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## Biological carbon dioxide utilisation in food waste anaerobic digesters



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### ABSTRACT

Carbon dioxide (CO<sub>2</sub>) enrichment of anaerobic digesters (AD) was previously identified as a potential on-site carbon revalorisation strategy. This study addresses the lack of studies investigating this concept in up-scaled units and the need to understand the mechanisms of exogenous CO<sub>2</sub> utilisation. Two pilot-scale ADs treating food waste were monitored for 225 days, with the test unit being periodically injected with CO<sub>2</sub> using a bubble column. The test AD maintained a CH<sub>4</sub> production rate of  $0.56 \pm 0.13 \text{ m}^3 \text{ CH}_4 \cdot (\text{kg VS}_{\text{fed}} \text{ d})^{-1}$  and a CH<sub>4</sub> concentration in biogas of 68% even when dissolved CO<sub>2</sub> levels were increased by a 3 fold over the control unit. An additional uptake of 0.55 kg of exogenous CO<sub>2</sub> was achieved in the test AD during the trial period. A 2.5 fold increase in hydrogen (H<sub>2</sub>) concentration was observed and attributed to CO<sub>2</sub> dissolution and to an alteration of the acidogenesis and acetogenesis pathways. A hypothesis for conversion of exogenous CO<sub>2</sub> has been proposed, which requires validation by microbial community analysis.

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### 1. Introduction

Anaerobic digestion (AD) stabilises organic wastes while producing biogas with a 50–75% methane (CH<sub>4</sub>) and 50–25% carbon dioxide (CO<sub>2</sub>) concentration. The calorific value of the biogas can then be used by combustion in combined heat and power (CHP) engines or by selectively separating the CH<sub>4</sub> for its use as biomethane. The remaining CO<sub>2</sub> present in the biogas, however, is emitted to the atmosphere with exhaust gas streams. Biogenic emissions of CO<sub>2</sub> from ADs just in the UK have been estimated at 0.27 MtCO<sub>2</sub> per annum for the water sector (Byrns et al., 2013) and at 0.31 MtCO<sub>2</sub> per annum for ADs treating agricultural and community waste (industrial sites not accounted) (Bajón Fernández et al., Submitted for publication a). Emissions of CO<sub>2</sub> with biogenic origin are not considered within carbon accounting inventories, however, its reduction is considered as a negative carbon release. Therefore, implementation of revalorisation strategies for biogas CO<sub>2</sub> could be an option to counteract the increasing greenhouse gas (GHG) emissions of the water sector (Byrns et al., 2013) and to further reduce the carbon footprint of AD

in the organic waste sector.

Implementation of carbon capture and storage (CCS) is feasible in the energy sector (DECC, 2012). However, its implementation for handling biogas CO<sub>2</sub> is limited by the requirement to transport CO<sub>2</sub> from AD sites in scattered location. New biogenic carbon sequestration methods such as the enrichment of anaerobic processes with CO<sub>2</sub> (for its bioconversion into CH<sub>4</sub>) are therefore being investigated as a method to utilise on-site CO<sub>2</sub> concentrated gas streams. The capacity of upflow anaerobic sludge blanket (UASB) reactors (Alimahmoodi and Mulligan, 2008), single stage anaerobic digesters (ADs) (Bajón Fernández et al., 2014; Sato and Ochi, 1994) and two phase ADs (Salomoni et al., 2011) has previously been examined to utilise additional CO<sub>2</sub>. However, these previous studies have focussed on proving the concept of carbon uptake or assessing associated benefits in CH<sub>4</sub> formation at laboratory scale. In the case of ADs treating food waste, CO<sub>2</sub> enrichment has so far only been studied in batch 1 L units (Bajón Fernández et al., 2014). Therefore, there is limited information available for scaled-up units. As a consequence, the means by which CO<sub>2</sub> enrichment could be retrofitted into scaled-up units have not yet been addressed. There is a need to investigate suitable technologies for completing CO<sub>2</sub> enrichment of ADs without incurring any dilution of the headspace CH<sub>4</sub> content. The anticipated simplicity for implementation (Byrns

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et al., 2013) and the possibility of transferring gas to liquid technologies already used in other industrial sectors (e.g. bubble columns) have been suggested but not investigated. Furthermore, the mechanisms by which CO<sub>2</sub> can be bioconverted to CH<sub>4</sub> have not been fully elucidated. Several studies have hypothesised mechanisms of CO<sub>2</sub> utilisation (Alimahmoodi and Mulligan, 2008; Bajón Fernández et al., 2014; Francioso et al., 2010). However, only one has reported microbial community data in ADs enriched with CO<sub>2</sub> where conditions specifically favouring development of hydrogenotrophic methanogens were not applied (Bajón Fernández et al., Submitted for publication b). In this case, an increase in acetoclastic methanogenic activity (*Methanosaetaceae*) as a result of periodic CO<sub>2</sub> injections was reported. Nevertheless, the question of whether this increase is due to a direct impact of CO<sub>2</sub> in *Archaea* communities or to an indirect benefit through an alteration of previous stages of the digestion process (i.e. acidogenesis and acetogenesis) remains unclear.

This study investigated both the practicalities of an up-scaled implementation of CO<sub>2</sub> enrichment into ADs and the fate of exogenous CO<sub>2</sub>. For the first time, a pilot-scale AD rig (106 L) treating food waste and adapted for CO<sub>2</sub> enrichment through an external bubble column was operated and compared to a standard unit. Results are discussed in terms of digestate quality, biogas production and CO<sub>2</sub> uptake when comparing ADs with and without CO<sub>2</sub> enrichment. A comprehensive discussion on the mechanisms of CO<sub>2</sub> utilisation is included based on monitoring hydrogen (H<sub>2</sub>) levels and volatile fatty acid (VFA) speciation dynamics.

## 2. Materials and methods

### 2.1. Description and operation of the AD rig retrofitted with CO<sub>2</sub> enrichment

Two pilot-scale semi-continuous ADs treating food waste were operated for 225 days. Each unit consisted of a cylindrical section and a cone base with a total volume of 193 L, of which 106 L were liquid working volume (Fig. 1). Each AD was continuously stirred with an external peristaltic pump (series 600, Watson Marlow, Cornwall, UK) operated to achieve a recirculation rate of 30 min. The ADs were maintained at mesophilic conditions (38.5 °C) with a heating jacket over the cylindrical section (LMK Thermosafe,

Haverhill, UK).

Both ADs were inoculated with digestate collected from a full-scale UK AD site treating 48,000 tonnes of organic waste per year. The units were fed on a daily basis (Monday to Friday) with a mixture of organic waste collected from local supermarkets and catering facilities. This waste was manually segregated to remove any inorganic content and macerated on-site with a series 'A' Muncher macerator (Mono Pumps Ltd., Manchester, UK) connected to a progressing cavity pump (W range, Mono Pumps Ltd., Manchester, UK). A loop in the system allowed the material to be recirculated into the macerator until a homogenous mixture with a maximum particle size of 6 mm was achieved. The substrate solids content was varied by adding water. On day 122 of operation a change of substrate source was required. Feed material was then collected weekly from the full-scale UK AD site where the inoculum was previously sourced from. Substrate was stored for a maximum of 7 days at 4 °C until the day of its use, when it was warmed to 22–30 °C before feeding to the ADs. Consistent quality of the substrate was then ensured by sieving it through a 6.3 mm aperture size sieve (Endecotts Ltd., London, UK) and homogenising it with a T 25 DS 2 digital ultra-turrax disperser (IKA, Staufen, Germany). The material's pH was raised to ca. 5.7 by addition of sodium hydroxide or calcium carbonate (Fisher Scientific, Loughborough, UK). Micronutrients were added daily into both ADs at a dosing rate of 50 ml of TEA 310 solution per tonne of volatile solids (VS) fed (Omex Environmental Ltd., King's Lynn, UK). An organic loading rate (OLR) of  $2.8 \pm 0.3$  kg VS m<sup>-3</sup> d<sup>-1</sup> was applied with a hydraulic retention time (HRT) of ca. 29 days.

