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Effects of Ag-doped content on the microstructure and thermoelectric properties of CoSb₃ thin films

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Abstract: In this work, Ag doped CoSb₃ thin films were grown directly on the heated substrate by the simple and effective magnetron sputtering process. The micro-structure and thermoelectric properties of the doped thin films are found to strongly depend on the Ag doping content. High-resolution transmission electron microscope image and corresponding selected-area electron diffraction pattern confirm that the doped thin film has single CoSb₃ crystal structure with well-crystallized. The high crystal quality of the doped thin film will lead to good thermoelectric properties. As expected, both the electrical conductivity and Seebeck coefficient are greatly enhanced after Ag doping, resulting in the enhancement of power factor, suggesting that the proposed strategy can be applied to prepare high-performance thermoelectric thin films.

Keywords: Thermoelectricity; Thin films; Cobalt antimonide; Silver; Doping.

1. Introduction

Thermoelectric (TE) device can directly convert heat emanating from the Sun, radioisotopes, automobiles, industrial sectors, or even the human body to electricity [1]. Meanwhile, electricity also can drive a TE device to work as a solid-state heat pump [2]. Therefore, it has gained great interest for their potential applications in many fields, including direct conversion of waste heat from industrial sectors or automobile exhausts to electricity, and solid-state Peltier coolers. Performance of the TE device is related to the dimensionless figure of merit ZT values, which is determined by the Seebeck coefficient (S), electrical conductivity (σ), and thermal conductivity (κ), expressed using an equation of ZT= $S^2\sigma T/\kappa$, where T is the working temperature, $S^2\sigma$ is defined as the power factor (PF), and κ includes the electronic thermal conductivity $\kappa_{\rm ele}$ and the lattice thermal conductivity $\kappa_{\rm flat}$ (κ = $\kappa_{\rm ele}$ + $\kappa_{\rm flat}$) [3].

Recently, increasing demand in micro-scale energy harvesting for power supplies and miniaturized sensors has prompted the investigation of multifunctional micro-TE devices [4]. But traditional TE device manufacturing technique can not meet the requirements of the miniaturization device, such as the micro-size, light weight and high energy density, especially flexible and wearable [5]. Therefore, the demand to apply micro-TE device as an alternative power source for small electric devices and systems are constantly increasing and the research in the micro-TE device are more and more attracting interesting.

Thus, various studied in the preparation of micro-TE device has been done [5-10]. Among them, micro-TE device fabricated based on thin film technique are one of the

most widely studied research fields [11-15]. The reason for this is not only because of the thin films with nanostructure can control the phonon transport by the quantum confinement effect, leading to enhance the TE performance of the materials [16-17], but also due to thin film device offers many advantages, including flexibility, small volume, light weight, high integration, and enhanced compatibility. Moreover, thin film preparing technology is already compatible in the semiconductor manufactory industry. Therefore, thin film TE devices have been considered as the promising technology for realizing the practical of micro-TE application. Consequently, there is a significant demand to develop high-performance thin film TE materials and devices.

Skutterudite CoSb₃ and its related compounds have drawn extensive attention as one of the most promising thermoelectric materials in the intermediate temperature range over the past few decades [18-24]. It is a narrow band gap semiconductor, owning larger carrier mobility, high Seebeck coefficient and complex crystal structure. Especially, CoSb₃ crystallizes in a body-centered cubic structure with space group Im3 and a unit cell consists of 32 atoms, in which Co atoms form eight sub-cubes with pnicogen rings occupying six of them, leaving the final two voids or cages empty. Thus, if the doped-element can fill into these voids, the Seebeck coefficient would be enhanced and the thermal conductivity reduced, resulting in dramatically increase of ZT value due to the "rattling" effect of filling which can effectively scatter wider-range amplified phonons [25-30]. For instance, Wang et al. demonstrated that the introduction of Yb in CoSb₃ can raise the ZT value from 0.5 to 1.5 at 850 K and a high average value of ~1.27 within 500 K - 850 K was obtained with the optimized

Yb content of 0.29 % [27]. Shi et al. reported that the multi-filling of CoSb₃ can lead to a maximum ZT of 1.7 at 850 K, which is several times higher than the pure sample [28]. G. Rogl et al. demonstrated that In-doped multi-filled n-type skutterudites can exhibit a high power factor and a very low thermal conductivity, leading to an outstanding ZT value of 1.8 at 823 K [29]. Zhou et al. shown that the ZT is 30% higher in comparison to the Ba filled CoSb₃ after inclusion of Ag nanoparticle [30]. Although filling is an efficient way for improving the properties of CoSb₃ materials, reports on the thermoelectric properties of filled-CoSb₃ thin films were scarce.

