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THz spectroscopy and first ISM detection of excited torsional states of ¹³C-methyl formate*

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ABSTRACT

Context. An astronomical survey of interstellar molecular clouds needs a previous analysis of the spectra in the microwave and sub-mm energy range of organic molecules to be able to identify them. We obtained very accurate spectroscopic constants in a comprehensive laboratory analysis of rotational spectra. These constants can be used to predict the transitions frequencies very precisely that were not measured in the laboratory.

Aims. We present the experimental study and its theoretical analysis for two 13 C-methyl formate isotopologues to detect these two isotopologues for the first time in their excited torsional states, which lie at 130 cm⁻¹ (200 K) in Orion-KL.

Methods. New spectra of HCOO¹³CH₃ ($^{13}C_2$) methyl formate were recorded with the mm- and submm-wave spectrometer in Lille from 50 to 940 GHz. A global fit for $v_t = 0$ and 1 was accomplished with the BELGI program to reproduce the experimental spectra with greater accuracy.

Results. We analysed 5728 and 2881 new lines for $v_t = 0$ and 1 for HCOO¹³CH₃. These new lines were globally fitted with 846 previously published lines for $v_t = 0$. In consequence, 52 parameters of the RAM Hamiltonian were accurately determined and the value of the barrier height ($V_3 = 369.93168(395)$ cm⁻¹) was improved. We report the detection of the first excited torsional states ($v_t = 1$) in Orion-KL for the ¹³C₂ and ¹³C₁ methyl formate based on the present analysis and previously published data. We provide column densities, isotopic abundances, and vibrational temperatures for these species.

Conclusions. Following this work, accurate prediction can be provided. This permits detecting 135 features of the first excited torsional states of ¹³C-methyl formate isotopologues in Orion-KL in the 80–280 GHz frequency range, without missing lines.

Key words. line: identification – astronomical databases: miscellaneous – ISM: molecules – submillimeter: ISM – astrochemistry – ISM: individual objects: Orion KL

1. Introduction

Methyl formate is one of the most abundant complex organic molecules in the interstellar medium (ISM). It was identified in different sources since 1975 (Brown et al. 1975; Churchwell et al. 1975); nearly one thousand lines were detected in the ground-torsional state $v_t = 0$. Its abundance is particularly high in Orion-KL: it was detected up to 900 GHz (Comito et al. 2005) far from the maximum of absorption situated around 300 GHz; some lines from the first (Kobayashi et al. 2007), and more recently, lines from the second torsional state were detected (Takano et al. 2012). The abundance is also fairly high in W51e2, where lines from methyl formate in its first torsional states were found (Demyk et al. 2008). Molecules in the ISM can be used to probe the physical conditions of sources, as illustrated recently by Favre et al. (2011, 2014) using methyl formate to probe the temperature structure and spatial distribution of this species in Orion-KL; depending on the molecular species, the distribution of the molecular gas can be different in this source because of its complexity (which is a result of massive star formation processes) different cloud components coexist there. Organic saturated O-rich molecules such as methyl formate generally trace the compact ridge component (for a description of the different cloud components of Orion-KL see e.g. Blake et al. 1987; Schilke et al. 2001; Persson et al. 2007; Tercero et al. 2010, 2011; Neill et al. 2013).

Around 180 molecules have been detected in the ISM or circumstellar shells, but in high mass-forming regions such as Orion-KL, 30% of the lines remain unidentified (Esplugues et al. 2013a; Tercero et al. 2010). Its important to identify all the lines from the most abundant species. These molecules are often called "weeds", and methyl formate is one of them.

^{*} Full Table A.1 and the IRAM spectra as FITS files are only available at the CDS via anonymous ftp to cdsarc.u-strasbg.fr (130.79.128.5) or via

http://cdsarc.u-strasbg.fr/viz-bin/qcat?J/A+A/568/A58

Identifying them is necessary in order to discover new species in the ISM without ambiguity. As shown recently by Tercero et al. (2013), the detection of methyl acetate (CH₃COOCH₃) and the gauche conformer of ethyl formate (g-CH₃CH₂OCOH) was only possible with the assignment of 4400 lines coming from several isotopologues and vibrational levels of different weeds: CH₃CH₂CN, CH₂CHCN, HCOOCH₃, and NH₂CHO.

Most of the times, accurate spectroscopic data in the millimeter and submillimeter-wave domain are not available for the isotopologues and vibrational levels. While the normal species of methyl formate were extensively studied (Ilyushin et al. 2009, and references therein) of only the two commercially available isotopologues H¹³COOCH₃ (Maeda et al. 2008a,b) and DCOOCH₃ (Oesterling et al. 1995) were studied in the millimeter-wave region. Therefore we decided to investigate all the mono-isotopic species of methyl formate in the submillimeter-wave domain a few years ago. For each of them, this permits generating a new accurate prediction and allows detecting the ¹³C species (Willaert et al. 2006; Carvajal et al. 2007, 2009, 2010), DCOOCH₃ (Margulès et al. 2010), ¹⁸O species (Tercero et al. 2012) in Orion-KL and recently HCOOCH₂D (Coudert et al. 2013). This confirms the assumption that most of the U-lines come from known molecules. The lines detected for the ¹⁸O species gave us the idea that it might also be possible to find lines related to the first excited torsional states that lie at 130 cm^{-1} (200 K) of the most abundant isotopologues ¹³C.

Our motivations for studying of the first excited torsional state of HCOO¹³CH₃ is also linked to a spectroscopic interest. Methyl formate, like other complex molecules, exhibits largeamplitude motion: torsion of the methyl group related to the rest of the molecules. A specific code that takes into account the interaction of the torsion and the overall rotation of the molecule is needed. Most of the codes were developed to treat data in the centimeter not in millimeter-wave domain. Recent efforts from different theoretical groups were recently make to improve the codes to reproduce the experimental accuracies with high quantum numbers that are reached at submillimeter frequencies. As shown for the normal species (Ilyushin et al. 2009) and the $^{13}C_1$ species (Carvajal et al. 2010), the global treatment of the ground- and first torsional states permits removing the high correlation between torsional parameters. This improves the quality of the fit and finally the accuracy of the predictions for the two states.

These works permit the first detection of the excited torsional state of the two ¹³C-methyl formate isotopologues. As in Carvajal et al. (2009), we publish the spectroscopic work about HCOO¹³CH₃ and the detection of the two ¹³C isotopologues in Orion together. In the previous paper about ¹³C₂ (Carvajal et al. 2009), our measurements were limited to 660 GHz for the ground-torsional states. To provide an accurate prediction of this state in the ALMA range, we also reinvestigated this state up to 940 GHz.

2. Experiments

2.1. Synthesis of the methyl formate isotopologue HCOO¹³CH₃

The details about synthesis and identification by NMR spectroscopy were described in Carvajal et al. (2009).

2.2. Lille – submillimeter spectra

The millimeter- and submillimeter-wave spectra were recorded using the Lille spectrometer that is based on solid-state sources (Motiyenko et al. 2010; Haykal et al. 2013a). The sample pressure was in the range $20-30 \times 10^{-6}$ bars. Spectra were recorded at room temperature (T = 294 K) in the 150–210, 225–315, 400–500, 500–630, and 780–940 GHz regions with frequency steps of 30, 36, 48, 54, and 76 kHz and an acquisition time of 35 ms per point. Absorption signals were detected either by a Schottky diodes detector (Virginia Diodes Inc.) below 315 GHz or by an InSb liquid He-cooled bolometer (QMC Instruments Ltd.) above 400 GHz, and were processed on a computer. The absolute accuracy of the line-centre frequency is estimated to be better than 30 kHz (50 kHz above 700 GHz) for isolated lines and can be as low as 100 kHz (150 kHz above 700 GHz) for blended or very weak lines.

3. Assignments and fit of the HCOO¹³CH₃ spectrum

The theoretical model and the so-called RAM method (rho-axis method) used for the present spectral analyses and fits of the ground- and first torsional-excited states of HCOO13CH3 species have also been used previously for a number of molecules that contain an internal methyl rotor (see for example Ilyushin et al. 2003, 2008) and in particular for the normal species of the cismethyl formate (Carvajal et al. 2007; Ilyushin et al. 2009), the two ¹³C species (Carvajal et al. 2009, 2010), for DCOOCH₃ (Margulès et al. 2010), and for ¹⁸O species (Tercero et al. 2012). The RAM Hamiltonian we used is based on previous works (Kirtman 1962; Lees & Baker 1968; Herbst et al. 1984). Because this method has been presented in great detail in the literature (Lin & Swalen 1959; Hougen et al. 1994; Kleiner 2010), we do not describe it here. The principal advantage of the RAM general approach used in our code BELGI, which is publicly available¹, is its general approach that simultaneously takes into account the A- and E-symmetry states in our fit. All the torsional levels up to a truncation limit of $v_t = 8$ are carefully tested and the interactions within the rotation-torsion energy levels are also included in the rotation-torsion Hamiltonian matrix elements (Kleiner et al. 1996). The various rotational, torsional, and coupling terms between rotation and torsion that we used for the fit of the HCOO¹³CH₃ species were defined previously for the normal methyl formate species (Carvajal et al. 2007). The labelling scheme of the energy levels and transitions used for methyl formate was also described in the same reference, as was the connection with the more traditional J_{K_a,K_c} labelling.