Both ADs were operated in an identical manner until day 148, when enrichment with CO<sub>2</sub> commenced on the test AD. Enrichment with CO<sub>2</sub> was performed by installing a 1 m tall and 10 cm diameter bubble column in the recirculation loop of the test AD (Fig. 1). The column was operated with a 7 L working volume and CO<sub>2</sub> and digestate contact was in a co-current mode (Fig. 1). Injection of CO<sub>2</sub> (g) was performed at the bottom of the column through a perforated plate connected to a manifold that divided the incoming gas stream into seven inlets. A metallic mesh with 0.5 mm hole size was placed on top of the perforated plate, which allowed generation of smaller gas bubbles in order to enhance gas to liquid mass transfer and hence CO<sub>2</sub> dissolution into the digesting material. The CO<sub>2</sub> flowrate into the column was controlled by

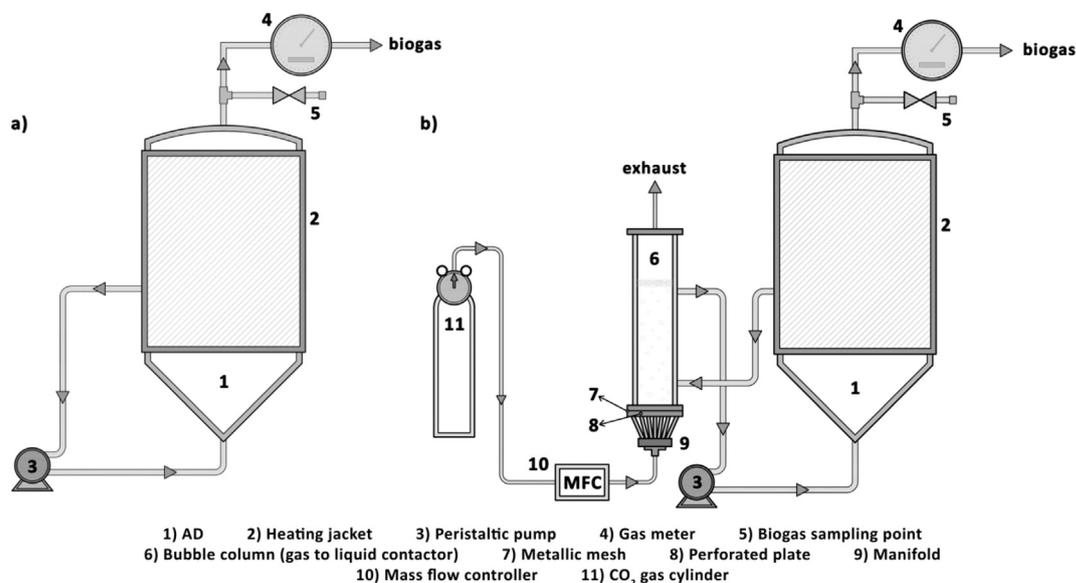


Fig. 1. Schematic representation of the pilot-scale experimental rig. (a) Conventionally operated AD and (b) AD retrofitted with an external bubble column for CO<sub>2</sub> injection. MFC: mass flow controller.

means of a mass flow controller (MFC) (Premier Control Technologies, Norfolk, UK), fixed at  $1.5 \text{ L} \cdot \text{min}^{-1}$  and supplied from gas cylinders (BOC, Manchester, UK). The external bubble column was retrofitted as a side process and connected for each  $\text{CO}_2$  injection, whereas the test AD operated similarly to the control AD during the rest of the time. Injection of  $\text{CO}_2$  was done three times per week, because up to 48 h may be required for the AD to recover from any acidification associated with  $\text{CO}_2$  injection (Bajón Fernández et al., 2014). During each injection the speed of the pump in the AD's recirculation loop was reduced in order to increase the gas to liquid contact time in the column. The whole AD content was contacted with the incoming  $\text{CO}_2$  within an hour. Concentrations of  $\text{CO}_2$  and  $\text{CH}_4$  in the column exhaust (Fig. 1) were monitored with online sensors (BCP sensors, Bluesens, Herten, Germany), which were connected to a logging computer using BacVis software (Bluesens, Herten, Germany) for data recording.

## 2.2. Analytical methods

The volume of biogas produced by the ADs was measured continuously with a TG05/5 gas meter equipped with a totaliser (Ritter, Bochum, Germany) connected to each of the units (Fig. 1). Data of biogas produced were logged daily. The biogas composition was analysed on a daily basis by means of a LMSXi multifunction gas analyser (Gas Data, Coventry, England), which had individual measurement cells for  $\text{CH}_4$ ,  $\text{CO}_2$  and  $\text{H}_2$  concentration. The  $\text{H}_2$  measurement cell was suitable for ranges of 0–1000 ppm. The digestate of each unit was analysed daily for pH and up to twice per week for total solids (TS), VS, ammonia, VFAs and alkalinity. Statistically significant differences were identified by performing *F*-test and *t*-test in order to confirm the rejection of the null hypothesis. Analysis of TS and VS were completed according to the American Public Health Association (2005).

Ammonia and VFAs were analysed in the solid free fraction of the samples. To obtain this fraction, samples were centrifuged in a Falcon 6/300 centrifuge (MSE UK Ltd., London, UK) for 20 min at 4700 g and  $19^\circ\text{C}$ . The supernatant was centrifuged again under the same conditions for 40 min and vacuum filtered through  $1.2 \mu\text{m}$  pore size microfiber filters GF/C (Whatman™, Kent, UK). The solid free fraction was then obtained with a last filtration stage through  $0.45 \mu\text{m}$  pore size syringe-drive filters (Millipore™, Billerica, United States). High performance liquid chromatography (HPLC) (Shimadzu VP Series unit, Milton Keynes, UK) was utilised to quantify the concentration of acetic acid, propionic acid, n-butyric acid, iso-butyric acid, n-valeric acid and iso-valeric acid, whose sum was reported as total VFA (TVFA) concentration ( $\text{TVFA} = \sum \text{individual acids}$ ). An equivalent methodology to that reported by Soares et al. (2010) was used with the exception of the HPLC run time, which was set at 60 min.

