatory drugs on cyanobacteria and algae in
  laboratory strains and in natural algal assemblages. Environ. Pollut. 212, 508–518.
  https://doi.org/10.1016/j.envpol.2016.02.031
- 393 Ben Ali, R., Ben Ouada, S., Chouchene, L., Messaoudi, I., Ben Ouada, H., Othmane, A.,
- **2017.** Cadmium effect on physiological responses of the tolerant Chlorophyta specie
- 395 *Picocystis* sp. isolated from Tunisian wastewaters. Environ. Sci. Pollut. Res. 24, 1803–
  396 1810. https://doi.org/10.1007/s11356-016-7950-0
- 397 Ben Ouada, S., Ben Ali, R., Leboulanger, C., Zaghden, H., Choura, S., Ben Ouada, H.,
- 398 Sayadi, S., 2018. Effect of Bisphenol A on the extremophilic microalgal strain *Picocystis*
- sp. (Chlorophyta) and its high BPA removal ability. Ecotoxicol. Environ. Saf., 158, 1-8.
- 400 https://doi.org/10.1016/j.ecoenv.2018.04.008
- Bischoff, H.W., Bold, H.C., 1963. Some soil algae from Enchanted Rock and related algal
  species, in: Austin, T. (Ed.), Phycological Studies IV. University of Texas Publication
  6318, pp. 1–95.
- 404 Campbell, D., Hurry, V., Clarke, K., Gustafsson, P., Oquist, G., 1998. Chlorophyll
  405 fluorescence analysis of cyanobacterial photosynthesis and acclimation. Microbiol. Mol.
  406 Biol. Rev. 62, 667–83.
- 407 Choi, C.J., Berges, J.A., Young, E.B., 2012. Rapid effects of diverse toxic water pollutants
  408 on chlorophyll a fluorescence: variable responses among freshwater microalgae. Water
  409 Res. 46 (8), 2615–2626. http://dx.doi.org/10.1016/j.watres.2012.02.027

410 Cooper, W.J., Song, W., 2012. Advanced oxidation degradation of diclofenac. IAEA Rep.
411 46, 168–179.

# Cuellar-Bermudez, S.P., Aleman-Nava, G.S., Chandra, R., Garcia-Perez, J.S., Contreras-Angulo, J.R., Markou, G., Muylaert, K., Rittmann, B.E., Parra-Saldivar,

- 414 **R., 2017.** Nutrients utilization and contaminants removal. A review of two approaches of
- 415 algae and cyanobacteria in wastewater. Algal Res. 24, 438–449.
  416 https://doi.org/10.1016/j.algal.2016.08.018
- de Orte, M.R., Carballeira, C., Viana, I.G., Carballeira, A., 2013. Assessing the toxicity
  of chemical compounds associated with marine land-based fish farms: The use of miniscale microalgal toxicity tests. Chem. Ecol. 29, 554–563.
  https://doi.org/10.1080/02757540.2013.790381
- 421 DeLorenzo, M.E., Fleming, J., 2008. Individual and mixture effects of selected
  422 pharmaceuticals and personal care products on the marine phytoplankton species
  423 *Dunaliella tertiolecta*. Arch. Environ. Contam. Toxicol. 54, 203–210.
  424 https://doi.org/10.1007/s00244-007-9032-2
- 425 Domaradzka, D., Guzik, U., Wojcieszyńska, D., 2015. Biodegradation and
  426 biotransformation of polycyclic non-steroidal anti-inflammatory drugs. Rev. Environ. Sci.
  427 Biotechnol. 14, 229–239. https://doi.org/10.1007/s11157-015-9364-8
- 428 Elser, J.J., Fagan, W.F., Denno, R.F., Dobberfuhl, D.R., Folarin, A., Huberty, A.,
- 429 Interlandi, S., Kilham, S.S., McCauley, E., Schulz, K.L., Siemann, E.H., Sterner,
- **R.W., 2000.** Nutritional constraints in terrestrial and freshwater food webs. Nature 408,
  578–580. https://doi.org/10.1038/35046058
- 432 Escapa, C., Coimbra, R.N., Paniagua, S., García, A.I., Otero, M., 2017. Comparative
  433 assessment of pharmaceutical removal from wastewater by the microalgae *Chlorella*434 sorokiniana, *Chlorella vulgaris* and *Scenedesmus obliquus*. In: Biological Wastewater

435

traetment and resource recovery, Robina Farooq and Zaki Ahmad (Eds.), Intech, pp. 99-

436 117 (DOI: 10.5772/66772).