We started with the assignment of the ground state. Predictions were provided by the BELGI program based on the set of 27 parameters previously determined in Carvajal et al. (2009). In this reference, only 658 lines were assigned from 150 to 700 GHz. Our new spectrometer has a wider coverage range than that used in 2009. Based on the previous work, we easily assigned the spectra up to J = 63 and $K_a = 34$ in the range 75–630 GHz. In the highest frequency range of the mm-and submm- wave spectrometer (700–940 GHz), the assignment was tentative because of the shift of the predicted lines from the experimental lines. The upper limit in frequency of the 936 previously fitted lines was 700 GHz. As a result, we assigned for the ground-state 5728 lines up to J = 60 and $K_a = 35$ from 50 to 940 GHz. Next, we started up again, but simultaneously including data of states $v_t = 0$ and $v_t = 1$. The fit was carried out in

¹ The source code for the fit, an example of input data file, and a readme file are available at the web site (http://www.ifpan.edu. pl/~kisiel/introt/introt.htm#belgi) managed by Zbigniew Kisiel. Extended versions of the code made to fit transitions with higher *J* and *K* are also available from the authors (I. Kleiner and M.C.).



Fig. 1. A-E splitting of selected branches (R branches for $K_a = 0$; 1 in blue and 1;2 in black) for $v_t = 1$. The *x*-axis represents the situation without perturbation. The A transitions (∇ signs) are twice as far as the E transitions (+ signs). For $K_a = 0$; 1 the A-E splitting changes sign at J = 44 (Ilyushin et al. 2009).

these two steps to avoid any possible misassignments. The improvement of the fit is then based on the variety and richness of the data set, and the addition of lines belonging to $v_t = 1$ is crucial. The analysis of the first excited torsional state was challenging because of a larger splitting of the A and E component by the tunnelling effect. In a recent work by Tudorie et al. (2012) the band origins ($E_A = 132.4303 \text{ cm}^{-1}$ and $E_E = 131.8445 \text{ cm}^{-1}$) for $v_t = 1$ of the parent molecule were precisely determined by a combination of microwave, mm-wave, submm-wave, and far infrared data. The value of height of the barrier was as much improved and V_3 was determined to be equal to 370.7398 (58) cm⁻¹. In this work, for HCOO¹³CH₃, the first torsional excited substates are predicted to be $E_A = 131.0565 \text{ cm}^{-1}$ and $E_{\rm E} = 130.4856 \text{ cm}^{-1}$. We estimated the rotational parameters using the same difference of the ground- and the first-torsional excited states rotational parameters of the ¹²C species. In the assignment procedure, the a-type transitions with A symmetry of R-branches were first analysed, in particular, the most intense lines with $K_a = 0$ and 1.

Methyl formate is a fairly asymmetric near-prolate top for which the dipole moment is non-zero along the a and b principle axis. For the parent specie, μ_a and μ_b were measured by Margulès et al. (2010) and their values are 1.648(8) and 0.706(12) D. We tracked these transitions based on the intensity criterion. Next we assigned the E type of transitions with the same benchmarks as those adopted for the A type of transitions. For K_a the splitting between the A transitions, as well as between the E transitions, decreases with the increase of the J quantum number. Four μ_a and μ_b rotational transitions with the same symmetry (A or E) become blended at J = 24. Furthermore, in the mm-wave ranges the splitting between the A and E transitions decreases with the increase of J and at a specific value of Jin the submm-wave ranges this splitting changes sign (Fig. 1). Following this behaviour of the A and E type of transition for $K_a = 0$ and 1, we assigned them up to J = 50 at ~520 GHz. We fitted these for $v_t = 1$ with the ERHAM (Groner 1997, 2012) program. The predictions based on this first fit were accurate enough to continue and to search for higher quantum number lines. We collected 800 A lines and 596 E lines with $K_{a,\text{max}} = 19$ and 11 up to J = 50 for the A and E lines. ERHAM² treats each torsional state separately, and to proceed with the global analysis of $v_t = 0$ and 1, we used BELGI. The final data set with the upper limits of the quantum numbers and the experimental uncertainties are summarized in Table 1. The 6574 lines belonging to $v_t = 0$ and 2881 lines belonging to $v_t = 1$ were fitted with root-mean-square deviations of 38.3 and 41.9 kHz. The fit of the total of 9455 lines for $v_t = 0$ and 1 of A and E symmetries resulted in the determination of 52 parameters of the RAM Hamiltonian. Table 3 presents the values of the set of 52 newly determined parameters. In this table we also provide a comparison with the previous work. Principally, we note a striking change in the value of the barrier height. The value of the parameter V_3 changed from 407.1549(147) to 369.93168(395) cm⁻¹.

The reason for this is the fact that this parameter is an effective value. In the previous fit (Carvajal et al. 2009), the data set of lines concerned only the $v_t = 0$ state and, therefore, neither the second-order parameter (V_6) of the torsional potential function nor the kinetic parameter F could be determined. Then, F was fixed to a relatively high value estimated by ab initio methods, and in the fitting procedure, the effective value of V_3 absorbed the error of F and contained the contribution of V_6 :

$$V(\gamma) = \frac{V_3}{2} \times (1 - \cos 3\gamma) + \frac{V_6}{2} \times (1 - \cos 6\gamma) + \dots$$

The new combination of data from $v_t = 0$ and 1 improved the quality of the fit and allowed determining of 25 new parameters, particularly F and V_6 .

4. Line strengths

The intensity calculations for the two ¹³C species of methyl formate were performed using the same method described in Hougen et al. (1994) and Ilyushin et al. (2008), that was applied for the normal species by Carvajal et al. (2007), the ¹³C species (Carvajal et al. 2009, 2010), DCOOCH₃ (Margulès et al. 2010), and ¹⁸O species (Tercero et al. 2012). For this reason we do not repeat it here. The line strength calculation for the ¹³C species was performed using the same value for the dipole moment as for the parent methyl formate (Margulès et al. 2010). This assumption was made because the angle between the internal axis of the methyl top and the principal *a*-axis does not change much upon substitution (see the $\langle (i, a)$ angle values in Table 3). All other structural parameters such as the angle between RAM and PAM axes (θ_{RAM}) and the ρ parameter do not change much upon substitution (see Table 3). Because the changes in the structure in the various isotopomers are not important, we assumed that the dipole moment does not change considerably from one species to another. The dipole moment components of H¹³COOCH₃ (Carvajal et al. 2010) and HCOO¹³CH₃ in RAM axis system that we used for the intensity calculations are $\mu_a^{\text{RAM}} = 1.792 \text{ D}$; $\mu_b^{\text{RAM}} = -0.044 \text{ D}$ and $\mu_a^{\text{RAM}} = 1.793 \text{ D}$; $\mu_b^{\text{RAM}} = -0.022 \text{ D}$, respectively. We provide Table A.1, which is part of the supplementary Table S (available on CDS). It contains all included lines from our fit for the HCOO¹³CH₃ methyl formate species. They show the line assignments, the observed frequencies with measurement uncertainty (in parenthesis), the computed frequencies, observed-calculated values, the line strength, and the lower-state energy relative to the J = K = 0 A-species level taken as zero-energy level. The prediction line-list of all transitions in $v_t = 0$ and 1 will be transmitted to the SPLATALOG and CDMS database (available at www.splatalogue.net).

2 http://www.ifpan.edu.pl/~kisiel/introt/introt.htm#
erham

	Numb	er of parameters		52	
	Nu	mber of lines		9455	
	rms of the 65	574 MW $v_t = 0-0$	lines	0.0383 MH	Iz
	rms of the 28	881 MW $v_t = 1 - 1$	lines	0.0419 MH	Iz
	rms of the 5	184 A symmetry	lines	0.0429 MH	Iz
	rms of the 4	271 E symmetry l	lines	0.0349 MH	Iz
Source ^b	Range ^c (GHz)	$v_t, J_{\max}, K_{\max}^d$	Number of lines ^e	Uncertainties ^f (MHz)	rms ^g (MHz)
AA09	4-20	0,7,3	27	0.005	0.0056
AA09	150–700 ^{<i>j</i>}	0,63,35	607	0.030	0.0252
NEW	50–940 ^k	0,60,35	4864	0.030	
NEW	$50-940^{k}$	1,60,26	1731	0.030	
AA09	150–700 ^j	0, 52, 27	2	0.050	
NEW	50–940 ^k	0, 52, 35	577	0.050	
NEW	50–940 ^k	1, 58, 26	814	0.050	
AA09	8–700 ^{<i>j</i>}	0, 58, 29	43	0.100^{h}	0.0644
NEW	$50-940^{k}$	0, 55, 33	170	0.100^{i}	
NEW	50–940 ^k	1, 57, 26	203	0.100^{i}	
AA09	8–700 ^{<i>j</i>}	0, 33, 11	122	0.150^{h}	0.1345
NEW	$50-940^{k}$	0, 54, 33	49	0.150^{i}	
NEW	50–940 ^k	1, 55, 24	68	0.150^{i}	
AA09	$8-700^{j}$	0, 58, 25	45	0.200^{h}	
NEW	$50-940^{k}$	0, 55, 31	68	0.200^{i}	
NEW	50–940 ^k	1, 55, 20	65	0.200^{i}	

Table 1. Root-mean-square (rms) deviations from the global fit^{*a*} of transitions involving $v_t = 0$ and $v_t = 1$ torsional energy levels of ${}^{13}C_2$ -methyl formate (HCOO ${}^{13}CH_3$).

