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# Copper-Catalyzed Oxidative Dehydrogenative C(sp³)–H Bond Amination of (Cyclo)Alkanes using *NH*-Free Heterocycles as Amine Sources

Chang-Sheng Wang, [a] Xiao-Feng Wu, [b] Pierre H. Dixneuf [a] and Jean-François Soulé\*[a]

**Abstract:** A copper-catalyzed oxidative C(sp³)–H/N–H coupling of *NH*-heterocycles with affordable (cyclo)alkanes was developed. This protocol involved C(sp³)–N bond formation *via* a radical pathway generated by a homolytic cleavage of di-*tert*-butyl peroxide and trapping of the radical(s) by copper catalysts. The reaction tolerated a series of functional groups, such as bromo, fluoro, ester, ketone, nitrile, methyl and methoxy. *NH*-free indoles, pyroles, pyrazoles, indazoles and benzotriazoles have been successfully *N*-alkylated.

#### Introduction

*N*-alkyl heterocycle derivatives constitute an important class of organic compounds, which have been widely used in biology, synthetic and polymer chemistry.<sup>[1]</sup> As example, The *N*-alkyl indole **I** displays biological activity for the treatment of neurodegenerative diseases.<sup>[2]</sup> The *N*-cycloalkyl pyrrole **II** is an analogue of BN1212 exhibiting antitubercular properties.<sup>[3]</sup> The *N*-cyclopentylindazole **III** was used as a selective inhibitor of EZH2.<sup>[4]</sup> The *N*-(*tert*-butoxymethyl)indole **IV** has shown inhibition properties on vascular endothelial growth factors, which are important signaling proteins involved in both vasculogenesis and angiogenesis (Figure 1).<sup>[5]</sup>

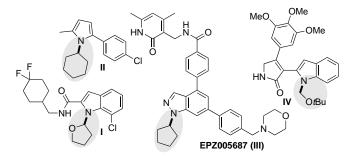


Figure 1. Examples of *N*-Alkyl heterocycles Displaying Biological Activities.

[a] Mr. C.-S. Wang, Prof. Dr. P.H. Dixneuf and Dr. J.-F. Soulé Institut des Sciences Chimiques de Rennes, UMR 6226 CNRS-Université de Rennes "Organométalliques: Matériaux et Catalyse", Campus de Beaulieu, 35042 Rennes, France E-mail: jean-francois.soule@univ-rennes1.fr

[b] Prof. Dr. X.-F. Wu Leibniz-Institut für Katalyse e.V. an der Universität Rostock, Albert-Einstein-Straße 29a, 18059 Rostock, Germany. Supporting information for this article is given via a link at the end of the document Despite this wide applicability, ecofriendly routes for the synthesis of N-alkyl heterocycles are limited. One of the most common synthetic pathway involves SN<sub>2</sub> reaction using a stoichiometric amount of a strong base and an appropriate alkylating agent (e.g., alkyl halide, alkyl triflate, etc.) generation a lot of waste (Figure 2a).[6] Since the discovery of transition-metal catalyzed C-H bond functionalizations,[7] catalytic C-H amination provides an appealing method for the straightforward construction of C-N bonds.[8] General protocols for the transition-metal catalyzed C-H amination required electrophilic amines, via an internal oxidant strategy (e.g., azides, [9] *N*-haloamines [10] or oximes [11]). However, such protocols could not be employed for the N-functionalization of NH-heterocycles. Amines can be directly used, although an external oxidant is required.[12] However, only rare examples of C–H amination with *NH*-heterocycles have been reported. [13] This lack of exploitation is probably due to the presence of competitive C(sp<sup>2</sup>)-H bonds on NH-heterocycles, which are generally more reactive.[14] Although synthetic tools enabling C(sp<sup>2</sup>)-N bond formation have been well established,[15] facile intermolecular amination of C(sp3)-H bonds under mild conditions still remains a challenge in synthetic chemistry. [4, 11d, 16]

Recently, cross-dehydrogenative coupling reactions have attracted increasing attention owing to the use of abundant hydrocarbons (i.e., arenes, olefins, or cycloalkanes) as unreactive coupling partners for the C-C, C-O, and C-N bond formation.[17] Warren and co-workers reported the oxidative dehydrogenative amination of alkanes with unactivated amines using copper(II) as catalyst associated to di-tert-butyl peroxide (DTBP) (Figure 2b).[18] Outstanding achievements are emerging in this area<sup>[19]</sup> although only amines with low nucleophilicity were successfully alkylated. For example, Hartwig and co-workers reported C(sp3)-H amination of cyclohexane with phthalimide and sulfonamides.[19c] Cai and co-workers have reported that with NH-indole, using nickel catalyst, regiospecific C2- or C3-products were obtained without the formation of N-alkylated indole (Figure 2c).[20] In addition. Yi and co-workers have reported that under metal-free conditions in the presence of DTBP, the cross-dehydrogenative coupling between NH-indoles and (cyclo)alkanes took place at the C7 position of the indole unit, and in some cases C2 or C4 regioselectivity was overserved (Figure 2d).[21] To the best of our knowledge, oxidative dehydrogenative amination (cyclo)alkanes using electron-rich NH-heteroarenes as amine sources has not been reported yet owing to the chemoselectivity issue between C-H or N-H reactivity. [22] Inspired by previous works on the generation of alkyl radical in the presence of coordinated amines, [18, 19c] we herein report a general method for the C-N bond formation via C-H/N-H cross-coupling of NH-free

indoles, pyrroles, benzotriazoles, indazoles, and pyrazoles with (cyclo)alkanes to afford the corresponding *N*-alkylated heterocycles in good yields with a high chemoselectivity (Figure 2e).

**a.** Synthesis of N-Alkylheterocycles  $\emph{via} \ S_{N2} \ reaction.^{[6]}$ 

$$\begin{array}{c} Y = Z \\ NH \\ + X - Alk \\ X = Cl, Br, I, OTf \end{array}$$

$$\begin{array}{c} BM \\ -(BH + MX) \\ Z, Y = CH \text{ or } N \end{array}$$

**b.** Copper-Catalyzed Oxidative Dehydrogenative Amination of Electron-Deficient Amine Derivatives (Warren). [18]

CI

CI

c. Ni-Catalyzed CDC C2 or C3 Alkylations of Indole Derivatives (Cai).  $^{\text{[19]}}$ 

d. Metal-Free C7 Alkylation of Indole Derivatives with Cyclohexanes  $\emph{via}$  CDC (Yi). $^{[21]}$ 

**e.** Copper-Catalyzed Oxidative Dehydrogenative Amination with *NH*-Heterocycles (this work)

Figure 2. N-Alkylation of amines and NH-Heterocycles

#### **Results and Discussion**

We started our investigation using NH-free indole and cyclohexane as model substrates. The reaction was initially studied with DTBP as an oxidant, in tBuOH without catalyst. As reported by Yi, [21] we have not observed the formation of the amination product 1 after heating at 135 °C over 15 h, but we did observe the formation of C-alkylated products (Table 1, entry 1). When FeCl<sub>3</sub> (5 mol%) in the presence of phenanthroline (L1) (10 mol%) was used as the catalytic system, the amination product 1 was obtained in low yield with predominant formation of other C2, C3 and C7-alkylated products (Table 1, entry 2). FeCl2, in concert with L1, gave a similar result (Table 1, entry 3). contrast, when copper (II or I) oxides, combined with L1, were used as catalysts, 1 was selectively obtained in 41% and 50% yield, respectively without the formation of other C2, C3 and C7alkylated products (Table 1, entries 4 and 5). Encouraged by these results, a number of other copper sources, including CuBr, Cul, [MeCN]<sub>4</sub>Cu(I)PF<sub>6</sub>, Cu(I)Br(Me<sub>2</sub>S), and Cu(OAc)<sub>2</sub>.n-H<sub>2</sub>O were evaluated in the presence of L1 (Table 1, entries 6-10). The results showed that Cu(I)Br(Me<sub>2</sub>S) displayed the highest catalytic activity for this reaction and *N*-cyclohexylindole (1) was isolated in 67% yield (Table 1, entry 10). Next, we evaluated without success the influence of other *N*,*N*-bidentate ligands, such as 4,7-diphenyl-1,10-phenanthroline (**L2**), 2,2'-bipyridine (**L3**), 4,4'-di*tert*-butyl-2,2'-bipyridine (**L4**) and 2,2':6',2"-terpyridine (**L5**) (Table 1, entries 11-14). When the reaction is performed in absence of phenanthroline (**L1**), the yield in 1 dropped to 10% concurrent with the formation of other C2, C3 and C7–alkylated products (Table 1, entry 15). Other oxidants such as TBHP, BPO were inefficient for the coupling of cyclohexane bonds with *NH*-indole; whereas, DCP give a lower yield of 32% in the desired product 1 (Table 1, entries 16–18). It is important to note that using a lower amount of cyclohexane the yield of 1 decreased (Table 1, entry 19). The reaction is less efficient in other solvents such as benzene, trifluorotoluene, or 1,2-dichloroethane (Table 1, entries 20-22).

