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Investigation on Tm³⁺/Yb³⁺ codoped germanate-tellurite glasses for efficient 2µm mid-infrared laser materials

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Abstract:

The Tm³⁺/Yb³⁺ co-doped germanate-tellurite glasses with good thermal properties were prepared. Based on the absorption spectra and the Judd-Ofelt theory, the J-O intensity parameters (Ω_t), radiative transition probability (276.78s⁻¹), fluorescence lifetime (3.89ms) and absorption and emission cross sections ($\sigma_e = 1.35 \times 10^{-20} \text{cm}^2$) were calculated. The ~2µm mid-infrared emission resulting from the ${}^{3}\text{F}_{4} \rightarrow {}^{3}\text{H}_{6}$ transition of Tm³⁺ sensitized by Yb³⁺ has been observed pumped by 980nm LD. Besides, the energy transfer mechanism between Yb³⁺ and Tm³⁺ was thoroughly discussed. Meanwhile, the measured ~2µm emission lifetime of Tm³⁺/Yb³⁺ co-doped glass can reach as high as 2.38ms. The above results showed that Tm³⁺/Yb³⁺

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co-doping glass could be expected to be a promising material to achieve high efficient $\sim 2\mu m$ lasing with a 980nm LD pumping.

1. Introduction

Over the past several years, increasing efforts have been made to develop $\sim 2.0 \mu m$ fiber lasers driven by their wide applications such as remote sensing, laser surgery, environmental monitoring, and eye-safe light detection and ranging (LIDAR) [1-4].

Among the rare earth ions, Tm^{3+} and Ho^{3+} are well-known for the generation of ~2.0µm emission, due to Tm^{3+} : ${}^{3}F_{4}\rightarrow {}^{3}H_{6}$ and Ho^{3+} : ${}^{5}I_{7}\rightarrow {}^{5}I_{8}$ transitions, respectively. In comparison with Ho^{3+} , Tm^{3+} is of special interests for its unsurpassed advantages, such as broad emission about 800nm suitable for tunable lasers [5] and high quantum efficiency be beneficial from the cross relaxation energy transfer (${}^{3}H_{4}+{}^{3}H_{6}\rightarrow {}^{3}F_{4}+{}^{3}F_{4}$) [6]. What is more, the ${}^{3}F_{4}$ level of Tm^{3+} can be populated using sensitizer ions such as Yb³⁺ and Er^{3+} [7, 8]. Yb³⁺ ion can be used as a sensitizer and it can be pumped by a more powerful and relatively inexpensive laser diode (LD) at 980nm. Because fast diffusion among the Yb³⁺ to $Tm^{3+}({}^{2}F_{5/2}+{}^{3}H_{6}\rightarrow {}^{2}F_{7/2}+{}^{3}H_{5})$. Yb³⁺ has a high absorption cross section and can effectively absorb excitation of commercial 980nm LD emission, and then transfer energy to $Tm^{3+}({}^{3}H_{5}$ level) via non-resonant energy transfer process.

In order to research luminescence properties of rare-earth ions, it is very significant to select glass hosts for researchers. Since the first successfully Tm^{3+} doped silica fiber laser was demonstrated in 1988 by Southampton [9]. In 1999, Stuart

