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1	Tuning the Nonlinear Optical Properties of BODIPYs by
2	Functionalization with Dimethylaminostyryl Substituents
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24 Abstract

Two difluoroboradiazaindacene (BODIPY) scaffolds with attached dimethylaminostyryl 25 26 substituents were synthesized. Guest-host polymeric films were produced by incorporating these chromophores into polymethylmethacrylate matrices. The second and third nonlinear optical 27 properties of the resulting polymer composites were studied by means of the Maker fringe 28 29 technique using a laser generating at 1064 nm with a 30 ps pulse duration. The macroscopic and 30 microscopic nonlinearities were found to be comparatively high and dependent on the number of 31 dimethylaminostyryl substituents attached to BODIPY core. The development of integrated 32 optics makes such nonlinear films of particular interest, since they can be used in the creation of 33 efficient nonlinear devices.

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35 Keywords: BODIPY dye, polymer composites, second/third harmonic generation, nonlinear
 36 susceptibility, hyperpolarizability

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38 **1. Introduction**

In recent years, difluoroboradiazaindacene (BODIPY) unit has attracted much interest because of valuable spectroscopic and photophysical properties such as its high molar absorptivity coefficient [1], high fluorescence quantum yield [2,3] and good stability towards light and chemicals [4]. Recently many BODIPY derivatives have been synthesized and utilized as fluorescent dyes [5], chemosensors [6-10], polarity sensitive probe [11], light harvesting systems [12,13], and for laser applications [14] and photodynamic therapy [15,16].

This research work is devoted to nonlinear optical (NLO) investigations of BODIPY model 45 46 compounds by means of SHG and THG (second/third harmonic generation) techniques, which 47 allow the determination of second and third order NLO parameters. Until now a small set of 48 BODIPY derivatives have been investigated for nonlinear optics [17-23]. Recently, much 49 attention has been devoted to functionalized BODIPY dyes that exhibit two-photon absorption 50 (TPA) properties due to their related application in TPA imaging [24-26]. To this end, extending 51 conjugation and forming Intramolecular Charge Transfer (ICT) by means of appropriate electron 52 donor or acceptor groups are effective strategies for the synthesis of highly efficient two-photon 53 active BODIPY compounds. Thus, understanding how incorporating electron donor and/or acceptor groups in the BODIPY architecture affect NLO response remains a challenge. 54

Our research work is focused on two BODIPY-based models which have one or two 55 dimethylaminostyryl branches, with high ICT strength from the dialkylamino-groups electron 56 57 donors to the electron acceptor boradipyrrin core. Interestingly, B1 and B2 models not only present intrinsic "push-pull" character, but also differ by their inherent D-A (B1) and D-A-D (B2) 58 59 structures that potentially govern distinct second and third order optical nonlinearities. Recently, 60 we have shown that dimethylaminostyryl arms affect the nonlinear refraction and absorption activity of the BODIPY structure [27]. Meanwhile, the SHG and THG study of 61 dimethylaminostyryl substituted BODIPY-based derivatives has not been done yet, so the results 62 63 of this investigation may extend the application range of the dye. The second and third nonlinear 64 optical properties of BODIPY-based materials, as guest-host polymeric films produced by 65 incorporating the chromophores into PMMA matrices, were studied by means of the Maker 66 fringe technique. The NLO susceptibilities and hyperpolarizabilities were extracted, which can give a better understanding on the origin of NLO phenomena and the structure-property 67 relationships in BODIPY chromophores. 68

69 **2. Experimental section**

70 2.1. Materials

The BODIPY models have been synthesized in good yield by the established Konoevenagel condensation of 3,5 dimethyl-BODIPY dyes with 4-dimethylaminobenzaldehyde, as it is previously described by Rurack [28] and Akkaya [29]. The structures of the resulting difluoroboradiazaindacene chromophores with dimethylaminostyryl substituents (namely B1 and B2) are shown in Fig. 1. The monocondensation was controlled by adjusting the mole ratio and stopping the reaction after few hours, while a larger excess of 4-dimethylaminobenzaldehyde and a longer reaction time increased the yield of by-condensed product B2 (Supporting Information).

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Fig. 1. Structures of the conjugated difluoroboradiazaindacene chromophores with
dimethylaminostyryl substituents: a) B1; b) B2.