Ammonia was quantified with Spectroquant test kits. Alkalinity was measured in the supernatant resulting from a double centrifugation process in which digestate samples were centrifuged at 4700 g for 20 min and the supernatant centrifuged again for 40 min under similar conditions. Partial alkalinity (PA) and intermediate alkalinity (IA) were monitored by titrating to a pH of 5.75 and of 4.30, respectively. Ripley ratio ( $\text{RR} = \text{IA}/\text{PA}$ ) was used as an indicator of digestion stability, with a value lower than 0.3 considered indicative of stable operation (Ripley et al., 1986). When higher values were measured, the OLR was temporarily reduced or the feed's pH further increased and normal feeding resumed after few days. The feed substrate was also characterised for pH, TS, VS, ammonia and VFA on a regular basis. Dissolved levels of  $\text{CO}_2$  were monitored in the control AD, the test AD and the exit of the  $\text{CO}_2$  injection column by utilising an InPro5000(i) dissolved  $\text{CO}_2$  sensor (Mettler Toledo Ltd., Leicester, UK) connected to a multi-parameter

transmitter M400 (Mettler Toledo Ltd., Leicester, UK).

## 3. Results

### 3.1. Substrate characterisation

The pre-conditioned feed substrate (after maceration, homogenisation and solids adjustment) was characterised by an ammonia concentration of  $361 \pm 20 \text{ mg L}^{-1} \text{ NH}_4\text{-N}$ , a TVFA content of  $21,114 \pm 1723 \text{ mg L}^{-1}$ , of which acetic acid was  $81 \pm 5\%$ ; and a pH of  $3.6 \pm 0.5$  (Table 1). The pH was raised to  $5.7 \pm 0.6$  units by addition of sodium hydroxide or calcium carbonate and, when required, water was added to adjust the material's solids concentration. The final substrate contained  $9.7 \pm 1.4\%$  TS, of which the VS were  $83.1 \pm 9.8\%$ .

### 3.2. Assessment of digestion performance (digestate quality and renewable energy enhancement) and process resilience

A stable digestion process was observed in both control and test ADs, with digestate characterised by a pH of  $7.9 \pm 0.2$  and  $7.8 \pm 0.2$  for control and test ADs, respectively, and a total ammonia concentration of  $1.8\text{--}1.9 \text{ g L}^{-1} \text{ NH}_4\text{-N}$  in both cases. Prior to  $\text{CO}_2$  injection a TVFA concentration of  $10,707 \pm 313 \text{ mg L}^{-1}$  with  $67 \pm 3\%$  acetic acid and of  $9470 \pm 739 \text{ mg L}^{-1}$  with  $57 \pm 6\%$  acetic acid were recorded in the control and test ADs, respectively (day 134 of operation). The high VFA levels did not hinder process performance, as previously reported by other authors (Banks et al., 2011) and verified by the  $\text{CH}_4$  output of this study. A specific  $\text{CH}_4$  production rate of  $0.53 \pm 0.16 \text{ m}^3 \text{ CH}_4 \cdot (\text{kg VS}_{\text{fed}} \cdot \text{d})^{-1}$  and a  $\text{CH}_4$  concentration of  $68.3 \pm 5.7\%$  for the control AD and of  $0.45 \pm 0.05 \text{ m}^3 \text{ CH}_4 \cdot (\text{kg VS}_{\text{fed}} \cdot \text{d})^{-1}$  and a  $\text{CH}_4$  content of  $68.8 \pm 3.4\%$  for the test unit were recorded. The variation between units was attributed to the biological nature of the process but was not statistically significant (*p*-value of 0.280).

Control and test ADs were operated in a similar manner until day 148, when  $\text{CO}_2$  injection into the test unit commenced. After  $\text{CO}_2$  enrichment of the test AD commenced,  $\text{CH}_4$  production was  $0.56 \pm 0.13 \text{ m}^3 \text{ CH}_4 \cdot (\text{kg VS}_{\text{fed}} \cdot \text{d})^{-1}$  with a  $\text{CH}_4$  content of  $68.5 \pm 3.4\%$  (Table 1). The average VS reduction of the substrate in the control AD was of 74% and of 76% in the test AD from the time of commencement of  $\text{CO}_2$  enrichment until completion of the trials. Concentration of  $\text{H}_2$  in the headspace was quantified at  $84 \pm 5$  ppm and  $75 \pm 15$  ppm for control and test ADs, respectively, as average up to day 148. However, after four enrichments with  $\text{CO}_2$  (12 days of operation after commencement of  $\text{CO}_2$  enrichment), an increase in the  $\text{H}_2$  content on the test AD was observed, with a value of 173 ppm recorded on day 160 of operation. An increasing trend in  $\text{H}_2$  content of biogas was observed with subsequent  $\text{CO}_2$  injections (Fig. 2), with an average value of the daily recordings of  $320 \pm 153$  ppm between days 149 and 225 of operation (Table 1). This implied a statistically significant 2.5 fold increase (*p*-value  $< 0.001$ ) when compared to the average  $\text{H}_2$  content of the control AD over the entire trial period ( $129 \pm 44$  ppm) (Table 1). A stable operation of the test AD, assessed with RR and  $\text{CH}_4$  production measurements, was observed in spite of the  $\text{H}_2$  concentration reaching values of over 600 ppm on several occasions (Fig. 2).

Interestingly, two failures of the heating system occurred during the experimental period, which enabled investigation of process resilience to disturbances. A failure in the heating jacket of the control AD on day 136 of operation for  $\leq 23$  h led to a temperature drop of  $11^\circ\text{C}$ . An immediate alteration in biogas production was observed, with the  $\text{CH}_4$  production rate dropping by over an order of magnitude and oscillating between  $2.8 \cdot 10^{-2}$  and  $6.1 \cdot 10^{-2} \text{ m}^3$

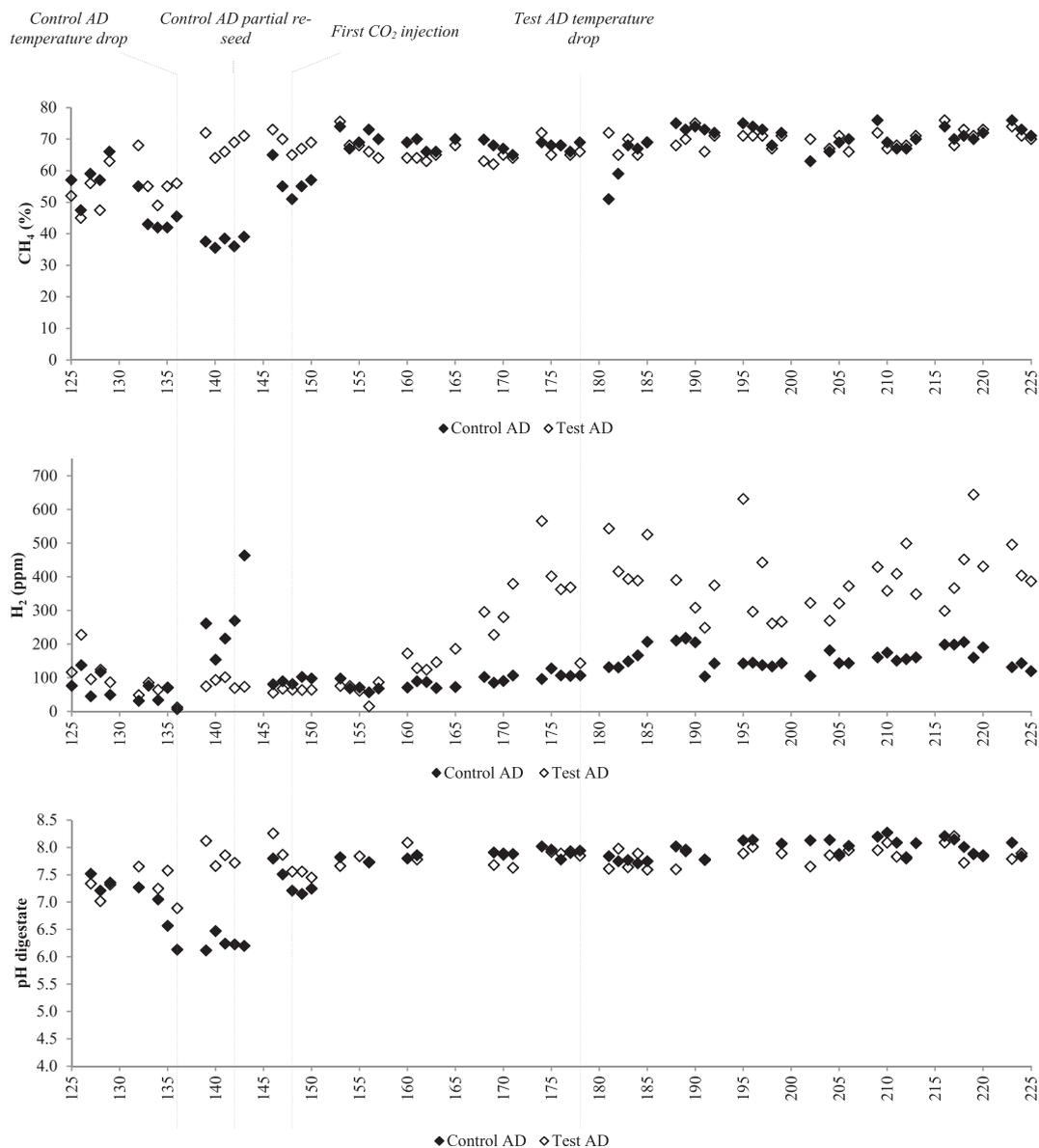
**Table 1**  
Characterisation of digester feed, digestate and headspace of the control and tests ADs.

