

Defect formation during chlorine-based dry etching and their effects on the electronic and structural properties of InP/InAsP quantum wells

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Defect formation during chlorine-based dry etching and their effects on the

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electronic and structural properties of InP/InAsP quantum wells 2 Jean-Pierre Landesman^{a)} 3 Institut de Physique de Rennes, CNRS-UMR 6251, Université Rennes 1, F-35042 Rennes, France 4 Juan Jiménez 5 GdS Optronlab, Dpto. Fisica de la Materia Condensada, Universidad de Valladolid, 47011 Valladolid, Spain 6 Christophe Levallois UMR FOTON, CNRS, INSA-Rennes, 20 avenue des buttes de Coësmes, F-35708 Rennes, France 8 Frédéric Pommereau 9 III-V Lab, Route de Nozay, F-91461 Marcoussis, France 10 Cesare Frigeri 11 CNR-IMEM Istituto, Parco area delle Scienze 37/A, 43010 Parma, Italy Alfredo Torres 13 GdS Optronlab, Dpto. Fisica de la Materia Condensada, Universidad de Valladolid, 47011 Valladolid, Spain 14 Yoan Léger and Alexandre Beck 15 UMR FOTON, CNRS, INSA-Rennes, 20 avenue des buttes de Coësmes, F-35708 Rennes, France 16 Ahmed Rhallabi 17 Institut des Matériaux Jean-Rouxel, CNRS-UMR 6502, Université Nantes 1, F-44322 Nantes, France 18 (Received 27 April 2016; accepted 3 May 2016; published xx xx xxxx) 19 The general objective is the investigation of the defects formed by dry etching tools such as those 20 involved in the fabrication of photonic devices with III-V semiconductors. Emphasis is put on 21 plasma exposures with chlorine-based chemistries. In addition to identifying these defects and 22 describing their effects on the electro-optic and structural properties, the long-term target would be 23 to predict the impact on the parameters of importance for photonic devices, and possibly include 24 these predictions in their design. The work is first centered on explaining the experimental 25 methodology. This methodology starts with the design and growth of a quantum well structure on 26 indium phosphide, including ternary indium arsenide/phosphide quantum wells with graded 27 arsenic/phosphor composition. These samples have then been characterized by luminescence 28 methods (photo- and cathodoluminescence), high-resolution transmission electron microscopy, and 29 secondary ion mass spectrometry. As one of the parameters of importance in this study, the authors 30 have also included the doping level. The samples have been exposed to the etching plasmas 31 for "short" durations that do not remove completely the quantum wells, but change their optical 32 signature. No masking layer with lithographic features was involved as this work is purely oriented 33 to study the interaction between the plasma and the samples. A significant difference in the 34 luminescence spectra of the as-grown undoped and doped samples is observed. A mechanism 35 describing the effect of the built-in electric field appearing as a consequence of the doping profile 36 is proposed. This mechanism involves quantum confined Stark effect and electric-field induced 37 carrier escape from the quantum wells. In the following part, the effects of exposure to various 38 chlorine-based plasmas were explored. Differences are again observed between the undoped and 39 doped samples, especially for chemistries containing silicon tetrachloride. Secondary ion mass 40 spectrometry indicates penetration of chlorine in the structures. Transmission electron microscopy 41 is used to characterize the quantum well structure before and after plasma bombardment. By 42 examining carefully the luminescence spectral properties, the authors could demonstrate the 43 influence of the etching plasmas on the built-in electric field (in the case of doped samples), and 44 relate it to some ionic species penetrating the structures. Etching plasmas involving both chlorine 45 and nitrogen have also been studied. The etching rate for these chemistries is much slower than 46 for some of the silicon tetrachloride based chemistries. Their effects on the samples are also very 47 different, showing much reduced effect on the built-in electric field (for the doped samples), but 48 significant blue-shifts of the luminescence peaks that the authors attributed to the penetration of 49 nitrogen in the structures. Nitrogen, in interstitial locations, induces mechanical compressive stress 50 that accounts for the blue-shifts. Finally, from the comparison between secondary ion mass 51 spectrometry and luminescence spectra, the authors suggest some elements for a general 52

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53 mechanism involved in the etching by chloride-chemistries, in which a competition takes place 54 between the species at the surface, active for the etching mechanism, and the species that penetrate

55 the structure, lost for the etching process, but relevant in terms of impact on the electro-optic and

56 structural features of the exposed materials. © 2016 American Vacuum Society.

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57 I. INTRODUCTION

58 The semiconductor equipment industry has developed over the years a number of tools and processes which are 59 today used at very large scales in fabrication throughout the 60 world. Dry etching, also designated as reactive ion etching 61 (RIE) or plasma etching, is one of these tools for the fabrica-62 tion of ultralarge scale integrated circuits in the silicon 63 microelectronics industry as well as in the fabrication of 64 photonic integrated circuits with compound semiconductors 65 based on gallium arsenide (GaAs) or indium phosphide 66 (InP). It is used, in particular, to precisely define features 67 with submicron dimensions, for example, transistor gates in 68 the silicon world.¹ Very deep features, such as "through sili-69 con vias," are also realized using different RIE setups.² The 70 reactive source gases used in plasma etching of silicon are 71 usually molecules containing fluorine atoms, while chloride 72 or bromide based plasmas are mainly used for the etching of 73 III-V semiconductor compounds like GaAs or InP. 74

Plasma etching is considered a mature technology. 75 Nevertheless, research and development efforts on this 76 technique are going on in a large number of places, both aca-77 demic and industry centers, to adapt the processes and equip-78 79 ment to the challenges occurring constantly on the different roadmaps, both for silicon devices and III-V compound 80 semiconductor devices. For example, pushing the limits of 81 the "more Moore" approach for CMOS technologies requires 82 plasma etching processes allowing gate design for the next 83 technology nodes, etching of materials newly introduced 84 into the gates, better control of the gate dimensions, etc.³ As 85 far as compound semiconductor device technologies are con-86 cerned, the trend to reduce critical dimensions-especially 87 88 for plasma etching operations-seems less emphasized, the efforts being rather focused on developing plasma etching 89 processes applicable to a wide variety of materials which are 90 nowadays introduced in the devices. However, some devices 91 also require plasma etching processes which can generate 92 features with critical dimensions in the nanometer range 93 and/or aspect ratios difficult to achieve (good examples of 94 such devices are the optic or opto-electronic components 95 based on photonic crystals). 96

97 Developing plasma etching processes involves research98 in different directions:

- 99 (1) plasma physics and chemistry, as well as reactor technol-00 ogy (for the part related to design of equipment);
- 101 (2) study of the plasma/surface interaction for each specific102 material;
- (3) accurate metrology for the optimization of the expected fea-tures; this is mainly performed using electron microscopy;
- 105 (4) materials and device physics to check the overall compat-
- ¹⁰⁶ ibility of the plasma process with device performances.