Notes. ^(a) Parameter values are given in Table 2. The complete list of observed minus calculated residuals of $v_t = 0$ lines are given in the supplementary Table A.1. ^(b) Sources of data: AA09 stands for Carvajal et al. (2009) and NEW data come from the present work. The spectroscopic instruments are described in Sect. 2.2. ^(c) Spectral range of the measurements. ^(d) Torsional state v_t , maximum J and K_a for each group of measurements. ^(e) Number of MW lines in each uncertainty group. ^(f) Uncertainty in MHz used in the fit. ^(g) Root-mean-square deviation in MHz for each group. ^(h) The details of the accuracy of spectral measurements are given in Carvajal et al. (2009). ⁽ⁱ⁾ The accuracy of spectral measurements for the Lille spectrometer is about 0.030 MHz for isolated lines but some lines are either blended or present a poor S/N ratio and therefore their measurement accuracy is taken as either 0.100 MHz, 0.150 MHz or 0.200 MHz according to the broadening of the experimental line. ^(j) The spectrometer spectral ranges for these measurements are 8–80 GHz, 150–245 GHz, 320–350 GHz, 510–525 GHz, and 580–661 GHz (for more details, see Carvajal et al. 2009). ^(k) The spectrometer spectral ranges for the measurements of present work are: 50–100 GHz and 150–940 GHz, using a multiplication chain of solid state sources, and 100–150 GHz, using a backward wave oscillator.

5. Detection of ¹³C-HCOOCH₃ $v_t = 1$ in Orion KL

In this section we use the notation 13 C-HCOOCH₃ when we refer to the two isotopologues. Following the procedure of our previous papers, we searched for the present species in the molecular line survey of Orion KL performed with the IRAM 30 m telescope. We detected both 13 C isotopologues of methyl formate in their first vibrationally excited state. In Sect. 5.1 we summarize the observations, data reduction, and overall results of this survey. We present the new detection in Sect. 5.2. In Sect. 5.3 we explain the assumed model and the excitation and radiative transfer calculations for vibrationally excited 13 C and 12 C methyl formate. We also compute our obtained 12 C/ 13 C ratio and the vibrational temperature for 13 C-HCOOCH₃ $v_t = 1$.

5.1. Observations, data reduction, and overall results

The molecular line survey towards Orion KL-IRc2 (α (J2000) = $5^{h}35^{m}14.5^{s}$, δ (J2000) = $-5^{\circ}22'30.0''$) was performed between September 2004 and January 2007 in five observing sessions with the IRAM 30 m telescope. All frequencies allowed by the A, B, C, and D receivers were covered (80–115.5, 130–178, and 197–281 GHz) with a spectral resolution of 1–1.25 MHz. System temperatures, image side-band rejections, and halfpower beam widths were in the ranges 100–800 K, 27–13 dB, and 29–9'', from 80 to 281 GHz. The observations were performed in the balanced wobbler-switching mode. Pointing and focus were checked every 1–2 h on nearby quasars. The intensity scale was calibrated with two absorbers at different temperatures using the ATM package (Cernicharo 1985; Pardo et al. 2001).

The data were processed using the GILDAS software³. The data reduction consisted of removing lines from the image sideband and fitting and removing baselines. We observed each frequency setting twice, with the second one shifted in frequency by 20 MHz. This allowed us to remove lines from the image sideband down to a level of 30 dB (this means that lines in the image side-band of 30 K will contribute with less than 0.03 K to the final reduced spectrum). Figures are shown in units of main-beam antenna temperature, $T_{\rm MB} = T_{\rm A}^*/\eta_{\rm MB}$, where $\eta_{\rm MB}$ is the mainbeam efficiency ranged from 0.82 to 0.48 from the lowest to the highest frequencies. For a detailed description of the observations and data reduction procedures see Tercero et al. (2010).

Up to date, we performed a deep analysis of the data (15 papers have been already published): identifying more than 15000 spectral features (Tercero et al. 2010; Tercero 2012), detecting new molecules in space (CH₃COOCH₃ and g-CH₃CH₂OCOH in Tercero et al. 2013; NH₃D⁺ in Cernicharo et al. 2013; CH₃CH₂SH in Kolesniková et al. 2014; also tentative detections of c-C₆H₅O in Kolesniková et al. 2013 and upper limit calculations for the column density of non detected molecules –cis-CH₂CHCH₂CN and cis-CH₂CHCH₂NC– in Haykal et al. 2013b) and several new isotopologues and

³ http://www.iram.fr/IRAMFR/GILDAS

Table 2. Torsion-rotation parameters needed for the global fit of transitions involving $v_t = 0$ and $v_t = 1$ torsional energy levels of ¹³C₂-methyl formate (H¹²COO¹³CH₃) provided in this work and comparison with the torsion-rotation parameters obtained in the former global fit of transitions only involving $v_t = 0$ torsional states (Carvajal et al. 2009).