D. Jackson et al. proposed a theoretical model that describes the ion-pair dynamics relevant to the Tm-doped silica system, which can deduce equations for the steady-state intracavity photon density and for the steady-state population densities of the isolated and paired ions [10]. In 2009, Peter F. Moulton et al. used a piece of 25-µm-diameter and 0.08 numerical aperture (NA) core thulium-doped silica fiber to derive single mode laser output of the output powers of 300W and the slope efficiency as high as 64.5% for launched pump power [11]. Various host materials including fluoride [12], tellurite [13], germanate [6] and silicate glasses [14] have been successfully applied to achieve ~2.0µm fiber lasers with different efficiency. The obvious defects of silicate glasses are the low rare-earth solubility and high phonon energy. The fluoride glasses have the drawbacks of inferior mechanical and chemical properties. Germanate glasses have been intensively studied as promising candidates in mid-infrared optical applications, and high energy laser system [15-17]. Particularly, one system, Germanate-tellurite (GT) glass which has been practically applied in large infrared (IR) windows, because of the excellent combinations of superior IR transparency, comparatively low phonon energy, high rare-earth solubility, and good mechanical characteristics [18]. Tellurite glass has attracted a great deal of interest not only for its relatively lower maximum phonon energy (\sim 760cm⁻¹) among all the oxide glasses, but also for the improvement in chemical and mechanical stability as well as the high refractive index, excellent infrared transmission [19]. Compared with germanate glass (~900cm⁻¹) [20], tellurite glass (~760cm⁻¹) [19] has much lower phonon energy, which is favorable for 2µm emission and that is helpful to reduce the

multi-phonon relaxation rate for Tm^{3+} : ${}^{3}\text{F}_{4}\rightarrow{}^{3}\text{H}_{6}$ transition. However, tellurite glass has lower glass transition temperature (~350°C) than germanate glass (~600°C). Thus, tellurite glass has poor thermal stability to resist thermal damage at high pumping power. On the contrast, the thermal stability, chemical durability, resistance to thermal damage of germanate glass are superior. Therefore, germanate-tellurite glass combines the advantages from both germanate and tellurite glasses, i.e., good thermal stability, chemical durability, lower phonon energy, high rare earth solubility and high transparency in a wide wavelength range [21,22]. These features render germanate-tellurite glass as an ideal host for mid-infrared laser material. To the best of our knowledge, the fluorescence corresponding to the ${}^{3}\text{F}_{4}\rightarrow{}^{3}\text{H}_{6}$ transitions of the Tm³⁺ has been under-reported sensitized by Yb³⁺ in germanate-tellurite glasses [23-25].

In this work, we report Tm^{3+}/Yb^{3+} codoped germanate-tellurite glasses for generating emission in the ~2.0µm wavelength band. Spectroscopic properties of Tm^{3+} and energy transfer mechanism between Tm^{3+} and Yb^{3+} ions were discussed in detail. Energy transfer microscopic parameters were calculated based on absorption spectra. Furthermore, the thermal properties are investigated and Judd-Ofelt intensity parameters have been determined to evaluate the radiative properties. In this paper, the lifetime and gain spectra have also been discussed. The results could verify that the Tm^{3+}/Yb^{3+} co-doped germanate-tellurite glasses could be potential materials for mid-infrared lasers.

2. Experiment

2.1. Material synthesis

The sample glasses were prepared by traditional melt-quenching method with using high-purity chemicals (99%-99.99+%). The investigated glasses of molar compositions:80(GeO₂+TeO₂)-(19.5-y)(K₂CO₃+Nb₂O₅+La₂O₃)-xYb₂O₃-yTm₂O₃(x=0. 5; y=0.1, 0.25, 0.4, 0.5 and x=0; y=0.1) were placed in Al₂O₃ crucible and heated with a SiC-resistance electric furnace at 1200°C for 25min. The melts were then poured onto a preheated stainless mold, followed by annealing at 530°C for 5h to relinquish the inner stresses. Then it was allowed to cool slowly to room temperature. The cooled samples had been cut and polished to the size of $20 \times 20 \times 1.5$ mm³ carefully, prepared for the optical property measurements.