# 437 Escapa, C., Torres, T., Neuparth, T., Coimbra, R. N., García, A. I., Santos, M. M., Otero,

- M., 2018. Zebrafish embryo bioassays for a comprehensive evaluation of microalgae
  efficiency in the removal of diclofenac from water. Sci. Total Environ. 640-641, 1024–
  1033. doi:10.1016/j.scitotenv.2018.05.353
- Flemming, H.-C., Wingender, J., 2010. The biofilm matrix. Nat. Rev. Microbiol. 8(9), 623633.
- Ji, M.-K., Kabra, A.N., Choi, J., Hwang, J.-H., Kim, J.R., Abou-Shanab, R. a. I., Oh, Y.-
- K., Jeon, B.-H., 2014. Biodegradation of bisphenol A by the freshwater microalgae *Chlamydomonas mexicana* and *Chlorella vulgaris*. Ecol. Eng. 73, 260–269.
  https://doi.org/10.1016/j.ecoleng.2014.09.070
- Kummerová, M., Zezulka, S., Babula, P., Tríska, J., 2016. Possible ecological risk of two
  pharmaceuticals diclofenac and paracetamol demonstrated on a model plant *Lemna minor*. J. Hazard. Mater. 302, 351–361. https://doi.org/10.1016/j.jhazmat.2015.09.057
- Langenhoff, A., Inderfurth, N., Veuskens, T., Schraa, G., Blokland, M., KujawaRoeleveld, K., Rijnaarts, H., 2013. Microbial removal of the pharmaceutical
  compounds ibuprofen and diclofenac from wastewater. Biomed Res. Int. 2013, 1–9.
  https://doi.org/10.1155/2013/325806
- 454 Li, R., Chen, G.Z., Tam, N.F.Y., Luan, T.G., Shin, P.K.S., Cheung, S.G., Liu, Y., 2009.
- 455 Toxicity of bisphenol a and its bioaccumulation and removal by a marine microalga
  456 *Stephanodiscus hantzschii*. Ecotoxicol. Environ. Safe. 72, 321–328.
- 457 Lonappan, L., Brar, S.K., Das, R.K., Verma, M., Surampalli, R.Y., 2016. Diclofenac and
- 458 its transformation products: Environmental occurrence and toxicity A review. Environ.
- 459 Int. 96, 127–138. https://doi.org/10.1016/j.envint.2016.09.014

Matamoros, V., Gutiérrez, R., Ferrer, I., García, J., Bayona, J.M., 2015a. Capability of 460 microalgae-based wastewater treatment systems to remove emerging 461 organic contaminants: pilot-scale study. J. Hazard. Mater. 288, 34-42. 462 А https://doi.org/10.1016/j.jhazmat.2015.02.002 463

- Matamoros, V., Uggetti, E., García, J., Bayona, J.M., 2015b. Assessment of the
  mechanisms involved in the removal of emerging contaminants by microalgae from
  wastewater: A laboratory scale study. J. Hazard. Mater. 301, 197–205.
  https://doi.org/10.1016/j.jhazmat.2015.08.050
- Mezhoud, N., Zili, F., Bouzidi, N., Helaoui, F., Ammar, J., Ben Ouada, H., 2014. The
  effects of temperature and light intensity on growth, reproduction and EPS synthesis of a
  thermophilic strain related to the genus *Graesiella*. Bioprocess. Biosyst. Eng.
  37(11),2271-80. doi: 10.1007/s00449-014-1204-7.
- 472 Michael, I., Achilleos, A., Lambropoulou, D., Torrens, V.O., Pérez, S., Petrović, M.,
  473 Barceló, D., Fatta-Kassinos, D., 2014. Proposed transformation pathway and evolution
  474 profile of diclofenac and ibuprofen transformation products during (sono)photocatalysis.

