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Article

Improvement of Power Factor of CoSb₃ Thermoelectric Thin Films via Microstructure Optimization

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Abstract: Skutterudite CoSb₃ has emerged as one of the most studied candidate materials for thermoelectric applications. In this work, CoSb₃ thin films were prepared by radio frequency sputtering, and their microstructure was investigated with emphasis on the effect of target composition and deposition temperature. The goal was to enhance the thermoelectric properties of CoSb₃ thin films via microstructure optimization. Results showed that the Sb content of films gradually decreased with increasing deposition temperature. Although the thin films prepared by the target with a Co and Sb element ratio of 1:3.5 approached the ideal stoichiometric ratio, they showed poor thermoelectric properties due to the formation of an additional Sb phase. By contrast, the thin films obtained with insufficient Sb showed a single CoSb₃ phase and good thermoelectric properties. The mechanism behind this difference was studied accordingly. The power factor of the thin films was enhanced due to their dense structure and good crystallization.

Keywords: films; structure; thermoelectric properties

1. Introduction

Thermoelectric materials, which can directly convert thermal energy to electric energy, have attracted a significant amount of attention due to their promising applications in solid state cooling and power generation from waste heat [1–4]. The efficiency of thermoelectric generators is determined by the thermoelectric performance of such materials; it is determined by the dimensionless figure of merit, *ZT*, which is defined as $ZT = S^2 \sigma T / \kappa$, where *S* corresponds to the Seebeck coefficient, σ refers to the electrical conductivity, *T* represents the absolute temperature, and κ is the thermal conductivity [5]. High-performance thermoelectric materials require high Seebeck coefficient and electrical conductivity as well as low thermal conductivity [6]. Thin films have been demonstrated as low-dimensional systems that can enhance the thermoelectric performance of materials due to their dense interfaces and grain boundaries, which can effectively scatter phonons [7–10]. For example, a high room temperature figure of merit of *ZT* = 2.4 has been reported for p-type Bi₂Te₃/Sb₂Te₃ superlattices [11], and such a result is attributed to quantum confinement effects.

 $CoSb_3$ is one of the numerous promising thermoelectric materials in the intermediate temperature range [12–15]. This material possesses a special skutterudite structure with two lattice voids, which can be filled with rare elements that can scatter a wide range of phonons and enhance thermoelectric



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properties. In recent years, several studies on the synthesis of $CoSb_3$ -based thin films have been investigated [16–18]; for instance, Kumar et al. [17] obtained a power factor (PF) of 0.97 mW·m⁻¹·K⁻² for In- and Yb-co-doped CoSb₃ films deposited by pulsed laser deposition. However, the thermoelectric properties of CoSb₃ thin films remain lower than those of bulk materials due to the difficulty in controlling element composition from raw materials to thin films during preparation.

Radio frequency (RF) magnetron sputtering is an attractive technique for preparing thin films for industrial development [19]. We observed that CoSb₃ thin films can be obtained by RF magnetron sputtering after annealing [20–22], but they contain an excess amount of Sb and display poor thermoelectric properties; unlike in bulk materials, excess Sb exerts a negative influence on thin films [23,24]. This difference may be attributed to the amorphous state of excess Sb in thin films [20,23,24]. Conversely, thin films show an unnecessary additional CoSb₂ phase that positively affects their thermoelectric properties. Therefore, in the current work, CoSb₃ thin films were deposited by RF magnetron sputtering, and their composition was regulated along with substrate temperature during thin film growth. The microstructure of thin films was investigated and compared with that of thin films under annealing [21,22]. The influence of microstructure on thermoelectric properties was also interpreted to improve the thermoelectric performance of thin films via microstructure optimization.