2.2. Performance measurements

In this paper, the refractive index (1.73) and density ($4.27g/cm^3$) of $0.25Tm^{3+}/0.5Yb^{3+}$ co-doped sample were tested by prism minimum deviation method and Archimedes principle using distilled water as an immersion liquid. The maximum phonon energy of present germanate-tellurite glass is ~785cm⁻¹. The maximum phonon energy (~785cm⁻¹) is less than tellurite glasses (~900cm⁻¹) [16]. The characteristic temperatures of glass transition (T_g), onset crystallization peak (T_x) and top crystallization (T_p) were performed with 10K/min using the NETZSCH DTA 404 PC differential scanning calorimeter. Fluorescence spectra of the samples in the range of 1500-2400nm were measured by a liquid-nitrogen-cooled PbS detector using a

980nm laser diode (LD) as an excitation source. The absorption spectra measurements were performed by PerkinElmer Lambda 900 UV/VIS/NIR double beam spectrophotometer (Waltham, MA) in the range from 300 to 2100nm. The lifetime decay curves were measured and shown by TDS 3012C type Digital Phosphor Oscilloscope (100MHz, 1.25GS/S). For different samples, they were tested under the same experimental conditions. In addition, all the test results were carried out at room temperature.

3. Results and discussions

3.1. Thermal stability analysis

To correctly evaluate the thermal properties of the glasses, as shown in Fig.1, the DSC measurement of sample glasses were displayed. The figures of T_g , T_x , and T_p are 538°C, 726°C, and 750°C, respectively. T_g is a significant factor for laser glasses. The transition temperature (T_g) of the samples (538°C) is higher than those of bismuth (269°C) [26], tellurite (354°C) [27], tellurite(300°C) [13], germanate (526°C) [28] and fluoride glasses (332°C) [29]. The higher T_g is generally considered to possess better thermal properties to resist thermal damage at high pumping intensities. The ΔT (T_x - T_g) can usually be used to evaluate the thermal properties of glasses. A bigger ΔT reveals that the glass possesses an excellent thermal ability against the nucleation and crystallization as well. From Table 1, it is obvious that ΔT is 188°C, which is significantly larger than those of bismuthate (67°C) [26], tellurite (141°C) [27], germanate (129°C) [28] and fluoride glasses (76°C) [29], which reveals that

germanate-tellurite glasses could be potential materials for mid-infrared lasers. The above results indicate that germanate-tellurite glasses have better thermal performance in this paper. Hence, the prepared germanate-tellurite glasses have good anti-crystallization properties and could be selected as potential laser material.



Fig.1. DSC curves of germanate-tellurite glass.

		glass host	s.		
Glass samples	T _g (°C)	$T_x(^{\circ}C)$	$T_p(^{\circ}C)$	ΔT	References
GT	538	726	750	188	This work
Germanate	526	747	769	129	[28]
Tellurite	354	511	527	141	[27]
Bismuthate	269	336	358	67	[26]
Fluoride	332	408	426	76	[29]

Table 1. The glass transition temperature (T_g), onset crystallization temperature (T_x), temperature of crystallization peak(T_p), thermal stability parameters ΔT in various

3.2. Absorption spectra and J-O analysis

As shown in Fig.2, absorption spectra of 0.25Tm^{3+} , 0.5Yb^{3+} single doped and $0.25 \text{Tm}^{3+}/0.5 \text{Yb}^{3+}$ co-doped glass samples of thickness of 1.5mm is depicted in the wavelength range of 300-1750nm. For single Tm^{3+} doped the sample, in the range from 300 to 1750nm, It could be discerned five absorption bands centered at around 468、684、791、1208 and 1713nm, which correspond to the transitions from the ³H₆ ground state to excited levels ¹G₄, ³F_{2,3}, ³H₄, ⁻³H₅ and ³F₄, respectively, according to the absorption spectrum of Tm^{3+} single doped germanate glass. The shapes and the peak positions of each transition in Tm^{3+} single doped germanate-tellurite glasses nearly do not change when compared with the $\text{Tm}^{3+}/\text{Yb}^{3+}$ co-doped glasses [30]. The phenomenon could be explained by the reason that the Tm^{3+} ions are

homogeneously incorporated into the glass network without clustering and any changes in the local ligand field [31]. In this paper, the 980nm LD is used as the pumping source. As we already know, the center of absorption peak of Yb^{3+} ions at 976nm is indeed more intense than all the absorption peaks of Tm^{3+} . The radiation wavelength of high power commercial 980nm LD can be matched with 976nm absorption peak of Yb^{3+} ions, showing that codoping with Yb^{3+} and Tm^{3+} is an efficient approach to enhance ~1.8µm emission from $Tm^{3+}[32,33]$.