 $\textbf{Table 1} \ \, \textbf{Optimization of Oxidative Dehydrogenative Amination of Cyclohexane} \\ \text{with } \textit{NH-} \textbf{free Indole}.$ 

[a] Determined by GC-analysis using *n*-dodecane as internal standard, isolated yield is shown in parentheses. [b] Formation of of other C2, C3 and C7-alkylated products. [c] Reaction performed at 0.5 mol/L of cyclohexane.

With the optimized reaction conditions in hands, we then explored the substrate scope. Firstly, we investigated the reactivity of *NH*-heterocycles with cyclohexane using Cu/L1/DTBP system (Scheme 1). *NH*-6-Fluoroindole and *NH*-5-bromoindole were successfully *N*-alkylated to deliver the corresponding *N*-cyclohexylindoles 2 and 3 in 64% and 94% yield, respectively. It should be mentioned that for these reactions, no cleavage of the C-halogen bonds were observed, allowing further transformations. An electron-donating group on the indole moiety,

such as 5-methoxyl, slightly affected the yield in the coupling product 4; while an electron-withdrawing group such as nitrile or methyl ester allowed the formation of the desired N-alkylindoles 5 and 6 in 54% and 62% yields, respectively. The reaction conditions tolerate a C3 substituent on the indole moiety such as ester. Indeed, from methyl NH-indole-3-carboxylate the optimized reaction conditions gave exclusively the N-alkylation product 7 in 81%. Whereas, indole bearing a C2-substituent such as methyl or phenyl did not allow the formation of the corresponding Nalkylation product. Under these reaction conditions, NH-pyrrole was also alkylated affording N-cyclohexylpyrrole 8 in 65% yield, without the formation of C2 or C3 alkylation products. Cu/L1/DTBP-catalyzed oxidative dehydrogenative amination tolerated functional groups such as ketone and nitrile. Indeed, 2acetyl, 2-propionyl, and 2-nitrile pyrroles successfully reacted with cyclohexane to afford the N-cyclohexylpyrroles 9-11 in 84-97% yields. A trisubstituted pyrrole including ethyl ester group was also N-alkylated to afford 12 in 55% yield. Polv-nitrogen heterocyclic compounds are ubiquitous structural elements important to a wide range of fine and bulk chemical fields.[23] Therefore, we investigated their reactivity in oxidative dehydrogenative amination. NH-3,5-Dimethylpyrazole and NHindazole smoothly reacted in the presence of cyclohexane to give the corresponding N-alkylated products 13 and 14 in moderate vields, but without the formation of C-alkylation products. NH-Benzotriazole displayed poor reactivity, as the N-alkylated products 15-17 were isolated in moderate to low yields. In the case of 5-methylbenzo[1,2,3]triazole, the product was isolated as mixture of two regioisomers 17a and 17b in 55:45 ratio. Finally, our catalytic system displayed similar performance than other protocols<sup>[19a]</sup> for the N-alkylation of sulfonamides, as 18 and 19 were obtained in 96% and 86% yields. It is important to note that some functional groups on the heterocycle moiety such as nitro, amino, or hydroxyl are not compatible with the reaction conditions.

**Scheme 1.** Scope of *NH*-Heterocycle in Cu-catalyzed C(sp³)–H Amination of Cyclohexane.

Substrate scope studies were also carried out on other hydrocarbons using NH-pyrrole or NH-indole as amine sources (Scheme 2). As expected, the reaction between NH-indole with linear alkanes (e.g., pentane or hexane) leads to a variety of Nalkylated product at C1, C2, C3 carbons. The direct amination of other cycloalkanes (e.g., cyclopentane and cyclooctane) with NHindole allowed the formation of 20 and 21 in 58% and 55% yields, respectively. Tetrahydrofuran can be also used as suitable alkylating agent through the regioselective C(sp3)-H bond activation. [24] The amination occurred at the  $\alpha$ -position of the oxygen atom to afford 22 and 23 in 69% and 61% yields from NH-2-cyanopyrrole and *NH*-indole. Similarly, 1,4-dioxane was successfully coupled with 2-cyanopyrrole and indole giving the Nalkylated products 24 and 25 in 68% and 43% yield, respectively. Interestingly, the reaction of 2-cyanopyrrole with 2,2dimethoxypropane led selectively to the formation of N-(2methoxypropan-2-yl)-2-cyanopyrrole (26) in 57% yield. product results from the substitution of one methoxy group. Methyl tert-butyl ether reacted at the methoxy position to afford regioselectively the *N*-alkylated heterocycle **27** in 46% yield. In contrast, cyclopentyl methyl ether provided a mixture of two regioisomers **28a** and **28b** in 1:1 ratio. These regioselectivities are in line with a radical pathway, in which the radical is stabilized by the oxygen atom. Finally, from toluene, pyrrole was *N*-benzylated in moderate yield.

**Scheme 2.** Scope of (Cyclo)Alkanes in Cu-catalyzed C(sp³)—H Amination with *NH*-Heterocycle. [a] Using 2,2-dimethoxypropane as alkane source; [b] Using methyl *tert*-butyl ether as alkane source; [c] Using methyl cyclopentyl ether as alkane source

To obtain a better understanding of the mechanism, a radical control experiment was carried out in the presence of a radical scavenger (Scheme 3). When 2 equivalents of 2,2,6,6-tetramethylpiperidine1-oxyl (TEMPO) was added to the reaction under the standard conditions, no desired product 1 was detected. However, we have observed the formation of the adduct cyclohexane—TEMPO 30 (determined by GC-MS), which implied that a radical process is involved in the oxidative dehydrogenative amination of (cyclo)alkanes with NH-heterocycles.

Scheme 3. Radical Trapping Experiment with TEMPO.

Based on our experimental results and previous reports, [19] a possible mechanism is described in Scheme 4. First, homolytic

cleavage of the DTBP peroxide under heating allowed the formation of *tert*-butoxyl radical (*t*BuO') which is trapped by the copper(I) **A** to generate the copper(II) alkoxide **B**. σ-Bond metathesis of **B** with an *NH*-heterocycle regenerates intermediate **C**. The coordination of the *N*-heterocycle to the copper centre renders the N atom particularly reactive toward reaction with C-based radicals (R'), which are generated by abstracting a hydrogen atom from (cyclo)alkane by *tert*-butoxyl radical (*t*BuO'). Reaction of the alkyl radical (R') with Cu(II) **C** should allow the formation of the Cu(III) intermediate **D**, which undergoes a reductive elimination to release the desired *N*-alkylated heterocycle and reduce the Cu center to regenerate active copper(I) **A**.

Scheme 4. Proposed Mechanism.

#### Conclusions

In summary, we have described a copper-catalyzed oxidative dehydrogenative amination of unactivated alkanes with NHheterocycles to produce exclusively the corresponding Nalkylated products without formation of a C-C bond. Choice of copper source as well as the phenanthroline ligand were found to be critical to suppress the formation of C-alkylated heterocycles. A wide variety of NH-heterocycles including poly-nitrogen heterocyclic compounds were successfully N-alkylated. Moreover, the reaction conditions tolerate a broad variety of substituents of the heterocycles such as methyl, methoxy, fluoro, bromo, ketone, ester, and nitrile. This method employs abundant and cheap feedstock under oxidative conditions, without any prefunctionalization or the production of stoichiometric metal salt waste, making it very attractive for practical use in the preparation of useful N-alkylated heterocyclic derivatives.

#### **Experimental Section**

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**General Remarks:** All reactions were run in Schlenk tubes without the use of an inert atmosphere. Commercial *NH*-free heterocycles and (cyclo)alkanes were used without purification. The reactions were followed by GC and NMR. <sup>1</sup>H and <sup>13</sup>C spectra were recorded with a Bruker 400 MHz spectrometer in CDCl<sub>3</sub> solutions. Chemical shifts are reported in ppm relative to CDCl<sub>3</sub> (7.26 for <sup>1</sup>H NMR and 77.0 for <sup>13</sup>C NMR). Flash chromatography was performed on silica gel (230–400 mesh).

