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Palladium–Catalyzed Non-Directed C–H bond Arylation of Difluorobenzenes and Dichlorobenzenes Bearing Benzoxazole or Benzothiazole

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Abstract— We report, herein, on palladium-catalyzed direct arylation of difluorobenzenes and dichlorobenzenes bearing benzoxazole or benzothiazole moieties, which don't act as directing groups. With moderate electron-withdrawing substituents on the aryl bromides as coupling partners, the reaction proceeds nicely using phosphine-free PdCl₂ catalyst, and potassium pivalate/dimethylacetamide (PivOK/ DMA) as catalytic system. The reaction was regioselective and occurred at the less hindered *ortho*-positions of fluorine or chlorine atoms.

Keywords: Arylation, Palladium, C-H Bond Activation, Heterocycles, Fluorine Chemistry.

1. Introduction

Difluorobenzenes substituted by benzoxazole or benzothiazole units are an important class of ligands. As example, the cyclometalated iridium complex **I** is a photoluminescent complex involved in the construction of organic light emitting diodes.¹⁻³ The blue-light-emitting zinc material **II** has been synthetized from 2-(5-fluoro-hydroxyphenyl)benzothiazole.⁴ The iridium catalyst **III**, with two 2-(3,5-difluorophenyl)benzoxazole ligands, displays a high activity in water splitting.⁵ In addition, the motifs benzothiazole and benzoxazole are present in many pharmaceuticals. As example, 2-(benzothiazol-2-yl)-N,N-bis(2-chloroethyl)-5- fluorobenzenamine (**IV**) displays high activity against human cervical cancer cell lines,^{6, 7} and 2-(3-fluorophenyl)-1-(1-(3-fluorophenyl-1,2,3-triazol-4-yl)methylbenzoimidazole (**V**) exhibits antitubercular properties.⁸



Figure 1. Relevant Structures Containing Difluorobenzenes Substituted by a Benzoxazole or a Benzothiazole

Nowadays, the functionalization of C–H bonds is an important research area, as such methodologies provide an atom economic access to complex molecules.⁹⁻¹³ Since the pioneering work of Fagnou and co-workers on palladium-catalyzed direct arylation of electron-deficient polyfluorobenzenes, this methodology proved as one of the most reliable and easiest access to (poly)fluorobiphenyls (Figure 2a).¹⁴ Their work induced emulation in direct arylation of electron-deficient arenes.¹¹ Among other improvements, a particular attention has focused on the use of alternative coupling partners to aryl halides (*e.g.*, tosylates,¹⁵⁻¹⁷ diaryliodonium salts,¹⁸ boronic acids,^{19, 20} ArSO₂Na,²¹ carboxylic acid²² or simple arenes under oxidative condi-

tions²³), or the use of other catalytic systems.^{24, 25} In order to address the lower reactivity of less-electron deficient fluorobenzenes,^{14, 26, 27} directing group (e.g., carboxylic acid) strategy have been employed.^{28, 29} On other hand, benzoxazole unit has been employed as directing group for the regioselective C2 arylation of phenyl using a palladium catalyst in the presence of silver carbonate in trifluoroacetic acid (Figure 2b).^{30, 31} A similar procedure was reported by Ding, Peng and co-workers for the direct C2 arylation of 2-arylbenzothiazole (Figure 2c).³² To the best of our knowledge, the reactivity of dihalogenophenyls bearing a benzoxazole or a benzothiazole as potential directing groups in palladium-catalyzed direct arylation has never been reported. Furthermore, the reactivities and regioselectivities in palladium-catalyzed direct arylation of such substrates needed to be investigated (Figure 2d).

a. Pd-Catalyzed Direct Arylation of Fluorobenzene (Fagnou)



c. Pd-Catalyzed Direct Arylation of 2-Phenylbenzothiazole (Ding and Peng)



d. Pd-Catalyzed Direct Arylation of Dihalogenophenyl-benzoxazoles and -



Figure 2. Previous Palladium-Catalyzed Intermolecular Direct Arylations of Fluorobenzenes, 2-Arylbenzothiazoles or 2-Arylbenzoxazoles.