Fig.2. Absorption spectra of Tm³⁺,Yb³⁺ single doped and Tm³⁺/Yb³⁺ co-doped germanate-tellurite glasses.

The Judde-Ofelt (J-O) theory [34,35] plays a significant role in analyzing the spectroscopic properties (radiative lifetimes, intensity parameters Ω_t (t=2, 4, 6), spontaneous emission probabilities and so on) for rare earth ions doped glasses.

From Table 2, it compares the Judde-Ofelt parameters of Tm^{3+} ions in germanate, tellurite and bismuthate glasses. The Ω_2 parameter is intensively dependent on local

environment of the rare earth ions. The value of Ω_2 (4.45×10⁻²⁰ cm²) of Tm³⁺ is relatively large, which is higher than those of silicate (3.70×10⁻²⁰ cm²) [36], tellurite (3.30×10⁻²⁰ cm²) [37] glasses. On the one hand, it is straightly related to the symmetry of rare earth ions with ligand formation. On the other hand, the polarization of the localstructure and the covalent nature of the chemical bonds result in the formation of rare earth ions with ligands. A large field strength is mainly attributed to a high polarizability of oxygen at the rare earth site. For the crystal field parameter, the value of Ω_2 might also be affected by the asymmetry of the rare earth sites. The Ω_2 is an important factor to obtain good spectral properties of the host materials. From Table 2, the value of Ω_6 is larger than silicate glasses (0.60×10^{-20} cm²) [36], but smaller than tellurite (1.28×10^{-20} cm²) [37] and germanate (1.92×10^{-20} cm²) [38] glasses. At the same time, comparison with Ω_2 , Ω_6 doesn't depend on the environment but is more dependent on the overlap integrals of the 4f and 5d orbits [39].

Glass	Ω_2	Ω_4 Ω_6		Reference	
	$\times 10^{-20} \text{ cm}^2$)				
GTT	4.45	0.84	0.93	This work	
Germanate	8.13	2.32	1.92	[38]	
Silica	3.70	2.30	0.60	[36]	
Tellurite	3.30	2.38	1.28	[37]	
Bismuthate	6.51	2.11	1.52	[40]	

Table 2. Comparison of J-O intensity parameters of Tm³⁺ ions in various glass hosts.

From the Table 3, full understanding of the radiation characteristics of the prepared glasses, the radiative lifetimes (τ_{rad}), spontaneous transition probabilities (A_{rad}) and fluorescence branching ratios (β) were calculated. The radiative transition probability (A_{rad}) of Tm³⁺:³F₄ \rightarrow ³H₆ transition is calculated to be 276.78s⁻¹. The higher spontaneous radiative transition probability provides larger opportunity to achieve better laser action, so the prepared glasses could be a promising candidate for mid-infrared laser [41]. It is known from other papers that the radiative transition probability (A_{rad}) of Tm³⁺ in sample is higher than those of tellurite and germanate glass [42,43]. We can see that the radiative lifetime of ${}^{3}F_{4}\rightarrow {}^{3}H_{6}$ transition is as long as 3.89ms, which is longer than that of tellurite (1.64ms) and germanate (0.859ms) glass [42,43]. The high efficient ~2µm radiations can be achieved in the prepared glass.

