

Hyperfine coupling and slow magnetic relaxation in isotopically enriched Dy III mononuclear single-molecule magnets

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J. Flores Gonzalez, Fabrice Pointillart, O. Cador. Hyperfine coupling and slow magnetic relaxation in isotopically enriched Dy III mononuclear single-molecule magnets. Inorganic Chemistry Frontiers, 2019, 6 (4), pp.1081-1086. 10.1039/c8qi01209a. hal-02122161

HAL Id: hal-02122161 https://univ-rennes.hal.science/hal-02122161

Submitted on 1 Apr 2020

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INORGANIC CHEMISTRY

FRONTIERS







RESEARCH ARTICLE

View Article Online
View Journal | View Issue



Cite this: Inorg. Chem. Front., 2019, 6 1081



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The four main stable isotopes of the [A Dy(tta) $_{3}$ (L)]·C $_{6}$ H $_{14}$ (A Dy with A = 161-164) Single-Molecule Magnet (SMM) (tta $^{-}$ = 2-thenoyltrifluoroacetylacetonate and L = 2-{[2-methylpyridiyl]-4,5-[4,5-bis(propylthio)-tetrathiafulvalenyl]-1 H -benzimidazol-2-yl}pyridine) have been magnetically investigated and structurally characterized by single crystal X-ray diffraction. The two nuclear spin-free $^{162/164}$ Dy behave the same from a dynamic magnetic point of view and with slower magnetic relaxation than the two pure magnetically active isotopically substituted $^{161/163}$ Dy. In addition, this paper demonstrates that 161 Dy and 163 Dy relax differently even if they both carry I = 5/2 nuclear spin. After the release of the dipolar interactions through magnetic dilution, ([161 Dy $_{0.05}$ Y $_{0.95}$ (tta) $_{3}$ (L)]·C $_{6}$ H $_{14}$) (161 Dy@Y) relaxes four times slower than 163 Dy@Y due to different hyperfine coupling constants.

Received 7th November 2018, Accepted 31st January 2019 DOI: 10.1039/c8qi01209a

rsc.li/frontiers-inorganic

Introduction

The possibility to observe a magnetic memory at the single-ion level for a lanthanide ion was demonstrated in 2003 for a $[TbPc_2]$ (Pc = phthalocyaninato) complex. Since that discovery, lanthanide-based mononuclear Single-Molecule Magnets (SMMs) have been designed for their potential applications in high-density data storage and spintronic devices.

On one hand, the integration of lanthanide SMMs in devices for data storage is limited due to the low-temperature range at which such molecular systems highlight magnetic memory. To overcome such a limitation, few strategies have been explored: (i) a molecular engineering strategy to induce an adequate symmetry and crystal field around the lanthanide ion leading to an optimal crystal splitting of the ground multiplet state. Such a strategy has recently driven chemists to design a hexa-tert-butyldysprosocenium complex with an exceptional magnetic axiality and a blocking temperature of 60 K;^{5,6,7-10} (ii) the minimization of the magnetic moment perturbation by magnetic dilution by adding the complex to a frozen solution¹¹ or doping it with a diamagnetic isomorphous matrix¹² which leads to the isolation of the electronic spin carriers; and (iii) the minimization of the hyperfine interactions by isotopic enrichment in nuclear spin-free lanthanides which

leads to a diminution of the Quantum Tunnelling of Magnetization (QTM) efficiency. ^{13,14}

On the other hand, lanthanide SMMs could be suitable to perform nuclear spin qubits/qudits^{4,15–18} due to the several DiVincenzo characteristics^{19,20} already observed for such systems. In other words, two nuclear isomers of a lanthanide SMM could be used to enhance the magnetic performance for data storage (nuclear spin-free isomer) while the pure nuclear spin isomer could be used as a multilevel nuclear spin qubit (qudit) because of the nuclear-spin-driven QTM events.²¹ Nevertheless the Dy^{III} ion is composed of seven isotopes with nuclear spin values of 0 and 5/2. The four main ¹⁶¹Dy, ¹⁶²Dy, ¹⁶³Dy and ¹⁶⁴Dy isotopes represent 97.5% of the isotopic mixture but until now only two of them have been studied on the same system.^{13,14,21}