2. Materials and Methods

CoSb₃ alloy targets with Co:Sb atomic ratios of 1:3 and 1:3.5 were set in a JGP-450B double chamber magnetron sputtering ultrahigh vacuum system (Shenyang Science Instrument of Chinese Academy of Sciences, Shenyang, China). BK7 glass measuring 20 mm \times 20 mm \times 2 mm was used as a substrate and erased cleanly by air-laid paper after ultrasonic cleaning for 20 min in acetone, alcohol, and deionized water. A 5-min pre-sputtering process was carried out to remove contaminants on the target surface. The chamber was pumped down to 6.0×10^{-4} Pa, and working pressure measured 0.4 Pa with 40.0 sccm Ar. Before deposition, the substrate was heated to a setting temperature and kept stable. The temperature was set from 225 to 325 °C at 25 °C intervals. Then, thin films were grown at the substrate by RF magnetron sputtering with 50 W power. The thin films prepared with 1:3 target were labeled as T-series (T1–T5), and samples prepared with 1:3.5 target were labeled as S-series (S1–S5).

Structural phase formation in films was investigated by X-ray diffraction (XRD; D/max 2500, Rigaku Corporation, Tokyo, Japan) with the prescriptive θ –2 θ mode and a determined incident angle of 10–80° using Cu K α radiation (λ = 0.15406 nm). Surface morphology was analyzed by scanning electron microscopy (SEM) (Zeiss supra 55, Carl Zeiss, Oberkochen, Germany), whereas component analysis was performed via energy dispersive spectroscopy (EDS, Quantax, Bruker, Karlsruhe, Germany) with the error of <10%. Electrical conductivity σ and Seebeck coefficient *S* were measured by the simultaneous determination of the Seebeck coefficient and electrical conductivity system (Netzsch SBA458, Netzsch, Shanghai, China) in the interval of 323–573 K.

3. Results

Sb content was determined by EDS, along with the as-deposited thin films. Figure 1 presents the results. The figure shows that the samples (prepared with Co:Sb = 1:3.5 target) approached the 75% stoichiometric ratio, and that the Sb content decreased slightly with increasing deposition temperature due to the evaporation of Sb. Nevertheless, most of the thin films contained abundant Sb. The as-deposited thin films (prepared with Co:Sb = 1:3 target) showed a minimal lack of Sb, but Sb content decreased remarkably when the deposition temperature increased. The Sb content increased at high deposition temperatures in both samples prepared with different targets. This characteristic is presumably related to the enhancement of the Co evaporation effect at high temperatures.



Figure 1. Actual Sb contents for CoSb₃ thin films prepared by Co:Sb atomic ratios of 1:3 and 1:3.5 alloy target.

Figure 2a,b show the XRD patterns of thin films prepared with different targets, including the characteristic pattern of CoSb₃ (PDF#780976). As shown in Figure 2a, the strongest diffraction peak of sample S1 was located at \sim 31°, which is related to the (310) plane of the CoSb₃ phase. Most peaks also correspond to the characteristic diffraction peak, indicating that the sample featured a primary CoSb₃ phase. A few peaks of impurity can be observed in the pattern, and these peaks are related to the hexagonal Sb structure as the secondary phase. With increasing the deposition temperature, a similar result can be obtained, with CoSb₃ as the primary phase and Sb as the secondary phase, as indicated in XRD patterns of S2–S5. However, the intensity of the impurity peaks increased with increasing temperature, and the intensity of the peak located at 29° became the strongest when deposition temperature reached 325 °C. This result indicates that the Sb phase transformed into the primary phase of sample S5. Figure 2b shows that all thin films prepared with Co:Sb = 1:3 target yielded three major diffraction peaks, which were indexed as reflection planes from (013), (123), and (211) of the CoSb₃ phase. No impurity peaks could be observed from the patterns of these thin films, thereby indicating that all samples comprised a single CoSb₃ phase and were well-crystallized. Crystallite size D was calculated from XRD data using the Debye–Sherrer Equation: $D = \kappa \cdot \lambda / \beta_{2\theta} \cdot \cos\theta$, where λ represents the wavelength, $\beta_{2\theta}$ refers to the full width at half maximum, and θ is the angle [25]. Figure 2c shows the calculation results of the peak related to the (013) plane. Results showed that the crystallite size of all thin films increased with increasing deposition temperature. Such an increase suggests that the thermal effects can efficiently promote thin film growth. The grain size of samples prepared with the 1:3.5 target was larger than that of thin films prepared with the 1:3 target. Lattice parameter and crystal density were calculated by using the full-profile Rietveld refinement method and are shown in Figure 2d,e, respectively. Samples with excess Sb featured a large lattice parameter, possibly due to excess Sb filling the CoSb₃ cubic void during the growth process. This condition caused the expansion of the unit cell and large grain size. In general, thin films with a large grain size showed good electrical conductivity due to lower interface scattering. However, the lattice density of T1–T5 was much higher than that of S1–S5, and it followed an increasing trend when the deposition temperature increased. This result indicates that thin films possess a more compact structure and fewer defects than that of S-series samples. These properties benefit electron transport and should lead to improved thermoelectric properties.