Transition	Final state	S _{ed} (10 ⁻²⁰ cm ²)	$A_{ed} \left(s^{-1} \right)$	$A_{md}(s^{-1})$	β	τ _{rad} (ms)
$^{3}F_{4}$	$^{3}H_{6}$	3.353	276.780	0	1	3.89
³ Н ₅	$^{3}H_{6}$	1.266	224.027	77.898	98.57%	3.26
115	${}^{3}F_{4}$	0.561	4.386		1.43%	
	$^{3}H_{6}$	1.588	1240.053		90.58%	0.73
³ H ₄	${}^{3}F_{4}$	0.862	99.750		7.29%	
	$^{3}H_{5}$	0.474	15.134	14.009	2.13%	
	$^{3}H_{6}$	1.048	1635.344		74.58%	0.46
36	${}^{3}F_{4}$	0.169	57.322	93.308	6.87%	
13	$^{3}H_{5}$	3.089	403.103		18.38%	
	${}^{3}\text{H}_{4}$	0.939	3.741		0.17%	
	$^{3}H_{6}$	0.241	634.031		36.14%	0.57
³ F ₂	${}^{3}F_{4}$	1.356	874.099		49.82%	
	$^{3}H_{5}$	0.792	217.764		12.41%	
	${}^{3}\text{H}_{4}$	1.610	28.417		1.62%	
	${}^{3}F_{3}$	0.080	0.046	0.155	0.01%	
	$^{3}H_{6}$	0.279	1032.915		46.20%	0.45
${}^{1}G_{4}$	${}^{3}F_{4}$	0.089	125.596		5.62%	
	$^{3}H_{5}$	0.874	743.923		33.27%	
	$^{3}\text{H}_{4}$	1.119	275.795		12.34%	
	${}^{3}F_{3}$	0.422	47.050		2.10%	
	$^{3}F_{2}$	0.148	10.439		0.47%	

 Table 3. The spontaneous emission probabilities, branching ratios and radiative

lifetimes	of Tm ³⁺	in the	samples
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3.3. Emission spectra and cross section

In the past studies [44], the energy transfer process between Yb^{3+} and Tm^{3+} is recognized by researchers, and the efficiency of the process is very high. As shown in Fig.3.(a), with the increase of Tm^{3+} concentration, the Tm^{3+} :~1.8µm emission is also enhanced as well. And the fluorescence spectra of Tm^{3+}/Yb^{3+} co-doped germanate-tellurite glass pumped by 980nm LD is displayed. From Fig.3.(b), we can find that there is no fluorescence when Tm^{3+} singly doped in germanate-tellurite glasses. With the assistance of Yb³⁺ ions, the ~1.8µm emission peak of Tm³⁺/Yb³⁺ co-doped germanate-tellurite glasses are excited by 980nm LD. This confirms that the lack of Tm³⁺ absorption band near 976nm and the existence of energy transfer from Yb³⁺ to Tm³⁺. In addition, with the increase of Tm³⁺ ions concentration, the fluorescence intensities of Tm³⁺/Yb³⁺ co-doped glass samples increases initially, and then rapidly decreases. When the concentration of Tm³⁺ is 0.25mol%, the intensity of fluorescence emission reaches the maximum in these four sets of glass samples. So the optimum concentration of Tm³⁺ is 0.25mol%. The reason is that the energy transfer probability between Yb³⁺:²F_{5/2} and Tm³⁺:³H₅ ions is enhanced. Subsequently, this emission intensity decreases gradually with the increase of Tm³⁺ ions concentration, because of the concentration quenching [45]. Therefore, the optimum concentration ratio between Tm³⁺ ions and Yb³⁺ ions is 1:2, the results can show that the Tm³⁺/Yb³⁺ co-doped germanate-tellurite glasses could be potential materials for mid-infrared lasers.



Fig.3.(a).Emission spectra of Tm^{3+}/Yb^{3+} co-doped germanate-tellurite glasses; (b). Emission spectra of Tm^{3+} doped and Tm^{3+}/Yb^{3+} co-doped germanate-tellurite glasses.