|  | Feed          | Control AD                | Test AD                           |                                 |
|--|---------------|---------------------------|-----------------------------------|---------------------------------|
|  |               |                           | Before CO <sub>2</sub> enrichment | With CO <sub>2</sub> enrichment |
| <i>Parameter monitoring</i>  |               |                           |                                   |                                 |
| pH   | 5.7 ± 0.6     | 7.9 ± 0.2                 | 7.8 ± 0.2                         | 7.8 ± 0.2                       |
| TS (%)   | 9.7 ± 1.4     | 4.5 ± 0.9                 | 3.1 ± 0.4                         | 4.5 ± 0.8                       |
| VS (% of TS)   | 83.1 ± 9.8    | 51.0 ± 8.9                | 65.8 ± 7.3                        | 44.6 ± 5.3                      |
| VS (% of wet matter)   | 8.0 ± 1.2     | 2.3 ± 0.4                 | 2.0 ± 0.1                         | 2.0 ± 0.5                       |
| Ammonia (mg L <sup>-1</sup> NH <sub>4</sub> -N)  | 361 ± 20      | 1855 ± 205                | 1798 ± 124                        | 1807 ± 166                      |
| VFA concentration (mg L <sup>-1</sup> )  | 21,114 ± 1723 | 10,707 ± 313 <sup>a</sup> | 9470 ± 739 <sup>a</sup>           | 3662 ± 44 <sup>b</sup>          |
| <i>Headspace monitoring</i>  |               |                           |                                   |                                 |
| CH <sub>4</sub> production rate (m <sup>3</sup> CH <sub>4</sub> · (kg VS <sub>fed</sub> · d) <sup>-1</sup> ) | –             | 0.53 ± 0.16               | 0.45 ± 0.05                       | 0.56 ± 0.13                     |
| CH <sub>4</sub> concentration (%)  | –             | 68.3 ± 5.7                | 68.8 ± 3.4                        | 68.5 ± 3.4                      |
| H <sub>2</sub> concentration (ppm)   | –             | 129 ± 44                  | 75 ± 15                           | 320 ± 153                       |

Data corresponding to days of temperature drop have not been considered for average values.

<sup>a</sup> Value on day 134. For VFA dynamics refer to Fig. 3.

<sup>b</sup> Value on day 153. For VFA dynamics refer to Fig. 3.



**Fig. 2.** Evolution of CH<sub>4</sub> and H<sub>2</sub> biogas concentration and digester pH in the control and test ADs during the pilot scale digestion trials.

$\text{CH}_4 \cdot (\text{kg VS}_{\text{fed}} \cdot \text{d})^{-1}$  over the six days following the disturbance. When mesophilic conditions resumed a sudden increase of  $\text{H}_2$  was experienced, which ranged from 262 ppm on the day of temperature correction to 464 ppm 7 days afterwards. A RR of 11.5 was measured on day 139 of operation, in spite of substrate addition being suspended immediately after the temperature drop. Analysis of VFA on day 139 evidenced an accumulation of acids, with TVFA reaching values of  $17,235 \pm 147 \text{ mg L}^{-1}$  as opposed to the  $10,707 \pm 313 \text{ mg L}^{-1}$  recorded on day 134 and the  $9435 \pm 686 \text{ mg L}^{-1}$  obtained in the test AD (Fig. 3). Individual VFA increases when comparing day 139 over day 134 of operation of 1.4, 1.4, 3.8 and 4.3 fold were recorded for acetic, propionic, iso-butyric and n-butyric acids, respectively (Fig. 3). Calcium carbonate was dosed in order to increase the pH, which had dropped to ca. 6 units. However, the high  $\text{CO}_2$  content of the headspace (up to 65%), the inhibited  $\text{CH}_4$  production rate, the RR considerably over the desired value of 0.3 and the rise in  $\text{H}_2$  content indicated that methanogenesis activity was severely inhibited. Acetogenesis and methanogenesis were hence considered to be completely decoupled and a partial re-seed (80%) of the control AD was required on day 142 of operation. Signs of recovery were observed from day 146, when a  $\text{H}_2$  content in biogas of 81 ppm and a pH in digestate of 7.8 were recorded. A performance comparable to that previous to the temperature disturbance was achieved within 11 days from the partial re-seed (day 154 of operation), when a  $\text{CH}_4$  production rate of  $0.46 \text{ m}^3 \text{ CH}_4 \cdot (\text{kg VS}_{\text{fed}} \cdot \text{d})^{-1}$  was obtained. Alkalinity tests confirmed the recovery of digester stability since a RR of 0.82 on day 153 and of 0.30 on day 161 were recorded. Concentration of TVFA was  $6969 \pm 591 \text{ mg L}^{-1}$  on day 161.