In this paper, we explore the last direction, with a focus 107 on plasma etching of InP-based structures. We investigate 108 the effects of dry etching on the etched materials. We chose 109 processes commonly used in the photonic device industry 110 for etching InP and related materials (in our case, the reac- 111 tive source gases are chlorine-based mixtures with either 112 $SiCl_4$ or Cl_2). In the present stage of the study, our goal is to 113 identify the mechanisms by which the reactive gases interact 114 with the material. No specific geometric feature is designed 115 at this stage of the study. We simply expose the material to 116 the gases, under the conditions typically occurring during 117 plasma etching, and thereafter, we analyze the changes 118 induced in the materials using different experimental techni- 119 ques. Therefore, the issues concerning the etching mask 120 materials and design are not considered, but will be the 121 object of forthcoming work. 122

The methodology involves the design of heterostructures 123 which incorporate specific features [quantum wells (QWs)], 124 role of which is to provide an in-depth signature of the 125 effects induced by the etching gases, through changes in 126 their luminescence spectral parameters. This is an approach 127 used for a long time by groups trying to understand the 128 effects of energetic ions or reactive radicals on compound 129 semiconductors. The first studies on this topic were reported 130 for GaAs and InP.^{4,5} These pioneer studies dealt with the 131 effects of ion bombardment (mainly rare gas ions) on sam- 132 ples with AlGaAs QWs in GaAs or GaInAsP QWs in InP. It 133 was demonstrated that under the effect of ion bombardment 134 even at low energy (400–500 eV or even lower), defects are 135 produced at depths in the order of 100 to a few hundred 136 nanometers. Channeling was proposed as the mechanism 137 generating these defects.⁶ Even if the ion bombardment 138 experiment is not performed along directions where channel- 139 ing is expected [e.g., etching (100) surfaces of GaAs or InP 140 does not involve ion bombardment along a major channeling 141 direction in these crystals], the fact that a very small fraction 142 of the incident ions end-up travelling the crystal along other 143 directions such as (110) can lead to this depth of penetration, 144 and therefore promote the creation of defects across the 145 structure. In these early studies, occurrence of defects at a 146 specific depth below the surface is assumed from the reduc- 147 tion of photoluminescence (PL) efficiency of the QW located 148 at such specific depth. From these investigations, the authors 149 showed that ion bombardment experiments simulating the 150 conditions occurring during plasma etching affect the mate- 151 rials on depths much larger than 100 nm, even when the ions 152 energy lies between 100 and 500 eV. However, studying the 153 effects of rare gas ions does not allow for the extrapolation 154 to the case of industrial plasma etching processes. At least 155 one paper reports on similar studies involving chlorine 156 atomic and molecular ions derived from the decomposition 157

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scheme expected for $SiCl_4$ inside a RIE reactor.⁷ Again, the decrease in PL yield from the QW is seen to extend deeper than 100 nm.

Similar experiments could not be carried out for the etch-161 ing of Si, since there is no simple QW structure (with an effi-162 cient PL) that could be grown by epitaxy on Si. However, a 163 fundamental paper was published by Winters and collabora-164 tors,⁸ where the authors demonstrated the penetration of flu-165 orine into the Si lattice under conditions that simulate the 166 exposure to etching plasmas. The measurements, performed 167 using x-ray photoelectron spectroscopy, demonstrated a pen-168 etration depth of at least 20 nm, but the actual depth of pene-169 tration could be larger (the technique used in this study does 170 171 not allow the probing of larger depths).

Our aim in this study is to establish a scheme for the inter-172 action of the ions penetrating during the process of exposure 173 to the RIE plasma and the semiconductor materials under 174 consideration (InP/InAsP). It is clear from the published 175 176 results that this interaction leads to the creation of defects accounting for the reduction of the PL yield. However, the 177 detailed mechanisms have not yet been described. The pene-178 trating ions can have an impact on the electronic properties, 179 180 both the charge carrier lifetime and the radiative recombination rate. The approach followed consists in analyzing all the 181 PL spectral properties affected by the RIE process, and to 182 use samples with similar design of the QW structure but dif-183 184 ferent doping profiles (in order to see if the presence of internal built-in electric fields affects the response to the RIE 185 186 plasma). Additional characterizations are used to complete the investigations: secondary ion mass spectrometry (SIMS) 187 to probe directly the penetration of chlorine and transmission 188 189 electron microscopy (TEM) to assess accurately the etch depths through the QW structure and also to check the crys-190 tal and QW interfaces quality. In a preliminary paper,⁹ 191 reporting our first results for exposure of the InP/InAsP 192 samples to different chlorine-based plasmas, we have shown 193 evidence of the chlorine penetration in parallel with signifi-194 195 cant modifications of the PL and cathodoluminescence (CL) spectra. The present paper describes the extensive study that 196 we have performed along the general objective stated above. 197

198 II. EXPERIMENT

We first describe here the design and growth of the InP/ InAsP heterostructure samples. In a second part, we explain the conditions for the RIE process. Finally, we give the necessary details on the conditions which have been used for the examination of the samples before and after exposure to the RIE plasmas.

205 A. Sample structure and growth

The QW structures were grown on n-doped (100) InP substrates (doping level: 10^{18} cm⁻³). The whole structure contains nine QWs. QW thickness is constant for each sample (7–10 nm). The QWs are made of InAs_xP_{1-x} (with InP barriers), with As composition "x" graded along the QW series. In order to avoid reabsorption of the PL, the QW with the smallest energy bandgap (highest As content) was grown deepest—i.e., closest to the substrate—while the QW with 213 the highest As content was grown closest to the sample sur-214 face. Typical values for x (As composition) are between 215 0.35 (QW closest to the surface) and 0.49 (deepest QW). A 216 buffer layer (400 nm InP) is first grown prior the QW series. 217 The InP barrier thickness is 100 nm. An InP cap (300 nm) is 218 finally grown after the last QW. The different QWs will be 219 denoted as QW 1 to QW 9 (from surface to bulk). Therefore, 220 the bandgap energy decreases from QW 1 to QW 9. 221

Figure 1 gives a schematic picture of this structure.

The samples were grown by gas source molecular beam 223 epitaxy, using AsH_3 and PH_3 precursors, and a solid source 224 for In. The layers could be grown either nonintentionally 225 doped, or with Si or Be dopants for n-type and p-type, 226 respectively. Two different samples will be investigated 227 here: 228

- (1) Sample 1 was grown without intentional doping.
- (2) Sample 2 was doped: n-type $(1 \times 10^{18} \text{ cm}^{-3})$ for the InP 230 buffer layer, and p-type $(7 \times 10^{17} \text{ cm}^{-3})$ for the InP bar-231 rier and cap layers. QWs were not intentionally doped. 232

B. Reactive ion etching

Commercial inductively coupled plasma RIE systems 234 were used. These reactors have two RF power sources: one 235 for the plasma excitation, and one for the control of the bias 236 at the substrate.¹⁰ For the present study, we focused on dif-237 ferent etching processes available at III–V Lab for the fabri-238 cation of InP-based photonic devices. No specific process 239 was designed for the experiments carried here. Three differ-240 ent etching chemistries were investigated, with the parame-241 ters described in Table I.