475 Appl. Catal. B Environ. 147, 1015–1027. https://doi.org/10.1016/j.apcatb.2013.10.035

- 476 Müller, P., Li, X.-P., Niyogi, K.K., 2001. Non-photochemical quenching a response to
  477 excess light energy. Plant Physiol. 125, 1558–1566.
  478 https://doi.org/10.1104/pp.125.4.1558
- 479 Oral, O., Kantar, C., 2019. Diclofenac removal by pyrite-Fenton process: Performance in
  480 batch and fixed-bed continuous flow systems. Sci. Total Environ. 664, 817-823.
  481 https://doi.org/10.1016/j.scitotenv.2019.02.084.
- Otto, B., Beuchel, C., Liers, C., Reisser, W., Harms, H., Schlosser, D., 2015. Laccase-like
  enzyme activities from chlorophycean green algae with potential for bioconversion of
  phenolic pollutants. FEMS Microbiol. Lett. 362 (11), fnv072.

485

http://dx.doi.org/10.1093/femsle/fnv072.

486 Peeples, T.L., 2014. Bioremediation using extremophiles, in: Microbial biodegradation and
487 bioremediation. Elsevier, pp. 251–268. https://doi.org/10.1016/B978-0-12-800021488 2.00010-8

Ratha, S.K., Babu, S., Renuka, N., Prasanna, R., Prasad, R.B.N., Saxena, A.K., 2013.
Exploring nutritional modes of cultivation for enhancing lipid accumulation in microalgae. J. Basic Microbiol. 53, 440–450. http://dx.doi.org/10.1002/jobm.201200001.

492 Rehman, A., Shakoori, A.R., 2004. Tolerance and uptake of cadmium and nickle by
493 *Chlorella* sp., isolated from tannery effluents. Pakistan J. Zool. 36, 327–331.

494 Rigobello, E.S., Dantas, A.D.B., Di Bernardo, L., Vieira, E.M., 2013. Removal of
495 diclofenac by conventional drinking water treatment processes and granular activated
496 carbon filtration. Chemosphere 92, 184–191.
497 https://doi.org/10.1016/j.chemosphere.2013.03.010

Sophia A., C., Lima, E. C., 2018. Removal of emerging contaminants from the environment
by adsorption. Ecotoxicol. Environ. Safe. 150: 1–17. doi:10.1016/j.ecoenv.2017.12.026

Strasser, R.J., Srivastava, A., Tsimilli-Michael, M., 2000. The fluorescence transient as a
tool to characterize and screen photosynthetic samples, in: Yunus, M., Pathre, U.,
Mohanty, P. (Eds.), Probing photosynthesis: Mechanism, regulation and adaption.
London, pp. 445–483.

Valavanidis, A., Vlachogianni, T., Loridas, S., Fiotakis, C., 2014. An emerging
environmental problem disposed medicinal active products pharmaceuticals, antibiotics,
and disinfectants in the aquatic environment and toxicological considerations.
Pharmakeftiki 26(III), 78-98.

Varshney, P., Mikulic, P., Vonshak, A., Beardall, J., Wangikar, P.P., 2014. Extremophilic
 micro-algae and their potential contribution in biotechnology. Bioresour. Technol.

510 https://doi.org/10.1016/j.biortech.2014.11.040

# 511 Villar-Navarro, E., Baena-Nogueras, R. M., Paniw, M., Perales, J. A., Lara-Martín, P.

- A., 2018. Removal of pharmaceuticals in urban wastewater: High rate algae pond
  (HRAP) based technologies as an alternative to activated sludge based processes. Water
- 514 Res. 139: 19-29.doi:10.1016/j.watres.2018.03.072
- Xiong, J.-Q., Kurade, M. B., Jeon, B.-H., 2018. Can microalgae remove pharmaceutical
  contaminants from water? Trends Biotechnol. 36(1): 30–44.
  doi:10.1016/j.tibtech.2017.09.003
- 518 Xu, M., Huang, H., Li, N., Li, F., Wang, D., Luo, Q., 2019. Occurrence and ecological risk
- of pharmaceuticals and personal care products (PPCPs) and pesticides in typical surface
  watersheds, China. Ecotoxicol. Environ. Safe. doi:10.1016/j.ecoenv.2019.01.131
- 521 Zarrouk, C., 1966. Contribution à l'étude d'une Cyanophycée, influence de divers facteurs
  522 physiques et chimiques sur la croissance et la photosynthèse de "*Spirulina maxima*"
  523 (Setch et Gardner) Geitler. PhD thesis, University of Paris, France.
- Zhou, G.-J., Ying, G.-G., Liu, S., Zhou, L.-J., Chen, Z.-F., Peng, F.-Q., 2014.
  Simultaneous removal of inorganic and organic compounds in wastewater by freshwater
  green microalgae. Environ. Sci. Process. Impacts 16, 2018.
  https://doi.org/10.1039/C4EM00094C
- 528Zili, F., Mezhoud, N., Trabelsi, L., Chreif, I., Ben Ouada, H., 2015.Fatty acid composition
- 529 of the thermophilic *Gloeocapsa gelatinosa* under different combinations of temperature,
- 530 light intensity, and NaNO<sub>3</sub> concentration. J. Appl. Phycol. 27, 97.
  531 https://doi.org/10.1007/s10811-014-0296-4
- 532