In this paper, we propose to go a step further in studying four stable isotopically enriched dysprosium complexes $([^{A}Dy(tta)_{3}(L)]\cdot C_{6}H_{14}^{22}$ where $tta^{-} = 2$ -thenoyltrifluoroacetylacetonate, L = 2-{[2-methylpyridiyl]-4,5-[4,5-bis(propylthio)tetrathiafulvalenyl]-1*H*-benzimidazol-2-ylpyridine) and A = 161, 162, 163 and 164 (abbreviated as ^ADy here after). It is worth noting that 161Dy and 164Dy have been previously studied by some of us. 13 The aim of this study was to pursue the investigations on an identical system for which all the synthetic steps are under control. The two ¹⁶¹Dy and ¹⁶³Dy isotopes possess the same nuclear spin value (I = 5/2) but not the same nuclear magnetic moment while the 162Dy and 164Dy isotopes possess no nuclear spins. Magnetic dilution of the dysprosium complexes in an isomorphous diamagnetic matrix based on yttrium (ADy@Y) has been systematically performed to cancel dipolar interactions in the condensed phase.

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 $\dagger\,\text{Electronic}$ supplementary information (ESI) available. See DOI: 10.1039/c8qi01209a

Materials and methods

The elemental analyses of the compounds were performed at the Centre Régional de Mesures Physiques de l'Ouest, Rennes. The Dy/Y ratio of the compounds 162Dy@Y and 163Dy@Y was determined using SEM (Scanning Electron Microscopy). All observations and measurements were carried out with a JEOL ISM 6400 scanning electron microscope (JEOL Ltd, Tokyo, Japan) with an EDS (Energy Dispersive Spectrometry) analysis system (Oxford Link INCA). The voltage was kept at 9 kV, and the samples were mounted on carbon stubs and coated for 5 min with a gold/palladium alloy using a sputter coater (Jeol IFC 1100). This analysis has been performed by the "Centre de Microscopie Electronique à Balayage et microAnalyse (CMEBA)" from the University of Rennes 1 (France). The dc magnetic susceptibility measurements were performed on solid polycrystalline samples with a Quantum Design MPMS-XL SQUID magnetometer between 2 and 300 K in an applied magnetic field of 200 Oe for temperatures between 2 and 20 K, 2 kOe between 20 and 80 K and 10 kOe above 80 K. Making measurements at various applied fields allows us to verify that the samples are free from ferromagnetic impurities and also avoid saturation phenomena at low temperature. These measurements were all corrected for the diamagnetic contribution as calculated with Pascal's constants.

Experimental

All solvents were dried using standard procedures. 4,5-Bis(propylthio)-tetrathiafulvalene-2-(2-pyridyl)benzimidazole-methyl-2-pyridine ligand (L) was synthesized following a published procedure. 23 The isotopically enriched $\mathrm{Dy}_2\mathrm{O}_3$ oxides are commercially available from Eurisotop company. All other reagents were purchased from Aldrich Co. Ltd and were used without further purification.

Isotopically enriched Dy(tta) $_3\cdot 2H_2O$ compounds are prepared according to the following modified literature procedure: On one side, 47.1 mg (0.126 mmol) of Dy $_2O_3$ enriched at 94.4% in 162 Dy III or at 95.0% in 163 Dy III were placed in the presence of 130 μ L of concentrated HCl. After 30 min of stirring at 50 °C, the resulting solution of $^{162/163}$ DyCl $_3\cdot 6H_2O$ was diluted with 1 mL of water. On the other side, 224 mg of Htta (1.010 mmol) were dissolved in 180 mL of water at 60 °C under intense stirring. The pH of the solution was adjusted to 6–6.5 with NH $_4$ OH. To the obtained solution was added an aqueous solution of isotopically enriched DyCl $_3\cdot 6H_2O$ (0.252 mmol). The pH of the solution was adjusted to 7–7.5 during the stirring and then flocculation of Dy(tta) $_3\cdot 2H_2O$ took place and was filtered. Yield 162 Dy(tta) $_3\cdot 2H_2O$: 115 mg (53%). Yield 163 Dy(tta) $_3\cdot 2H_2O$: 110 mg (51%).