Etching rates differ strongly from one process to the other 243 and will be estimated from the SIMS measurements and 244 TEM images. The first process (SiCl₄/H₂/Ar) produces an 245 etch rate in the order of 500 nm/min. The Cl₂/N₂ process 246 with the conditions above was chosen for its very low etch 247



InP buffer (d = 400 nm) + substrate

Fig. 1. QW structure used for the study. *x* denotes the As composition for each QW, *e* the QW thickness, and λ its corresponding PL wavelength.

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TABLE I. Plasma parameters used for the three different etching chemistries investigated.

| Etching chemistry | Gas flow rates | Total pressure | RF power plasma excitation | RF power sample stage | Sample bias |
|---------------------------------------|---|----------------|----------------------------|-----------------------|-------------|
| SiCl ₄ /H ₂ /Ar | 5 sccm SiCl ₄ /2 sccm H ₂ /10 sccm Ar | 5 mTorr | 1200 W | 250 W | 60 V |
| Cl ₂ /N ₂ | 3 sccm Cl ₂ /20 sccm N ₂ | 5 mTorr | 900 W | 25 W | 30 V |
| SiCl ₄ /Ar | 4 sccm SiCl ₄ /5 sccm Ar | 5–10 mTorr | 1000–1200 W | 110–350 W | 40–90 V |

248 rate. The third process (SiCl₄/Ar) was adjusted to provide 249 different etch rates with the same chemistry.

The temperature at sample surface during etching was 250 estimated using temperature labels. This estimation showed 251 that the surface temperature is always below 200 °C for the 252 253 conditions of the present study. Care was taken to wait long enough for stationary operation of the reactor during the 254 temperature estimates. Such temperature rise is far from the 255 threshold for atom diffusion in InP-based materials. 256 Therefore, no cooling of the samples was provided. 257

258 C. Photoluminescence and cathodoluminescence

The PL experiments were performed at 15K, with 259 AQ2 260 1064 nm YAG laser excitation. This energy is below the band gap of InP substrate and barriers (thus avoiding excita-261 262 tion in these parts of the structure). Excitation power density was in the range of few hundreds to 1000 W/cm^2 . The stand-263 ard excitation power density for the spectra presented in this 264 paper is 500 W/cm². This power density is slightly higher 265 than standard PL excitation ($\sim 100 \text{ W/cm}^2$), but since the 266 barriers are transparent at the pump wavelength, only a small 267 fraction of this power is absorbed by the nine QWs. 268 Furthermore, we checked that all the PL experiments have 269 been conducted in a linear absorption regime and at a low 270 271 excitation power.

For the CL measurements, we used a Carl Zeiss (LEO 272 1530) field emission scanning electron microscope, equipped 273 with a Gatan mono-CL2 system with a cooled nitrogen 274 275 InGaAs array detector. The acceleration voltage for the electrons was set to 20 kV [which gives a penetration depth for 276 InP in the order of $1 \mu m$ (Ref. 11)]. The measurement 277 temperature for CL spectra shown in this paper is systemati-278 cally 80 K. 279

D. Transmission electron microscopy 280

TEM observations were made in a field emission gun 281 JEOL 2200FS machine working at 200 keV under different 282 operation modes: 283

- (1) High resolution TEM (HRTEM). This is the usual mode 284 to get atomic resolution of the sample, allowing, for 285 example, the investigation of the QW interfaces quality. 286
- (2) High angle annular dark field (HAADF) which, in asso-287 ciation with the scanning operation mode of the TEM 288 (STEM), allows getting information on the composition 289 of each layer, thanks to the Z^2 dependence of the 290 HAADF intensity, with Z the atomic number.¹² 291
- 292 (3) High resolution scanning TEM, where atomic resolution 293 can in principle be coupled with contrast related to the local chemical composition. 294

Sample preparation for the TEM observations was done 295 by standard thinning by Ar⁺ ion bombardment of cross 296 sectional samples, whose thickness had been previously 297 reduced by mechanical grinding. To avoid decomposition of 298 the samples as much as possible, these were kept at liquid 299 nitrogen temperature during the Ar⁺ ion thinning process. 300

E. Secondary ion mass spectrometry

SIMS profiles were produced by Cs^+ primary ion bom- 302 bardment at 16 keV. The secondary species detected were Cl 303 (mass 35), P_2 (mass 31), and As (mass 75). We did not 304 perform any kind of calibration to evaluate absolute atomic 305 concentrations. However, quite reproducible signals for P 306 and As were obtained from sample to sample (around 6% 307 reproducibility for the P₂ signal from sample to sample, and 308 15% for the As signal). Based on the assumption that count 309 rates with a similar magnitude are measured for the As and P 310 profiles in different samples, we took the raw data for the Cl 311 profile as a good indication of the actual Cl concentration 312 profile, without performing any calibration. As for the depth 313 scales and depth resolution of the SIMS profiles, no calibra- 314 tion was required since the samples include their own inter- 315 nal scales consisting in the series of nine InAsP QWs whose 316 depth is known. The SIMS profiles are given as a function of 317 sputter time by Cs⁺ ions. The conditions for ion sputtering 318 (ion energy, ion beam current) were kept constant for all the 319 measured profiles. This ensures the possibility to directly 320 compare the secondary ion signals from the different sam- 321 ples, assuming that the sputter yields for the different ele- 322 ments do not change from each other sample. 323

III. RESULTS

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A. As-grown samples

Figure 2 shows the PL spectra recorded for the as-grown 326 samples. The spectra for the two samples look very different, 327 the PL lines being much broader on the doped sample. This 328 is not related to the crystal quality, e.g., to the more or less 329 abrupt character of the QW interfaces. The crystal quality 330 was checked by the TEM observations, in particular high 331 resolution TEM as illustrated on Fig. 3. 332

The CL spectra for the two samples are shown on Fig. 4. 333 They display the same trend as the PL spectra, the difference 334 in the shape of the luminescence lines being even more pro- 335 nounced for the CL. In fact, on the CL spectrum of the doped 336 sample [Fig. 4(b)], one can only distinguish eight peaks. 337 Since the QWs themselves are not doped, we do not have a 338 straightforward explanation for the observed difference. To 339 investigate this, we performed some luminescence measure- 340 ments on the samples under an electric bias. Our PL setup 341

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FIG. 2. PL spectra measured at 15 K with 1064 nm laser excitation of the asgrown samples: (a) Undoped sample and (b) doped sample.

did not allow biasing the sample; therefore, these measurements were performed with the CL setup only. The idea
behind this investigation was to check the influence of the
built-in electric field present in the samples (due to the doping profile) on the luminescence line shape. The electric field
can induce a quantum confined Stark effect (QCSE) and can
also move the photogenerated carriers away from the QW

where they are generated.¹³ These can strongly affect the PL ³⁴⁹ line shape. QCSE can produce red-shift to longer wavelengths, lowering of the intensity, and spectral broadening.¹³ ³⁵¹ Electric field-induced tunneling of the photogenerated carriers out of the QWs induces quenching of the PL.¹³ ³⁵³

In order to probe these two effects (QCSE and luminescence quenching by electric field-induced tunneling), it is 355 necessary, in principle, to start from the flat-band situation;¹³ 356 however, the band bending of our samples is complex: 357

- (1) Sample 1 (grown nominally undoped) is nevertheless 358 grown on a doped substrate (n-type). 359
- (2) Sample 2 involves a p-n junction (at the transition 360 between the buffer layer and QW 9). Thus, the electric 361 field reaches its maximum at thermal equilibrium close 362 to QW 9.