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83 2.2. Film fabrication

For the polymeric film preparation a solution of PMMA (Sigma-Aldrich, Mw=350,000 g/mol) dissolved in 1,1,2-trichloroethane at concentration of 50 g/L was used as a host system.

The concentration of the compounds was 100 µmol towards 1 g of PMMA. The glass plates were washed in distilled water using ultrasonic bath, acetone, and ethanol and then dried. The solutions were deposited on glass substrates using the spin-coater (SCS G3) at 1000 rpm. Obtained guesthost polymer films were kept at room temperature during few days in order to eliminate any remaining of solvent. The thickness of deposited films was measured with the profilometer (Dektak 6M, Veeco) to be about 700 nm.

92 2.3. Optical absorption measurements

Absorption spectra of the B1 and B2 films were measured by means of Lambda 950
UV/Vis/NIR spectrophotometer (PerkinElmer) in the range 300-1200 nm. The pure PMMA film
on glass plate was used on the way of reference beam in spectrophotometer for the measurements
of absorption spectra of chromophores.

97 2.4. SHG and THG measurements

98 The phenomena of second and third harmonic generation (SHG and THG) may take place in 99 a material the nonlinear polarization of which, caused by intensive laser irradiation, stimulates the 100 coherent light emission at double and triple laser frequencies. The schematically energy-level 101 diagram of SHG and THG processes is shown in Fig. 2a. SHG and THG measurements were 102 performed by means of the rotational Maker fringe technique [30] in the transmission scheme for 103 the s- and p-polarized fundamental laser beams (Fig. 2b). A y-cut crystalline quartz plate has been 104 used as a reference material for SHG measurements and fused silica plate for THG 105 measurements. The output beam of a mode-locked Nd:YAG/YVO₄ laser (EKSPLA) generating 106 at λ =1064 nm with 30 ps pulse duration and 10 Hz repetition rate was employed as a fundamental 107 beam. The input energy of laser pulses was controlled by laser power/energy meter (LabMax

- 108 TOP, COHERENT) to be 90 µJ for SHG and 150 µJ for THG measurements. The detailed setups
- 109 description can be found elsewhere [31].

110



Fig. 2. a) Energy-level diagram of SHG and THG processes and b) the illustration of input-output
polarization scheme in Maker fringe experiment.

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115 **3. Results and discussion**

116 3.1. Optical absorption properties

117 The absorption spectra of B1 and B2 compounds embedded in PMMA films at the 118 concentration 100 µmol/g are given in Fig. 3. As expected, attaching the electron donating 119 dimethylamino groups at the 3 and 5 positions, through a conjugated bridge, leads to considerable 120 shift of the absorption band towards the red region, by comparison with the parent BODIPY 121 compound [2, 20]: UV-Vis (CH₂Cl₂) $\lambda_{max}(abs.) = 598$ nm (B1), 692 nm (B2). The UV-Vis 122 absorption and fluorescent emission spectra of the THF solutions displayed the maximal absorbance at $\lambda_{max}(abs.) = 601$ nm (B1), 694 nm (B2) with a molar extinction coefficient of about 123 90,000 M⁻¹ cm⁻¹ and the maximal emission at λ_{max} (em.) = 643 nm (B1), 722 nm (B2) with the 124 125 fluorescence quantum yield of 0.24 and 0.15 for B1 and B2, respectively [27]. Similarly, the 126 spectra of both films exhibit wide absorption bands with maximum at 605 and 703 nm for B1 and

B2, respectively, which correspond to $S_0 \rightarrow S_1$ (π - π^*) transitions. The short-wavelength vibronic shoulder of these bands can be assigned to a C-H out-of-plane vibration [32]. The absorption peaks at the wavelength less than 500 nm could be attributed to a partially forbidden $S_0 \rightarrow S_2$ (π -130 π^*) transitions in BODIPY core [33]. As the absorption spectra indicates, introduction of second electron-donor dimethylaminostyryl group exerts almost 100 nm red-shift of absorption band without considerable change of its intensity. At the wavelengths of more than 800 nm the samples show high optical transparency.





135

Fig. 3. UV-vis absorption spectra of the B1 and B2 compounds embedded in PMMA films. In
inset – the photos of the deposited films.