Furthermore, emission cross section is an important parameter to estimate the possibility for achieving laser action. According to the above measured absorption spectra, the absorption cross section can be expressed as Beer–Lambert equation

$$\sigma_{abs} = \frac{2.303}{N \times l} OD(\lambda) \tag{1}$$

where *N* is the density of rare-earth ion (ions/cm³), *l* is the thickness of the polished sample glass, $OD(\lambda)$ is the optical density, the value of the optical density is equal to the spectra intensity.

$$\sigma_{em} = \frac{\lambda^4 A_{rad}}{8\pi cn^2} \times \frac{\lambda I(\lambda)}{\int \lambda I(\lambda) d\lambda}$$
(2)

where λ is the wavelength, A_{rad} is the spontaneous transition probability, I(λ) is the emission intensity, c and n is the light speed in vacuum and refractive index, respectively.

As shown in figure Fig.4(a), the values of emission and absorption cross sections of Yb³⁺ are 0.87×10^{-20} cm² and 0.59×10^{-20} cm² in the sample glasses, respectively. From Fig.4(b), the value of emission and absorption cross section of Tm³⁺ are 1.35×10^{-20} cm² and 0.56×10^{-20} cm² at 1.8µm, respectively. We can see that the emission cross section (1.35×10^{-20} cm²) of Tm³⁺ is larger than those of silicate (0.2×10^{-20} cm²) [46], tellurite (0.86×10^{-20} cm²) [42], tellurite (0.96×10^{-20} cm²) [13] and bismuthate glasses (0.71×10^{-20} cm²) [47]. For a laser material, in generally, the high stimulated emission cross section is helpful to increase the gain. A larger gain results in better amplification behavior.



Fig. 4.(a)Absorption and emission cross sections corresponding to Yb^{3+} : ${}^{2}F_{5/2} \rightarrow {}^{2}F_{7/2}$.(b) Absorption and emission cross sections corresponding to Tm^{3+} :

$$^{3}F_{4} \rightarrow ^{3}H_{6}$$

The process of energy transfer between Tm^{3+} and Yb^{3+} ions is presented in Fig.5. As shown in Fig.5, the efficient energy transfer between Tm^{3+} and Yb^{3+} can be obtained through the transfer of phonons, and a main energy transfer process in the Tm^{3+}/Yb^{3+} co-doped germanate-tellurite glass has was shown. The Yb^{3+} is pumped by 980nm LD and excited from the ground state (${}^{2}F_{7/2}$) to the upper level (${}^{2}F_{5/2}$), following a energy transfer process from Yb^{3+} : ${}^{2}F_{5/2}$ to Tm^{3+} : ${}^{3}H_{5}$ (ET: Yb^{3+} : ${}^{2}F_{5/2}+Tm^{3+}$: ${}^{3}H_{6}\rightarrow Yb^{3+}$: ${}^{2}F_{7/2}+Tm^{3+}$: ${}^{3}H_{5}$). And the process is assisted by phonons. The Tm^{3+} ion in the ${}^{3}H_{5}$ level relaxes very quickly to the ${}^{3}F_{4}$ level, then particle is transferred from ${}^{3}F_{4}$ level to ${}^{3}H_{6}$ level. Yb^{3+} ion as a sensitizer can greatly improve upconversion efficiency through energy transfer (ET), which is due to strong absorption of Yb^{3+} ion around 980 nm when Tm^{3+}/Yb^{3+} -codoped glasses are excited by 980 nm laser. The simplified energy level diagram exhibit distinct the upconversion centered at 476nm, 650nm and 689 nm which are assigned to the ${}^{1}G_{4} \rightarrow {}^{3}H_{6}$, ${}^{1}G_{4} \rightarrow {}^{3}F_{4}$ and ${}^{3}F_{2,3} \rightarrow {}^{3}H_{6}$ transitions of Tm³⁺ ions, respectively.