General Procedure for Copper-Catalyzed Oxidative Dehydrogenative C(sp³)—H Bond Amination of (Cyclo)Alkanes with NH-free Heterocycles: In a dried 25 mL Schlenk tube equipped with a magnetic stirrer, CuBr(Me<sub>2</sub>S) (5.0 mg, 0.025 mmol, 5 mol%) and 1,10-phenanthroline (9.0 mg, 0.05 mmol, 10 mol%) were dissolved in *t*-BuOH (1 mL) and (cyclo)alkane (1 mL). Then, NH-free heterocycles (0.5 mmol) and additional (cyclo)alkane (2 mL) were successfully added before addition of DTBP (292 mg, 2.0 mmol, 4 equiv.). The resulting solution mixture was stirred at 135 °C over 15 h. After evaporation of solvent, the product was purified using flash column chromatography on silica gel.

**N-Cyclohexylindole (1):** Following the general procedure using *NH*-indole (58 mg, 0.5 mmol) and cyclohexane (3 mL), **1** was isolated in 67% yield (66 mg) as a colorless oil after purification using flash column chromatography with pentane: ethyl acetate = 40 : 1.  $^{1}$ H NMR (300 MHz, CDCl<sub>3</sub>) δ (ppm) 7.70 (d, J = 7.8 Hz, 1H), 7.45 (d, J = 8.2 Hz, 1H), 7.30 − 7.22 (m, 2H), 7.16 (d, J = 7.2 Hz, 1H), 4.28 (tt, J = 3.7, 11.8 Hz, 1H), 2.28 − 2.12 (m, 2H), 2.08 − 1.92 (m, 3H), 1.74 (td, J = 3.1, 12.3 Hz, 2H), 1.56 (tdd, J = 3.3, 8.1, 12.9 Hz, 3H), 1.44 − 1.25 (m, 2H).  $^{13}$ C NMR (75 MHz, CDCl<sub>3</sub>) δ (ppm) 135.6, 128.5, 124.1, 121.1, 121.0, 119.3, 109.5, 101.0, 55.1, 33.6, 26.1, 25.8. MS(EI) m/z(%) = 199 ([M]<sup>+</sup>,49), 156 (30), 130 (10),117 (100), 90 (14), 55 (13).

**N-Cyclohexyl-6-fluoroindole (2)**: Following the general procedure using 6-Fluoro-*NH*-indole (68 mg, 0.5 mmol) and cyclohexane (3 mL), **2** was isolated in 64% yield (69 mg) as a light yellow oil after purification using flash column chromatography with pentane: ethyl acetate = 40:1. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 7.55 (dd, J = 8.6, 5.5 Hz, 1H), 7.22 (d, J = 3.2 Hz, 1H), 7.08 (dd, J = 10.2, 1.7 Hz, 1H), 6.89 (dd, J = 9.4, 2.2 Hz, 1H), 6.51 (d, J = 3.1 Hz, 1H), 4.12 (tt, J = 11.8, 3.6 Hz, 1H), 2.23 – 2.06 (m, 2H), 2.04 – 1.90 (m, 2H), 1.76 – 1.64 (m, 2H), 1.60 – 1.44 (m, 2H), 1.40 – 1.24 (m, 2H). <sup>19</sup>F NMR (376 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) -121.6. <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 159.6 (d, J = 236.6 Hz), 135.5 (d, J = 10.0 Hz), 124.9, 124.5 (d, J = 3.6 Hz), 121.5 (d, J = 10.2 Hz), 108.0 (d, J = 24.6 Hz), 101.2, 95.9 (d, J = 26.3 Hz), 55.4, 33.4, 25.9, 25.6. MS(EI) m/z(%) = 217 ([M]\*,36), 174 (11), 135 (100), 108 (14), 55 (19).

**5-Bromo-***N***-cyclohexylindole (3)**: Following the general procedure using 5-bromo-*NH*-indole (97 mg, 0.5 mmol) and cyclohexane (3 mL), **3** was isolated in 94% yield (130 mg) as a light yellow oil after purification using flash column chromatography with pentane : ethyl acetate = 40:1. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 7.75 (dd, J=1.4, 1.0 Hz, 1H), 7.28 – 7.24 (m, 2H), 7.22 (d, J=3.2 Hz, 1H), 6.45 (d, J=3.2 Hz, 1H), 4.17 (tt, J=11.8, 3.7 Hz, 1H), 2.19 – 2.03 (m, 2H), 2.01 – 1.86 (m, 2H), 1.87 – 1.75 (m, 1H), 1.75 – 1.62 (m, 2H), 1.61 – 1.40 (m, 2H), 1.40 – 1.21 (m, 1H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 134.2, 130.1, 125.2, 123.9, 123.4, 112.4, 110.9, 100.6, 55.4, 33.5, 25.9, 25.6. MS(EI) m/z (%) = 277 ([M]+,100), 236 (20), 197 (82), 168 (8), 155 (37), 116 (62), 89 (35), 55(73).

**N-Cyclohexyl-5-methoxyindole** (4). Following the general procedure using 5-methoxy-NH-indole (74 mg, 0.5 mmol) and cyclohexane (3 mL), 4 was isolated in 41% yield (46 mg) as a light yellow oil after purification using flash column chromatography with

pentane : ethyl acetate = 40 : 1.  $^{1}$ H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 7.29 (d, J = 8.9 Hz, 1H), 7.21 (d, J = 3.2 Hz, 1H), 7.12 (d, J = 2.4 Hz, 1H), 6.91 – 6.85 (m, 1H), 6.44 (d, J = 3.1 Hz, 1H), 4.17 (tt, J = 11.7, 3.6 Hz, 1H), 3.87 (s, 3H), 2.23 – 2.10 (m, 2H), 2.01 – 1.87 (m, 2H), 1.85 – 1.75 (m, 1H), 1.75 – 1.65 (m, 2H), 1.54 – 1.43 (m, 2H), 1.38 – 1.23 (m, 1H).  $^{13}$ C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 153.9, 130.9, 128.7, 124.6, 111.5, 110.1, 102.6, 100.5, 55.9, 55.3, 33.6, 26.0, 25.7. MS(EI) m/z(%) = 229 ([M]+,100), 214 (8), 186 (26), 147 (57), 132 (63), 104 (14), 55(18).

**N-Cyclohexylindole-5-carbonitrile (5):** Following the general procedure **A**. using *NH*-indole-5-carbonitrile (71 mg, 0.5 mmol) and cyclohexane 3 mL, **5** was isolated in 54 % yield (60 mg) as a light yellow oil, after purification using flash column chromatography with pentane: ethyl acetate = 40:1.  $^{1}$ H NMR (300 MHz, CDCl<sub>3</sub>) δ (ppm) 7.96 (s, 1H), 7.43 – 7.39 (m, 2H), 7.34 (d, J = 3.3 Hz, 1H), 6.58 (d, J = 3.3 Hz, 1H), 4.23 (tt, J = 3.7, 11.9 Hz, 1H), 2.16 – 2.08 (m, 2H), 2.01 – 1.92 (m, 2H), 1.86 – 1.78 (m, 1H), 1.72 (ddd, J = 3.2, 12.5, 24.7 Hz, 2H), 1.57 – 1.44 (m, 2H), 1.36 – 1.23 (m, 1H).  $^{13}$ C NMR (75 MHz, CDCl<sub>3</sub>) δ (ppm) 137.0, 128.2, 126.6, 126.5, 124.1, 120.9, 110.3, 102.3, 102.2, 55.6, 33.5, 25.8, 25.5. MS(EI) m/z(%) = 224 ([M]\*,41), 181 (18), 142 (100),115 (9), 83 (10), 55 (26).