In our previous work, we found that the Ag can fill into the lattice voids and efficiently regulate the performance of the CoSb₃ thin film. However, there are many defect states on the thin film surface impeding the further improvement of thin film thermoelectric properties [31]. Recently, we have found that the CoSb₃ thin film had less surface defects when heated the substrate during the thin film deposition, rather than annealing after deposition process [32]. Therefore, in this work, Ag doped thin films were grown directly on the heated substrate with fixed temperature, and the microstructure and thermoelectric properties of the thin films were investigated. As expected, the thin film with great crystallinity and enhanced thermoelectric properties were obtained.

2. Experimental details

Ag target (99.99 %) and $CoSb_3$ (99.95%) alloy target with Co/Sb atomic ratio of 1:3 were fixed in the magnetron sputtering facility, equipped with a three-position rotatable target system. The sputtering angle was 45 ° and the target-substrate distance

was 10 cm. Glass substrates were used and cleaned in a ultrasonic bath for 10 minutes in acetone, 10 min in absolute ethyl alcohol and 10 min in deionized water. The vacuum chamber was firstly pumped down to 8.0×10^{-4} Pa and the working pressure was kept at 0.4 Pa with argon flow of 40 sccm. As shown in Figure 1, the glass substrate was heated to be 250 °C before the deposition process. Then the CoSb₃ layer was deposited onto the substrate with a sputtering power of 50 W and the deposition time is 15 min. During the middle of the CoSb₃ layer deposition process, the Ag was co-deposited by using a sputtering power of 1 W. For this co-sputtering layer, the deposition time was adjusted from 10 s to 60 s with interval of 10 s for controlling the Ag content. The content of Ag was analyzed by energy dispersive spectroscopy and was respectively 0.3 %, 1.6 %, 2.2 %, 3.8 %, 4.6 % and 5.1 % for different co-deposition time. Some details of the thin films were shown in Table 1.

The crystalline phases were determined by X-ray diffraction (XRD) (D/max2500, Rigaku Corporation) with the 20 angle range of 10 °-80 °, using Cu K_{α} radiation (λ = 0.15406 nm). Raman scattering measurements were performed by using a spectra system Lab Ram Xplora (Horiba Jobin Yvon). The laser excitation was the 514.5 nm line of an Ar⁺ ion laser. The micro-structure was characterized by using scanning electron microscopy (SEM, Zeiss supra 55) and transmission electron microscopy (TEM, JEM-3200FS) with the operating voltage of 3 KV and 300 KV, while the composition analysis was performed by energy dispersive spectroscopy (EDS). The TEM sample was prepared by following process: thin film was firstly scratched down from the substrate and put into the alcohol. Then, the solution was dripped onto the

copper mesh after stirred for ten minutes. The electrical conductivity and Seebeck coefficient were simultaneously measured in the interval of 298 K \sim 573 K by using a Nezsch- SBA458 equipment.

3. Results and discussion

The crystalline phases were determined by X-ray diffraction (XRD) with the 2θ angle range of 10 °-80 °, using Cu K α radiation (λ = 0.15406 nm). Figure 2 shows the XRD patterns of Ag doped CoSb₃ thin films, including the un-doped sample. After comparison to the characteristic CoSb₃ pattern, three main diffraction peaks located at ~32°, ~37°, and ~46° from Ag-doped CoSb₃ thin films are defined as the CoSb₃ (013) (123) and (420) planes. There are many weak peaks in all the patterns and they also can be attributed to the CoSb₃ skutterudite structure, indicating that all the thin films are composed of pure CoSb₃ structure. As shown in the inset of Figure 2, the peak slightly shifts to smaller angle after Ag doping and the shift increases with increasing Ag content, suggesting crystal cell expansion. To further confirm the Ag state in the films. Raman scattering measurements were used and the results are shown in Figure 3. Four modes are observed in the patterns of all the samples at the positions around ~106 cm⁻¹, ~131 cm⁻¹, ~146 cm⁻¹ and ~179 cm⁻¹, corresponding respectively to the F_{g2}, E_{g1}, A_{g1} and A_{g2} Raman sensitive phonon vibration modes in CoSb₃ predicted by theoretical simulation (The positions of CoSb₃ theoretical standard Raman peaks are around ~83 cm⁻¹, ~97 cm⁻¹, ~139 cm⁻¹,~150 cm⁻¹, ~157 cm⁻¹, ~178 cm⁻¹ and ~179 cm⁻¹, ~182 cm⁻¹, corresponding respectively to the Fg1, Fg2, Eg1, Ag1, Fg3, Fg4, Ag2 and Eg2 Raman sensitive phonon vibration modes) [33]. No