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þ										$\times 10$	$\times 10$					< 10 ⁻¹³					$\times 10^{-1}$								
$v_t = 0$	0.0^{f}	0.0^{f}	0.0^{f}	0.0^{f}	0.0^{f}	0.0^{f}	0.0^{f}	0.0^{f}	0.0	0.9060(708)	0.3349(260)	0.0^{f}	0.0^{f}	0.0^{f}	0.0^{f}	$-0.31(1) \times$	0.0^{f}	0.0	0.0	0.0^{f}	0.1130(25)	0.0^{f}	0.0^{f}	0.0^{f}	0.0				
$v_t = 0, 1^c$	$-0.312260(869) \times 10^{-3}$	$0.84669(480) \times 10^{-3}$	$-0.161265(889) \times 10^{-3}$	$-0.29738(147) \times 10^{-3}$	$0.7960(365) \times 10^{-8}$	$0.13176(197) \times 10^{-7}$	$0.91080(149) \times 10^{-8}$	$0.4484(183) \times 10^{-7}$	$-0.21242(275) \times 10^{-6}$	$0.46277(344) \times 10^{-7}$	$-0.83350(642) \times 10^{-7}$	$0.47452(372) \times 10^{-7}$	$0.4462(154) \times 10^{-9}$	$0.18733(536) \times 10^{-9}$	$-0.12682(268) \times 10^{-7}$	$-0.755(12) \times 10^{-12}$	$0.10555(83) \times 10^{-10}$	$-0.34983(244) \times 10^{-10}$	$0.50362(252) \times 10^{-10}$	$-0.364(6) \times 10^{-12}$	$0.1996(22) \times 10^{-11}$	$-0.1811(68) \times 10^{-11}$	$0.2077(52) \times 10^{-11}$	$-0.10787(148) \times 10^{-10}$	$0.14903(534) \times 10^{-10}$				
Parameter	N_v	K_2	c_{11}	dd_{ab}	M_v	c_3	f_v	k_{5J}	k_{5K}	d_{abJ}	d_{abK}	c_{2K}	k_{2J}	c_{1K}	D_{bcJ}	H_J	H_{JK}	H_{KJ}	H_K	h_J	h_{JK}	h_K	D_{abJJ}	D_{abJK}	l_v				
Operator	$(1 - \cos 6\gamma)P^2$	$(1 - \cos 6\gamma)P_a^2$	$(1-\cos 6\gamma)(P_h^2-P_c^2)$	$(1 - \cos 6\gamma)(P_a P_b + P_b P_a)$	$P_{\gamma}^4 P^2$	$2 \ P_{\gamma}^4(\dot{P}_h^2 - P_c^2)$	$(1 - \cos 3\gamma)\tilde{P}^4$	$(1-\cos 3\gamma)P_a^2P^2$	$(1 - \cos 3\gamma)\tilde{P}_a^4$	$(1-\cos 3\gamma)(P_aP_b+P_bP_a)P^2$	$(1 - \cos 3\gamma)(P_a^3 P_b + P_b P_a^3)$	$(1 - \cos 3\gamma)\{P_a^2, (P_h^2 - P_c^2)\}$	$P_{\gamma}^2 P_a^2 P^2$	$2P_{\chi}^{2}\{P_{a}^{2'},(P_{b}^{2}-P_{c}^{2})\}$	$\sin 3\gamma(P_b^{-}P_c^{-}+P_cP_b^{-})P^2$	P^{6}	$P^4 P_a^2$	$P^2 P_a^4$	P_a^6	$2P^4(P_h^2 - P_c^2)$	$P^2\{P_a^2,(\check{P}_b^2-\check{P}_c^2)\}$	$\{P_a^4, (P_h^2 - P_c^2)\}$	$P^4(\tilde{P}_aP_b+P_bP_a)$	$P^2(P_a^3P_b+P_bP_a^3)$	$P_{\gamma}P_aP^4$				
nlm	642						624									606									615				
$v_t = 0^d$	$407.1549(147)^{g}$	5.69168218^{e}	0.0845207(106)	0.5857484(245)	0.2959971(182)	0.1729010(134)	-0.1573691(747)	0.0^{f}	$0.4682(101) \times 10^{-4}$	0.0^{f}	-0.0040623(839)	0.0010563(383)	-0.0007557(425)	0.0124250(566)	0.0^{f}	0.0^{ℓ}	-0.013852(236)	$-0.16792(565) \times 10^{-3}$	$-0.2045(247) \times 10^{-5}$	0.0^{f}	$0.39452(688) \times 10^{-5}$	$0.1707(107) \times 10^{-4}$	$0.15312(163) \times 10^{-6}$	$0.21223(170) \times 10^{-5}$	$-0.17369(181) \times 10^{-5}$	$0.39739(813) \times 10^{-7}$	$0.108794(621) \times 10^{-5}$	$0.24287(221) \times 10^{-6}$	$0.13608(491) \times 10^{-3}$
$v_t = 0, 1^c$	369.93168(395)	5.4785214(464)	0.084337567(345)	0.58658923(178)	0.29682136(198)	0.173496850(137)	-0.15976497(192)	25.40353(193)	$0.27105(240) \times 10^{-5}$	$0.10489(207) \times 10^{-5}$	0.0^{f}	0.00074120(917)	-0.00229820(140)	0.01046256(804)	$-0.3062(148) \times 10^{-4}$	$-0.43892(170) \times 10^{-4}$	-0.00573380(154)	0.0^{f}	$0.1824(134) \times 10^{-6}$	$0.127972(329) \times 10^{-4}$	$0.94726(730) \times 10^{-6}$	$-0.343262(938) \times 10^{-5}$	$0.24016(201) \times 10^{-6}$	$-0.235426(785) \times 10^{-5}$	$0.576749(422) \times 10^{-5}$	$0.8245(100) \times 10^{-7}$	$-0.41365(637) \times 10^{-7}$	$-0.48884(361) \times 10^{-6}$	$0.1345545(843) \times 10^{-3}$
Parameter	V_3	F	θ	$A^{ m RAM}$	$B^{ m RAM}$	$C^{\rm RAM}$	D_{ab}	V_6	G,	c_1	D_{ac}	D_{bc}	F_v	k_5	c_2	k_2	d_{ab}	Δ_{ab}	L_v	k_1	c_4	δ_{ab}	Δ_J	Δ_{JK}	Δ_K	δ_J	δ_K	D_{abJ}	D_{abK}
Operator ^b	$(1 - \cos 3\gamma)/2$	P_{γ}^2	$P_{\gamma}\dot{P}_{a}$	P_a^2	P_h^2	$P_c^{\tilde{2}}$	$(P_aP_b + P_bP_a)$	$(1 - \cos 6\gamma)/2$	$P_{\gamma}^2 P^2$	$2P_{\gamma}^{2}(\dot{P}_{h}^{2}-P_{c}^{2})$	$\sin 3\gamma (P_a P_c + P_c P_a)$	$\sin 3\gamma (P_bP_c+P_cP_b)$	$(1 - \cos 3\gamma)P^2$	$(1-\cos 3\gamma)P_a^2$	$(1 - \cos 3\gamma)(P_h^2 - P_c^2)$	$P_{\gamma}^2 P_a^2$	$(1 - \cos 3\gamma)(P_aP_b + P_bP_a)$	$P_{\gamma}^2(P_aP_b+P_bP_a)$	$P_{\gamma}P_aP^2$	$P_{\gamma}P_a^3$	$P_{\gamma}\{P_{a},(P_{h}^{2}-P_{c}^{2})\}$	$P_{\gamma}\{P_a^2,P_b\}$	$-\tilde{P}^4$	$-P^2P_a^2$	$-P_a^4$	$-2P^2(P_b^2 - P_c^2)$	$-\{P_a^2,(P_b^2-P_c^2)\}$	$P^2(P_aP_b+P_bP_a)$	$(P_a^5P_b + P_bP_a^5)$
nlm^a	220		211	202				440	422								<u> </u>		413				404						

parameter values are in cm⁻¹ except for ρ which is unitless. Statistical uncertainties are given in parentheses in units of the last quoted digit. ^(d) Values of the parameters from the fit for the ground torsional state $v_t = 0$ given by Carvajal et al. (2009). The 27 parameters were fitted for 936 MW lines from $v_t = 0$ with a root mean square of 97 kHz. All parameter values are in cm⁻¹ except for ρ et al. (2003). $\{A, B\} = AB + BA$. The product of the parameter and operator from a given row yields the term actually used in the vibration-rotation-torsion Hamiltonian, except for F, ρ and A, which occur in the Hamiltonian in the form $F(P_y - \rho P_a)^2 + A^{RAM}P_a^2$ (c) Values of the parameters from the present fit, for the ground torsional state $v_i = 0$ and the first excited torsional state $v_i = 1$. All which is unitless. Statistical uncertainties are given in parentheses in units of the last quoted digit. (e) The internal rotation constant F of HCOO¹³CH₃ was kept fixed to the ab initio value calculated in the equilibrium structure (see Carvajal et al. 2009). ⁽¹⁾ Kept fixed. ^(a) Very large effective value obtained as a consequence of fixing F parameter to a large ab initio value and because F, V_3 and V_6 were not able to fit them at the same time only from the ground torsional state (Carvajal et al. 2009). **Notes**. ^(d) Notation from Ilyushin et al. (2003); n = l + m, where n is the total order of the operator, l is the order of the torsional part and m is the order of the rotational part.

	HCOOCH ₃ ^{<i>a</i>} Global fit ($v_t = 0, 1$)	$HCOO^{13}CH_3{}^b$ Global fit ($v_t = 0$)	$HCOO^{13}CH_3^{c}$ Global fit ($v_t = 0, 1$)
A(MHz)	19924.4249(433)	19 629.65(176)	19707.8000(625)
B(MHz)	6947.9458(414)	6804.42(171)	6776.1833(660)
C(MHz)	5318.10685(450)	5183.442(402)	5201.30471(410)
$\langle (i,a)^d$	53.06	51.68	52.18(25)
$\theta_{\mathrm{RAM}}{}^{e}$	24.87951(13)	23.68	23.89824(22)
ρ	0.08427227(25)	0.0845207(106)	0.084337567(345)
$F(\mathrm{cm}^{-1})$	5.497592(59)	5.69168218 ^f	5.4785214(464)
$V_3 ({\rm cm}^{-1})$	370.7398(58)	407.1549(147)	369.93168(395)
σ^{g}	0.67	1.08	0.77
Nº of transitions	15 840	936	9455
Nº of fitted parameters	53	27	52

Table 3. Rotational constants in the principal axis system (PAM), angles between the principal axis and the methyl top axis, and internal rotation parameters upon isotopic substitution.

Notes. ^(a) Rotation-torsion parameters for the normal species HCOOCH₃ obtained in Tudorie et al. (2012) with the RAM-axis system, after transforming the RAM values for the A, B, C rotational parameters into PAM values following the procedure described in Carvajal et al. (2007). The RAM to PAM transformation is given in Carvajal et al. (2010). ^(b) Rotation-torsion parameters for ¹³C₂-methyl formate from Carvajal et al. (2009) transformed to the principal axis system. ^(c) Rotation-torsion parameters for ¹³C₂-methyl formate from the present work transformed into the principal axis system. ^(d) Angle in degrees between the a-principal axis and the methyl top axis (*i*). ^(e) The angle θ_{RAM} between the *a*-principal axis and the *a*-RAM axis is given in degrees and obtained from Eq. (1) from Carvajal et al. (2007), with the parameters A^{RAM} , B^{RAM} , C^{RAM} , and D_{ab} of Table 2. ^(f) Fixed to the ab initio value (Carvajal et al. 2009). ^(g) Unitless standard deviation.

Table 4. Kinematic parameters and physical conditions of the Orion-KL spatial components.