2. Results and discussion

We selected 2-(3,5-difluorophenyl)benzoxazole and 4-bromobenzonitrile as model substrates, and we used our previous optimized reaction conditions described for the direct arylation of 3-substituted fluorobenzenes, namely, 5 mol% PdCl(C₃H₅)(dppb) catalyst in the presence of KOAc as base in DMA at 150 °C (Table 1, entry 1).³³ We were pleased to find that the reaction occurred regioselectively at the C-H bond flanked by the two fluorine atoms to afford the regioisomer 1a in 53% yield. It is important to note that the regioisomer **1b**, resulting from a benzoxazolyl directed C-H bond arylation, was not detected. This result (i.e., the most acidic C–H bond reacts preferentially) suggests that, under these reaction conditions, a concerted metalation-deprotonation mechanism is operative instead of a directed C-H bond activation process. Then, we employed other palladium sources. Phosphine-free Pd(OAc)₂ gave a lower 45% yield, whereas using 5 mol% PdCl₂ a higher yield of 57% was obtained (Table 1, entries 2 and 3). $Pd_2(dba)_3$ catalyst was also effective for this reaction, albeit 1a was isolated in a lower yield (Table 1, entry 4). Then, we investigated the effect of the base on this transformation. Potassium carbonate led to a very low yield in 1a due to a low conversion (Table 1, entry 5). The highest yield in 1a was obtained using PivOK as base with 75% isolated yield of **1a** (Table 1, entry 6). As proposed by Fagnou and co-workers, ¹⁴ PivOK certainly acts as a proton-shuttle in the CMD mechanism and facilitates the C-H bond cleavage. PivONa exhibits a slightly lower efficiency, probability because of its lower solubility in DMA (Table 1, entry 7). Notably, a lower amount of catalyst (i.e., 2 mol% PdCl₂) gave lower yield in **1a** (Table 1, entry 8). Finally, other solvents were tested for the direct arylation of 2-(3,5difluorophenyl)benzoxazole. DMF and DMSO were completely ineffective for this reaction as only starting materials and bromoarene homocoupling were detected (Table 1, entries 9 and 10). Then, we used xylene as non-polar and noncoordinating solvent in order to favor the benzoxazolyl directed C-H bond activation process; however no reaction occurred (Table 1, entry 11). It should be mentioned that when the reaction was performed from 3,5-difluorobenzaldehyde instead of 2-(3,5-difluorophenyl)benzoxazole, no coupling product was detected (Table 1, entry 12).

F F F O N + (1.5 equiv.)	CN [Pd] (5 n Base (2 e Solvent, 150	nol%) F rquiv.) °C, 16h			F
Entry	[Pd]	Base	Solvent	Yield 1a (%)	
1	PdCl(C ₃ H ₅)(dppb)	KOAc	DMA	53	
2	$Pd(OAc)_2$	KOAc	DMA	45	
3	PdCl ₂	KOAc	DMA	57	
4	Pd ₂ (dba) ₃	KOAc	DMA	32	
5	PdCl ₂	K_2CO_3	DMA	12	
6	PdCl ₂	PivOK	DMA	75	
7	PdCl ₂	PivONa	DMA	67	
8 ^a	PdCl ₂	PivOK	DMA	56	
9	PdCl ₂	PivOK	DMF	0	
10	PdCl ₂	PivOK	DMSO	0	
11	PdCl ₂	PivOK	xylene	0	
12 ^b	PdCl ₂	PivOK	DMA	0	

Table 1. Optimization of the Reaction Conditions

CN

^aThe reaction was performed using 2 mol% of PdCl₂.

^bThe reaction was performed from 3,5-difluorobenzaldehyde.