**Methyl** *N*-cyclohexylindole-5-carboxylate (6): Following the general procedure **A**. using methyl *NH*-indole-5-carboxylate (88 mg, 0.5 mmol) and cyclohexane 3 mL, **6** was isolated in 62 % yield (80 mg) as a light yellow oil, after purification using flash column chromatography with pentane: ethyl acetate = 30:1.  $^{1}$ H NMR (300 MHz, CDCl<sub>3</sub>) δ (ppm) 8.40 (d, J = 1.3 Hz, 1H), 7.91 (dd, J = 1.6, 8.7 Hz, 1H), 7.38 (d, J = 8.7 Hz, 1H), 7.28 (d, J = 3.3 Hz, 1H), 6.60 (d, J = 3.2 Hz, 1H), 4.24 (tt, J = 3.7, 11.9 Hz, 1H), 3.93 (s, 3H), 2.13 (d, J = 11.3 Hz, 2H), 1.95 (d, J = 13.5 Hz, 2H), 1.81 (d, J = 13.1 Hz, 1H), 1.71 (qd, J = 3.2, 12.5 Hz, 2H), 1.58 – 1.44 (m, 2H), 1.36 – 1.22 (m, 1H).  $^{13}$ C NMR (75 MHz, CDCl<sub>3</sub>) δ (ppm) 168.3, 138.0, 128.0, 125.5, 124.0, 122.5, 121.3, 109.0, 102.8, 55.4, 51.8, 33.5, 25.9, 25.6. MS(EI) m/z(%) = 257 ([M]<sup>+</sup>,100), 226 (14), 214 (19),175 (45), 144 (97), 116 (31), 89 (11), 55 (31).

**Methyl 1-cyclohexyl-***1H***-indole-3-carboxylate (7):** Following the general procedure **A**. using methyl *1H*-indole-3-carboxylate (88 mg, 0.5 mmol) and cyclohexane 3 mL, **7** was isolated in 81% yield (104 mg) as a light yellow oil, after purification using flash column chromatography with pentane: ethyl acetate = 30:1.  $^1$ H NMR (300 MHz, CDCl<sub>3</sub>) δ (ppm) 8.24 – 8.19 (m, 1H), 7.97 (s, 1H), 7.45 – 7.38 (m, 1H), 7.31 – 7.25 (m, 2H), 4.23 (tt, J = 3.6, 11.8 Hz, 1H), 3.93 (s, 3H), 2.16 (d, J = 11.5 Hz, 2H), 2.00 – 1.91 (m, 2H), 1.81 (d, J = 13.1 Hz, 1H), 1.76 – 1.64 (m, 2H), 1.58 – 1.43 (m, 2H), 1.36 – 1.23 (m, 1H).  $^{13}$ C NMR (75 MHz, CDCl<sub>3</sub>) δ (ppm) 165.6, 136.2, 131.0, 126.7, 122.4 121.8, 121.8, 110.0, 106.9, 55.6, 50.9, 33.4, 25.8, 25.5. MS(EI) m/z(%) = 257 ([M]<sup>+</sup>,88), 226 (14), 175 (43),144 (100), 116 (12), 55 (23).

**N-Cyclohexylpyrrole (8).** Following the general procedure using NH-pyrrole (34 mg, 0.5 mmol) and cyclohexane (3 mL), **8** was isolated in 65% yield (48 mg) as a light yellow oil after purification using flash column chromatography with pentane: ethyl acetate = 40 : 1.  $^{1}$ H NMR (300 MHz, CDCl<sub>3</sub>) δ (ppm) 6.79 – 6.74 (m, 2H), 6.20 – 6.15 (m, 2H), 3.83 (tt, J = 11.6, 3.6 Hz, 1H), 2.20 – 2.06 (m, 2H), 1.97 – 1.87 (m, 2H), 1.81 – 1.70 (m, 1H), 1.70 – 1.57 (m, 2H), 1.52 – 1.37 (m, 2H), 1.33 – 1.19 (m, 1H).  $^{13}$ C NMR (101 MHz, CDCl<sub>3</sub>) δ (ppm) 118.4, 107.4, 58.7, 34.7, 25.8, 25.5. MS(EI) m/z(%) = 149 ([M]\*,67), 106 (19), 94 (23), 67 (100), 55 (23).

**2-Acetyl-N-Cyclohexylpyrrole (9).** Following the general procedure using 2-acetyl-*NH*-pyrrole (55 mg, 0.5 mmol) and cyclohexane (3

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mL), **9** was isolated in 84% yield (80 mg) as a light yellow oil after purification using flash column chromatography with pentane: ethyl acetate = 40:1.  $^{1}$ H NMR (400 MHz, CDCl<sub>3</sub>)  $\bar{o}$  (ppm) 7.11 – 7.07 (m, 1H), 6.96 (dd, J = 4.0, 1.6 Hz, 1H), 6.14 (dd, J = 4.0, 2.7 Hz, 1H), 5.17 – 5.06 (m, 1H), 2.43 (s, 3H), 2.17 – 1.95 (m, 2H), 1.92 – 1.81 (m, 2H), 1.77 – 1.68 (m, 1H), 1.54 – 1.42 (m, 4H), 1.27 – 1.12 (m, 1H).  $^{13}$ C NMR (101 MHz, CDCl<sub>3</sub>)  $\bar{o}$  (ppm) 188.4, 130.0, 125.5, 120.4, 108.1, 56.6, 34.6, 27.8, 25.9, 25.7. MS(EI) m/z(%) = 191 ([M]\*,38), 176 (62), 148 (10), 120 (7), 109 (46), 94 (100) 55 (25).

**N-Cyclohexyl-2-propionylpyrrole (10).** Following the general procedure using 2-propionyl-*NH*-pyrrole (62 mg, 0.5 mmol) and cyclohexane (3 mL), **10** was isolated in 78% yield (80 mg) as a light yellow oil after purification using flash column chromatography with pentane: ethyl acetate = 40:1.  $^{1}$ H NMR (300 MHz, CDCl<sub>3</sub>) δ (ppm) 7.10 – 7.07 (m, 1H), 6.98 (dd, J = 1.6, 4.0 Hz, 1H), 6.14 (dd, J = 2.7, 4.0 Hz, 1H), 5.22 – 5.05 (m, 1H), 2.82 (q, J = 7.4 Hz, 2H), 2.14 – 2.00 (m, 2H), 1.92 – 1.78 (m, 2H), 1.79 – 1.67 (m, 1H), 1.54 – 1.44 (m, 4H), 1.20- – 1.17 (m, 1H), 1.17 (t, J = 7.4 Hz, 3H).  $^{13}$ C NMR (75 MHz, CDCl<sub>3</sub>) δ (ppm) 191.9, 129.7, 125.2, 119.3, 108.0, 56.7, 34.6, 32.7, 25.9, 25.7, 9.0. MS(EI) m/z(%) = 205 ([M]+,24), 176 (42), 162 (3), 148 (8), 123 (14), 94(100), 81 (5),55(16).

*N*-Cyclohexylpyrrole-2-carbonitrile (11). Following the general procedure using *NH*-pyrrole-2-carbonitrile (46 mg, 0.5 mmol) and cyclohexane (3 mL), **11** was isolated in 97% yield (84 mg) as a light yellow oil after purification using flash column chromatography with pentane: ethyl acetate = 40:1.  $^{1}$ H NMR (300 MHz, CDCl<sub>3</sub>) δ (ppm) 6.91 (dd, J = 2.6, 1.6 Hz, 1H), 6.74 (dd, J = 3.9, 1.6 Hz, 1H), 6.15 (dd, J = 3.9, 2.8 Hz, 1H), 4.08 (tt, J = 11.9, 3.8 Hz, 1H), 2.14 – 2.03 (m, 2H), 1.96 – 1.84 (m, 2H), 1.80 – 1.69 (m, 1H), 1.69 – 1.57 (m, 2H), 1.52 – 1.35 (m, 2H), 1.30 – 1.18 (m, 1H).  $^{13}$ C NMR (75 MHz, CDCl<sub>3</sub>) δ (ppm) 123.1, 119.6, 114.1, 109.2, 102.8, 58.4, 34.2, 25.6, 25.2. MS(El) m/z(%) = 174 ([M]<sup>+</sup>,59), 93 (84), 83 (38), 67 (52), 55 (100).

Ethyl-*N*-cyclohexyl-3,5-dimethylpyrrole-2-carboxylate (12). Following the general procedure using ethyl 3,5-dimethyl-*NH*-pyrrole-2-carboxylate (84 mg, 0.5 mmol) and cyclohexane (3 mL), 12 was isolated in 55% yield (68 mg) as a light yellow oil after purification using flash column chromatography with pentane: ethyl acetate = 40 : 1.  $^{1}$ H NMR (300 MHz, CDCl<sub>3</sub>)  $\bar{o}$  (ppm) 5.73 (s, 1H), 5.08 – 4.77 (m, 1H), 4.28 (q, J = 7.1 Hz, 2H), 2.33 (s, 3H), 2.25 (s, 3H), 2.14 – 1.96 (m, 2H), 1.91 – 1.80 (m, 4H), 1.76 – 1.62 (m, 2H), 1.36 (t, J = 7.1 Hz, 3H), 1.30 – 1.16 (m, 2H).  $^{13}$ C NMR (75 MHz, CDCl<sub>3</sub>)  $\bar{o}$  (ppm) 162.3, 135.4, 129.6, 119.5, 112.5, 59.4, 57.1, 31.9, 26.6, 25.5, 15.2, 14.7, 14.5. MS(EI) m/z(%) = 249 ([M]+,96), 204 (23), 176 (68), 167 (72), 138 (82), 121 (100), 95 (76), 55 (36).