The microscopic parameter based on the evaluation of the overlap integral between the absorption and emission cross sections can be obtained using the following equations:

$$C_{DA} = \frac{6cg_{low}^{D}}{(2\pi)^{4}n^{2}g_{up}^{D}} \sum_{m=0}^{\infty} e^{(2\pi+1)^{*}_{0} \frac{S_{0}^{m}}{m!}(\bar{n}+1)^{*}_{0} \int \sigma_{ems}^{D}(\lambda_{m}^{+}) \sigma_{abs}^{A}(\lambda) \lambda \lambda}$$
(3)

The g^{D}_{low} and g^{D}_{up} are the degeneracies of the respective lower and upper levels of the donor, respectively. $\bar{n} = 1/(e^{\hbar w_0/KT} - 1)$ is the average occupancy of the phonon mode at T. S₀ is Huang-Rhys factor.

The Table 4 has listed the energy transfer microscopic parameters of Tm^{3+}/Yb^{3+} co-doped germanate-tellurite glass. It's clearly shown that effective energy transfer takes place between Tm^{3+} and Yb^{3+} ions. As shown in Table 4, the energy transfer coefficient between Yb^{3+} ; ${}^{2}F_{5/2}$ level and Tm^{3+} ; ${}^{3}H_{5}$ level can be as high as $1.3826 \times 10^{-40} cm^{6}/s$. It is worth mentioning that the above energy transfer process is non-resonant and mainly assisted by one phonon (39.076%), two phonons (27.785%) and three phonons (1.696%). This circumstance mainly results from low phonon energy of germanate-tellurite glass. Besides, with increasing Tm^{3+} concentration, the energy transfer process can difficultly happen from Yb^{3+} ; ${}^{2}F_{5/2}$ to Tm^{3+} ; ${}^{3}H_{5}$. As shown in Fig.6, the decay curves can confirm this process. Hence, it is very important for mid-infrared materials to select suitable Tm^{3+} concentration.



Fig.5. Simplified energy level diagram of the Tm^{3+}/Yb^{3+} co-doped system.

Energy transfer	N(numb	Coefficient(cm ⁶ /s)			
$Yb^{3+}(^{2}F_{5/2}) \rightarrow Tm^{3+}(^{3}H_{5})$	0	1	2	3	1.3826×10 ⁻⁴⁰
	30.708	39.076	27.785	1.696	

Table 4.Energy transfer parameters between Tm^{3+} and Yb^{3+} in glass sample

3.4. Fluorescence lifetimes and gain coefficient

As shown in Fig.6, the 1.8µm highest lifetime (2.38ms) of Tm^{3+}/Yb^{3+} co-doped germanate-tellurite glasses sample is obtained, and is significantly larger than those of silicate (1.42ms) [48] and germanate (0.36ms) glass [49]. The lifetime of Tm^{3+}/Yb^{3+} in co-doped is longer than that in Tm^{3+} single doped germanate-tellurite glass. It indicates that there is an effective energy transfer between Tm^{3+} and Yb^{3+} . We can see that the lifetime of ${}^{3}F_{4}$ level decrease gradually with increasing Tm^{3+} concentration. The reduced lifetime isn't beneficial for upconversion process. The decreased

upconversion process contributes to population aggregation of $Tm^{3+}:{}^{3}F_{4}$ and might enhance $Tm^{3+}:{}^{3}H_{5}\rightarrow{}^{3}F_{4}$. Here the stronger upconversion process is due in part to the excited state absorption (ESA), while the excited state absorption process reduces the ${}^{3}F_{4}$ level number of particles, which is detrimental to the 1.8µm lifetime. Therefore, the decrease in the conversion process is beneficial to the 1.8µm lifetime of Tm^{3+} . In addition, the relatively long lifetime generally enhances the luminous efficiency of glasses and it generally reduces the laser oscillation threshold. Hence, the emission lifetimes (2.38ms) indicate that Tm^{3+}/Yb^{3+} co-doped germanate-tellurite glass is a promising materials for ~2.0µm fiber laser.