 $[^{162/163}$ Dy(tta)₃(L)]·C₆H₁₄ (162 Dy and 163 Dy): a CH₂Cl₂ solution of L (24.4 mg, 0.04 mmol) and $^{162/163}$ Dy(tta)₃·2H₂O (34.5 mg, 0.04 mmol) is stirred for 1 hour. n-Hexane is layered to the solution and red prismatic single crystals are obtained after two weeks of diffusion and evaporation at room tempera-

ture. Yield 162 Dy: 47.7 mg (83%). Yield 163 Dy: 50.0 mg (87%). Anal. calcd (%) for $C_{52}H_{38}$ Dy $F_{9}N_{4}O_{6}S_{9}$: C 43.47, H 2.65, N 3.90; found 162 Dy: C 43.61, H 2.72, N 3.79. found 163 Dy: C 43.66, H 2.77, N 3.83.

[$^{162/163}$ Dy_{0.05}Y_{0.95}(tta)₃(L)]·C₆H₁₄ ($^{162/163}$ Dy@Y): a CH₂Cl₂ solution (15 mL) of L (61.0 mg, 0.1 mmol) is added to a CH₂Cl₂ solution (10 mL) containing $^{162/163}$ Dy(tta)₃·2H₂O (3.4 mg, 0.004 mmol) and Y(tta)₃·2H₂O (75.7 mg, 0.096 mmol) and stirred for 1 hour. n-Hexane is layered to the solution and red prismatic single crystals are obtained after two weeks of diffusion and evaporation at room temperature. Yield 162 Dy@Y: 47.7 mg (80%). Yield 163 Dy@Y: 50.0 mg (85%). C₅₂H₃₈Dy_{0.05}Y_{0.95}F₉N₄O₆S₉: C 45.69, H 2.78, N 4.10; found 162 Dy: C 45.60, H 2.82, N 3.99; found 163 Dy: C 45.66, H 2.87, N 4.03.

Results and discussion

Structural description

The four compounds 162 Dy, 163 Dy, 162 Dy@Y and 163 Dy@Y are isostructural (Table S1, Fig. S1–S4†). The ratio Dy/Y (0.05/0.95) for the two doped compounds $[^{162}$ Dy_0.05Y_0.95(tta)_3(L)]·C₆H₁₄ (162 Dy@Y) and $[^{163}$ Dy_0.05Y_0.95(tta)_3(L)]·C₆H₁₄ (163 Dy@Y) has been determined by EDS analysis and then it is used to refine the X-ray structure on single crystals. The X-ray structures for the natural isotopes, 161 Dy and 164 Dy, have been studied in our previous work. 13,22 Thus a single and rapid structural description is given below to understand the structural characteristic of the system.

The X-ray structures consist of a mononuclear complex of $\mathrm{Dy^{III}}$ which lies in a $\mathrm{N_2O_6}$ coordination environment with a square antiprism symmetry. Three $\mathrm{tta^-}$ anions and the ligand L provide the six oxygen and two nitrogen atoms, respectively (Fig. 1). As expected, the Dy–N bond length is longer than the Dy–O ones due to the oxophilic character of the lanthanide. The central C=C bond length and the non-planar TTF core confirm the neutrality of L. The crystal packing highlighted the formation of "head-to-tail" dimers of complexes (Fig. S5†).

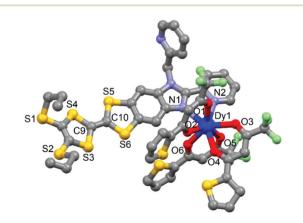


Fig. 1 Molecular structure of 163 Dy. Hydrogen atoms and CH₂Cl₂ molecule of crystallization are omitted for clarity.