The heating system of the test AD failed for  $\leq 23 \text{ h}$  on day 178 (31 days after  $\text{CO}_2$  enrichment was started), leading to a temperature drop of  $12.5 \text{ }^\circ\text{C}$ . Similar to the experience in the control AD, an immediate reduction in  $\text{CH}_4$  production rate and a drop in the pH of the digesting media were recorded. Nevertheless, alterations were significantly lower than those previously recorded for the control AD, with a  $\text{CH}_4$  production rate decrease from 0.67 to  $0.46 \text{ m}^3 \text{ CH}_4 \cdot (\text{kg VS}_{\text{fed}} \cdot \text{d})^{-1}$  and a pH drop of 0.24 units. An increase of TVFA concentration from  $2473 \pm 153 \text{ mg L}^{-1}$  to  $4764 \pm 145 \text{ mg L}^{-1}$  between days 175 and 178 was recorded, which was within the normal variability observed during the entire digestion period (Fig. 3). No specific sign of VFA accumulation associated with methanogenic activity inhibition was obtained. The immediate restart of the heating jacket and the suspension of substrate addition during two days sufficed to recover the initial digestion performance, without any re-seeding required. A  $\text{CH}_4$  production rate of 0.71 and  $0.75 \text{ m}^3 \text{ CH}_4 \cdot (\text{kg VS}_{\text{fed}} \cdot \text{d})^{-1}$  was recorded on days 182 and 183 of operation, respectively. The different behaviour of the control and test ADs to a situation of stress for methanogenic communities could be associated with a higher resistance of these *Archaea* in ADs retrofitted with  $\text{CO}_2$  injection. This potential benefit of  $\text{CO}_2$  injection in process resilience could prove beneficial in food waste ADs, which are commonly associated with operational problems (Banks et al., 2011), and should be further investigated. Concentration of  $\text{H}_2$  in the test AD on recovery from the temperature drop stabilised again to a baseline higher than that of the control unit, with values oscillating between 380 and 550 ppm of  $\text{H}_2$  during the week following recovery (Fig. 2).

### 3.3. Impact of $\text{CO}_2$ enrichment on dissolved $\text{CO}_2$ and digestate's ammonia concentration

The dissolved  $\text{CO}_2$  and ammonia levels recorded in the digestate of both control and test ADs are presented in Fig. 4. An average dissolved  $\text{CO}_2$  concentration of  $2.0\text{E-}3 \pm 5.3\text{E-}4 \text{ kmol m}^{-3}$  was recorded in the liquid phase of the control AD during the digestion

trial. A similar value of  $2.0\text{E-}3 \pm 5.9\text{E-}4 \text{ kmol m}^{-3}$  was obtained for the test AD when measuring dissolved  $\text{CO}_2$  inside of the unit or in the inlet to the bubble column used for mass transfer. Each enrichment with  $\text{CO}_2$  led to dissolved  $\text{CO}_2$  levels of  $6.1\text{E-}3 \pm 1.4\text{E-}3 \text{ kmol m}^{-3}$  in the material exiting the bubble column (Fig. 4). Hence, all the content of the test AD was enriched with an additional  $4.0\text{E-}3 \text{ kmol CO}_2 \cdot \text{m}^{-3}$ , implying an input of ca.  $4.0\text{E-}4 \text{ kmol}$  of exogenous  $\text{CO}_2$  ( $18,455 \text{ mg CO}_2$ ) per enrichment when considering the working volume of the unit (106 L). When a frequency of injection of three times per week is considered, it is calculated that 0.55 kg of exogenous  $\text{CO}_2$  were assimilated by the test AD during the trial period. Monitoring the dissolved  $\text{CO}_2$  concentration confirmed the rapid utilisation of additional  $\text{CO}_2$ , since levels dropped from  $6.1\text{E-}3 \pm 1.4\text{E-}3 \text{ kmol m}^{-3}$  obtained after enrichment to  $2.0\text{E-}3 \pm 5.9\text{E-}4 \text{ kmol m}^{-3}$  within 24 h. This utilisation rate of  $\text{CO}_2$  matched the overcome of the slight acidification due to  $\text{CO}_2$  enrichment in the 24 h following a  $\text{CO}_2$  injection. During each  $\text{CO}_2$  injection a pH drop of 0.4–0.6 units was experienced between the inlet and outlet of the bubble column. This pH reduction was consistently overcome within 24 h, with an average pH of  $7.9 \pm 0.2$  and  $7.8 \pm 0.2$  maintained in control and test ADs, respectively. This implied no alteration of the pH with respect to the period when  $\text{CO}_2$  injection was not applied.

The suitability of utilising a co-currently operated bubble column for gas to liquid mass transfer was assessed by examining the concentration of  $\text{CO}_2$  in the AD's headspace and the amount of  $\text{CH}_4$  stripped from the digesting material during the mass transfer process. The  $\text{CO}_2$  content in the produced biogas, which was recorded daily, did not increase; with non-dissolved  $\text{CO}_2$  being released with the exhaust of the bubble column only. The extent to which  $\text{CH}_4$  was degassed during the mass transfer process was quantified by measuring on-line the  $\text{CH}_4$  content of the gas exhaust of the bubble column (Fig. 4). Concentrations between 0.8 and 2.1% of  $\text{CH}_4$  were measured, which implied a release of 0.72–1.89 L  $\text{CH}_4$  per  $\text{CO}_2$  enrichment (every 48 h) when considering the incoming  $\text{CO}_2$  flowrate of  $1.5 \text{ L min}^{-1}$ . When compared to the average of  $235 \pm 49 \text{ L CH}_4$  produced per day by the test AD, the loss of  $\text{CH}_4$  in the mass transfer system accounted for  $\leq 0.4\%$  and was hence considered to be negligible.

Periodic injections of  $\text{CO}_2$  in the test AD did not vary the concentration of ammonia in the digesting material to a significant extent (Table 1). Average total ammonia concentration was  $1798 \pm 124 \text{ mg L}^{-1} \text{ NH}_4\text{-N}$  before  $\text{CO}_2$  enrichment and  $1807 \pm 166 \text{ mg L}^{-1} \text{ NH}_4\text{-N}$  during the rest of the digestion trials. This seems to indicate that injection of  $\text{CO}_2$  did not have a significant positive benefit in controlling ammonia inhibition, which agrees with previous literature stating that increased pH and temperatures are required for an efficient free ammonia removal in ADs by stripping it with biogas (Serna-Maza et al., 2014; Walker et al., 2011).

## 4. Discussion

### 4.1. Suitability of injecting $\text{CO}_2$ into ADs with an external bubble column

The majority of previous studies investigating  $\text{CO}_2$  injection into ADs have been completed at laboratory scale only, without the suitability of injecting  $\text{CO}_2$  through existing gas mixing systems or by means of external mass transfer units having been investigated for scaled-up systems. This study provides an insight into the effectiveness of using an external bubble column to inject  $\text{CO}_2$  in ADs through examining biogas quality, amount of  $\text{CH}_4$  lost and mass transfer efficiency. Non-dilution of AD headspace and the low amount of  $\text{CH}_4$  degassed during the enrichment ( $\leq 0.4\%$ ), indicated