Parameter	Extended ridge (ER)	Compact ridge (CR)	Plateau (P)	Hot core (HC)
Source diameter (")	≃120 (120)	5-15 (15)	20-30 (30)	5-10 (10)
Offset (from IRc2) (")	0 (0)	7 (7)	0 (0)	2-3 (2)
$n({\rm H}_2) ({\rm cm}^{-3})$	$\simeq 10^5$	$\simeq 10^{6}$	$\simeq 10^{6}$	$10^{7} - 10^{8}$
T_{k} (K)	≃60 (60)	80-150 (110)	100-150 (125)	150-300 (225)
$\Delta v_{FWHM} \ (\mathrm{km} \mathrm{s}^{-1})$	≃ 4 (4)	3-5 (4)	≃25 (25)	5-15 (10)
$v_{\rm LSR}~({\rm kms^{-1}})$	8-10 (8)	7-9 (7.5)	6-11 (9)	3-6 (5.5)

Notes. Parameters compiled from different authors: Blake et al. (1987); Schilke et al. (2001); Persson et al. (2007); Tercero et al. (2010); Neill et al. (2013). The values between parenthesis are those adopted for the model of HCOOCH₃ $v_t = 1$ and ¹³C-HCOOCH₃ $v_t = 0$.

vibrationally excited states of well-known molecules in this source (¹³C-CH₃CH₂CN in Demyk et al. 2007; ¹³C-HCOOCH₃ in Carvajal et al. 2009; CH₃CH₂¹⁵N, CH₃CHDCN, and CH₂DCH₂CN in Margulès et al. 2009; DCOOCH₃ in Margulès et al. 2010; ¹⁸O-HCOOCH₃ in Tercero et al. 2012; NH₂CHO $v_{12} = 1$ in Motiyenko et al. 2012; CH₃CH₂CN $v_{10} = 1$ and $v_{12} = 1$ in Daly et al. 2013; HCOOCH₂D in Coudert et al. 2013; CH₂CHCN $v_{10} = 1 \Leftrightarrow (v_{11} = 1, v_{15} = 1)$ in López et al. 2014), and constraining physical and chemical parameters by means of the analysis of different families of molecules (OCS, CS, H₂CS, HCS⁺, CCS, CCCS species in Tercero et al. 2010; SiO and SiS species in Tercero et al. 2011; CH₃CH₂CN species in Daly et al. 2013; SO and SO₂ species in Esplugues et al. 2013a; HC₃N and HC₅N species in Esplugues et al. 2013b; CH₃CN in Bell et al. 2014; CH₂CHCN species and the isocyanides in López et al. 2014). In addition and to analyze some of these results, we modelese the following molecules: CH₃SH, CH₃OH, CH₃CH₂OH in Kolesniková et al. (2014), HCOOCH₃ in Margulès et al. (2010), and NH₂CHO in Motiyenko et al. (2012). Nevertheless, the study of this line survey is still open and several groups are working simultaneously based on the scope of these observations⁴.

In agreement with numerous works of this region (see e.g. Blake et al. 1987; Schilke et al. 2001; Persson et al. 2007; Neill et al. 2013), at least four cloud components could be identified in the line profiles of our low-resolution spectral lines, which are characterized by different radial velocities and line widths

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(Tercero et al. 2010, 2011). Each component corresponds to specific region of the cloud that overlaps in our telescope beam. Table 4 summarizes the kinematic and physical parameters of the different components in Orion-KL.

5.2. Results

The two ¹³C isotopologues of HCOOCH₃ in their first vibrationally excited state ($v_t = 1$) have been detected in Orion-KL by means of 135 spectral features (77 of them practically unblended) without missing transitions in the range 80-280 GHz. Figure 2 shows selected detected lines of the two isotopologues together with our best model (see below). The black histogram line corresponds with the observed spectra. The model of the $H^{13}COOCH_3 v_t = 1$ and $HCOO^{13}CH_3 v_t = 1$ lines is shown in red and dark blue, respectively. In all boxes unblended or slightly blended detected lines of these species are depicted. The cyan blue line represents the total model compiled from our already published work (see Sect. 5.1). In Table 5 we list all detected features in our line survey. These detections are based on an inspection of the data and the modelled synthetic spectrum of the studied species and all the species already identified in our previous papers (see above). The observed main-beam temperature and the radial velocity where obtained from the peak channel of our spectra, therefore, errors in the baselines -up to ± 0.15 K for spectra at 1.3 mm due to the extremely high line density (line confusion limit) - and contribution from other species might affect the $T_{\rm mb}^{\rm obs}$ value. Adding species to our total model allows reducing the uncertainty in the baselines, so, in general, the $T_{\rm mb}^{\rm obs}$

⁴ The data of the IRAM 30 m line survey of Orion KL are available in ascii format on request to B. Tercero.

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	12	+	1 1	11	+	130 348.229	229.4	32.1	130 349.3	4.6 ± 1.2	0.06	0.03	
$E^{-13}C_1$ 12 -1	12	1	1 –1	11		130424.730	229.5	32.2	130 424.3	8.1 ± 1.2	0.04	0.03	
$A^{-13}C_2$ 11 3	6	+	0 3	8	+	130674.060	231.2	26.8	130 673.6	8.0 ± 1.1	0.07	0.03	CH ₃ CH ₂ Cl ⁵ N
$E^{-13}C_2$ 11 5	9	1	0 5	ŝ		131 497.371	340.9	23.2	131 497.9	5.7 ± 1.1	0.03	0.02	
$E^{-13}C_2$ 11 -3	6	1	0 -3	8		131 600.450	329.8	26.6	131 600.6	6.6 ± 1.1	0.03	0.02	
$A^{-13}C_2$ 11 4	8	- 1	0 4	Γ	Ι	131 626.157	235.9	25.2	131 626.6	6.0 ± 1.1	0.07	0.02	(CH ₃) ₂ CO
$A^{-13}C_2$ 11 4	L	+	0 4	9	+	132401.079	236.0	25.2	132400.5	8.3 ± 1.1	0.04	0.02	CH ₃ CHDCN
$A^{-13}C_1$ 11 3	6	+	0 3	8	+	132 807.823	231.8	27.4	$132\ 806.7$	9.6 ± 1.1	0.07	0.03	
$A^{-13}C_1$ 11 9	0	-	0 9	-	Ι	132945.107	279.3	9.86	132946.6	3.7 ± 1.1	0.07	0.02	CCCS
$A^{-13}C_1$ 11 9	С	+	0 9	0	+	132945.107	279.3	9.86	+	3.7 ± 1.1			
$E^{-13}C_2$ 11 3	8	1	0 3	L		136042.486	330.6	26.8	136 042.9	6.2 ± 1.1	0.03	0.03	
$E^{-13}C_1$ 11 3	8	1	0 3	2		138640.538	232.8	27.6	$138\ 640.6$	6.9 ± 1.1	0.04	0.03	
$A^{-13}C_2$ 13 1	13	+	2	12	+	138 830.565	235.5	34.0	138 829.7	8.9 ± 1.1	0.06	0.04	
$E^{-13}C_1$ 12 -2	11	-	1 -2	10		139 241.862	234.5	31.5	139 239.2	12.6 ± 1.1	0.06	0.03	CH ₂ DCH ₂ CN
$E^{-13}C_1$ 13 -1	13	1	2 –1	12		140917.742	236.2	35.0	140918.6	5.3 ± 1.1	0.04	0.04	
$A^{-13}C_1$ 13 0	13	+	2 0	12	+	140937.135	236.2	34.8	140937.6	6.1 ± 1.1	0.13	0.04	CH ₂ DOH
$E^{-13}C_1$ 13 0	13	1	2 0	12		$141\ 008.471$	236.2	35.0	141 007.8	8.3 ± 1.1	0.09	0.04	
$E^{-13}C_1$ 12 1	11	1	1	10		141 542.530	234.2	31.6	141 543.0	6.1 ± 1.1	0.05	0.04	
$A^{-13}C_2$ 12 3	10	+	1 3	6	+	142240.894	238.0	29.5	$142\ 240.3$	8.2 ± 1.1	0.04	0.03	
$E^{-13}C_2$ 12 4	8	-	1 4	7		144894.428	341.8	28.2	144 895.6	4.6 ± 1.0	0.06	0.03	
E- ¹³ C ₂ 12 -4	6	-	1 4	8		144984.322	341.4	28.2	144 983.7	8.4 ± 1.0	0.05	0.03	
$E^{-13}C_1$ 12 7	5	1	1 7	4		145 262.992	266.0	21.5	145 262.9	7.1 ± 1.0	0.07	0.02	U
$A^{-13}C_1$ 12 7	9	+	1 7	S	+	145 282.393	265.1	21.5	145 282.9	5.9 ± 1.0	0.07	0.04	
$A^{-13}C_1$ 12 7	S	-	1 7	4	I	145 282.519	265.1	21.5	÷	6.2 ± 1.0			
$E^{-13}C_1$ 12 5	L	1	1 5	9		146170.669	250.0	27.0	146 170.3	7.7 ± 1.0	0.15	0.03	H ¹³ COOCH ₃
$E^{-13}C_1$ 12 4	8	1	1 4	L		147 500.782	244.1	28.9	147 501.6	5.4 ± 1.0	0.15	0.07	
$E^{-13}C_1$ 12 -4	6		1	8		147501.246	243.7	28.7		6.3 ± 1.0			
$A^{-13}C_1$ 13 2	12	-	2	11	I	149 734.053	241.5	34.0	149 734.5	6.0 ± 1.0	0.05	0.04	
$A^{-13}C_1$ 12 2	10	+	1 2	6	+	150951.028	237.3	31.4	150950.6	7.9 ± 1.0	0.04	0.04	
$E^{-13}C_1$ 14 0	14	1	3 –1	13		151 328.687	243.5	6.07	151 331.6	1.3 ± 1.0	0.11	0.05	U
$A^{-13}C_1$ 14 1	14	+	3 1	13	+	151 329.124	243.4	37.5	÷	2.2 ± 1.0			
$E^{-13}C_1$ 14 0	14	-	3 0	13		151 452.158	243.5	37.7	151451.6	8.1 ± 1.0	0.04	0.05	
$E^{-13}C_2$ 13 -3	11		2 -3	10		154 267.593	344.1	32.4	154 267.7	6.8 ± 1.0	0.08	0.04	CH ₃ CHO
$A^{-13}C_2$ 13 7	2	+	2 7	9	+	154 902.510	272.0	24.4	154 902.4	7.2 ± 1.0	0.24	0.05	$SO_2 v_2 = 1$
$A^{-13}C_2$ 13 7	9	-	2 7	5	Ι	154 902.822	272.0	24.4	- 	7.8 ± 1.0			
$A^{-13}C_1$ 13 3	11	+	2 3	10	+	156 106.621	246.2	33.2	156 107.9	4.6 ± 1.0	0.15	0.04	HCOOCH ₃ $v_t = 1$
$E^{-13}C_1$ 13 -3	11	1	2 –3	10		156705.635	246.5	33.3	156 705.5	7.3 ± 1.0	0.11	0.04	CH ₃ CH ₂ C ¹⁵ N
$E^{-13}C_1$ 13 10	З	1	2 10	0		156879.965	307.6	14.4	$156\ 880.5$	6.0 ± 1.0	0.05	0.01	
$E^{-13}C_1$ 13 9	4	1	2 9	3		157 026.174	294.9	18.4	157 026.5	6.4 ± 1.0	0.09	0.02	HCOOCH ₃