extra peak can be observed, confirming that the films have single cubic structure. Additionally, the disorder Raman peak shifts are due to the lattice expansion. Comparing to our previous work [31], similar results can be found, suggesting that the Ag is most likely filling into the vacant lattice void.

Figure 4 shows surface morphology of thin films, observed using SEM. The results reveal that all the samples exhibit a smooth and dense surface across the entire surface with uniform particles forming more or less large grains. The morphology of the films do not have obvious change after Ag doping. These findings indicate that all the thin films have no obvious crystal imperfection, which should be favorable for electronic transport. The elemental mappings of the sample with Ag content of 3.8 % are shown in Figure 5(a) and the results indicate that all the elements are homogenously distributed in the thin films. High-resolution (HR) TEM image and corresponding selected-area electron diffraction (SAED) pattern are shown in Figure 5(b). It indicates that the film has single crystal structure and the interplanar d-spacings of 0.294 nm and 0.255 nm correspond respectively to the (013) and (123) planes of the skutterudite CoSb₃. The SAED result further confirms the CoSb₃ crystalline phase. The different shades of the grey caused by the different thickness of the position. Therefore, the SEM and HRTEM results indicate that the thin film has high crystal quality with few grain boundaries or other interfaces for electrons scattering, which should lead to good electronic transport property.

Figure 6(a) shows the electrical conductivity, σ , as function of temperature. It can be found that the electrical conductivities of un-doped sample is $0.16 \times 10^4 \text{ S} \cdot \text{m}^{-1}$ at

room-temperature, and is enhanced after Ag doping. Comparatively, samples which Ag contents are > 4.6 % yield high electrical conductivity over 0.26×10^4 S·m⁻¹. The carrier concentration n of the samples with Ag content of 4.6 % and 5.1 % has been measured at room temperature, together with the un-doped sample. The carrier concentration was shown in Table 1. The σ of all the samples increases with the increasing temperature which shows their thermal stability due to the dense structure with good crystallinity. The σ of Ag-doped samples have similar value over 0.50×10^4 S·m⁻¹ and the film with Ag content of 2.2 % has the highest value of 0.65×10^4 S·m⁻¹ at 623 K among these samples.

Figure 6(b) illustrates the temperature dependence of the Seebeck coefficient, S, of the thin films. As shown in Figure 6(b), the Seebeck coefficient of all the samples is negative, exhibiting n-type conduction. S of un-doped sample yields a value of -24.1 μ V·K⁻¹ at 298 K which is an extremely low value compared to that of the CoSb₃ bulk materials, which is about 220 μ V·K⁻¹ [29]. The absolute S value is obviously enhanced when the Ag doping content is ≤ 2.2 %, especially for the sample with Ag content of 0.3%. It has the absolute S value of 43.3 μ V·K⁻¹ which is two times of the un-doped thin film. But the S value of Ag doped thin films is close to the value of the un-doped sample when the films have high Ag doping content of >3.8%. The absolute S values of all the samples increases with increasing temperature and the S of samples with low Ag doping content sharply increases in their absolute value with increasing temperature, which reach to 241.8 μ V·K⁻¹ and 174.9 μ V·K⁻¹ at 623 K when the Ag content is 0.3 % and 1.6 %, respectively. But samples will have close and lower S

values at high measuring temperatures with increasing Ag doping content.

From figures 6(a) and (b), it is worth noting that both of the Seebeck coefficient and the electrical conductivity are enhanced by appropriate Ag doping content. Generally, the S and σ are both dependent on the carrier concentration with inverse relationship. Thus, it demonstrates that Ag is an efficient dopant that can resolve the conflict between S and σ which might attribute to the increased carrier concentration and electron effective mass after the Ag-doping [31]. Additionally, the room-temperature Seebeck coefficient and electrical conductivity of the thin films were measured for three times and the results show that their have few change, indicating high thermal stability of the films.