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139 *3.2. SHG response study*

The rotational Maker fringe technique was implemented for the measurements of SHG in B1 and B2 guest-host polymeric films for *s*- and *p*-polarized fundamental laser beam. Since the dipole moments of the chromophores are initially randomly oriented within the film, the second order nonlinear optical response of the film is negligible due to its macroscopic centrosymmetry. In order to break the centrosymmetry and induce the uniaxial orientation of NLO chromophores

145 in the polymeric films, provoking their manifestation of second order NLO behavior, the samples 146 were corona poled just before SHG measurements. The poling temperature was 95°C, a little 147 lower than the glass transition temperature and applied electric field was 5 kV/cm. In Fig. 4 the 148 dependences of the second harmonic intensity generated as the function of incident angle are 149 presented. The obtained angular dependencies are typical for induced $C_{\infty p}$ symmetry view with a 150 maximum signal at 60°-65° and zero intensity at normal incidence of the fundamental beam. The 151 polarization of the second harmonic signal was found to be always *p*-polarized regardless of the incident beam polarization due to the peculiarities of symmetry dependent second order 152 153 susceptibility tensor components.

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155

Fig. 4. SHG intensity as a function of incident angle in B1 and B2 guest-host films at the *s*-, and*p*-polarized fundamental beam.

158

As it can be seen for *s*-*p* or *p*-*p* polarizations, the intensity of SHG from B2 compound which has two dimethylaminostyryl units is almost two-to-three times higher than the intensity of B1 with one dimethylaminostyryl unit. In contrast to B1 film, the SHG response from B2 film for *s*-*p* (input-output) polarization is higher than for *p*-*p* polarization. For B1 the higher SHG signal can 163 be obtained when the polarization of incident beam is close to parallel to its dipole moment or 164 direction between donor (D) amine group and acceptor (A) BODIPY core, meanwhile the higher 165 SHG signal in B2 can be obtained when the polarization of incident beam is rather perpendicular 166 to D-A direction. In the Fig. 5 the dependences of SHG intensity on polarization of incident beam measured at 60° are presented, which confirm the contrary directions in the molecules where 167 168 highest response occurs. The second order nonlinearity of BODIPY with attached 169 dimethylaminostyryl substituents is more efficient in D-A-D architecture (B2), in which two ICT 170 processes contribute to the NLO response, than in D-A (B1).





172

173 Fig. 5. The dependences of SHG intensity on incident beam polarization in B1 and B2 films.174

Since the films have the thickness much less than their coherence length, the intensities of SHG from films were compared to that from a quartz plate and the quadratic NLO susceptibilities were calculated using following equation taking into account linear optical absorption [34]:

179
$$\chi^{(2)} = \chi^{(2)}_{Quartz} \left(\frac{2}{\pi}\right) \left(\frac{l^{coh}_{Quartz}}{d}\right) \left(\frac{\frac{ad}{2}}{1 - e^{-\frac{ad}{2}}}\right) \left(\frac{l^{2\omega}}{l^{2\omega}_{Quartz}}\right)^{1/2},\tag{1}$$

180

where $\chi^{(2)}_{Quartz}$ is the quadratic susceptibility of quartz, l^{coh}_{Quartz} is the coherent length of quartz, α 181 is the absorption coefficient at doubled laser wavelength, d is the film thickness, $I^{2\omega}$ and $I^{2\omega}_{ouartz}$ 182 are the SHG intensities from the sample and quartz under the same conditions of measurement, 183 184 respectively. The obtained effective quadratic NLO susceptibilities are presented in Table 1. Here again, the effective values of $\chi^{(2)}$ for B2 compound were found to be higher than that for B1 due 185 186 to the doubled amount of ICT which plays a crucial role in nonlinear behavior of a molecule. The 187 obtained values of quadratic susceptibility are quite high and comparable with those which we 188 have recently obtained for spin-deposited thin films of azobenzene-based push-pull compounds 189 [35]. Meanwhile, in the microscopic approach, the first hyperpolarisabilities of B1 and B2 extracted form $\chi^{(2)}$ assuming parallel orientations of dipole moment of molecules towards polar 190 191 axis induced by the poling field are found to be one order of magnitude lower than for other 192 BODIPY-based derivatives [19].

193

Table 1. The values of $\chi^{(2)}_{eff}$, $\chi^{(3)}_{elec}$, β_{eff} and γ_{elec} obtained for B1 and B2.