Blend	U			CH ₃ OH		¹³ CH ₃ OH		U	CH ₃ CH ₂ OCOH	CH ₃ CH ₂ OCOH	n -	CH ₃ C ¹³ CH				U			HCOO ¹³ CH ₃	U		CH ₁ ¹³ CH ₂ CN	,				CH ₃ CH ₂ OH	U			U				DCOOCH ₃	SO ¹⁷ O	HCCC ¹⁵ N	U		U		HCOOCH ₃ $v_t = 1$
$T^{ m mod}_{ m mb}({ m K})$	0.05		0.02	0.04	0.04	0.04	0.05	0.03	0.04	0.05	0.04	0.05	0.04		0.03	0.04	0.04	0.05	0.04	0.06		0.06	0.06	0.06	0.08	0.07	0.08	0.07	0.06		0.08	0.09		0.08	0.09	0.07		0.09	0.09	0.09	0.09	0.08
${T_{ m mb}^{ m obs}}{({ m K})}$	0.15			0.11	0.10	0.14	0.07	0.13	0.10	0.06	0.10	0.15	0.08		0.08	0.12	0.10	0.07	0.16	0.14		0.22	0.10	0.05	0.11	0.08	0.21	0.33	0.15		0.25	0.16		0.14	0.04	0.25		0.16	0.10	0.14		0.23
$v_{\rm LSR}$ (km s ⁻¹)	1.4 ± 1.0	1.4 ± 1.0	13.6 ± 1.0	8.9 ± 1.0	6.9 ± 0.9	5.7 ± 0.9	6.8 ± 0.9	7.1 ± 0.9	6.9 ± 0.9	8.4 ± 0.9	8.6 ± 0.9	5.1 ± 0.9	8.2 ± 0.9	8.2 ± 0.9	7.4 ± 0.9	8.9 ± 0.9	5.6 ± 0.9	7.7 ± 0.9	6.1 ± 0.9	9.1 ± 0.9	11.1 ± 0.9	7.6 ± 0.9	8.1 ± 0.9	5.9 ± 0.9	8.9 ± 0.7	5.9 ± 0.7	8.9 ± 0.7	7.7 ± 0.7	6.7 ± 0.7	6.7 ± 0.7	7.1 ± 0.7	6.9 ± 0.7	6.9 ± 0.7	7.5 ± 0.7	13.5 ± 0.7	5.3 ± 0.7	1	6.4 ± 0.7	5.6 ± 0.7	9.5 ± 0.7	14.0 ± 0.7	7.2 ± 0.7
Obs. freq. (MHz)	157 224.1	+	+	157 231.6	158 076.5	158 311.6	158 376.5	158 492.8	159 175.2	159 620.4	159 906.5	165 796.4	166 581.5	+	166 627.5	$167\ 330.3$	167 356.5	$168\ 050.1$	168 270.3	$169\ 802.4$	+	169839.4	$169\ 880.5$	171 157.6	200011.0	200351.0	201 244.8	202 078.1	202 460.6	+	202 695.6	202 701.7	+-	202 854.4	203 099.3	203 146.8		203 679.3	203 683.2	203 733.1	+	203 884.2
$S_{ij\mu}^2$ (D ²)	14.4	14.4	22.0	30.8	31.1	30.0	36.1	27.8	30.1	36.1	31.4	33.6	18.1	18.1	25.0	30.3	30.3	36.1	32.1	28.4	28.4	41.7	42.0	35.9	46.5	40.9	44.4	40.7	26.1	26.1	44.4	32.4	32.4	47.6	47.8	42.2	1	51.0	51.0	51.3	51.3	44.4
$E_{\rm u}$ (K)	306.4	306.4	283.6	348.9	250.5	257.0	346.9	264.3	257.1	346.8	251.3	248.1	313.8	313.8	389.3	271.4	271.4	254.5	264.3	280.8	280.8	258.4	356.9	351.0	382.3	372.6	281.9	278.5	355.2	355.2	279.0	328.6	328.6	285.1	285.3	285.9		287.3	287.3	287.4	287.4	281.3
Calc. freq. (MHz)	157 221.146	157 221.146	157 227.514	157 232.535	158 076.398	158310.862	158376.417	158492.862	159 175.181	159 621.190	159 907.354	165 795.305	166 582.134	166 582.134	166 627.702	167331.309	167355.684	168050.461	168269.790	169803.580	169804.698	169 839.735	169881.101	171 156.971	200 012.238	200350.331	201 246.113	202 078.548	202 460.334	202460.334	202 695.684	202 701.664	202 701.650	202 854.699	203 103.686	203 145.664		203 678.875	203 682.196	203 734.744	203 737.849	203 884.346
p'	I	+			+	+							+	Ι		I	+		Ι	+	Ι	+						+	+	I		Ι	+	Ι		I		+	+			+
K_c'	3	0	4	6	×	×	12	L	×	12	6	6	ŝ	4	S	×	2	11	8	L	9	15	15	11	16	12	14	11	9	S	11	L	8	16	16	13	2	18	18	18	18	14
K_a'	10	10	×	4	4	S	4	9	ŝ	-	4	Э	10	10	×	9	9	ς	2	2	Г	0	ī	0	4	С	Ϋ́	4	11	11	4	6	6	-	-	4	-	-	0	ī	0	2
J'	12	12	12	12	12	12	13	12	12	13	12	12	13	13	13	13	13	13	13	13	13	15	15	13	17	15	16	15	16	16	15	16	16	17	17	16		18	18	18	18	16
d	Ι	+			+	+							+	I		I	+		I	+	I	+						+	+	Ι		I	+	Ι		I		+	+			+
K_c	4	З	S	10	6	6	13	8	6	13	10	10	4	2	9	6	~	12	6	8	2	16	16	12	17	13	15	12	2	9	12	8	6	17	17	1	•	19	19	19	19	15
K_{a}	10	10	8	4	4	ŝ	c_1^-	9	ŝ	-	4	Э	10	10	×	9	9	ŝ	S	Г	Г	0	ī	0	4	З	Ϋ́	4	11	11	4	6	6	1	-	4	-	-	0	ī	0	2
7	13	13	13	13	13	13	14	13	13	14	13	13	14	14	14	14	14	14	14	14	14	16	16	14	18	16	17	16	17	17	16	17	17	18	18	17	-	19	19	19	19	17
Spec.	$A^{-13}C_{1}$	$A^{-13}C_{1}$	$E^{-13}C_{1}$	$E^{-13}C_{2}$	$A^{-13}C_{2}$	$A^{-13}C_{1}$	$E^{-13}C_{2}$	$E^{-13}C_1$	$E^{-13}C_{1}$	$E^{-13}C_{2}$	$E^{-13}C_1$	$E^{-13}C_{1}$	$A^{-13}C_{2}$	$A^{-13}C_{2}$	$E^{-13}C_{2}$	$A^{-13}C_{2}$	$A^{-13}C_{2}$	$E^{-13}C_{1}$	$A^{-13}C_{2}$	$A^{-13}C_{1}$	$A^{-13}C_{1}$	$A^{-13}C_{2}$	$E^{-13}C_{2}$	$E^{-13}C_{2}$	$E^{-13}C_{2}$	$E^{-13}C_{2}$	$E^{-13}C_{1}$	$A^{-13}C_{1}$	$A^{-13}C_{2}$	$A^{-13}C_{2}$	$E^{-13}C_{1}$	$A^{-13}C_{2}$	$A^{-13}C_{2}$	$A^{-13}C_{1}$	$E^{-13}C_{1}$	$A^{-13}C_{2}$	20	$A^{-13}C_{1}$	$A^{-13}C_{1}$	$E^{-13}C_{1}$	$E^{-13}C_{1}$	$A^{-13}C_{1}$