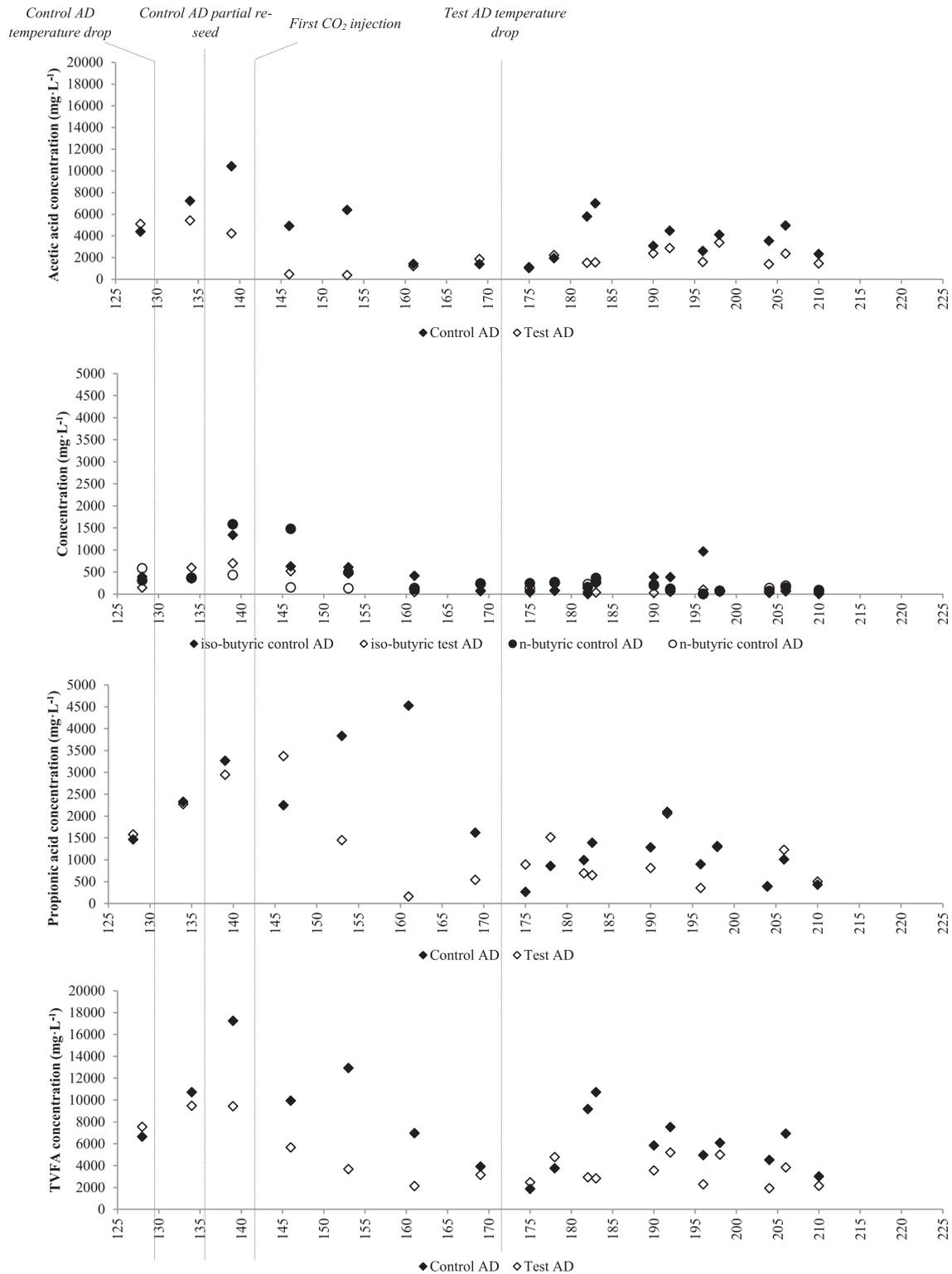
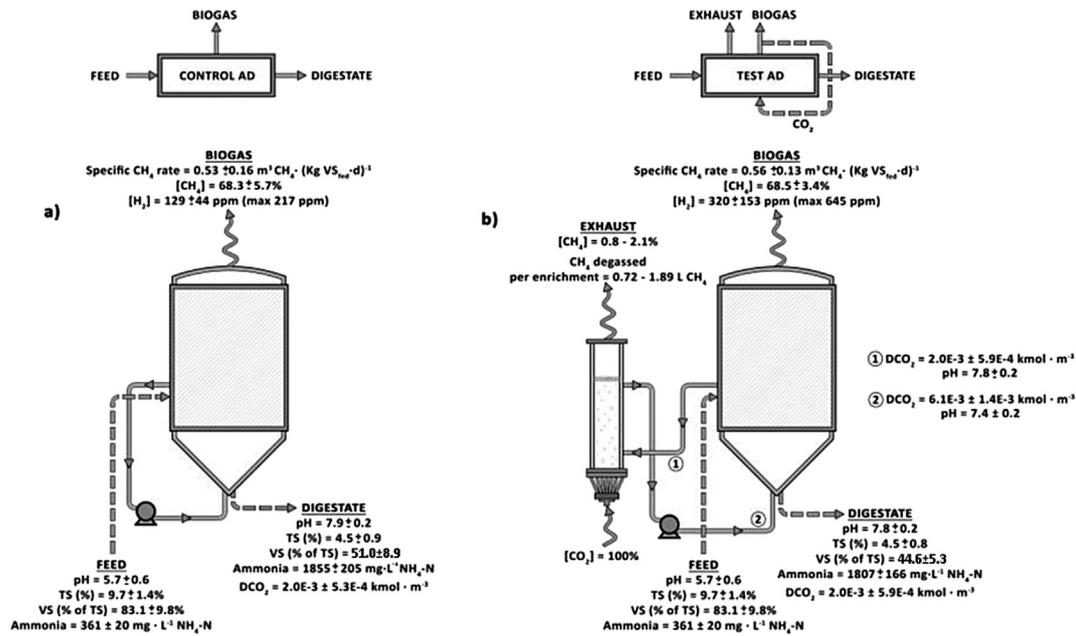


Fig. 3. Dynamics of total and individual VFA digestate concentrations for control and test ADs during the pilot scale digestion trials.

the suitability of employing an external bubble column for performing the required gas to liquid mass transfer. As far as efficiency of the system is concerned, operation of the bubble column increased the dissolved  $\text{CO}_2$  levels by a 3 fold (from  $2.0\text{E-}3 \pm 5.9\text{E-}4 \text{ kmol m}^{-3}$  to  $6.1\text{E-}3 \pm 1.4\text{E-}3 \text{ kmol m}^{-3}$ ). However, the solubility of  $\text{CO}_2$  at  $38.5^\circ\text{C}$  in aqueous solutions with a  $\text{CO}_2$  partial pressure

( $p_{\text{CO}_2}$ ) of  $1\text{atm}$  is  $2.4\text{E-}2 \text{ kmol m}^{-3}$  ( $1071 \text{ mg L}^{-1}$ ) (Green and Perry, 2008). Therefore, the operated bubble column achieved only ca. 25% of the  $\text{CO}_2$  that could have been dissolved at  $p_{\text{CO}_2}$  of  $1\text{atm}$ . This indicates the important role that  $\text{CO}_2$  gas to liquid mass transfer plays in the amount of  $\text{CO}_2$  which can be dissolved in an anaerobic process when implementing  $\text{CO}_2$  enrichment. In turn the amount of



**Fig. 4.** Schematic summary of performance of (a) control and (b) test ADs in terms of digestate quality and biogas production. Recorded dissolved  $\text{CO}_2$  concentrations are also included.

$\text{CO}_2$  dissolved determines the contribution towards reduction of carbon footprint that can be achieved (negative carbon release if dissolving  $\text{CO}_2$  with biogenic origin) and the potential increase in renewable energy production. The complex rheology of anaerobically digested material (Baudez et al., 2011; Eshtiaghi et al., 2012) and the strong impact of viscosity on mass transfer retardation (Ozbek and Gayik, 2001) requires a better understanding in order for mass transfer systems involving these fluids to be designed and operated in an efficient manner. The use of bubble columns for dissolving exogenous  $\text{CO}_2$  into anaerobic digesting media is considered suitable because of a lower risk of clogging than other technologies. Besides, efficiency of mass transfer could be increased by a greater gas to liquid contact time, a reduced bubble size or a higher incoming gas flowrate (Kantarci et al., 2005), which would increase dissolved  $\text{CO}_2$  levels and hence potential for carbon assimilation.