The temperature dependence of PF at the temperature range from 298 K to 623 K is shown in Figure 6(c) and the inset shows the PF value between 298 K to 498 K. The PF value shows increasing trend as function of temperature for all the samples and the PF value of the Ag-doped samples is higher than the un-doped sample when the temperature is under 498 K as shown in the inset. Although some films have slightly lower value than that of the un-doped sample at higher temperature, the Ag-doped samples generally have larger PF value on the whole. The maximum PF value of $2.97 \times 10^{-4} \text{ Wm}^{-1} \text{K}^{-2}$ is obtained from the film with Ag content of 0.3%, which is two times of the un-doped sample's PF value.

4. Conclusion

The micro-structure and thermoelectric properties of Ag doped CoSb₃ thin films which grew directly on the heated substrate with various Ag contents were

investigated. The microstructure analysis confirms that the doped thin films have well-crystallized, and the Ag have entered into the lattice and most likely occupy the lattice voids. Due to the improved crystallinity and Ag filling effect, the Seebeck coefficient and electrical conductivity are significantly enhanced. As expected, the power factor is increased to a high value of $2.97 \times 10^{-4} \, \text{W} \, \text{m}^{-1} \, \text{K}^{-2}$ which is almost 100 % enhancement after Ag doping.

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Table 1 Thickness and carrier concentration of the thin films as function of Ag content

Ag-doped contents	0	0.3	1.6	2.2	3.8	4.6	5.1
(at.%)	Ü	0.5	1.0	۷.۷	5.0	4.0	5.1
Thickness of film	170	177	175	180	180	105	107
(nm)	170	1//	173	180	180	185	187
Carrier concentration	0.17	1.33	1.55	1.67	127		2.55
$(\times 10^{21}/\text{cm}^3)$					1.77 1.98	1.98	2.55

Figure Captions

- Figure 1 Preparing detail of the Ag doped CoSb₃ thin films.
- Figure 2 XRD patterns of the samples.
- Figure 3 The results of Raman spectrum for $CoSb_3$ doped with different Ag content.
- Figure 4 SEM surface morphology of Ag-doped CoSb₃ thin films.
- Figure 5 (a) The TEM image and elemental mappings, (b) HRTEM image of the Ag-doped sample.
- Figure 6 Temperature dependence of (a) Electrical conductivity, (b) Seebeck coefficient, (c) Power factor PF.