Blend	H ¹³ COOCH ₃	CH3COUCH3 H.CS	00711	CH ₃ COOCH ₃	U							NH_2CHO		CH ₂ CH ¹³ CN/U			¹³ CH ₃ CH ₂ CN	$CH_2 CH CN v_{11} = 2$					$^{13}C^{17}O$		U			$c-C_3H_2$	U			$CH_2 CH CN v_{11} = 2$	n	C	CH ₃ COOCH ₃		$CH_2CHCN \nu_{15} = 1$			CH ₃ CH ₂ CN	U	
$T^{ m mb}_{ m mb}({ m K})$		0.06	71.0	0.06	0.07	0.09	0.09	0.10				0.09		0.08	0.08	0.09	0.09		0.13				0.09		0.09	0.08	0.07	0.08	0.06	0.06	0.09	0.08	000	V.U	0.08	0.09	0.10		0.06	0.09	0.08	0.08
$T_{ m mb}^{ m obs}({ m K})$		90.0 0.30	10.0	0.04	0.14	0.18	0.18	0.19				0.24		0.16		0.13	0.23		0.34				0.27		0.11		0.16	0.20	0.10	0.10	0.13	0.13	000	0.09	0.14	0.14	0.42		0.18	0.24	0.30	0.11
$v_{\rm LSR}$ $({\rm km~s^{-1}})$		1.0 ± 0.7	7.2 ± 0.7	5.1 ± 0.7	6.5 ± 0.7	8.0 ± 0.7	6.8 ± 0.7	0.0 ± 0.7	4.1 ± 0.7	7.3 ± 0.7	11.4 ± 0.7	2.4 ± 0.7	6.1 ± 0.7	1.8 ± 0.7	8.8 ± 0.7	6.7 ± 0.7	3.2 ± 0.7		1.9 ± 0.7	4.9 ± 0.7	7.3 ± 0.7	10.2 ± 0.7	6.8 ± 0.7	6.8 ± 0.7	3.3 ± 0.7	9.5 ± 0.7	6.0 ± 0.7	11.4 ± 0.7	8.9 ± 0.7	7.6 ± 0.7	7.2 ± 0.7	4.3 ± 0.7		/0 ± C/	7.9 ± 0.7	7.0 ± 0.7	6.4 ± 0.7	6.4 ± 0.7	7.3 ± 0.7	6.0 ± 0.7	6.5 ± 0.7	6.2 ± 0.7
Obs. freq. (MHz)		204 024.3	1	206 867.3	207 553.6	210 097.2	210346.0	211 087.2	+	+	+	211 137.3	+	211 376.0	+	213 262.2	213 884.7		214 197.2	+	- -	+	214 572.2	+	215 324.7		217 441.0	218057.1	218 647.0	219 017.2	219 092.2	219 885.9		4.001 222	224 160.9	225 917.1	226 614.6	+	227 459.6	229 247.1	229 294.6	229 952.1
$\sum_{ij\mu^2 (D^2)}^{S_{ij}\mu^2}$		39.3 20.3	30.3	38.3	40.4	48.9	49.2	8.83	52.3	52.3	8.83	8.60	52.6	43.3	46.0	50.2	47.0		9.04	54.0	54.0	9.04	32.9	32.9	44.6	45.2	42.5	46.2	39.3	36.8	48.4	43.3	0.01	49.0	44.4	45.7	36.4	36.4	39.1	46.5	45.2	45.2
$E_{\rm u}$ (K)		242.4	342.4	308.8	300.4	293.7	392.4	296.0	296.0	296.0	296.0	394.5	394.5	284.0	388.8	295.3	291.5		297.6	297.6	297.6	297.6	351.5	351.5	286.0	395.1	408.2	297.8	330.1	340.8	300.8	401.8	0 000	0.700	305.0	397.1	362.4	362.4	448.7	299.9	419.7	419.2
Calc. freq. (MHz)		204 024.731 205 072 338	205 972.338	206 865.964	207 553.192	210097.941	210345.827	211 082.205	211 085.095	211 087.344	211 090.234	211 134.083	211 136.716	211 372.277	211 377.266	213 261.995	213 882.009		214 193.599	214 195.705	214 197.429	214 199.535	214 572.059	214572.060	215 322.039	215 326.475	217 440.211	218060.270	218648.438	219017.617	219 092.328	219883.920		601.001 222	224 161.641	225 917.084	226614.150	226614.152	227 459.769	229 246.345	229 294.250	229 951.441
p'			+	+	I	I		+	+	+	+			Ι		Ι	+						Ι	+				I			+						Ι	+		+		
K_c'	Ţ	11	9	10	11	17	17	19	19	19	19	19	19	13	15	17	15	(-	19	19	19	19	~	Г	13	14	12	4	6	6	16	12	21	10	12	13	6	×	10	13	12	13
K_a'	`	0 0	10	Г	9	0	4	-	-	0	0	ī	ī	ŝ	0	-	0		ī	-	0	0	10	10	ŝ	4	9	4	×	6-	ŝ	S	ſ	n N	S	4	10	10	6-	4	9	9-
J'		10	16	16	16	18	18	19	19	19	19	19	19	16	17	18	17	(-	19	19	19	19	17	17	16	17	17	17	17	17	18	17	01	10	17	17	18	18	18	17	18	18
d			+	+	Ι	Ι		+	+	+	+			Ι		I	+						Ι	+				Ι			+						Ι	+		+		
K_c	ç	<u>7</u> ×		11	12	18	18	20	20	20	20	20	20	14	16	18	16	0	20	20	20	20	6	×	14	15	13	15	10	10	17	13	1	1/	13	14	10	6	11	14	13	14
K_a	`	0 0	10	2	9	0	4	0	-	0	-	0	Τ	З	0	1	0	¢	0	Τ	0	Τ	10	10	З	4	φ	4	×	6	З	S	¢	ſ	2	4	10	10	6-	4	9	9
J	ţ	17	17	17	17	19	19	20	20	20	20	20	20	17	18	19	18	Ċ	20	20	20	20	18	18	17	18	18	18	18	18	19	18	0	IУ	18	18	19	19	19	18	19	19
Spec.	- - -	A- ¹³ C ²	$A^{-13}C_1$	$A^{-13}C_1$	$A^{-13}C_{1}$	$A^{-13}C_2$	$E^{-13}C_2$	$A^{-13}C_{2}$	$A^{-13}C_2$	$A^{-13}C_2$	$A^{-13}C_2$	$E^{-13}C_2$	$E^{-13}C_2$	$A^{-13}C_2$	$E^{-13}C_2$	$A^{-13}C_{1}$	$A^{-13}C_{1}$	2 2 1	E-C-	E- ^{L2} C1	$E^{-13}C_{1}$	$E^{-13}C_{1}$	$A^{-13}C_2$	$A^{-13}C_2$	$E^{-13}C_{1}$	$E^{-13}C_2$	$E^{-13}C_2$	$A^{-13}C_{1}$	$E^{-13}C_1$	$E_{13}C_{1}$	$A^{-13}C_2$	$E^{-13}C_2$	13.2 1		E- ¹³ C1	$E^{-13}C_2$	$A^{-13}C_2$	$A^{-13}C_2$	$E^{-13}C_2$	$A^{-13}C_{1}$	$E^{-13}C_2$	$E^{-13}C_2$