**N-Cyclohexyl-3,5-dimethylpyrazole (13).** Following the general procedure using 3,5-dimethyl-*1H*-pyrazole (48 m, 0.5 mmol) and cyclohexane (3 mL), **13** was isolated in 44% yield (39 mg) as a light yellow oil after purification using flash column chromatography with pentane: ethyl acetate = 40:1.  $^{1}$ H NMR (300 MHz, CDCl<sub>3</sub>) δ (ppm) 5.75 (s, 1H), 3.96 – 3.74 (m, 1H), 2.23 (s, 3H), 2.22 (s, 3H), 1.93 – 1.81 (m, 6H), 1.74 – 1.64 (m, 1H), 1.43 – 1.27 (m, 3H).  $^{13}$ C NMR (75 MHz, CDCl<sub>3</sub>) δ (ppm) 146.8, 137.6, 104.5 (d, J = 11.9 Hz), 57.2 (d, J = 10.7 Hz), 32.9, 25.9, 25.2, 13.6 (d, J = 8.9 Hz), 11.0 (d, J = 9.1 Hz). MS(EI) m/z(%) = 178 ([M]<sup>+</sup>,30), 163 (9), 135 (9), 123 (14), 109 (63), 97 (100), 81 (9).

**N-Cyclohexylindazole (14).** Following the general procedure using NH-indazole (59 mg, 0.5 mmol) and cyclohexane (3 mL), **14** was isolated in 56% yield (56 mg) as a light yellow oil after purification using flash column chromatography with pentane: ethyl acetate = 40: 1.  $^{1}$ H NMR (400 MHz, CDCl<sub>3</sub>)  $\bar{\delta}$  (ppm) 8.02 (s, 1H), 7.73 (d, J = 8.1

Hz, 1H), 7.47 (d, J = 8.4 Hz, 1H), 7.36 (t, J = 7.9 Hz, 1H), 7.13 (t, J = 7.5 Hz, 1H), 4.60 – 4.26 (m, 1H), 2.11 – 1.99 (m, 2H), 1.99 – 1.92 (m, 2H), 1.84 – 1.74 (m, 1H), 1.63 – 1.59 (m, 1H), 1.54 – 1.41 (m, 2H), 1.41 – 1.30 (m, 2H).  $^{13}$ C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 138.6, 132.5, 125.8, 124.0, 121.2, 120.4, 109.2, 58.2, 32.5, 25.9, 25.5. MS(EI) m/z(%) = 200 ([M]<sup>+</sup>,37), 171 (13), 157 (37), 131(59), 118 (100), 91 (23), 77 (13).

**N-Cyclohexylbenzo[d][1,2,3]triazole (15).** Following the general procedure using *NH*-benzo[d][1,2,3]triazole (60 mg, 0.5 mmol) and cyclohexane (3 mL), **15** was isolated in 35% yield (35 mg) as a light yellow oil after purification using flash column chromatography with pentane: ethyl acetate = 40:1. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ (ppm) 8.06 (d, J = 8.3 Hz, 1H), 7.57 (d, J = 8.3 Hz, 1H), 7.45 (t, J = 6.6 Hz, 1H), 7.34 (d, J = 6.9 Hz, 1H), 4.72 – 4.59 (m, 1H), 2.23 – 2.15 (m, 2H), 2.06 – 1.94 (m, 2H), 1.87 – 1.77 (m, 1H), 1.62 – 1.34 (m, 5H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ (ppm) 146.1, 132.2, 126.7, 123.7, 120.1, 109.8, 59.1, 32.6, 25.6, 25.3. MS(EI) m/z(%) = 201 ([M]\*,26), 147 (12), 133 (27), 120 (100), 91 (57), 64(29), 55(25). mp 102-104 °C.

*N*-Cyclohexyl-5,6-dimethylbenzo[*d*][1,2,3]triazole (16). Following the general procedure using 5,6-dimethyl-*NH*-benzo[*d*][1,2,3]triazole (74 mg, 0.5 mmol) and cyclohexane (3 mL), 16 was isolated in 30% yield (35 mg) as a light yellow solid after purification using flash column chromatography with pentane: ethyl acetate = 40 : 1.  $^{1}$ H NMR (300 MHz, CDCl<sub>3</sub>) δ (ppm) 7.79 (s, 1H), 7.33 (s, 1H), 4.68 – 4.52 (m, 1H), 2.44 (s, 3H), 2.41 (s, 3H), 2.21 – 2.14 (m, 4H), 2.07 – 1.96 (m, 2H), 1.86 – 1.77 (m, 1H), 1.61 – 1.34 (m, 3H).  $^{13}$ C NMR (75 MHz, CDCl<sub>3</sub>) δ (ppm) 145.4, 136.9, 133.4, 131.2, 119.1, 109.3, 58.9, 32.5, 25.6, 25.3, 20.9, 20.4. MS (EI) m/z(%) = 229 ([M]<sup>+</sup>,26), 158 (9), 119(100), 91 (13), 55 (36). mp 122-124 °C.

N-Cyclohexyl-5-methylbenzo[d][1,2,3]triazole (17a) and Ncyclohexyl-6-methylbenzo[d][1,2,3]triazole (17b). Following the general procedure using 5-methyl-NH-benzo[d][1,2,3]triazole (74 mg, 0.5 mmol) and cyclohexane (3 mL), 17a and 17b was isolated in 41% yield (44 mg) in 45:55 ratio as a light yellow oil after purification using flash column chromatography with pentane: ethyl acetate = 40 : 1. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 7.92 (d, J = 8.5 Hz, 1H, **17b**), 7.81 (s, 1H, 17a), 7.46 (d, J = 8.5 Hz, 1H, 17a), 7.34 (s, 1H, 17b), 7.31 - 7.26 (m, 1H, 17a), 7.17 (dd, J = 1.4, 8.5 Hz, 1H, 17b), 4.76 -4.47 (m, 1H, 17a+17b), 2.54 (s, 3H, 17b), 2.51 (s, 3H, 17a), 2.23 -2.09 (m, 4H, 17a+17b), 2.08 - 1.92 (m, 2H, 17a+17b), 1.87 - 1.75 (m, 1H, **17a+17b**), 1.61 – 1.46 (m, 2H, **17a+17b**), 1.45 – 1.33 (m, 1H, 17a+17b). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) δ (ppm) 146.7 (17a), 144.7 (17b), 137.2 (17b), 133.7 (17a), 132.7 (17a), 130.7 (17b), 129.1 (17a), 126.0 (17b), 119.4 (17b), 118.9 (17a), 109.3 (17b), 109.0 (17a), 59.0 (17a), 58.8 (17b), 32.5 (17a), 32.5 (17b), 25.6 (17a and 17b), 25.3 (17a and 17b), 22.0 (17a or 17b), 21.4 (17a or 17b). MS (EI) m/z(%) = 215 ([M]+,74), 186 (15), 172(28), 158 (75), 144 (53), 105 (100), 91 (38), 55 (76).

**N-Cyclohexylbenzenesulfonamide (18).** Following the general procedure using benzenesulfonamide (79 mg, 0.5 mmol) and cyclohexane (3 mL), **18** was isolated in 96 % yield (114 mg) as a light yellow oil after purification using flash column chromatography with pentane: ethyl acetate = 10:1.  $^{1}$ H NMR (400 MHz, CDCl<sub>3</sub>) δ (ppm) 7.93 – 7.86 (m, 2H), 7.57 – 7.51 (m, 1H), 7.51 – 7.43 (m, 2H), 5.14 (d, J = 7.5 Hz, 1H), 3.18 – 3.03 (m, 1H), 1.74 – 1.66 (m, 2H), 1.64 – 1.56 (m, 2H), 1.51 – 1.41 (m, 1H), 1.27 – 1.19 (m, 1H), 1.19 – 1.12 (m, 3H), 1.12 – 1.00 (m, 1H).  $^{13}$ C NMR (101 MHz, CDCl<sub>3</sub>) δ (ppm) 141.5, 132.4, 129.0, 126.9, 52.7, 33.8, 25.1, 24.6. MS (EI) m/z(%) = 239 ([M]+,32), 210 (7), 196 (87), 158 (9), 141(72), 98 (46), 77 (100), 51 (23).