The Fermi level is most probably pinned at the surface¹⁴ ³⁶⁴ for both samples. Since the QW structure is grown very close 365 to the surface, this Fermi level pinning also affects the distri- 366 bution of the electric field across the structure. Therefore, it 367 is quite difficult to assess how the electric field should 368 change throughout the structure for different applied biases. 369 For sample 1, we assumed that at thermal equilibrium the 370 electric field is very low in the QWs region. For sample 2, 371 we assumed that the maximum of the electric field occurs 372 close to QW 9 because of the p-n junction. Without being 373 able to determine the flat band situation, all we did was to 374 observe the changes in the CL spectra as a function of the 375 sample biasing, which was applied at the surface through a 376 contact made of silver paste. The backside of the substrate 377 was grounded. The CL measurements were made immedi- 378 ately next to the surface contact. 379

Figure 5 gives the results for the CL measurements under 380 positive surface bias for both samples. For the undoped sam-381 ple, this positive bias induces a red-shift and a quenching of 382 the luminescence, especially at 3 V. We interpret this as an 383 increase in the band-bending compared to the situation 384 where no voltage is applied, and it can be attributed to the 385 above mentioned QCSE. For the doped sample, there is no 386 significant change up to 2 V (modifications of the CL 387



Fig. 3. TEM observations on the as grown doped sample. (a) HAADF image of the whole structure (the substrate is at the bottom). (b)–(d) HRTEM images of QW 9 (a), QW 8 (b), and QW 3 (c). Note that the magnification for these three images is different. The substrate is at the bottom also for the HRTEM images.

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FIG. 4. CL spectra measured at 80 K of the as-grown samples: (a) undoped sample and (b) doped sample.

intensity observed for 0, 1, and 2 V are considered within the 388 reproducibility limits of the CL setup). However, a drastic 389 change starts at 3.2 V: QW 9, which was absent from CL 390 spectra at lower voltage, increases, even dominating the CL 391 emission at 5 V. A sharp line associated with QW 9, possibly 392 made of two contributions, is observed at this voltage. We 393 interpret this observation as evidence that the generated 394 charge carriers were initially (at 0 V) completely driven out 395 of QW 9 by the built-in electric field, resulting in the absence 396 of the CL emission in QW 9. Above 3.2 V, the effective elec- 397 tric field at the p-n junction is reduced so that QW 9 starts 398 being visible on the CL spectra. As for the other QWs, they 399 are probably in a region where the band bending is not sig- 400 nificantly modified by the applied bias, due to Fermi level 401 pinning at the surface, and to the fact that the p-n junction 402 absorbs most of the effects of the applied bias. 403

The difference in the response to surface positive bias of 404 the CL spectra allows us to conclude that these spectra are 405 strongly affected by the doping level through the QCSE and 406 luminescence quenching by electric field-induced tunneling. 407 Band bending in the undoped sample, at thermal equilib-408 rium, is small, and thus, the PL and CL spectra display the 409 intrinsic luminescence line spectra. In the doped sample, the 410 presence of a p-n junction affects the spectra even at thermal 411 equilibrium, in the sense that QW 9 is completely absent in 412 the CL spectrum (although it is seen in the PL with an asso-413 ciated broad line), and all the other QWs produced a broad-414 ened luminescence line shape. The different observations for 415 QW 9 on the CL and PL can be attributed to the difference 416 in the excitation processes: due to the limited penetration 417 depth of the electrons at 20 kV, this QW is more difficult to 418



FIG. 5. (Color online) CL spectra as a function of the bias applied: (a) undoped sample and (b) doped sample.

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excite than in the PL with 1064 nm excitation. From Ref. 11, 419 it can be estimated that the number of electron-hole pairs 420 generated by an incident electron with 20 kV kinetic energy 421 decreases rapidly after a depth of 0.5 μ m (although this paper 422 deals with AlGaAs instead of InP, we can calculate a similar 423 trend for our samples). On the other hand, 1064 nm corre-424 sponds to an energy below the band-gap for photons in InP, 425 and thus, its penetration is much larger than the depth of the 426 QW structure in our samples. 427

Finally, we also measured the SIMS profiles for the 428 as-grown samples. Figure 6 gives an example of such a 429 profile obtained on the doped sample. The surface peak 430 observed for Cl is an effect of the very high sensitivity of the 431 SIMS technique: the sample surface was contaminated by 432 some Cl compound. However, we can see on Fig. 6 that the 433 Cl profile drops to the background level after about 10 s sput-434 tering. As for the P and As profiles, the QW structure is 435 clearly identified. Figure 7 displays the As profile alone, 436 where the qualitative correlation with the As content in the 437 QW is seen (through an increasing background level 438 between the deepest QWs). 439

440 B. SiCl₄/H₂/Ar process

We now describe the results corresponding to the etching with the $SiCl_4/H_2/Ar$ plasma, first describing the PL and CL measurements, and then comparing with TEM and SIMS observations.

The SiCl₄/H₂/Ar process was optimized especially with respect to the very fast etching rate. In the first set of experiments, the samples were exposed 1 min to this plasma to avoid the risk of a complete removal of the QW structure.

Figure 8 is a superposition of the PL data for the asgrown samples, and the samples exposed 1 min to the SiCl₄/ H_2 /Ar plasma. Note that the two samples were not exposed in the same run. On the undoped sample, the signal for QW 1 and QW 2 has disappeared after exposure. On the dopedsample, only the signal for QW 1 disappeared.

The plasma takes a fraction of a minute to stabilize; therefore, 1 min is the shortest exposure that could be performed in a stationary regime; thus, we can expect slight differences



FIG. 6. (Color online) SIMS profiles measured on the as-grown, doped sample: ${}^{31}P_2$ (red), ${}^{75}As$ (black), and ${}^{35}Cl$ (blue). The scale on the left is for P and As, and the scale on the right is for Cl.