Compound	$\chi^{(2)}_{eff}$, s-p	$\chi^{(2)}_{eff}, p-p$	$eta_{\it eff}$,	$\chi^{(3)}_{elec}, s-s$	Yelec,
Compound	$(pm V^{-1})$	$(pm V^{-1})$	$(10^{-38} \text{ m}^4 \text{V}^{-1})$	$(10^{-20} \text{ m}^2 \text{V}^{-2})$	$(10^{-47} \text{ m}^5 \text{V}^{-2})$
B1	2.75	3.61	1.86	0.32	1.16
B2	5.86	5.42	2.79	1.79	6.37
	T.				

195

196 *3.3. THG response study*

197 The measurements of THG response in B1 and B2 films were performed for *s-s* incident-198 generated polarization scheme (Fig. 6). It can be seen that B2 film represents much higher THG

response compared to B1 film. As the THG originates from the film together with the glass
substrate, the impact of substrate was taken into account during data processing and the cubic
NLO susceptibility calculation.

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203

Fig. 6. THG intensities from B1 and B2 films as a function of incident angle for the *s-s*polarization of the fundamental beam – generated signal.

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207 Third order NLO susceptibility was calculated using the comparative model which takes into208 account linear optical absorption [36]:

209

210
$$\chi^{(3)} = \chi^{(3)}_{Silica} \left(\frac{2}{\pi}\right) \left(\frac{l^{Coh}_{Silica}}{d}\right) \left(\frac{\frac{\alpha d}{2}}{1 - e^{-\frac{\alpha d}{2}}}\right) \left(\frac{l^{3\omega}}{l^{3\omega}_{Silica}}\right)^{1/2},\tag{2}$$

211

where $\chi_{Silica}^{(3)}$ is the cubic susceptibility of silica, l_{Silica}^{coh} is the coherent length of silica, α is the absorption coefficient at triple laser wavelength, *d* is the film thickness, $I^{3\omega}$ and $I_{Silica}^{3\omega}$ are the THG intensities of the samples and silica at the same conditions, respectively. The calculated values of $\chi^{(3)}$ for the B1 and B2 films are given in Table 1 and they are one-to-two orders of magnitude higher than for silica $(2 \cdot 10^{-20} \text{ m}^2 \text{V}^{-2})$. The compound B2 is characterized with a much higher third order NLO susceptibility than B1 due to more effective electron delocalization caused by charge transfers from the dimethylamino groups to the BODIPY center together with its D-A-D structure, contributing to the microscopic nonlinearity.

220 The electronic contribution of second hyperpolarizability γ_{elec} for B1 and B2 molecules was deducted from third order susceptibility (Table 1), as THG measurements are only dealing with 221 electronic constituent, taking into account the concentration of the molecule and the local field 222 223 factor [37, 38]. The comparatively higher γ_{elec} for B2 is attributed to the presence of two dimethylaminostyryl substituents which imparts a greater intramolecular charge-transfer 224 225 character compared to B1 compound. It is worth noting that except for the intra-molecular electronic contribution some role may also play vibrational contribution to hyperpolarizability of 226 227 the systems.

228

4. Conclusions

In summary, B1 and B2 BODIPY-based derivatives with one and two conjugated 230 231 dimethylaminostyryl substituents exhibit considerable second and third order NLO responses at 232 1064 nm, which can be tuned by the number of these peripheral substituents. In particular, the expansion of the π -conjugated system by two electron donor groups, leading to a D-A-D 233 234 structure, results in significant increase of both SHG and THG responses comparing to the D-A 235 structure. Attaching strong electron-donating groups on the BODIPY core can thus promote 236 charge transfer character in these compounds and enhance their second and third order NLO properties. Varying the electron donating efficiency and the respective position of donor and 237

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238	acceptor groups in BODIPY conjugated systems will be the object of further studies with the aim
239	to establish applicable structure-property relationships.
240	
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245	
246	Appendix A. Supplementary data
247	Supplementary data related to this article can be found at http://dx.doi.org/
248	
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BODIPY models with attached dimethylaminostyryl substituents were synthesized.

The nonlinear optical properties of functionalized BODIPY models were studied. The nonlinearities are dependent on the number of dimethylaminostyryl substituents. Electron-donating groups enhance second and third order NLO response of BODIPY.