Static magnetic properties

The thermal dependence of the magnetic susceptibility and the first magnetization at 2 K are measured for the $^{162}\mathrm{Dy}$ and $^{163}\mathrm{Dy}$ samples and for $^{162}\mathrm{Dy}$ and $^{163}\mathrm{Dy}$ (Fig. S6 and S7†). The $\chi_{\mathrm{M}}T$ vs. T and M vs. H curves are identical for the two isotopes because the static magnetic properties are independent of the nuclear magnetic moment. The experimental values are in agreement with the expected values for a single $\mathrm{Dy^{III}}$ ion. 25

Dynamic magnetic properties

The dynamic magnetic properties of the two isotopologues 162 Dy (I=0) and 163 Dy (I=5/2) are determined by measuring the frequency dependences of the magnetic susceptibility (Fig. 2, S9 and S10†). Both isotopologues display a frequency dependence for the out-of-phase component, $\chi_{\rm M}"$, of the ac susceptibility in a zero external dc field, a sign of slow magnetic relaxation. As previously observed for 161 Dy and 164 Dy, the maximum on the $\chi_{\rm M}"$ vs. ν curve (where ν is the frequency of the oscillating field) occurs at a lower frequency for the nuclear spin-free isotope (16 Hz) than for the isotope with I=5/2 (160 Hz).

The relaxation times (τ) for both isotopologues have been extracted with an extended Debye model (see the ESI†). Their temperature dependences follow a modified Arrhenius law $(\tau^{-1} = \tau_0^{-1} \exp(\Delta/T) + \tau_{\text{TI}}^{-1})$ (where τ_0 and τ_{TI} are, respectively, the intrinsic and temperature-independent relaxation times

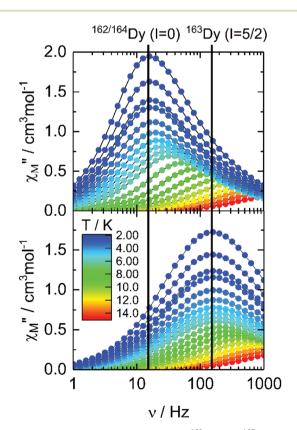


Fig. 2 Frequency dependences of $\chi_{\rm M}$ " of $^{162}{\rm Dy}$ and $^{163}{\rm Dy}$ in a zero magnetic field in the temperature range of 2–15 K.

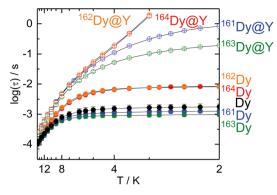


Fig. 3 Temperature dependences of the relaxation times τ of the four main stable isotopologues ^{161–164}Dy (full circles) and diluted samples ^{161–164}Dy@Y in a zero magnetic field in the temperature range of 2–15 K. Full lines are guides to the eye only.

and Δ is the energy barrier) (Fig. 3, Tables 1, S2 and S3†). Both Δ and τ_0 remain almost identical for ¹⁶²Dy and ¹⁶³Dy, suggesting that the thermally activated regime is not affected by the nucleus and thus by the hyperfine interactions (Table 1, Fig. S11 and S12†), while the thermally independent regime (QTM) is strongly affected by the isotopic enrichment with the cancellation of the nuclear spin of dysprosium (Fig. 3). Even if the QTM is reduced, it is still operative at a low temperature. To cancel this, an optimal magnetic field of 1000 Oe is applied (Fig. S13-S16†), leading to a drastic slowing down of the magnetic relaxation. Nevertheless, such an intense magnetic field also cancels the isotopic effect. In fact the two Arrhenius plots for both ¹⁶²Dy and ¹⁶³Dy are superimposed (Fig. S17†). Fig. 3 provides an easy comparison between all isotopologues. As expected, the two nuclear spin-free isotopes 162 Dy and 164 Dy behave the same whereas the two isotopes ¹⁶¹Dy and ¹⁶³Dy (I = 5/2) display faster magnetic relaxation. In the QTM regime, a significant difference exists between ¹⁶¹Dy and ¹⁶³Dy. Fig. 4 shows the $\chi_{\rm M}$ " vs. ν curves for the two ¹⁶¹Dy and ¹⁶³Dy with respective maxima centred at 106 Hz and 160 Hz, respectively. ¹⁶³Dy relaxes slightly faster than ¹⁶¹Dy. This can be compared with the very different (of opposite sign) hyperfine coupling constants of 161 Dy and 163 Dy. 26 Furthermore, the hyperfine interactions are expected to be stronger for the isotope 163Dy than for 161 Dy, 27 leading to a more efficient QTM for the former as experimentally observed (Fig. 3 and Table 1). The effect of the nuclear spin value and, a fortiori, the effect of the hyperfine constant value are very sensitive to any magnetic field (external or internal) because the hyperfine interaction, of the order of a few thousandths of wavenumber, is easily destroyed by the dipolar magnetic field generated by intermolecular interactions. In order to evaluate the exact effect of the hyperfine coupling constants for the two ¹⁶¹Dy and ¹⁶³Dy isotopes, the internal magnetic field has been released in the doped samples $[^{161}Dy_{0.03}Y_{0.97}(tta)_3(L)]\cdot C_6H_{14}$ $(^{161}Dy@Y$, the work¹³) data come previous from our $[^{163}Dy_{0.05}Y_{0.95}(tta)_3(L)]\cdot C_6H_{14}$ ^{163}Dy @Y. The frequency dependences, in the zero external field, of the two doped isotopoloResearch Article