With the optimized conditions in hands, we turned our attention to the scope and limitation of this palladium-catalyzed nondirected C–H bond arylation. Firstly, we performed the C4 direct arylation of 2-(3,5-difluorophenyl)benzoxazole with a set of aryl bromides (Scheme 1). Other aryl bromides containing electron-withdrawing *para*-substituents, such as 4bromoacetophenone or ethyl 4-bromobenzoate, smoothly reacted to afford the desired C4-arylated products **2** and **3** in 82% and 84% yields, respectively. Not surprisingly, 2-bromobenzonitrile, which has a low steric profile and higher Hammet constant that 4-bromobenzonitrile (0.71 vs 0.66),³⁴ exhibited a higher reactivity to give the desired product **4** in 89% yield. However, more bulky 2-bromobenzaldehyde led to the C4-arylated compound **5** in only 67% yield. Then, we investigated the reactivity of a set of six-membered ring heteroaryl bromides. 3-Bromopyridine reacts with 2-(3,5difluorophenyl)benzoxazole to give the C4-arylated product **6** in moderate 46% yield. The arylated products **7** and **8** – resulting from the cross-coupling with 3-bromoquinoline and 5-bromopyrimidine– were isolated in high 73% and 71% yields, respectively. However, it is important to note that ligand-free PdCl₂/PivOK system did not allowed the use of aryl bromides bearing electron-donating substituents due to a slower oxidative addition process.



 $\begin{array}{c} () \\ ()$

Scheme 1. Scope of Aryl Bromides for Pd-catalyzed Direct Arylation of 2-(3,5-Difluorophenyl)benzoxazole.

Next, we studied the reactivity of 2-(3,5-difluorophenyl)benzothiazole in palladium-catalyzed C–H bond arylation (Scheme 2). Interestingly, unlike benzoxazole, benzothiazole exhibits directing group properties. Indeed, the reaction between 2-(3,5-difluorophenyl)benzothiazole and 4-bromobenzonitrile afforded a mixture of the C4- and C2-arylated products **9a** and **9b** in 74:26 ratio. The major regioisomer **9a**, which results from the non-directed C–H bond activation (i.e., the C–H bond flanked by the two fluorine atoms), was isolated in 51% yield. A similar reactivity and regioselectivity trend was observed with 4-bromonitrobenzene. From 2-bromobenzaldehyde or 2-bromobenzonitrile, only the regioisomers **11a** and **12a** arising from C4-arylation were isolated in 58% and 69% yields, respectively. Using 5-bromopyrimidine as coupling partner, the reaction was not regioselective and a mixture of both regioisomers **13a** and **13b** was obtained in 63:37 ratio. The major regioisomer **13a**, resulting from the non-directed C–H bond activation, was isolated in pure form in 50% yield.



Scheme 2. Pd-catalyzed Direct Arylation of 2-(3,5-Difluorophenyl)benzothiazole

We investigated then the regioselectivity of 2-(2,4-difluorophenyl)benzoxazole and 2-(2,4-difluorophenyl)benzothiazole in palladium-catalyzed direct arylation with aryl bromides as coupling partners (Scheme 3). We found that using 2-(2,4difluorophenyl)benzoxazole as substrate, the direct arylation again occurred regioselectively at the C-H bond flanked by the two fluorine atoms. Using methyl 4-bromobenzoate, the desired C3-arylated product 14 was isolated in 71% yield. Metasubstituted aryl bromides such as 3-bromoacetophenone or 3-bromobenzonitrile also displayed high reactivities and the desired biphenyls 15 and 16 were isolated in 64% and 59% yields, respectively. Under these reaction conditions, bromobenzene smoothly reacted to provide 17 in only 21% yield. We imputed this lack of reactivity to the slow oxidative addition rate of this electron-rich aryl bromide. Heteroaryl bromides, such as 3-bromoquinoline, afforded the desired product 18 in 42% yield. We should note that in all cases, no arylation at C6-position of phenyl moiety was detected in the crude mixture. In contrast to the reaction with 3,5-difluorobenzene bearing a benzothiazole at C1 -in which, two regioisomers were obtained (Scheme 2)-, the reaction with 2,4-difluorobenzene bearing a benzothiazole at C1 afforded a single regioisomer. Indeed, when the reaction was performed from 2-(2,4-difluorophenyl)benzothiazole and ethyl 4-bromobenzoate, only the regioisomer 19 was obtained in 59% yield. The arylation took place at the C-H bond between the two fluorine atoms, whereas the benzothiazolyl directed C-H bond activation process is not operative. Similar regioselectivity and reactivity trend was observed with other 4- and 3-substituted aryl bromides affording regioselectively the desired arylated products 20, 21, and 22 in 71%, 63%, and 54% yields, respectively.