Fig. 1. Full scan spectrum of the DCF reference solution (DCF in ultra-pure water) in a scan
range of 60 to 500 m/z

**Fig. 2.** Variation in the PSII photochemical efficiency (Fv/Fm) of (a) *Picocystis* sp. and (b) *Graesiella* sp. during 5 days of DCF exposure. Data are represented as mean  $\pm$  SD (n=3). Same letters indicate no significant difference according to Post Hoc test (P $\ge$ 0.05) (uppercase letters for culture duration and lowercase letters for tested concentrations).

**Fig. 3.** Variation in the non-photochemical quenching (NPQ) of (a) *Picocystis* sp. and (b) *Graesiella* sp. under exposure to 0, 25, 50, 100 and 200 mg L<sup>-1</sup> DCF for 5 days. Data are represented as mean  $\pm$  SD (n=3). Same letters indicate no significant difference according to Post Hoc test (P $\ge$ 0.05) (uppercase letters for culture duration and lowercase letters for tested concentrations).

Fig. 4. LC-MS/MS ion chromatograms showing the degradation of 100 mg L<sup>-1</sup> initial DCF
concentration by *Picocystis* sp.: (a) DCF without microalgae, (b) microalgae without DCF and
(c) DCF in presence of *Picocystis* sp.

Fig. 5. ECI-MS2 spectra of DCF degradation products obtained in presence of *Picocystis* sp.:
(a) DP1 and (b) DP2.

## 551 **Tables caption**

**Table 1.** Growth inhibition percentage of *Picocystis* sp. and *Graesiella* sp., calculated from the area under the growth curves, as a function of DCF concentrations after 1, 2, 3, 4 and 5 days of exposure. Data are represented as mean  $\pm$  SD (n=3). Same letters indicate no significant difference according to Post Hoc test (P $\ge$ 0.05) (uppercase letters for culture duration and lowercase letters for tested concentrations).

**Table 2.** Total (Tr) and abiotic removal (Ab), accumulated/adsorbed amount (Ac/Ad), biodegradation percentage (Bp) and specific removal (RS) of DCF by *Picocystis* sp. and *Graesiella* sp. cells exposed during five days to initial concentrations of 25, 50 and 100 mg L<sup>-1</sup> DCF. Data are represented as mean  $\pm$  SD (n=3). Different letters within the same column indicate significant difference according to Duncan test (P≥0.05).



Retention time (min)



















Growth inhibition percentage (%)								
	Day 1	Day 2	Day 3	Day 4	Day 5			
Picocystis								
25 mg L <sup>-1</sup>	$-4.37 \pm 0.85$ Cb	$-6.55 \pm 0.91$ Ab	$-8.54 \pm 0.82$ ABb	$-8.18 \pm 0.62$ BCb	$-14.27 \pm 2.62$ Ab			
50 mg L <sup>-1</sup>	$-6.32 \pm 1.02$ Ca	-10.48 ± 1.32Aa	$-12.95 \pm 0.41$ Aba	-14.51 ± 1.03BCa	-21.36 ± 2.03Aa			
$100 \text{ mg } \text{L}^{-1}$	$8.41 \pm 3.24$ Cc	$7.61 \pm 1.69$ Ac	$11.77 \pm 0.61$ Abc	13.42 ± 1.06BCc	$16.20 \pm 1.06$ Ac			
200 mg L <sup>-1</sup>	$12.67 \pm 1.53$ Cd	$11.05 \pm 0.96$ Ad	$14.24 \pm 0.52$ Abd	$16.68 \pm 0.56$ BCd	$20.56 \pm 3.56$ Ad			
Graesiella								
25 mg L <sup>-1</sup>	$7.10 \pm 0.24$ Aa	$7.65 \pm 0.50$ Ba	$11.41 \pm 1.24$ Ca	13.67 ± 1.53Da	$16.48 \pm 1.38$			
50 mg L <sup>-1</sup>	$7.95 \pm 0.90$ Ab	9.31 ± 1.76вь	12.61 ± 1.69Cb	$16.05 \pm 0.96$ Db	$19.27 \pm 2.02$			
100 mg L <sup>-1</sup>	$9.30 \pm 1.15$ Ac	12.76 ± 0.26Вс	$16.77 \pm 0.61$ Cc	$20.24 \pm 1.52 \text{Dc}$	$27.48 \pm 0.99$			
200 mg L <sup>-1</sup>	$16.93 \pm 0.32$ Ad	$20.74 \pm 1.49$ Bd	$26.42 \pm 1.06$ Cd	$28.68 \pm 0.56$ Db	$36.33 \pm 2.18$			