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FIG. 7. SIMS profile measured on the as-grown, doped sample, for ⁷⁵As. This isolated profile is the same as that shown for ⁷⁵As in Fig. 6. It is displayed separately for clarity.

in the amount of etched material for two separate exposures, 458 even if the nominal reactor parameters are the same. 459

Apart from the number of QWs having apparently been 460 etched, there is another very significant difference. For the 461 undoped sample, the remaining QWs have similar spectral 462 characteristics to those of the corresponding QWs on the asgrown sample, except for a slight shift to shorter wavelength ("blue-shift") for QW 4, QW 6, and QW 8. For the doped 465 sample, a pronounced sharpening of the luminescence peaks 466 arising from QW 2 to QW 8 is observed; simultaneously, the 467 peaks present a significant shift to the blue. 468



FIG. 8. (Color online) PL spectra of the samples as-grown (line in black) and exposed to the $SiCl_4/H_2/Ar$ plasma for 1 min (line in red): (a) undoped sample and (b) doped sample.

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In order to specify more precisely the etching depth, we 469 measured the SIMS profiles. As an example, Fig. 9 displays 470 the results for the doped sample. The resolution of the pro-471 files is disturbed, probably because of the roughening that 472 occurs at the surface as the etching proceeds faster. Rough 473 surfaces induce a loss of depth resolution during SIMS 474 profiling. Anyhow, we can identify in Fig. 9 that the nine 475 QWs are still present in the sample (there are nine clearly 476 visible minima in the P profile, with the corresponding nine 477 478 maxima on the As profile). To complete the scheme concerning the etch depth, we have performed a TEM observation of 479 the same sample. This is shown in Fig. 10. From this image, 480 we can estimate, by comparison with the image in Fig. 3, 481 that 250 nm of the InP cap have been etched, but all the nine 482 QWs remain present. Therefore, we can conclude at this 483 stage that the PL yield of the QWs closest to the surface is 484 completely reduced after plasma exposure, even if these 485 QWs have not been etched at all. For the doped samples 486 (where we have the SIMS and TEM information to com-487 488 pare), QW 1 is affected by the reduction. For the undoped sample (for which we have not performed SIMS and TEM), 489 if we assume a similar etch depth, QW 1 and QW 2 were 490 affected. It should be noted (for the undoped sample) that for 491 the QWs remaining visible on the PL the peak intensity is 492 much higher than the corresponding intensity on the same 493 sample prior to etching. 494

The other information retrieved from the SIMS has to do 495 with the Cl profile: after exposure to the SiCl₄/H₂/Ar plasma 496 for 1 min, it takes more than 50 s sputtering for the Cl signal 497 498 to reach a background level slightly higher than for the asgrown sample. We deduce from this observation that some 499 Cl has penetrated through the sample. This is best visualized 500 in Fig. 11 where the Cl profiles for the as-grown sample and 501 502 the samples exposed 1 min to the SiCl₄/H₂/Ar plasma are represented on a log scale (sputtering from 0 to 100 s). 503

To complete this part on the 1 min exposure to the SiCl₄/ H₂/Ar plasma, the corresponding CL spectra measured on the same samples are shown in Fig. 12. For the undoped samples, we can notice that the magnitude of the changes is larger than for the PL measurements on the same samples. The changes for the CL yield must be related to the smaller



FIG. 9. (Color online) SIMS profiles measured on doped sample after 1 min exposure to the SiCl₄/H₂/Ar plasma: ${}^{31}P_2$ (red), ${}^{75}As$ (black), and ${}^{35}Cl$ (blue). The scale on the left is for P and As, and the scale on the right is for Cl.





Fig. 10. HAADF image of doped sample after 1 min exposure to the SiCl₄/ H_2 /Ar plasma.

probed depth for the 20 keV electrons than for the 1064 nm 510 photons: after etching 250 nm of the InP cap layer, the 511 deeper QWs become more efficiently excited. The blue-shift 512 observed on the remaining QW transitions is more pro- 513 nounced than the shift observed on the PL spectrum. This 514 has to be related to the excitation process, different for PL 515 and CL. For the doped sample, the most noticeable trends 516 are the strong increase in QW 9 CL emission after exposure 517 to the plasma, and the strong blue-shifts for the other QWs, 518 which is coherent with what is observed on the PL spectra, 519 although the shift is larger than the one measured by PL. The 520 spectral sharpening seen especially for QW 2, QW 3, QW 4, 521 QW 5, and QW 6 on the doped sample after exposure is also 522 observed on the CL, something similar to the PL. This will 523 be discussed with regard to the different band bending situa- 524 tions for the undoped and doped samples, and assumptions 525 on the nature of the penetrating species. 526

Finally, for the case of the $SiCl_4/H_2/Ar$ plasma, the doped 527 sample was also exposed for 2 min. We just show in Fig. 13 528 the PL, TEM, and SIMS data for this sample. Etching has 529



FIG. 11. (Color online) SIMS profiles measured on doped sample before (a) and after 1 min exposure to the SiCl₄/H₂/Ar plasma (b) for 35 Cl on a log scale, for the early etching between 0 and 100 s.

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FIG. 12. (Color online) CL spectra of the samples as-grown (1) and exposed to the $SiCl_4/H_2/Ar$ plasma for 1 min (2): (a) undoped sample and (b) doped sample.

stopped almost exactly at QW 4 (etch depth approximately
600 nm). QW 5, QW 6, QW 7, QW 8, and QW 9 still emit
brightly on the PL spectrum. For these QWs, the transitions
are significantly sharper (in terms of spectral width) than after 1 min exposure to the same plasma and the blue-shift
appears increased.

536 C. Cl₂/N₂ process

The PL spectra for both the undoped and the doped samples (unfortunately, the duration of the plasma exposures differ—4 min in the first case, 3 min in the second case, but this will not block the discussion) are shown in Fig. 14. The first

Fig. 14. (Color online) PL spectra of the samples as-grown (1) and exposed to the Cl_2/N_2 plasma (2). (a) Undoped sample, plasma exposure 4 min; (b) doped sample, plasma exposure 3 min.

point to note is that even after 4 min etching, QW 1 has not 541 been reached by the etching process, in agreement with the 542 slower rate of this etching process (as compared to SiCl₄/H₂/ 543 Ar, at least under the conditions used in this work). Indeed, 544 results from previous studies¹⁵ show that dissociation and 545 ionization processes of Cl₂ by electron impact are less efficient in Cl₂/N₂ plasmas because of the decrease in electron 547 density with the percentage of N₂. The precise etch depth 548 will be examined from the TEM images and SIMS profiles. 549 The second point is that the PL yield is not at all affected by 550 the Cl₂/N₂ exposure. This is in strong contrast with what was 551 observed for SiCl₄/H₂/Ar after 1 min and also with some 552 published results. The third point, for the doped sample, is 553



FIG. 13. (Color online) PL spectra (a), HAADF image (b) and SIMS profiles (c) of the doped sample exposed to the SiCl₄/H₂/Ar plasma for 2 min. PL spectra of as-grown sample (1) and exposed sample (2). In the SIMS plot, red is for P, black for As, and blue for Cl.