Table 1 Dynamic parameters of the four, pure and diluted, isotopologues of Dy

	Dy	I = 5/2	I = 0	I = 5/2	I = 0
Δ/K $ au_0/\mathrm{s}$ $ au_{\mathrm{TI}}/\mathrm{s}^a$	43 $6.7(1) \times 10^{-6}$ $1.8(1) \times 10^{-3}$	$45 \\ 6.0(8) \times 10^{-6} \\ 1.3(1) \times 10^{-3}$	$44 \\ 7.8(1) \times 10^{-6} \\ 8.1(1) \times 10^{-3}$	46 $5.6(7) \times 10^{-6}$ $1.0(1) \times 10^{-3}$	$43 7.7(9) \times 10^{-6} 8.3(1) \times 10^{-3}$
	Dy@Y	¹⁶¹ Dy@Y	¹⁶² Dy@Y	¹⁶³ Dy@Y	¹⁶⁴ Dy@Y
Δ/K $ au_0/\mathrm{s}$ $ au_{\mathrm{TI}}/\mathrm{s}^a$	Not measured	58 4.0(10) × 10 ⁻⁶ 0.83(6)	$ 59 3.5(9) \times 10^{-6} \approx 100^{b} $	$573.9(6) \times 10^{-6}0.19(0.4)$	$ \begin{array}{c} 60 \\ 3.8(8) \times 10^{-6} \\ \approx 100^{b} \end{array} $

^a Values extracted from the experimental data at 2 K. ^b Values obtained by extrapolation of the experimental data.

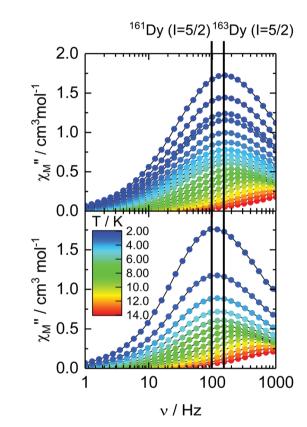


Fig. 4 Frequency dependences of $\chi_{\rm M}$ " of $^{161}{\rm Dy}$ and $^{163}{\rm Dy}$ in a zero magnetic field in the temperature range of 2–14 K.

gues are depicted in Fig. 5 and S18–S21.† The cancelling of the internal magnetic field has two consequences on the magnetic properties of the doped samples: (1) the maxima of the out-of-phase component of the magnetic susceptibility are shifted to lower frequencies for both ¹⁶¹Dy@Y (0.2 Hz) and ¹⁶³Dy@Y (0.8 Hz) with respect to the condensed phase (Fig. 3 and Tables 1, S7†). This is even true at the lowest temperatures where the relaxation time becomes almost 1000 times slower than that in the condensed phase. (2) The difference between the two doped isotopologues (¹⁶¹Dy@Y and ¹⁶³Dy@Y) is much more pronounced, with, at 2 K, the relaxation time of ¹⁶¹Dy@Y being four times slower than that of ¹⁶³Dy@Y. Considering