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**4-Chloro-***N***-Cyclohexylbenzenesulfonamide (19).** Following the general procedure using 4-chlorobenzenesulfonamide (95 mg, 0.5 mmol) and cyclohexane (3 mL), **19** was isolated in 86% yield (117 mg) as a light yellow solid after purification using flash column chromatography with pentane: ethyl acetate = 10:1.  $^{1}$ H NMR (400 MHz, CDCl<sub>3</sub>) δ (ppm) 7.83 (d, J = 8.4 Hz, 1H), 7.45 (d, J = 8.4 Hz, 1H), 5.15 (d, J = 7.6 Hz, 1H), 3.12 (tt, J = 6.8, 13.5 Hz, 1H), 1.76 – 1.67 (m, 2H), 1.64 – 1.56 (m, 2H), 1.53 – 1.42 (m, 1H), 1.26 – 1.05 (m, 5H).  $^{13}$ C NMR (75 MHz, CDCl<sub>3</sub>) δ (ppm) 140.1, 138.8, 129.3, 128.4, 52.8, 33.8, 25.1, 24.6. MS (EI) m/z(%) = 273 ([M]\*,34), 244 (7), 230 (100), 175 (70), 153 (8), 111(84), 75 (33), 55(21).

*N*-Cyclopentylindole (20). Following the general procedure using *NH*-indole (58 mg, 0.5 mmol) and cyclopentane (3 mL), 20 was isolated in 58 % yield (54 mg) as a light orange oil after purification using flash column chromatography with pentane : ethyl acetate = 40 : 1.  $^{1}$ H NMR (400 MHz, CDCl<sub>3</sub>) δ (ppm) 7.69 (d, J = 7.8 Hz, 1H), 7.46 (d, J = 8.2 Hz, 1H), 7.30–7.23 (m, 2H), 7.20–7.12 (m, 1H), 6.56 (d, J = 3.2 Hz, 1H), 4.91–4.79 (m, 1H), 2.32 – 2.21 (m, 2H), 2.06–1.90 (m, 4H), 1.86–1.77 (m, 2H).  $^{13}$ C NMR (75 MHz, CDCl<sub>3</sub>) δ (ppm) 136.2, 128.6, 124.5 (d, J = 6.9 Hz), 121.2 (d, J = 1.8 Hz), 120.9 (d, J = 4.2 Hz), 119.3, 109.9, 101.0, 57.0, 32.6, 24.1. MS (EI) m/z(%) = 185 ([M]\*,40), 156 (24), 117 (100), 90 (16).

**N-Cyclooctylindole (21).** Following the general procedure using NH-indole (58 mg, 0.5 mmol) and cyclooctane (3 mL), **21** was isolated in 55% yield (62 mg) as a light yellow oil after purification using flash column chromatography with pentane: ethyl acetate = 40 : 1.  $^{1}$ H NMR (400 MHz, CDCl<sub>3</sub>) δ (ppm) 7.69 (d, J = 7.9 Hz, 1H), 7.43 (d, J = 8.2 Hz, 1H), 7.28 (d, J = 3.3 Hz, 1H), 7.24 (d, J = 7.5 Hz, 1H), 7.15 (t, J = 7.5 Hz, 1H), 6.55 (d, J = 3.1 Hz, 1H), 4.63 – 4.53 (m, 1H), 2.21 – 1.99 (m, 2H), 1.94 – 1.83 (m, 2H), 1.82 – 1.65 (m, 10H).  $^{13}$ C NMR (75 MHz, CDCl<sub>3</sub>) δ (ppm) 135.3, 128.5, 124.7, 121.1, 121.0, 119.2, 109.6, 101.0, 56.0, 33.4, 26.9, 26.1, 24.9. MS (EI) m/z(%) = 227 ([M]<sup>+</sup>,67), 156 (54), 143 (37), 130 (23), 117 (100), 90 (14), 69 (15), 14 (55).

*N*-(Tetrahydrofuran-2-yl)pyrrole-2-carbonitrile (22). Following the general procedure using *NH*-pyrrole-2-carbonitrile (46 mg, 0.5 mmol) and tetrahydrofuran (3 mL), 22 was isolated in 69% yield (56 mg) as a light yellow oil after purification using flash column chromatography with pentane: ethyl acetate = 40:1.  $^{1}$ H NMR (400 MHz, CDCl<sub>3</sub>) δ (ppm) 6.97 (dd, J = 1.7, 2.5 Hz, 1H), 6.83 (dd, J = 1.5, 3.8 Hz, 1H), 6.17 (t, J = 3.4 Hz, 1H), 6.00 (dd, J = 3.7, 6.5 Hz, 1H), 4.21 (dd, J = 7.4, 14.1 Hz, 1H), 3.99 (dd, J = 7.2, 15.3 Hz, 1H), 2.54 – 2.35 (m, 1H), 2.28 – 2.17 (m, 1H), 2.16 – 2.00 (m, 2H).  $^{13}$ C NMR (101 MHz, CDCl<sub>3</sub>) δ (ppm) 123.6, 121.4, 113.7, 109.6, 101.6, 88.3, 69.5, 33.6, 24.5. MS(EI) m/z(%) = 162 ([M]\*,12), 131 (5), 92 (10), 71 (100).

**N-(Tetrahydrofuran-2-yl)indole** (23). Following the general procedure using *NH*-indole (58 mg, 0.5 mmol) and tetrahydrofuran (3 mL), 23 was isolated in 61% yield (57 mg) as a colorless yellow oil after purification using flash column chromatography with pentane: ethyl acetate = 40 : 1.  $^{1}$ H NMR (400 MHz, CDCl<sub>3</sub>) δ (ppm) 7.69 (d, J = 7.8 Hz, 1H), 7.53 (d, J = 8.3 Hz, 1H), 7.31 – 7.26 (m, 2H), 7.20 (t, J = 7.5 Hz, 1H), 6.60 (d, J = 3.1 Hz, 1H), 6.60 (d, J = 3.1 Hz, 1H), 6.29 (dd, J = 6.1, 4.3 Hz, 1H), 4.18 (td, J = 7.9, 6.0 Hz, 1H), 4.05 (dd, J = 15.3, 7.3 Hz, 1H), 2.47 – 2.39 (m, 2H), 2.26 – 2.11 (m, 2H).  $^{13}$ C NMR (101 MHz, CDCl<sub>3</sub>) δ (ppm) 135.7, 129.2, 124.0, 121.8, 121.0, 120.0, 110.0, 102.3, 85.8, 68.4, 31.6, 24.8. MS(EI) m/z (%) = 187 ([M]\*,21), 117 (100), 89 (10), 71 (22).

**N-(1,4-Dioxan-2-yl)pyrrole-2-carbonitrile (24).** Following the general procedure using *NH*-pyrrole-2-carbonitrile (46 mg, 0.5 mmol) and 1,4-dioxane (3 mL), **24** was isolated in 68% yield (60 mg) as a

light yellow oil after purification using flash column chromatography with pentane: ethyl acetate = 40:1.  $^{1}$ H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 7.20 (dd, J = 2.7, 1.5 Hz, 1H), 6.85 (dd, J = 3.9, 1.4 Hz, 1H), 6.23 (t, J = 3.6 Hz, 1H), 5.55 (dd, J = 6.7, 3.0 Hz, 1H), 4.01 (dd, J = 11.8, 3.0 Hz, 1H), 3.93 – 3.71 (m, 5H).  $^{13}$ C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 125.1, 121.2, 113.2, 110.1, 103.6, 80.6, 68.7, 66.0, 64.9. MS(EI) m/z(%) = 178 ([M]<sup>+</sup>,20), 118 (7), 86 (100), 59 (12).

*N*-(1,4-Dioxan-2-yl)indole (25). Following the general procedure A. using *NH*-indole (58 mg, 0.5 mmol) and 1,4-dioxane (3 mL), **25** was isolated in 43% yield (44 mg) as a white solid after purification using flash column chromatography with pentane : ethyl acetate = 40 : 1. 

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ (ppm) 7.68 (d, J = 7.8 Hz, 1H), 7.52 (d, J = 8.2 Hz, 1H), 7.47 (d, J = 3.4 Hz, 1H), 7.32 – 7.26 (m, 1H), 7.23 – 7.16 (m, 1H), 6.61 (d, J = 3.2 Hz, 1H), 5.80 – 5.74 (m, 1H), 4.14 – 4.10 (m, 2H), 3.97 – 3.81 (m, 4H). 

<sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ (ppm) 136.1, 128.8, 125.5, 122.2, 121.1, 120.4, 110.1, 103.0, 79.2, 68.7, 66.4, 64.6. MS (EI) m/z (%) = 203([M]+,18), 117(100), 90 (12). mp 76-78 °C.

*N*-(2-Methoxypropan-2-yl)pyrrole-2-carbonitrile (26). Following the general procedure using *NH*-pyrrole-2-carbonitrile (46 mg, 0.5 mmol) and 2,2-dimethoxypropane (3 mL), **26** was isolated in 57% yield (47 mg) as a light yellow oil after purification using flash column chromatography with pentane: ethyl acetate = 40:1.  $^{1}$ H NMR (400 MHz, CDCl<sub>3</sub>) δ (ppm) 6.93 (dd, J= 2.8, 1.7 Hz, 1H), 6.91 (dd, J= 3.8, 1.6 Hz, 1H), 6.15 (dd, J= 3.7, 3.0 Hz, 1H), 3.12 (s, 3H), 1.81 (s, 6H).  $^{13}$ C NMR (101 MHz, CDCl<sub>3</sub>) δ (ppm) 124.3, 123.4, 114.7, 108.7, 101.4, 89.8, 50.1, 27.0. MS(EI) m/z(%) = 164 ([M]<sup>+</sup>,4), 133 (11), 92(11), 73 (100).

*N*-(*Tert*-butoxymethyl)indole (27). Following the general procedure using *NH*-indole (58 mg, 0.5 mmol) and 2-methoxy-2-methylpropane (3 mL), 27 was isolated in 46% yield (47 mg) as a light yellow oil after purification using flash column chromatography with pentane : ethyl acetate = 40 : 1.  $^{1}$ H NMR (400 MHz, CDCl<sub>3</sub>) δ (ppm) 7.66 (d, J = 7.9 Hz, 1H), 7.53 (d, J = 8.2 Hz, 1H), 7.31 – 7.25 (m, 1H), 7.22 (d, J = 3.2 Hz, 1H), 7.20 – 7.14 (m, 1H), 6.56 (d, J = 3.2 Hz, 1H), 5.56 (s, 2H), 1.29 (s, 9H).  $^{13}$ C NMR (101 MHz, CDCl<sub>3</sub>) δ (ppm) 136.1, 129.1, 127.5, 121.9, 120.9, 119.9, 110.1, 102.4, 74.2, 70.3, 28.0. MS(EI) m/z(%) = 203 ([M]+,20), 130 (35), 117(100), 57 (22).

*N*-((Cyclopentyloxy)methyl)pyrrole-2-carbonitrile (28a) and 1-(1-methoxycyclopentyl)pyrrole-2-carbonitrile (28b). Following the general procedure using *NH*-pyrrole-2-carbonitrile (46 mg, 0.5 mmol) and methoxycyclopentane (3 mL), **28a** was isolated in 24% yield (23 mg) as pure form as a light yellow oil, and **28b** was isolated in 24% yield (23 mg) as pure form as a light yellow oil after purified using flash column chromatography with pentane : ethyl acetate = 40 : 1. **28a**  $^{1}$ H NMR (400 MHz, CDCl<sub>3</sub>) δ (ppm) 7.00 (dd, J = 2.6, 1.6 Hz, 1H), 6.83 (dd, J = 3.9, 1.5 Hz, 1H), 6.24 (dd, J = 3.8, 2.9 Hz, 1H), 5.35 (s, 2H), 4.01 – 3.91 (m, 1H), 1.68 (ddd, J = 6.9, 4.2, 1.7 Hz, 4H), 1.63 – 1.44 (m, 4H).  $^{13}$ C NMR (101 MHz, CDCl<sub>3</sub>) δ (ppm) 126.7, 120.9, 113.4, 110.4, 104.0, 80.2, 76.1, 32.3, 23.5. MS(EI) m/z(%) = 190 ([M]+,51), 159 (43), 145 (15), 132 (25), 106 (100), 97 (25), 71 (48), 67 (52).

**28 b** <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 6.97 (dd, J = 2.8, 1.7 Hz, 1H), 6.90 (dd, J = 3.8, 1.6 Hz, 1H), 6.13 (dd, J = 3.8, 2.9 Hz, 1H), 2.96 (s, 3H), 2.59 – 2.47 (m, 2H), 2.20 – 2.06 (m, 2H), 1.92 – 1.73 (m, 4H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 125.3, 123.2, 114.7, 108.4, 102.5, 99.1, 50.0, 35.9, 22.2. MS(EI) m/z(%) = 190 ([M]+,4), 119 (3), 106 (4), 92 (100), 78 (77), 69(44), 51(12).

N-Benzylpyrrole-2-carbonitrile (29). Following the procedure using NH-pyrrole-2-carbonitrile (46 mg, 0.5 mmol,) and toluene (3 mL), 29 was isolated in 24% yield (22 mg) as a light yellow oil after purification using flash column chromatography with pentane : ethyl acetate = 40 : 1.  $^{1}$ H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (m) 7.41 – 7.31 (m, 3H), 7.22 - 7.16 (m, 2H), 6.87 - 6.81 (m, 2H), 6.21 (dd, J = 3.9, 2.8 Hz, 1H), 5.21 (s, 2H). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) δ (ppm) 136.0, 129.0, 128.4, 127.4, 126.6, 120.3, 113.8, 109.9, 104.2, 52.4. MS(EI)  $m/z(\%) = 182 ([M]^+,18), 91 (100), 65 (14).$ 

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- L. L. Bambas, A. Weissberger, The Chemistry of Heterocyclic Compounds, Five Member Heterocyclic Compounds with Nitrogen and Sulfur or Nitrogen, Sulfur and Oxygen Except Thiazole, Vol. 4, John Wiley & Sons, 2009.
- S. R. Karra, Merck Patent GmbH, Germany . 2016, p. 153 pp.
- M. Biava, R. Fioravanti, G. C. Porretta, G. Sleiter, D. Deidda, G. Lampis, R. Pompei, Med. Chem. Res. 1999, 9, 19-34.
- [4] Q. Gou, Y.-W. Yang, Z.-N. Liu, J. Qin, Chem. Eur. J. 2016, 22, 16057-16061.
- C. Peifer, R. Selig, K. Kinkel, D. Ott, F. Totzke, C. Schaechtele, R. [5] Heidenreich, M. Roecken, D. Schollmeyer, S. Laufer, J. Med. Chem. 2008, 51, 3814-3824.
- (a) Z.-G. Le, Z.-C. Chen, Y. Hu, Q.-G. Zheng, Synthesis 2004, 1951-1954; (b) X.-J. Wang, K. Sidhu, L. Zhang, S. Campbell, N. Haddad, D. C. Reeves, D. Krishnamurthy, C. H. Senanayake, Org. Lett. 2009, 11, 5490-5493; (c) Z.-G. Le, Z.-C. Chen, Y. Hu, Q.-G. Zheng, Synthesis 2004, 208-212
- (a) F. Kakiuchi, T. Kochi, Synthesis 2008, 3013-3039; (b) L. Ackermann, R. Vicente, A. R. Kapdi, Angew. Chem. Int. Ed. 2009, 48, 9792-9826; (c) T. Satoh, M. Miura, Synthesis 2010, 3395-3409; (d) B.-J. Li, Z.-J. Shi, Chem. Soc. Rev. 2012, 41, 5588-5598; (e) P. B. Arockiam, C. Bruneau, Chem. Soc. Rev. 2012, 41, 3536-3536, (e) F. B. Alockalli, G. Büllead, P. H. Dixneuf, Chem. Rev. 2012, 112, 5879-5918; (f) B. Li, P. H. Dixneuf, Chem. Soc. Rev. 2013, 42, 5744-5767; (g) M. Zhang, Y. Zhang, X. Jie, H. Zhao, G. Li, W. Su, Org. Chem. Front. 2014, 1, 843-895. (a) F. Collet, R. H. Dodd, P. Dauban, Chem. Commun. 2009, 5061-5074;
- (b) K. Shin, H. Kim, S. Chang, Acc. Chem. Res. 2015, 48, 1040-1052; (c) Kim, S. Chang, Chem. 10.1021/acs.chemrev.1026b00644.
- (a) H. J. Kim, M. J. Ajitha, Y. Lee, J. Ryu, J. Kim, Y. Lee, Y. Jung, S. Chang, J. Am. Chem. Soc. **2014**, 136, 1132-1140; (b) S. H. Park, J. Kwak, K. Shin, J. Ryu, Y. Park, S. Chang, J. Am. Chem. Soc. **2014**, 136, 2492-2502; (c) J. Peng, Z. Xie, M. Chen, J. Wang, Q. Zhu, Org. Lett. **2014**, 16, 4702-4705; (d) P. Patel, S. Chang, ACS Catal. **2015**, 5, 853-858; (e) M. A. Ali, X. Yao, G. Li, H. Lu, Org. Lett. 2016, 18, 1386-1389; (f) I. T. Alt, B. Plietker, Angew. Chem. Int. Ed. 2016, 55, 1519-1522; (g) H. Kim, G. Park, J. Park, S. Chang, ACS Catal. 2016, 6, 5922-5929; (h) J. Wippich, N. Truchan, T. Bach, Adv. Synth. Catal. 2016, 358, 2083-2087; (i) T. Zhang,