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that the Cl_2/N_2 exposure does not seem to affect very signifi-554 cantly the spectral shape of the QW transitions, except for 555 QW 1 and maybe QW 2, whose associated luminescence 556 bands appear sharper after exposure. The difference appears 557 very important with the SiCl₄/H₂/Ar exposure [Figs. 8(b)] 558 where all the luminescence bands, except QW 9, were sharp-559 ened. Finally, concerning the undoped sample, we note that 560 the blue-shifts are now observed for all QW transitions, the 561 peak shift being larger than for the SiCl₄/H₂/Ar exposure. 562 Using a standard fitting procedure (using the Voigt profile, 563 consisting of the convolution of a Lorentzian function by a 564 Gaussian broadening, for each contribution) for PL experi-565 mental spectra, we could determine the following blue-shifts 566 (Cl₂/N₂—exposed sample compared to as-grown sample), 567 indicated on Table II. QWs 2, 4, and 6 have the maximum 568 569 shifts. Let us recall that these shifts are much more pronounced than for the SiCl₄/H₂/Ar-1 min exposure [the cor-570 responding values were estimated less than 2 nm for QWs 571 3–9 on Fig. 8(a)]. Fitting the PL spectra for the doped sample 572 to extract accurate spectral shifts was not fully reliable 573 because of the overlap between neighboring peaks. 574

A TEM-HAADF image of the doped sample after 3 min etching with Cl_2/N_2 is displayed in Fig. 15. InP cap layer of 120 nm still remain, while 180 nm were etched.

SIMS profiles for the doped samples exposed 1 and 3 min to the Cl_2/N_2 plasma are shown on Fig. 16. The Cl signal appears somewhat higher in the bulk part after 1 min than after 3 min, but this may be due to background contamination in the SIMS apparatus for this specific measurement.

583 D. SiCl₄/Ar process

This process was tested with the idea to investigate sam-584 585 ples etched with a single process, but strongly different etch rates. In order to change the etch rate, the RF power at the 586 plasma excitation source and at the sample biasing source 587 were changed, keeping the flow rates for SiCl₄ and Ar con-588 stant. The total pressure in the plasma chamber was also 589 adjusted. Slower etch rate was obtained by reducing the two 590 RF powers (from 1200 to 1000 W at the plasma excitation 591 stage and from 350 to 110 W at the sample stage), simultane-592 ously increasing the total pressure (from 5 to 10 mTorr) in 593 the plasma chamber (which helps decreasing the sample 594 595 bias).

TABLE II. Spectral shifts estimated from the PL data for the undoped sample in Fig. 14(a).

| Quantum well number | Spectral shift (nm |
|---------------------|--------------------|
| QW 1 | 1.7 |
| QW 2 | 7.2 |
| QW 3 | 3.3 |
| QW 4 | 6.2 |
| QW 5 | 4.1 |
| QW 6 | 5.6 |
| QW 7 | 3.9 |
| QW 8 | 3.9 |
| QW 9 | 2.8 |
| | |

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Fig. 15. HAADF image of doped sample after 3 min exposure to the Cl_2/N_2 plasma.

We could perform this part of the study only on the doped 596 sample. We discuss below the CL spectra and SIMS profiles. 597

The results in Fig. 17 were obtained with plasma tuning 599 (RF power, total pressure) to limit the etch rate. The sample 600 bias was 40 V. 601

The main change on the CL spectra concerns QW 1, 602 which displays a sharp and intense transition after plasma 603



FIG. 16. (Color online) SIMS profiles measured on doped sample after 1 min (a) and 3 min (b) exposure to the Cl_2/N_2 plasma, respectively: ${}^{31}P_2$ (red), ${}^{75}As$ (black), and ${}^{35}Cl$ (blue). The scale on the left is for P and As, and the scale on the right is for Cl.

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FIG. 17. (Color online) CL spectra (a) and SIMS profiles (b) for the dopedsample as-grown and exposed to the SiCl₄/Ar plasma—slow etching process (1 min). The spectrum (1) corresponds to the as-grown sample, and the spectrum (2) corresponds to the plasma exposed surface.

exposure. The other structures did not present relevant 604 changes in the CL spectra. While QW 9 starts to become 605 visible, the SIMS profiles tell us that the sample was almost 606 not etched. They also indicate that the Cl signal has 607 608 increased to about 7000 counts inside the sample after plasma exposure, compared to about 10-20 counts measured 609 for the as-grown sample (cf Fig. 6). The Cl profile has a clear 610 maximum at about 1 μ m below the surface. The distribution 611 of Cl inside the sample presents an implantationlike profile. 612

613 (2) SiCl₄/Ar fast etching process:

To measure the impact of the etch rate (keeping constant the SiCl₄ and Ar flow rates), the RF power for the plasma excitation and at the sample stage was increased. The sample bias reached 90 V. The corresponding results—CL spectra and SIMS profiles—are shown in Fig. 18, for a plasma exposure of 1 min.

The SIMS profiles show that the etching has reached QW (etch depth: 400 nm). From the PL spectra, comparing with the spectra in Fig. 17, we conclude that the luminescence bands associated with QW 3 and QW 4 transitions have strongly sharpened, and their intensities have increased. The transitions for deeper QWs also evidence a beginning of sharpening.

From the SIMS data, we see that the Cl signal in the bulk is very low, we do not see any more the Cl signal obtained after the slow etching plasma.

FIG. 18. (Color online) CL spectra (a) and SIMS profiles (b) for the dopedsample as-grown and exposed to the SiCl₄/Ar plasma—fast etching process (1 min). The spectrum (1) corresponds to the as-grown sample, and the spectrum (2) corresponds to the plasma exposed surface.

IV. DISCUSSION

We suggest in this section a discussion of the results 631 described in the previous paragraphs. The comparison 632 between the behaviors of the undoped and doped samples 633 under the same plasma exposures will be helpful to understand the changes induced by the plasma exposure in the two 635 samples and the different etching procedures. 636

The SiCl₄/H₂/Ar plasma was first investigated. The etch 637 rate with this plasma is 250 nm/min (and 600 nm/2 min; the 638 deviation from linearity is due to the time necessary for the 639 plasma to stabilize). After 1 min, we observe the lumines- 640 cence of QW 1 completely suppressed (and even for QW 2 641 in the case of the undoped sample). However, only a part of 642 the InP cap layer has been etched, meaning that the QWs 643 themselves have not been etched. This suppression of the 644 luminescence for the QWs closest to the surface is due to the 645 diffusion of defects, probably Cl-related defects formed 646 when Cl penetrates the structure below the surface, as shown 647 by the SIMS profiles. This result keeps resemblances with 648 some of the published work that we have mentioned in the 649 introduction. $^{5-8}$