Scheme 3. Pd-catalyzed Direct Arylation of 2,4-Difluorophenyls Bearing a Benzoxazole or a Benzothiazole

We had previously demonstrated that polychlorobenzenes react quite similarly to polyfluorobenzenes in palladium-catalyzed direct arylation.³⁵ Indeed, the arylation occurred at the most acidic position, albeit for some specific cases, in which the steric hindrance was the most important factor. Therefore, we investigated the reactivity of 3,5-dichlorobenzenes bearing benzoxazole or benzothiazole motifs. From 2-(3,5-dichlorophenyl)benzoxazole using 5 mol% of PdCl₂ in the presence of PivOK as base in DMA, the arylation occurred regioselectively at the most acidic position, namely, the C–H bond between the two chlorine atoms. The C–H bonds at *ortho*-positions to the benzoxazole remained untouched. Using a set of aryl bromides, the C4-arylated product **23-26** were isolated in 33-44% yields. On the other hand, the 3,5-dichlorophenyl substituted by a benzothiazole moiety at C1 was found to be unreactive under these reaction conditions.



Scheme 4. Pd-catalyzed Direct Arylation of 3,5-Dichlorophenyls Bearing a Benzoxazole or a Benzothiazole

3. Conclusion

In summary, we have demonstrated that palladium-catalyzed direct arylation of electron-deficient arenes such as 3,5difluorobenzene, 2,4-difluorobenzene or 3,5-dichlorobenzene bearing benzoxazole or benzothiazole units, reacted preferentially *via* the activation of the most acidic C–H bond, whereas the C–H bonds at *ortho*-position of benzoxazole was not activated. In the case of 2-(3,5-difluorophenyl)benzothiazole, the benzothiazole unit acts as a directing group and mixtures of both C2- and C4-regioisomers were obtained. The non-directed C-H bond functionalization proceeds with ligand-free PdCl₂ catalyst and PivOK as base in DMA. This procedure tolerates a wide variety of substituents on the aryl bromides such as nitro, cyano, ester, ketone, formyl and also pyridine derivatives. This strategy allows the regioselective *para-* or *meta*arylations without the use of directing groups. Moreover, it could find further applications in ligand synthesis for the tuning of organic light emitting diodes.