Tr (%)	Ab (%)	Ac/Ad (%)	<b>Bp</b> (%)	SR (mg g <sup>-1</sup> )
$73.04 \pm 3.82a$	$3.83 \pm 1.05a$	$0.04 \pm 0.01a$	69.17± 4.88a	$13.22 \pm 1.27a$
$42.46 \pm 2.08b$	$6.81 \pm 2.22b$	$0.12 \pm 0.02b$	$35.53 \pm 4.32b$	$12.39 \pm 1.66a$
24.81±1.58c	$4.28 \pm 1.32$ ab	$0.17 \pm 0.04 b$	20.36± 2.94c	18.96±3.55a
$52.21 \pm 3.22a$	7.53± 1.14a	$0.21 \pm 0.06a$	$44.47 \pm 4.42a$	$6.65 \pm 0.95a$
28.34± 1.17b	6.52± 1.27a	$0.44 \pm 0.04a$	$21.38 \pm 2.48b$	7.31±1.19a
23.5±1.82c	5.15± 1.53a	$0.31 \pm 0.03a$	18.04± 3.38b	$12.23 \pm 3.24$ b
	Tr (%) 73.04 ± 3.82a 42.46 ± 2.08b 24.81± 1.58c 52.21 ± 3.22a 28.34± 1.17b 23.5± 1.82c	Tr (%)Ab (%) $73.04 \pm 3.82a$ $3.83 \pm 1.05a$ $42.46 \pm 2.08b$ $6.81 \pm 2.22b$ $24.81 \pm 1.58c$ $4.28 \pm 1.32ab$ $52.21 \pm 3.22a$ $7.53 \pm 1.14a$ $28.34 \pm 1.17b$ $6.52 \pm 1.27a$ $23.5 \pm 1.82c$ $5.15 \pm 1.53a$	Tr (%)Ab (%)Ac/Ad (%) $73.04 \pm 3.82a$ $3.83 \pm 1.05a$ $0.04 \pm 0.01a$ $42.46 \pm 2.08b$ $6.81 \pm 2.22b$ $0.12 \pm 0.02b$ $24.81 \pm 1.58c$ $4.28 \pm 1.32ab$ $0.17 \pm 0.04b$ $52.21 \pm 3.22a$ $7.53 \pm 1.14a$ $0.21 \pm 0.06a$ $28.34 \pm 1.17b$ $6.52 \pm 1.27a$ $0.44 \pm 0.04a$ $23.5 \pm 1.82c$ $5.15 \pm 1.53a$ $0.31 \pm 0.03a$	Tr (%)Ab (%)Ac/Ad (%)Bp (%) $73.04 \pm 3.82a$ $3.83 \pm 1.05a$ $0.04 \pm 0.01a$ $69.17 \pm 4.88a$ $42.46 \pm 2.08b$ $6.81 \pm 2.22b$ $0.12 \pm 0.02b$ $35.53 \pm 4.32b$ $24.81 \pm 1.58c$ $4.28 \pm 1.32ab$ $0.17 \pm 0.04b$ $20.36 \pm 2.94c$ $52.21 \pm 3.22a$ $7.53 \pm 1.14a$ $0.21 \pm 0.06a$ $44.47 \pm 4.42a$ $28.34 \pm 1.17b$ $6.52 \pm 1.27a$ $0.44 \pm 0.04a$ $21.38 \pm 2.48b$ $23.5 \pm 1.82c$ $5.15 \pm 1.53a$ $0.31 \pm 0.03a$ $18.04 \pm 3.38b$