Apart from this effect, we observe a small blue-shift of 651 the luminescence peaks (less than 2 nm), in the undoped 652 sample, in contrast to the very strong change of the luminescence bands observed in the doped sample: their spectral 654 width is dramatically reduced, and also a noticeable blue-555 shift takes place. We noticed in addition that the blue-shift 656 on the undoped sample is more important when measured 657

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658 with CL than with PL, due to the different excitation condi-659 tions. Taking into account our preliminary study on the asgrown samples, we analyze the observation for the doped 660 sample as evidence that the SiCl₄/H₂/Ar plasma has an effect 661 662 on the built-in electric field initially present in the sample. The electric field is reduced on the whole thickness of the 663 QW structure, although the etching has affected only 250 nm 664 on the top surface. Reducing the electric field reduces the 665 666 QCSE which impacts QW 1 to QW 8 in this sample. According to the CL result, the charge carrier escape from 667 QW 9, which is an effect of the strong electric field in the 668 p-n junction, is also suppressed or reduced by the SiCl₄/H₂/ 669 Ar plasma etching. This reduction of the internal electric 670 field in the doped sample can only be the consequence of the 671 672 penetration of charged species during the plasma exposure. 673 We suggest that these charged species could be Cl⁻ ions, either directly generated in the plasma or resulting from Cl 674 675 atoms or excited radicals having captured an electron inside the sample. However, the concentration of these ions is not 676 677 high enough to be detected on the SIMS profiles. Only the higher Cl concentration immediately beneath the surface is 678 observed on the SIMS profiles. 679

Finally, we also interpret the blue-shifts on the undoped sample as a reduction of the built-in electric field (weak with respect to the doped sample) present in this sample after growth. The fact that this blue-shift is more important when measured with CL is a consequence of the different excitation schemes for CL and PL.

Then we investigated the Cl_2/N_2 plasma. The etch rate for 686 this plasma is very low under the conditions of our experi-687 ment. However, its impact on the undoped sample consists 688 in inducing much larger blue-shifts than the SiCl₄/H₂/Ar 689 plasma: a maximum of 7.2 nm is observed for QW 2. On the 690 contrary, on the doped sample, this plasma produces little 691 effect, only a sharpening of the line for QW 1 and, to a lesser 692 693 extent, QW 2. No blue-shift is observed on the doped sample. The QCSE is still present in the doped sample after this 694 plasma exposure; it is only reduced very close to the surface. 695 Altogether, this plasma has very little effect on the internal 696 built-in electric field. The explanation for the strong blue-697 698 shifts observed on the undoped sample cannot be related to suppression of the QCSE, since a similar suppression or 699 reduction does not take place in the doped sample. SIMS 700 profiles for this plasma indicate, as for the previous plasma, 701 702 only a limited penetration of Cl beneath the surface.

It seems logical to claim that the different effects observed 703 for the two plasmas would be due, at least in part, to the pres-704 ence of N in the Cl₂/N₂ plasma. We were not able to measure 705 the N profiles in SIMS; however, it is rather probable that N 706 707 penetrates the sample under such plasma exposure, since N does not contribute to the etching reactions at the surface. 708 Then, considering the possible penetration of N, we could 709 explain the blue-shifts observed on the undoped sample as a 710 consequence of the interaction between the N atoms and the 711 712 crystal matrix. The blue-shifts of the PL lines measured enter 713 within the margins expected for a compressive stress of the lattice,^{16,17} which would be induced by N interstitials. It is 714 interesting to point the fact that on the PL spectra the QW 715

lines do not show a conspicuous splitting corresponding to 716 the lh and hh related transitions.¹⁸ Actually, we could have 717 anticipated this seeing two different peaks resolved for each 718 QW transition in the as-grown sample (due to the intrinsic 719 stress of the $InAs_xP_{1-x}$ layers). This is not the case, either 720 because the magnitude of the splitting between heavy-hole 721 and light-hole transitions is too small to be resolved experi-722 mentally, even for the most strained QW, or because our ex- 723 citation power density for the PL measurement is not in the 724 adequate range. Even if we do not see the splitting on the as 725 grown sample, the mechanical stress induced by the N atoms 726 interacting with the structure could create the splitting if this 727 stress occurs in a biaxial geometry. We suggest that the pres- 728 ence of N, in an interstitial position, mainly induces a hydro-729 static stress which does not change the symmetry of the PL 730 transitions observed, since hydrostatic stress does not split 731 the lh and hh related bands. The magnitude of the blue-shifts 732 for the different QWs is dependent from their position. A 733 maximum is recorded for QW 2 with a PL peak blue-shifted 734 by 7.2 nm. Then, a decrease of the blue-shift is observed for 735 the deeper QWs. This variable blue-shift could be attributed 736 to the combination of the effect of interstitial N with the 737 native compressive biaxial stress related to As incorporation 738 changing in the different QWs. However, further investiga-739 tions would be required to confirm this hypothesis. 740

On the other hand, we did not observe the blue-shifts in 741 the case of the doped sample. This sample is strongly 742 affected by the QCSE, which is not suppressed nor even 743 reduced for most of the structure by the Cl_2/N_2 plasma etch-744 ing. Actually, we have no exact idea of the scenario resulting 745 from combined QCSE and compressive hydrostatic stress, 746 but certainly, the QCSE might dominate masking the obser-747 vation of the blue-shifts due to stress. 748

Finally, let us consider the SiCl₄/Ar plasma experiments, 749 which were designed to test the effects of the etch rate under 750 conditions of constant reactive gas injection. For the very 751 slow etch process, we observed a maximum penetration of 752 the Cl in the structure, with a distribution similar to that 753 resulting from an implantation mechanism. On the other 754 hand, very few changes appeared on the CL spectrum after 755 this slow etching plasma. Only a slight modification of the 756 electric field close to the surface may occur, explaining the 757 alone change of QW 1 related emission, while the emission 758 from the other QWs did not show significant changes. The 759 association of these trends (maximum penetration of Cl, lim- 760 ited change on the CL spectrum) points to the fact that prob-761 ably the Cl species penetrating the sample is now atomic Cl, 762 contrary to what we claim for the SiCl₄/H₂/Ar plasma. For 763 the fast etching process, we observe a much more limited 764 penetration of Cl than for the slow etching process. The CL 765 spectrum is more strongly affected, suggesting the presence 766 of Cl⁻ ions, rather than neutral Cl atoms. 767

From these combined observations, we now try to build up 768 a scheme for the interaction mechanisms of the etching plasma 769 and the QWs structure, at least for the SiCl₄/Ar plasma: 770

 different forms of Cl play a relevant role in this mecha- 771 nism, at least Cl⁻ and atomic Cl; 772

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| 773 | (2) only the Cl present at the surface contributes to the etch- |
|-----|---|
| 774 | ing process, while the Cl penetrating across the surface |
| 775 | inside the sample does not contribute to the etching |
| 776 | process; |
| | |

(3) the etching experiments with $SiCl_4/Ar$ plasmas give a

conspicuous anticorrelation between the amount of Cl inside the sample and the etching rate.