Figure 2. Powder XRD of thin films and calculated results as the function of deposition temperature: (a) XRD patterns of the S-series samples; (b) XRD patterns of the T-series samples; (c) crystallite size; (d) X-ray-refined lattice parameter; and (e) crystal density.

Figure 3a shows surface morphology of thin films S1–S5. The grain size increased with rising deposition temperature, and this condition was consistent with results of the XRD analysis. However, grains were not normalized and thus caused the formation a rough and loose structure. As shown in the illustration of S5 in Figure 3a, the grains size of the thin films increased with increasing temperature. However, the grains revealed nonnormalized size and shape, especially in sample S5, which presented numerous large hexagonal-like clusters; such clusters indicate the clustering of excess Sb atoms. The surface morphology of CoSb₃ thin films T1–T5 was also measured, and results are shown in Figure 3b. The results reveal that T1–T4 samples exhibited a smooth surface across all regions. The film surface became increasingly dense, and grains were normalized with increasing temperature. These findings signify that dense thin films with minimal crystal imperfection can be obtained. Although T5 features a dense structure, a number of impure large grains and a porous surface can be observed (marked as red region). These characteristics indicate that thin films exhibited a number of defects that would adversely affect electron transport when the deposition temperature was extremely high.



Figure 3. (a) Surface morphology images of S-series prepared by Co:Sb atomic ratios of 1:3.5 alloy target; (b) Surface morphology images of T-series prepared by Co:Sb atomic ratios of 1:3 alloy target.

Based on the thin film growth model [26,27], the mechanism considered in the present study is illustrated in Figure 4a,b. The substrate used in this work is an amorphous BK7 glass with a flat surface. In general, thin films followed the growth pattern shown in the Frank-van der Merwe model (layer growth). As presented in Figure 4a, grains crystallized preferentially based on substrate

morphology at an early stage. Then, layer fabrication followed that of the preformed layer during the growth process. Thus, the thin films with a single CoSb₃ phase exhibited less phase defects, and the cubic structure led to a dense and smooth surface, as shown in Figure 3b. Similarly, the thin films formed a CoSb₃ structure, along with some hexagonal Sb grains, as shown in the same process shown in Figure 4b. This growth mode caused the formation of progressive interface voids and a loose structure (Figure 3a). A particularly different growth phenomenon was obtained by comparison with the results of annealed samples in our early works [21,22]. The samples deficient in Sb showed a secondary CoSb₂ phase, whereas those with excess Sb phase formed only when the temperature was higher than 600 K. Thus, the mechanism behind such differences was also investigated. As presented in Figure 4c, the thin films exhibited a stable composition distribution before annealing, and the elements failed to diffuse extensively. When the thin films presented Sb deficiency, regions such as (1) and (3), which are distributed with sufficient Sb, grew $CoSb_3$ crystals, whereas some regions such as (2) lacked Sb and crystallized with a CoSb₂ structure. Grain crystallization occurred during sputtering when the substrate was heated, and Sb was continuously supplied to guarantee the fabrication of a single structure. Similarly, crystals grew with a CoSb₃ structure when the thin films contained superfluous Sb (Figure 4d). However, Sb crystals did not easily form at low temperatures by limited diffusion because of the existence of nucleating CoSb₃ acting as the barrier layer. Conversely, hexagonal Sb grains could easily be clustered and crystallized when all grains grew in the same process at a low temperature.