Besides, in order to further evaluate the performance of the samples, the fluorescence quantum efficiency has been estimated from the lifetime values by the following equation.

$$\eta = \frac{\tau}{\tau_{rad}} \times 100\% \tag{4}$$

where τ is the measured fluorescence lifetime of the sample (0.5Tm³⁺/0.1Yb³⁺), and τ_{rad} is the theoretical lifetime the sample (0.5Tm³⁺/0.1Yb³⁺). The measured ³F₄ lifetime (2.38ms) for Tm³⁺ is shorter than the calculated lifetime (3.89ms), which is due to non-radiative quenching. It can be found that the fluorescence quantum efficiency is high as 61.18%, which is higher than that of silicate glass (13.0%) [50]. Therefore, Tm³⁺/Yb³⁺ codoped germanate-tellurite glass is a more promising material for improving the Tm³⁺ 2.0µm fiber laser performance.



Fig.6. Decay curve of Tm^{3+}/Yb^{3+} co-doped germanate-tellurite glasses.

Another important parameter to evaluate the gain performances of prepared sample quantitatively is gain coefficient. In order to calculate the gain ability of Tm^{3+}/Yb^{3+} co-doped germanate-tellurite glasses, the gain coefficient $G(\lambda, P)$ of Tm^{3+}/Yb^{3+} co-doped germanate-tellurite glasses is calculated. The room temperature gain coefficient ($G(\lambda, P)$) can be simply denoted as

$$G(\lambda, P) = N[P \quad g_n(\lambda) - (-P \quad \partial_{abs} \lambda)]$$
⁽⁵⁾

where N is the doping concentration of the rare-earth ion, P is the population inversion.



Fig. 7. Gain coefficient of Tm^{3+} : ${}^{3}F_{4} \rightarrow {}^{3}H_{6}$ transitions in germanate-tellurite glasses.

The Fig.7 shows the calculated gain coefficients of Tm^{3+} and Yb^{3+} transitions as the function of wavelength. The gain coefficients results are calculated by setting the P ranging from 0 to 1 in interval of 0.1. We can see that the gain coefficient is increasing as the value of P continues to increase. As shown in Fig.7, the maximum gain coefficients of Tm^{3+} reach 3.02cm^{-1} , furthermore the concentrations of Tm^{3+} and Yb^{3+} are 0.1mol% and 0.5mol%, respectively. Besides when the P is more than 0.4, the value $G(\lambda, P)$ of the Tm^{3+} : ${}^{3}\text{F}_{4} \rightarrow {}^{3}\text{H}_{6}$ transitions becomes positive number. In addition when the P is less than 0.4, it becomes negative number. The above results show that $\text{Tm}^{3+}/\text{Yb}^{3+}$ co-doped GeO₂-TeO₂-K₂CO₃-Nb₂O₅-La₂O₃ glass has high gain and low pumping threshold for the laser, and is an outstanding gain host material for ~2µm mid-infrared laser.

4. Conclusions

In a word, efficient ~1.8µm emissions have been achieved in Tm³⁺/Yb³⁺ co-doped germanate-tellurite glasses. The energy transfer process between Yb³⁺ and Tm³⁺ has been investigated in detail. The thermal stability, absorption and emission cross sections, lifetimes have been analyzed. Results indicated that Tm³⁺ in the prepared glasses had not only superior thermo-mechanical stability (T_g =538°C, Δ T=188°C), but also large emission cross section (1.35×10⁻²⁰cm²). The measured lifetime of Tm³⁺/Yb³⁺ co-doped germanate-tellurite glasses is as high as 2.38ms. It also possesses superior gain performance for Tm³⁺:³F₄→³H₆ and Yb³⁺:²F_{5/2}→²F_{7/2} transitions. All of the results indicate that the prepared germanate-tellurite glass is a promising candidate for ~2.0µm mid-infrared laser materials applications.

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