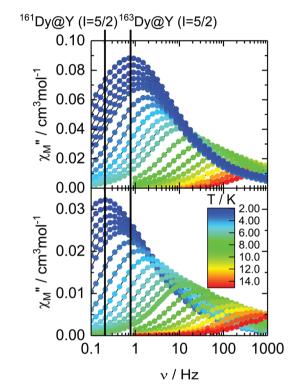


Fig. 5 Frequency dependences of $\chi_{\rm M}$ " of ¹⁶¹Dy@Y and ¹⁶³Dy@Y in a zero magnetic field in the temperature range of 2–15 K.

that the magnetic entities in ¹⁶¹Dy@Y and ¹⁶³Dy@Y are perfectly isolated systems, the difference observed is entirely due to hyperfine coupling. As expected, the relaxation tends to become thermally independent at low temperature. Therefore, a hyperfine coupling effect on the magnetic relaxation of a molecule can only be probed in diluted systems. Of course, the release of the hyperfine coupling in nuclear spin-free systems such as ¹⁶²Dy@Y and ¹⁶⁴Dy@Y produces slower relaxation (Fig. 3 and Table S6†) which never ends in a thermally independent regime at low temperature. At this stage, relaxation is governed by Orbach or Raman processes or a combination of these two processes.

The hysteresis loops for ¹⁶¹Dy@Y and ¹⁶³Dy@Y were compared and classical butterfly shaped hysteresis loops for the

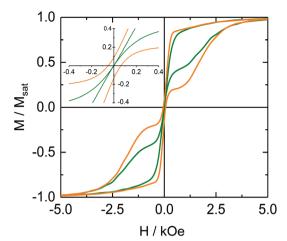


Fig. 6 Normalized magnetic hysteresis loops at 0.48 K and at a sweep rate of 16 Oe s $^{-1}$ for 162 Dy@Y (orange lines) and 163 Dy@Y (green lines). The inset is a zoomed view of the origin to visualize the remnant magnetization.

Dy^{III} mononuclear SMM are observed but no significant differences are observed between the two isotopes (Fig. S22†), whereas the hysteresis loops of the doped compounds with the fastest (¹⁶³Dy@Y) and the slowest (¹⁶²Dy@Y or ¹⁶⁴Dy@Y) relaxation of magnetization show significant differences with a clear magnetic bistability at a zero-magnetic field for the nuclear spin-free ¹⁶²Dy@Y (Fig. 6). In contrast, ¹⁶³Dy@Y is closed at zero field in agreement with the relaxation measurements.

Conclusions

The two 162 and 163 isotopes of the Dy^{III} mononuclear Single-Molecule Magnet [Dy(tta)3(L)]·C6H14 were investigated and then compared with the two 161 and 164 isotopes previously studied by some of us. The nuclear spin-free isotope (162Dy) displayed a slower magnetic relaxation in the thermally independent regime compared to the isotope 163 Dy (I = 5/2) due to the cancelling of the hyperfine interactions coming from the metal centre. As expected, the two nuclear spin-free ¹⁶²Dy and ¹⁶⁴Dy behaved as the same while a significant difference of the relaxation time of magnetization was observed between the two ¹⁶¹Dy and ¹⁶³Dy isotopic isomers (I = 5/2). At 2 K, the ¹⁶³Dy isotope relaxed 50% faster than 161Dy because of the greater hyperfine interaction constants for the former compared to the latter. Finally after removing the dipolar magnetic interactions by doping, the effect of the hyperfine interaction constants on magnetic relaxation is directly observed with ¹⁶³Dy@Y which relaxed 300% faster than ¹⁶¹Dy@Y.

Future Dy^{III} isotopic enrichment in which Dy^{III} is in a nuclear spin-free environment is under investigation in order to determine the effect of the hyperfine coupling arising from the first neighbouring atoms and the effect of the oblate *versus* prolate electronic distribution of the lanthanide ion.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

This work was supported by the CNRS, Université de Rennes and the European Research Council through the ERC-CoG 725184 MULTIPROSMM (project no. 725184).

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