#### 4.2. Impact of $\text{CO}_2$ injection in AD performance and mechanisms of utilisation based on VFA and $\text{H}_2$ dynamics

The test AD achieved an average  $\text{CH}_4$  production rate of  $0.45 \pm 0.05 \text{ m}^3 \text{ CH}_4 \cdot (\text{kg VS}_{\text{fed}} \cdot \text{d})^{-1}$  before any  $\text{CO}_2$  was applied, which is within the order of magnitude reported in the literature for domestic food waste (Banks et al., 2011). When this value is considered as a baseline, the  $\text{CH}_4$  production rate observed during the time when  $\text{CO}_2$  enrichment was applied ( $0.56 \pm 0.13 \text{ m}^3 \text{ CH}_4 \cdot (\text{kg VS}_{\text{fed}} \cdot \text{d})^{-1}$ ) implied a ca. 20% improvement ( $p$ -value of 0.058), which is in agreement with performances previously reported in the literature (Salomoni et al., 2011; Sato and Ochi, 1994). However, no significant benefit ( $p$ -value of 0.261) was recorded when comparing the performance of the test AD with  $\text{CO}_2$  enrichment ( $0.56 \pm 0.13 \text{ m}^3 \text{ CH}_4 \cdot (\text{kg VS}_{\text{fed}} \cdot \text{d})^{-1}$ ) with that of the control unit ( $0.53 \pm 0.16 \text{ m}^3 \text{ CH}_4 \cdot (\text{kg VS}_{\text{fed}} \cdot \text{d})^{-1}$ ). This suggests that any improvement was not appreciable due to the natural variability of the performance of the biological process (i.e. high standard deviation).

Of note was the impact observed in relation to the  $\text{H}_2$  content of the biogas produced, which reached a baseline 2.5 fold higher in

the test AD than in the control unit ( $p$ -value  $< 0.001$ ) during the period when  $\text{CO}_2$  was periodically injected. The observed increased  $\text{H}_2$  production can be used to further understand the mechanisms of  $\text{CO}_2$  utilisation because of the role of  $\text{H}_2$  as an electron carrier and intermediate product in several reactions of the digestion process (Cord-Ruwisch et al., 1997). Sudden increases in  $\text{H}_2$  concentration have been reported in response to process disturbances, such as changes in the feed quality or loading rate (Kidby and Nedwell, 1991; Mosey and Fernandes, 1989) and when feeding a digestion process with unfermented material of a labile nature (Kidby and Nedwell, 1991). The sudden increase in readily available substrate in turn leads to an active hydrolysis, acidogenesis and acetogenesis with an associated release of  $\text{H}_2$  (Guwy et al., 1997). The fast response to system destabilizations and the recovery of initial  $\text{H}_2$  levels shortly after the disturbance is overcome, has led several authors to study the possibility of using it as a control parameter in ADs (Rodríguez et al., 2006). Fluctuations in  $\text{H}_2$  with return to initial concentrations are hence considered indicative of specific events or transition phenomena, rather than of long-term alterations (Mosey and Fernandes, 1989).

During the pilot plant trials of this study two types of disturbances in biogas  $\text{H}_2$  levels were observed. An increase from  $84 \pm 5 \text{ ppm}$  to  $464 \text{ ppm}$  was recorded in the control AD when the temperature dropped by  $11 \text{ }^\circ\text{C}$  (Fig. 2) over a  $\leq 23 \text{ h}$  period, with  $\text{H}_2$  rapidly returning to initial levels once the disturbance was overcome. On the contrary, an increase in  $\text{H}_2$  concentration was observed in the test AD following four  $\text{CO}_2$  injections, which led to a new  $\text{H}_2$  baseline ( $320 \pm 153 \text{ ppm}$ ) to be maintained during the rest of the trial period and to sporadic peaks of up to  $645 \text{ ppm}$  (Fig. 2). The rapid variation in  $\text{H}_2$  level in the control AD was an indicator of process disturbance. This was overcome when normal operation conditions were re-established and agrees with the previously mentioned literature findings. The increased  $\text{H}_2$  production of the test AD, however, was maintained over 65 days of operation (until the experimental trials were concluded), and was assumed to be associated with  $\text{CO}_2$  injection affecting the microbial process in a more permanent manner. The different nature of both  $\text{H}_2$  alternations was further evident when attending to the

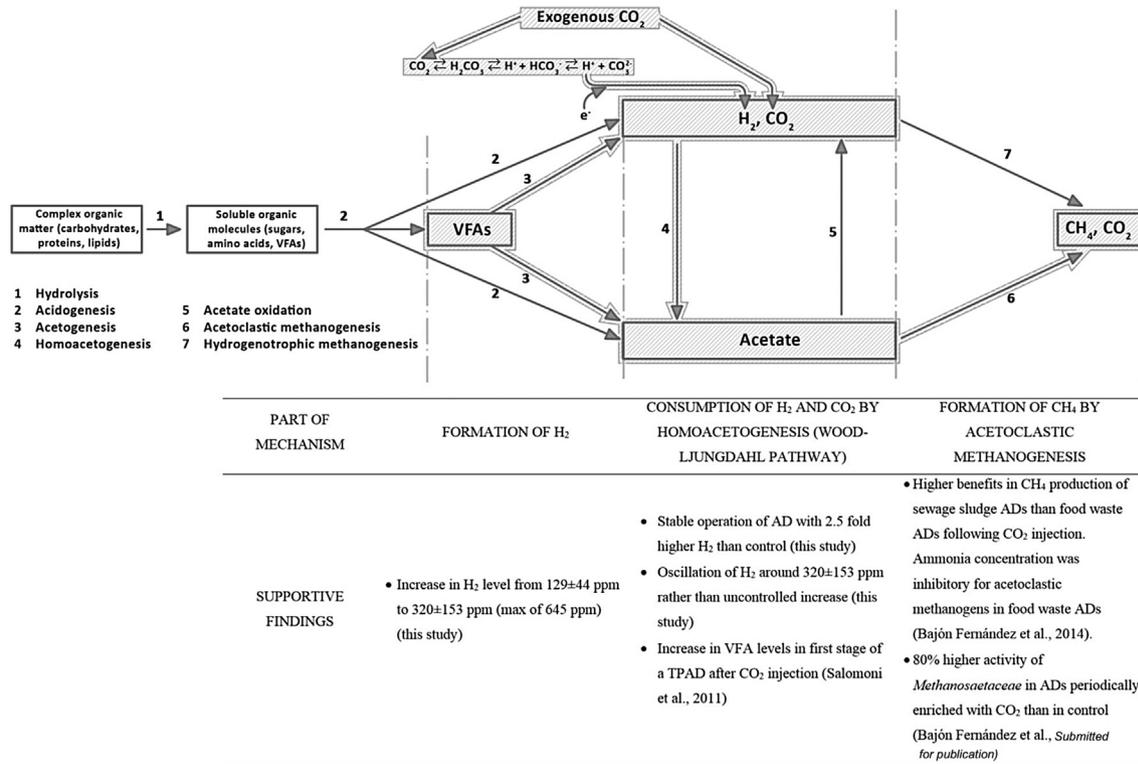


Fig. 5. Hypothesised mechanism of exogenous CO<sub>2</sub> utilisation in ADs, with findings supporting each of the proposed stages.