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References

- 1. H.-W. Hong and T.-M. Chen, Mater. Chem. Phys., 2007, 101, 170-176.
- 2. T.-R. Chen, H.-P. Lee, J.-D. Chen and K. H. C. Chen, Dalton Trans., 2010, 39, 9458-9461.
- 3. T.-R. Chen, J. Organomet. Chem., 2008, 693, 3117-3130.
- 4. X. Huixia, W. Hua, Y. Yan, C. Liuqing, H. Yuying and X. Bingshe, Synth. Met., 2012, 162, 775-780.
- 5. T.-R. Chen, H.-P. Lee and J.-D. Chen, *Inorg. Chem.*, 2011, **50**, 3645-3650.
- 6. S. Kini, S. Swain and A. Gandhi, Ind. J. Pharm. Sci., 2007, 69, 46-50.
- 7. R. M. Kumbhare and V. N. Ingle, Ind J Chem, 2009, 48, 996-1000.
- C. Gill, G. Jadhav, M. Shaikh, R. Kale, A. Ghawalkar, D. Nagargoje and M. Shiradkar, *Bioorg. Med. Chem. Lett.*, 2008, 18, 6244-6247.
- 9. F. Kakiuchi and T. Kochi, Synthesis, 2008, 3013-3039.
- 10. E. M. Beck and M. J. Gaunt, Top. Curr. Chem., 2010, 292, 85-121.
- 11. M. He, J.-F. Soulé and H. Doucet, ChemCatChem, 2014, 6, 1824-1859.
- 12. R. Rossi, F. Bellina, M. Lessi and C. Manzini, Adv. Synth. Catal., 2014, 356, 17-117.
- 13. K. Yuan, J.-F. Soulé and H. Doucet, ACS Catal., 2015, 5, 978-991.
- 14. M. Lafrance, C. N. Rowley, T. K. Woo and K. Fagnou, J. Am. Chem. Soc., 2006, 128, 8754-8756.
- 15. S. Fan, J. Yang and X. Zhang, Org. Lett., 2011, 13, 4374-4377.
- 16. L. Ackermann and S. Fenner, Chem. Commun., 2011, 47, 430-432.
- 17. J. W. W. Chang, E. Y. Chia, C. L. L. Chai and J. Seayad, Org. Biomol. Chem., 2012, 10, 2289-2299.
- 18. F. Guo, J. Han, S. Mao, J. Li, X. Geng, J. Yu and L. Wang, RSC Adv., 2013, 3, 6267-6270.

- 19. Y. Wei, J. Kan, M. Wang, W. Su and M. Hong, Org. Lett., 2009, 11, 3346-3349.
- 20. X. Fang, Y. Huang, X. Chen, X. Lin, Z. Bai, K.-W. Huang, Y. Yuan and Z. Weng, J. Fluorine Chem., 2013, 151, 50-57.
- 21. T. Miao and L. Wang, Adv. Synth. Catal., 2014, 356, 429-436.
- 22. K. Xie, Z. Yang, X. Zhou, X. Li, S. Wang, Z. Tan, X. An and C.-C. Guo, Org. Lett., 2010, 12, 1564-1567.
- 23. Y. Wei and W. Su, J. Am. Chem. Soc., 2010, 132, 16377-16379.
- 24. D. Yuan and H. V. Huynh, Organometallics, 2012, 31, 405-412.
- 25. H.-Q. Do and O. Daugulis, J. Am. Chem. Soc., 2008, 130, 1128-1129.
- 26. P. Ricci, K. Kramer, X. C. Cambeiro and I. Larrosa, J. Am. Chem. Soc., 2013, 135, 13258-13261.
- 27. M. He, J.-F. Soulé and H. Doucet, ChemCatChem, 2015, 7, 2130-2140.
- 28. H. A. Chiong, Q.-N. Pham and O. Daugulis, J. Am. Chem. Soc., 2007, 129, 9879-9884.
- 29. J. Cornella, M. Righi and I. Larrosa, Angew. Chem. Int. Ed., 2011, 50, 9429-9432.
- 30. F. Yang, Y. Wu, Z. Zhu, J. Zhang and Y. Li, *Tetrahedron*, 2008, **64**, 6782-6787.
- 31. F. Yang, Y. Wu, Y. Li, B. Wang and J. Zhang, Tetrahedron, 2009, 65, 914-919.
- 32. Q. Ding, H. Ji, D. Wang, Y. Lin, W. Yu and Y. Peng, J. Organomet. Chem., 2012, 711, 62-67.
- 33. T. Yan, L. Zhao, M. He, J.-F. Soulé, C. Bruneau and H. Doucet, Adv. Synth. Catal., 2014, 356, 1586-1596.
- 34. Y. Takahata and D. P. Chong, Int. J. Quantum Chem., 2005, 103, 509-515.
- 35. L. Zhao, T. Yan, C. Bruneau and H. Doucet, Catal. Sci. Technol., 2014, 4, 352-360.

Abstract