- X. Hu, Z. Wang, T. Yang, H. Sun, G. Li, H. Lu, Chem. Eur. J. 2016, 22,
- (a) F. Mo, J. M. Yan, D. Qiu, F. Li, Y. Zhang, J. Wang, *Angew. Chem. Int. Ed.* **2010**, *49*, 2028-2032; (b) C. Grohmann, H. Wang, F. Glorius, *Org.* Lett. 2012, 14, 656-659; (c) M. Shang, S.-H. Zeng, S.-Z. Sun, H.-X. Dai, J.-Q. Yu, Org. Lett. 2013, 15, 5286-5289; (d) H. Kim, T. Kim, D. G. Lee, S. W. Roh, C. Lee, *Chem. Commun.* **2014**, *50*, 9273-9276; (e) F.-N. Ng, Z. Zhou, W.-Y. Yu, *Chem. Eur. J.* **2014**, *20*, 4474-4480; (f) X.-H. Hu, X.-F. Yang, T.-P. Loh, *ACS Catal.* **2016**, *6*, 5930-5934; (g) W. Xie, J. H. Yoon, S. Chang, J. Am. Chem. Soc. 2016, 138, 12605-12614.
- (a) Y. Tan, J. F. Hartwig, *J. Am. Chem. Soc.* **2010**, *132*, 3676-3677; (b) N. J. Race, J. F. Bower, *Org. Lett.* **2013**, *15*, 4616-4619; (c) S. Yu, B. Wan, X. Li, Org. Lett. 2013, 15, 3706-3709; (d) J. He, T. Shigenari, J.-Q. Yu, Angew. Chem. Int. Ed. 2015, 54, 6545-6549; (e) K. Muralirajan, R. Kuppusamy, S. Prakash, C.-H. Cheng, Adv. Synth. Catal. 2016, 358, 774-783; (f) G. C. Senadi, T.-Y. Lu, G. K. Dhandabani, J.-J. Wang, Org. Lett. 2017, 19, 1172-1175.
- (a) H.-Y. Thu, W.-Y. Yu, C.-M. Che, J. Am. Chem. Soc. 2006, 128, 9048-9049; (b) T.-S. Mei, X. Wang, J.-Q. Yu, *J. Am. Chem. Soc.* **2009**, *131*, 10806-10807; (c) E. J. Yoo, S. Ma, T.-S. Mei, K. S. L. Chan, J.-Q. Yu, *J.*
- Am. Chem. Soc. 2011, 133, 7652-7655.
  (a) M.-L. Louillat, F. W. Patureau, Org. Lett. 2013, 15, 164-167; (b) M.-L. Louillat, A. Biafora, F. Legros, F. W. Patureau, Angew. Chem. Int. Ed. 2014, 53, 3505-3509; (c) S. Kaur, M. Kumar, V. Bhalla, Green Chem. **2016**, *18*, 5870-5883; (d) L. Niu, H. Yi, S. Wang, T. Liu, J. Liu, A. Lei, *Nat.* Commun. 2017, 8, 14226.
- C. B. Bheeter, L. Chen, J.-F. Soulé, H. Doucet, Catal. Sci. Technol. 2016, 6, 2005-2049.
- (a) M.-L. Louillat, F. W. Patureau, Chem. Soc. Rev. 2014, 43, 901-910; (b) J. Jiao, K. Murakami, K. Itami, *ACS Catal.* **2016**, *6*, 610-633; (c) P. Subramanian, G. C. Rudolf, K. P. Kaliappan, *Chem. Asian J.* **2016**, *11*, 168-192; (d) Y. Zhou, J. Yuan, Q. Yang, Q. Xiao, Y. Peng, ChemCatChem 2016, 8, 2178-2192.
- (a) R. H. Crabtree, Dalton Trans. 2001, 2437-2450; (b) Á. Iglesias, R. Álvarez, Á. R. de Lera, K. Muñiz, Angew. Chem. Int. Ed. 2012, 51, 2225-
- 2228; (c) G. Pandey, R. Laha, *Angew. Chem. Int. Ed.* **2015**, *54*, 14875-14879; (d) H. Kim, S. Chang, *ACS Catal.* **2016**, *6*, 2341-2351. (a) C.-J. Li, *Acc. Chem. Res.* **2009**, *42*, 335-344; (b) W.-J. Yoo, C.-J. Li, *Top. Curr. Chem.* **2010**, 281-302; (c) S. H. Cho, J. Y. Kim, J. Kwak, S. Chang, Chem. Soc. Rev. 2011, 40, 5068-5083; (d) I. B. Krylov, V. A. Vil', A. O. Terent'ev, Beilstein J. Org. Chem. 2015, 11, 92-146
- (a) S. Wiese, Y. M. Badiei, R. T. Gephart, S. Mossin, M. S. Varonka, M. M. Melzer, K. Meyer, T. R. Cundari, T. H. Warren, Angew. Chem. Int. Ed. 2010, 49, 8850-8855; (b) R. T. Gephart, D. L. Huang, M. J. B. Aguila, G. Schmidt, A. Shahu, T. H. Warren, *Angew. Chem. Int. Ed.* **2012**, *51*, 6488-6492; (c) E. S. Jang, C. L. McMullin, M. Kass, K. Meyer, T. R. Cundari, T. H. Warren, *J. Am. Chem. Soc.* **2014**, *136*, 10930-10940.
- (a) B. L. Tran, B. Li, M. Driess, J. F. Hartwig, *J. Am. Chem. Soc.* **2014**, *136*, 2555-2563; (b) L. Zhou, S. Tang, X. Qi, C. Lin, K. Liu, C. Liu, Y. Lan, A. Lei, *Org. Lett.* **2014**, *16*, 3404-3407; (c) B. L. Tran, M. Driess, J. F. Hartwig, *J. Am. Chem. Soc.* **2014**, *136*, 17292-17301; (d) F. Teng, S. Sun, Y. Jiang, J.-T. Yu, J. Cheng, *Chem. Commun.* **2015**, *51*, 5902-5905; (e) H.-T. Zeng, J.-M. Huang, *Org. Lett.* **2015**, *17*, 4276-4279; (f) Y. Li, F. Zhu, Z. Wang, X.-F. Wu, *ACS Catal.* **2016**, *6*, 5561-5564.
- L.-K. Jin, L. Wan, J. Feng, C. Cai, *Org. Lett.* **2015**, *17*, 4726-4729. J. Xiu, W. Yi, *Catal. Sci. Technol.* **2016**, *6*, 998-1002.
- R. Xia, H.-Y. Niu, G.-R. Qu, H.-M. Guo, Org. Lett. 2012, 14, 5546-5549.
- (a) T. Eicher, S. Hauptmann, The Chemistry of Heterocycles: Structure, Reactions, Syntheses, and Applications, 2nd ed., Wiley-VCH, New York, 2003; (b) L. D. Quin, J. A. Tyrell, Fundamentals of heterocyclic chemistry: importance in nature and in the synthesis of pharmaceuticals, Wiley, New York, **2010**.
- Q. Yang, P. Y. Choy, W. C. Fu, B. Fan, F. Y. Kwong, J. Org. Chem. 2015, 80, 11193-11199

#### Layout 2:

### **FULL PAPER**

A variety of *N*-alkylated heterocycles is facilely synthesized by direct C-H bond amination of low cost (cyclo)alkanes catalyzed by inexpensive copper salts. A broad *NH*-free heterocycles scope has been used as amine sources and the reaction displayed a good functional group compatibility.

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Copper-Catalyzed Oxidative Dehydrogenative C(sp3)-H Bond Amination of (Cyclo)Alkanes using NH-Free Heterocycles as Amine Sources