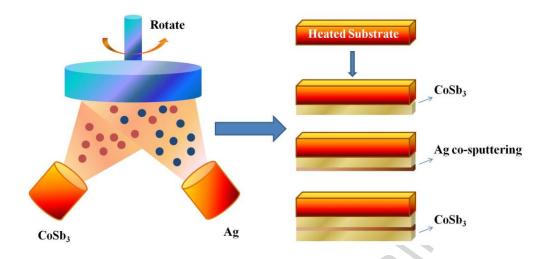


Figure 1

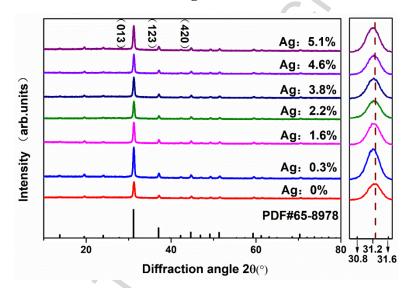


Figure 2

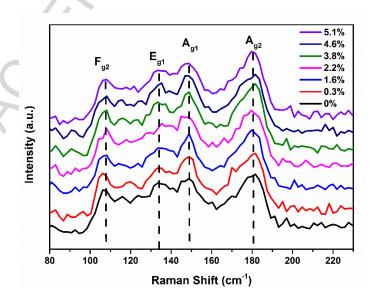


Figure 3

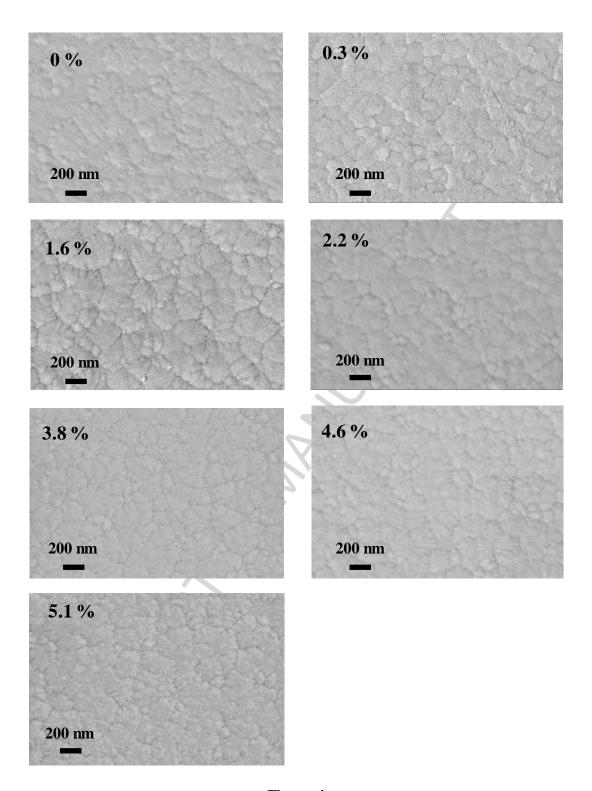


Figure 4

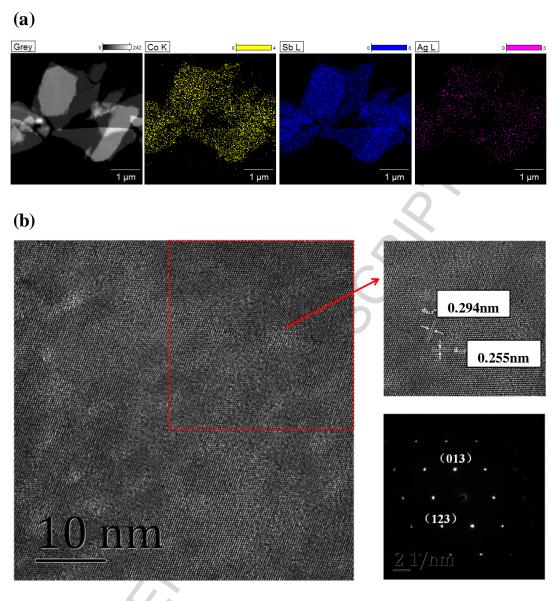
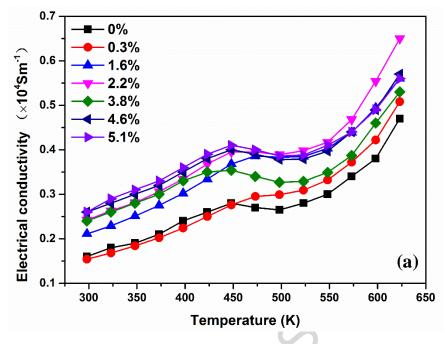
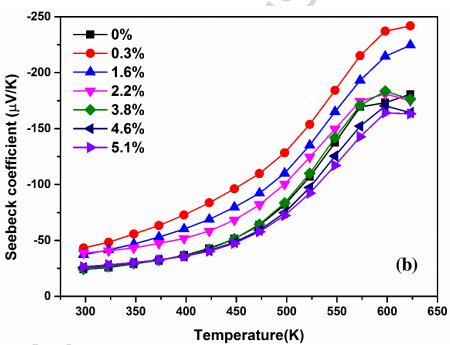


Figure 5





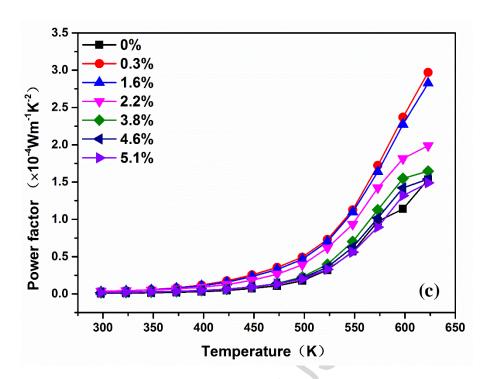


Figure 6

Highlights

- 1. Ag doped CoSb₃ films were grown directly on the heated substrate.
- 2.Both the Seebeck coefficient and electrical conductivity are significantly enhanced.
- 3. Power factor has almost 100 % enhanced after Ag doping

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