dynamics of VFA speciation within the AD. The increase in H<sub>2</sub> concentration of the control AD was simultaneous to a sudden increase in TVFA concentration (Fig. 3), which reached  $17,235 \pm 147 \text{ mg} \cdot \text{L}^{-1}$  on day 139. Accumulation of VFA indicated that hydrolysis, acidogenesis and acetogenesis were taking place in spite of the temperature drop, while the acid assimilatory capacity of methanogenic communities was inhibited. Progression of fermentation without an efficient assimilation of acetic acid and H<sub>2</sub> would have resulted in unfavourable conditions for acetogenesis itself, leading to accumulation of VFAs of higher number of carbons (Fig. 3). Propionic and butyric acid degradation reactions have been reported to be thermodynamically unfavoured at H<sub>2</sub> partial pressure ( $p_{\text{H}_2}$ ) over  $10^{-4}$  atm and  $10^{-3}$  atm, respectively (Cord-Ruwisch et al., 1997; Harper and Pohland, 1986; Kidby and Nedwell, 1991; Labatut et al., 2014). The  $p_{\text{H}_2}$  in the control unit reached these unfavourable conditions, with a value of  $4.6 \cdot 10^{-4}$  atm (atmospheric pressure considered inside the AD). This in turn lead to a hindered degradation of propionic and butyric acids, which accumulated on the system reducing the digester's pH (Fig. 2). Eventually process failure occurred (sour AD) and a partial re-seed for stability recovery was required.

On the contrary, the increase in H<sub>2</sub> concentration in the test AD was not related to a rising trend in TVFA or individual VFA concentrations (Figs. 2 and 3). In fact, TVFA and acetic acid were quantified at  $3662 \pm 44 \text{ mg} \cdot \text{L}^{-1}$  and  $369 \pm 18 \text{ mg} \cdot \text{L}^{-1}$ , respectively, on day 153, which was lower than average values maintained during the entire digestion trials (Fig. 3). The increase in H<sub>2</sub> was considered resulting from injection of CO<sub>2</sub> (only variable modified) and was attributed to a boost of H<sub>2</sub> producing mechanisms rather than to a reduced H<sub>2</sub> assimilatory capacity. Two mechanisms could have led to the increased H<sub>2</sub> production observed. On the one hand, dissolution of CO<sub>2</sub> in the aqueous media could have contributed to an increased H<sub>2</sub> concentration as a result of CO<sub>2</sub> forming carbonic acid that releases protons when dissociated into carbonate and

bicarbonate species. At the low oxidation reduction potential found in ADs ( $< -200 \text{ mV}$  (Gupta et al., 1994)) the protons could react to form H<sub>2</sub>. On the other hand, the H<sub>2</sub> increase could have resulted from its production by acetogenesis (Fig. 5). In this case, an increase in acetic acid would have been expected, similar to that recorded in the control unit, unless the acetic acid assimilatory capacity of the system was enhanced. The activity of *Methanosaetaceae* (obligate acetoclastic methanogen) has been reported to increase after periodic CO<sub>2</sub> injections in ADs (Bajón Fernández et al., Submitted for publication b), hence being likely to have had the capacity to assimilate additional acetate. Further investigation needs to be undertaken to determine the contribution of both pathways to the formation of additional H<sub>2</sub>. By either mechanism the additional H<sub>2</sub> would have been formed in the liquid phase. The limited mass transfer of H<sub>2</sub> between the liquid and gas phases (Guwy et al., 1997) explained that four injections of CO<sub>2</sub> were required before an impact in the headspace's H<sub>2</sub> content was evident and that pH was recovered between injections while H<sub>2</sub> levels did not drop to the baseline of the control AD.

It is of note that the H<sub>2</sub> concentration oscillated around  $320 \pm 153 \text{ ppm}$ , with peaks over 600 ppm but without a continuously increasing trend in spite of CO<sub>2</sub> being injected periodically. The fact that H<sub>2</sub> concentration did not increase further, suggests that additional H<sub>2</sub> produced was consumed in the AD. Assimilation of H<sub>2</sub> could occur by the Wood-Ljungdahl pathway of CO<sub>2</sub> fixation. This metabolic pathway can be stimulated by the availability of exogenous CO<sub>2</sub> (Misoph and Drake, 1996) and requires eight electrons and eight protons for each two molecules of CO<sub>2</sub> assimilated, which can be supplied by consumption of H<sub>2</sub> (Ragsdale and Pierce, 2008). This pathway leads to the generation of acetate, which in turn would have been assimilated by the enhanced acetoclastic methanogenesis.

It is then proposed that CO<sub>2</sub> leads to a boost of H<sub>2</sub> production, derived from the protons formed when dissolving CO<sub>2</sub> in the

aqueous media, from a boost of obligate acetogenesis or from a combination of both (Fig. 5). Part of the additional H<sub>2</sub> formed is then assimilated in the AD, leading to a steady operation as opposed to a continuously increasing H<sub>2</sub> level. Assimilation of additional H<sub>2</sub> is likely to occur through the Wood-Ljungdahl pathway, which has a preference for exogenous CO<sub>2</sub>. The additional acetic acid formed by this pathway would then be assimilated by acetoclastic methanogenesis, which has been reported to have an increased activity when subjected to periodic CO<sub>2</sub> injections. The proposed mechanism of CO<sub>2</sub> assimilation is summarised in Fig. 5, including previous findings that support the suggested hypothesis. Further work will be required to support or reject the proposed mechanism. In particular, microbial community analyses to understand the potential impact of CO<sub>2</sub> injection in acetogenesis are of great interest.

## 5. Conclusion

The capacity of ADs treating food waste to utilise exogenous CO<sub>2</sub> was tested and the practicalities of an up-scaled implementation and mechanisms of CO<sub>2</sub> utilisation were investigated. Injection of CO<sub>2</sub> through an external bubble column was suitable, as the headspace was not diluted and CH<sub>4</sub> loss during injection was negligible ( $\leq 0.4\%$ ). A CH<sub>4</sub> production rate of  $0.56 \pm 0.13 \text{ m}^3 \text{ CH}_4 \cdot (\text{kg VS}_{\text{fed}} \cdot \text{d})^{-1}$  was recorded for an AD periodically enriched with CO<sub>2</sub>. An additional uptake of 0.55 kg of exogenous CO<sub>2</sub> in the test AD during the trial period was calculated, which could be augmented if the bubble column mass transfer efficiency was increased, hence augmenting the potential benefits in CO<sub>2</sub> mitigation. A 2.5 fold increase in H<sub>2</sub> concentration was observed after four CO<sub>2</sub> injections, likely due to CO<sub>2</sub> dissolution or an alternation of acidogenesis/acetogenesis. Additional H<sub>2</sub> was believed uptaken by Wood-Ljungdahl pathway and the acetate generated by this in turn assimilated by an increased activity of obligate acetoclastic *Archaea*. This proposed hypothesis of exogenous CO<sub>2</sub> conversion requires verification with microbial community analysis.

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