In order to quantify this last point, the SIMS data were 780 also exploited to identify a possible trend on the concentra-781 tion on Cl at the surface at the different steps of the etching 782 process. The signal measured for ³⁵Cl for the first point 783 recorded (sputter time 5.4 \times 10⁻¹ s/sputter depth approxi-784 mately 2.5 nm) was used as an indication of the Cl level at 785 the surface before performing the sputter profile. This is 786 shown in Table III ("Surface Cl"). This analysis was carried 787 out for all the SIMS profiles measured, and Table III indi-788 789 cates also the corresponding etching rate. These data show that the surface Cl signal is between 1.84 and 6.15×10^{5} 790 counts/s for the as-grown surfaces and for the surfaces of 791 samples exposed to plasmas with a fast etching rate, while 792 793 for samples exposed to plasmas with a slow etching rate the Cl surface signal measured lies between 3.89 and 5.26×10^6 794 795 counts/s, i.e., approximately an order of magnitude higher. Concerning the bulk Cl concentration, one observes that for 796 797 the lowest etching rate, especially for the sample exposed to the SiCl₄/Ar plasma "slow etch rate," the bulk Cl signal dis-798 plays a maximum value. 799

Therefore, the tendency derived from the SIMS data is that 800 801 the high etch rates are observed for samples that display the smaller Cl signal at both the surface and the bulk, while the 802 slow etch rates are observed when the Cl signal is high at both 803 the surface and the bulk. This suggests that the Cl species that 804 are active in the etching process do not penetrate inside the 805 samples, whereas the forms that penetrate inside the samples 806 are inefficient for the etching process. Figure 19 illustrates in 807 more details this tendency observed for the different etch plas-808 mas investigated. 809

810 V. SUMMARY AND CONCLUSIONS

The experimental work detailed in this paper is based on the long-established methodology of dedicated QW structure for the investigation of defect production and consequences during plasma etching of III–V semiconductor materials. Using a combination of characterization techniques, including PL, CL, SIMS, and TEM, we tried to quantify the effects

TABLE III. SIMS data for Cl at the surface and etch rate. The surface data are taken as the SIMS count for the first point recorded (after 5.4×10^{-1} s sputtering).

| No. | Sample description | Surface Cl (counts/s) | Etch rate |
|-----|--|-----------------------|----------------|
| 1 | As-grown sample | 1.84×10^{5} | |
| 2 | SiCl ₄ /H ₂ /Ar —1 min | 4.13×10^{5} | 250-300 nm/min |
| 3 | Cl_2/N_2 —1 min | 3.89×10^6 | 60 nm/min |
| 4 | As-grown sample | 6.15×10^5 | _ |
| 5 | SiCl ₄ /Ar plasma—slow (1 min) | 5.26×10^{6} | 55 nm/min |
| 6 | SiCl ₄ /Ar plasma—fast (1 min) | $5.81 	imes 10^5$ | 400 nm/min |

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Fig. 19. (Color online) SIMS profiles measured for the different samples displayed in Table III, in log scale, for the early stages of the sputter process between 0 and 100 s.

of different plasma etching chemistries used in the context \$17 of fabrication of InP-based photonic circuits. These chemistries have in common the fact that they involve chlorine \$19 compounds. \$20

We first analyzed carefully the properties of the "as- 821 grown" samples (before any plasma exposure), observing 822 the importance of taking into account the type and amount 823 of doping in the samples. Indeed, doping is associated with 824 the presence of a built-in electric field whose impact on the 825 luminescence spectra can be very significant, through the 826 quantum confined Stark effect, and also the fact that gener- 827 ated carriers can escape from the QWs through tunneling 828 induced by this built-in electric field. Having understood 829 why the luminescence spectra differ in samples with differ- 830 ent doping profiles, we could perform a systematic study of 831 different etching chemistries. The first chemistry (SiCl₄/H₂/ 832 Ar) yields an etch rate in the order of 250–300 nm/min in 833 our conditions, which is interesting when trying to etch fea- 834 tures several micrometers deep. We have limited the expo- 835 sures to very short durations (1 and 2 min) to avoid etching 836 the whole QW structure. The luminescence of the QWs clos- 837 est to the etch front was observed to decrease and disappear, 838 similar to previous results in the literature; however, this is 839 the case only for the first and probably second QW, and the 840 deeper QWs are not affected. The other observation is a very 841 significant difference between the undoped and the doped 842 samples: in the case of the undoped sample, no change of the 843 spectral shape for the remaining luminescence peaks occurs, 844 while for the doped sample a very pronounced spectral 845 sharpening is seen. The two types of sample react differently 846 to the plasma. We tentatively explain this difference suggest-847 ing that some of the Cl species from the plasma penetrate 848 through the QW structure (which is confirmed by the SIMS 849 profiles), and that part of these species bares an electric 850 charge (probably Cl⁻ ions). Thus, they modify the built-in 851 electric field in a way for which the QCSE and associated 852 carrier escape are reduced. 853

This effect—sharpening of the luminescence peaks—is 854 much less pronounced in the case of the Cl₂/N₂ plasmas, 855

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which anyhow display a reduced etch rate (60 nm/min). The
main effect of these plasmas is a blue-shift of the luminescence peaks (especially observable on the undoped sample),
which we attribute to the penetration of N in the QW structure. N occupies interstitial sites, thus producing a hydrostatic compressive stress.

Finally, we also investigated a SiCl₄/Ar chemistry, for 862 which we were able to vary the etch rate between 55 and 863 400 nm/min keeping the flow rates constant (i.e., the 864 "chemistry" constant). This last experiment, apart from con-865 firming the spectral sharpening effect observed with the 866 SiCl₄/H₂/Ar chemistry, leads us to suggest a balance mecha-867 nism for the interplay between the Cl-species active for the 868 etching process at the surface, and the Cl-species that pene-869 trate the sample structure and are "lost" for the etching pro-870 cess, lowering the etching efficiency. From our observations, 871 a tendency appeared for the etch rate to be anticorrelated to 872 the amount of Cl-species present beneath the surface (as 873 detected by the SIMS profiles). 874

All the results described in this paper point to the impor-875 tance of checking the impact of reactive etching plasmas on 876 the remaining materials structures. Even though our experi-877 ments were performed in the absence of any mask structure, 878 we can claim that when using similar plasmas for the pro-879 duction of photonic features with a three-dimensional geom-880 etry (like a simple rectangular ridge structure), similar 881 effects would occur due to the LATERAL penetration of the 882 etching species through the structures being etched, produc-883 ing the same artifacts that we have described. Taking into 884 account these possible artifacts in the general design of pho-885 tonic devices is an issue that cannot be neglected. Thanks to 886 the lateral resolution of the CL technique, which we have 887 not used in the present study, we will be able to extend our 888 889 general methodology to the case where quantifying etching-930

induced defects within 3D structures (e.g., rectangular waveguides) is required. This will be the next step of this study.

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