Figure 4. Schematic illustration of thin film growth: (**a**) thin film with a single $CoSb_3$ phase; (**b**) thin film with additional Sb structure; (**c**) thin film deficient in Sb; and (**d**) thin film with excess Sb.

Figure 5a,b show electrical conductivity, σ , as the function of temperature. In Figure 5a, the electrical conductivities of samples S1–S5 shared similar values around 2.0×10^4 S·m⁻¹ at 300 K, and increased with increasing temperature. Comparatively, S5 yielded a slightly higher electrical conductivity at 300 K due to its large grain size; its maximum value measured 3.7×10^4 S·m⁻¹ at

473 K. However, the electrical conductivities of samples decreased dramatically at 450 K and dropped to extremely low values when the temperature reached 550 K. This result can be attributed to the loose structure, which can easily introduce oxidation defects and result in the decline of σ . As shown in Figure 5b, the electrical conductivity, σ , increased with increasing deposition temperature. This result was consistent with the enhancement of grain size. T5 presented the highest electrical conductivity among the samples, reaching a maximum value of 1.3×10^4 S·m⁻¹ at 473 K, and then decreased when the temperature continually increased. Although σ values of the T-series samples were lower than those of the S-series samples, they were steadier in terms of their thermal conductivities because of their dense structure, which prevents defects at high temperatures.



Figure 5. Electrical conductivity, Seebeck coefficient, and PF as the function of temperature: (**a**) electrical conductivity of S-series; (**b**) electrical conductivity of T-series; (**c**) Seebeck coefficient of S-series; (**d**) Seebeck coefficient of T-series; (**e**) PF of S-series; and (**f**) PF of T-series.

Figure 5c,d illustrate the influence of substrate temperature on the Seebeck coefficients, *S*, of the thin films. As shown in Figure 5c, the Seebeck coefficients of S1, S2, and S3 were negative at room temperature, whereas those of S4 and S5 were positive. This result is attributed to impure Sb crystallization, which constantly exhibited a p-type conduction according to the combined results of microstructure measurement. *S* of S1 to S3 increased with increasing temperature, and S1 yielded a

maximum value of $-12.87 \ \mu V \cdot K^{-1}$ at 550 K. However, the value was extremely lower than those of CoSb₃ bulk materials. Interestingly, *S* values of S4 and S5 decreased with increasing temperature and transformed into n-types when the measurement temperature reached a certain high point, possibly as a result of the anabatic oxidation effect. As presented in Figure 5d, the thin films presented negative Seebeck coefficients, and the values were close to each other at low temperatures. Absolute values of all samples increased when temperature increased. In particular, the Seebeck coefficients of T1, T2, and T3 showed a sharp increase in their absolute values, which reached $-170 \ \mu V \cdot K^{-1}$ at 573 K, possibly due to their structure being denser and more crystallized than that of samples fabricated at high temperatures.

PF, one of the important thermoelectric parameters, determines thermoelectric energy conversion and is defined as S² σ . The temperature dependence of PF is shown in Figure 5e,f. As shown in Figure 5e, PF of S1–S5 showed an increasing trend, but the values were extremely low and decreased with deposition temperature at high temperatures. Different from that of the S-series samples, PF of the T-series samples was much higher and sharply increased when temperature >450 K due to the high Seebeck coefficient in Figure 5f. T2 and T3 respectively obtained maximum PF values of 9.8 × 10⁻⁴ and 9.1 × 10⁻⁴ W·m⁻¹·K⁻², which were almost 20 times higher than those of the S-series samples.

4. Conclusions

CoSb₃ thin films were prepared with different ratios of alloy targets, and the influence of deposition temperature on the microstructure and thermoelectric properties of the thin films was investigated. Results showed that the thin films featured a well-crystallized and single skutterudite CoSb₃ structure when 1:3 alloy target was used. Growth mechanism was studied to determine the reason behind the different microstructures of the samples with sufficient and deficient Sb, and those subjected to annealing and deposition at the heating substrate. As expected, thin films with improved thermoelectric performance could be obtained with the enhancement of PF of $9.75 \times 10^{-2} \text{ mW} \cdot \text{m}^{-1} \cdot \text{K}^{-2}$ via microstructure optimization.

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