Blend	U																CH ₃ CH ₂ OCOH		$CH_3CH_2^{13}CN$ $H^{13}COOCH_3$								CH ₃ COOCH ₃															
$T^{ m mod}_{ m mb}({ m K})$	0.10	0.14		0.12		0.21				0.10	0.24				0.07	0.10	0.10	0.11	0.21	0.13		0.26				0.08	0.08	0.09	0.07				0.06	0.11	0.14			0.10	0.10		0.10	0.11
${T_{ m mb}^{ m obs}}{({ m K})}$	0.21	c1.0		0.15		0.22				0.29	0.35				0.24	0.24	0.12	0.21	0.68	0.13		0.44				0.18	0.14	0.15	0.15				0.22	0.15	0.07			0.43	0.18		0.14	0.10
$v_{\rm LSR}$ $(\rm km~s^{-1})$	5.7 ± 0.7	6.8 ± 0.7	0.0 ± 0.7 8.0 ± 0.7	7.7 ± 0.7	7.9 ± 0.7	5.2 ± 0.6	6.3 ± 0.6	7.3 ± 0.6	8.4 ± 0.6	6.2 ± 0.6	6.2 ± 0.6	7.0 ± 0.6	7.7 ± 0.6	8.5 ± 0.6	7.2 ± 0.6	5.2 ± 0.6	7.7 ± 0.6	8.5 ± 0.6	7.5 ± 0.6	7.1 ± 0.6	7.8 ± 0.6	7.6 ± 0.6	8.5 ± 0.6	7.9 ± 0.6	8.2 ± 0.6	6.3 ± 0.6	4.8 ± 0.6	6.4 ± 0.6	6.7 ± 0.6	6.7 ± 0.6	6.7 ± 0.6	6.7 ± 0.6	7.4 ± 0.6	6.8 ± 0.6	7.1 ± 0.6	7.9 ± 0.6	8.1 ± 0.6	6.2 ± 0.6	7.3 ± 0.6	7.3 ± 0.6	7.5 ± 0.6	7.4 ± 0.6
Obs. freq. (MHz)	230 064.6	230 100.8		230 719.6	÷	231 709.6	+	÷	+	234 054.0	235 114.0	+	+	+	240 213.9	241 082.0	241 373.5	244 774.9	244 799.9	251 115.9	+-	252 327.2	-;	-;	+	252 689.8	252 704.8	253 343.5	254 444.7	+	÷	÷	254 625.8	254 899.8	255 417.1	-;	-;	261 016.8	262 404.6		263 025.8	263 484.6
${S_{ij}\mu^2 \over ({ m D}^2)}$	46.8	23.6	46.0	40.1	40.1	9.76	57.7	57.7	9.76	52.3	10.0	59.4	59.4	10.0	44.6	56.9	51.9	58.6	58.6	45.4	45.4	10.7	10.7	63.0	63.0	49.4	49.4	51.0	25.5	25.5	6.22	6.22	44.1	58.0	46.5	2.44	46.5	52.3	40.9	40.9	56.1	56.1
$E_{\rm u}$ (K)	300.4	427.4	427.4 412.2	351.4	351.4	317.7	317.7	317.7	317.7	313.5	319.7	319.7	319.7	319.7	449.0	327.0	320.9	329.3	329.3	373.3	373.3	341.5	341.5	341.5	341.5	352.5	451.8	344.1	451.2	451.2	451.2	451.2	388.9	337.9	375.3	242.4	375.3	338.3	427.7	427.7	342.3	441.3
Calc. freq. (MHz)	230 063.581	230100.629	230 100.029	230720.161	230720.299	231 708.149	231 709.053	231709.768	231710.673	234 053.346	235 113.385	235 114.014	235 114.538	235 115.167	240 214.055	241 080.593	241 374.048	244 776.097	244 800.362	251 115.944	251 116.550	252 327.721	252 328.505	252 328.001	252 328.225	252 689.210	252703.017	253 342.936	254 444.432	254 444.432	254 444.432	254 444.432	254 626.092	254 899.597	255417.206	255 417.848	255 418.028	261 016.105	262404.906	262 404.906	263 026.207	263 484.906
p'		-	ł	+	Ι	+	+	+	+	+						I				+	Ι	+	+	+	+	Ι		I	I	+	I	+			+	+	Ι	Ι	+	I	Ι	
K_c'	13	∩ <i>-</i>	1 T	10	6	21	21	21	21	17	21	21	21	21	12	20	16	20	20	12	11	23	23	23	23	13	13	15	L	9	٢	9	10	19	12	9	11	15	6	10	18	18
K_a'	4;	4 -	- 1 1	6	6	1	-	0	0	0	ī	ī	0	0	ŝ	-	4	c_1^-	1	6	6	-	0	-	0	L	L	9	14	14	14	14	10	0	6	4	6	S	12	12	ß	ю
J'	17	<u>1 0</u>	0 81	18	18	21	21	21	21	19	21	21	21	21	19	21	19	21	21	20	20	23	23	23	23	20	20	20	20	20	20	20	20	21	20	10	20	20	21	21	21	21
d		-	ł	+	Ι	+	+	+	+	+						Ι				+	I	+	+	+	+	Ι		I	Ι	+	+	Ι			+	+	Ι	Ι	+	I	I	
K_c	14	0 V	0 15 15	11	10	22	22	22	22	18	22	22	22	22	13	21	17	21	21	13	12	24	24	24	24	14	14	16	×	Г	Г	×	11	20	13	Г	12	16	10	11	19	19
K_{a}	4;	4 -	 	6	6	0		0		0	0	ī	0	Ξ	ŝ		4	-2^{-}	1	6	6	0	-		0	2	Г	9	14	4	4	4	10	0	6	2	6	S	12	12	ю	ю
ſ	18	61 01	19	19	19	22	22	22	22	20	22	22	22	22	20	22	20	22	22	21	21	24	24	24	24	21	21	21	21	21	21	21	21	22	21	11	21	21	22	22	22	22
Spec.	$E^{-13}C_1$	A- ¹³ C1	E- ¹³ C,	$A^{-13}C_1$	$A^{-13}C_{1}$	$A^{-13}C_{2}$	$A^{-13}C_{2}$	$A^{-13}C_{2}$	$A^{-13}C_{2}$	$A^{-13}C_{1}$	$E^{-13}C_{1}$	$E^{-13}C_{1}$	$E^{-13}C_{1}$	$E^{-13}C_{1}$	$E^{-13}C_{2}$	$A^{-13}C_{2}$	$E^{-13}C_1$	$E^{-13}C_{1}$	$E^{-13}C_{1}$	$A^{-13}C_{2}$	$E^{-13}C_{2}$	$A^{-13}C_2$	$A^{-13}C_{1}$	$A^{-13}C_{1}$	$A^{-13}C_{1}$	$A^{-13}C_{1}$	$E^{-13}C_{1}$	$E^{-13}C_{1}$	$A^{-13}C_{1}$	$A^{-13}C_{1}$	$A^{-13}C_{1}$	$A^{-13}C_2$	$A^{-13}C_{2}$	$A^{-13}C_{2}$	$A^{-13}C_{2}$	$E^{-13}C_{2}$						

Blend			DCOOCH ₃	CH ₃ COOCH ₃															C					CH ₃ OCH ₃			CH ₃ CH ₂ OH				
T^{mod}_{mb} (K)	0.12		0.11	0.08	0.29				0.13		0.11	0.10	0.27				0.05		0.06		0.11	0.12		0.13		0.11	0.29				0.11
$\overset{Tobs}{\overset{\mathrm{obs}}{(\mathrm{K})}}$	0.13		0.11	0.14	0.35				0.19		0.17	0.15	0.17				0.11		0.21		0.23	0.24		0.35		0.24	0.31				0.22
$v_{\rm LSR}$ (km s ⁻¹)	7.0 ± 0.6	8.1 ± 0.6	8.5 ± 0.6	6.8 ± 0.6	6.3 ± 0.6	6.2 ± 0.6	6.0 ± 0.6	6.1 ± 0.6	6.9 ± 0.6	7.0 ± 0.6	5.6 ± 0.6	7.3 ± 0.6	7.0 ± 0.5	7.1 ± 0.5	6.9 ± 0.5	7.0 ± 0.5	7.0 ± 0.5	7.0 ± 0.5	5.4 ± 0.5	5.4 ± 0.5	6.6 ± 0.5	7.3 ± 0.5	7.3 ± 0.5	7.7 ± 0.5	7.9 ± 0.5	6.5 ± 0.5	8.5 ± 0.5	8.6 ± 0.5	8.5 ± 0.5	8.7 ± 0.5	7.9 ± 0.5
Obs. freq. (MHz)	264936.9	+	265442.1	265640.8	266483.3				267 355.8		270 152.0	272723.7	272941.8				274077.0	+	274 125.7		274 173.3	274 666.7		274 998.3		276849.6	276934.7				280722.7
$S_{ij\mu}^2$ (D ²)	60.7	3.22	63.8	53.7	11.5	67.6	11.5	67.6	47.3	47.3	58.5	58.5	68.3	11.7	11.7	68.3	31.4	31.4	35.1	35.1	60.0	47.0	47.0	49.4	49.4	60.1	70.3	70.3	12.0	12.0	61.1
$E_{\rm u}$ (K)	350.6	236.2	354.0	356.9	356.6	356.6	356.6	356.6	400.6	400.6	454.7	355.4	367.2	367.2	367.2	367.2	515.3	515.3	494.7	494.7	358.8	425.6	425.6	411.7	411.7	358.5	369.9	369.9	369.9	369.9	468.2
Calc. freq. (MHz)	264 936.868	264 937.868	265 443.450	265 640.606	266482.702	266482.602	266482.416	266482.517	267 355.733	267 355.808	270 150.708	272 723.937	272941.805	272941.891	272 941.649	272 941.735	274 076.953	274 076.953	274 124.220	274 124.220	274 172.892	274 666.966	274 666.961	274 998.925	274 999.054	276849.213	276 936.143	276936.189	276936.089	276936.244	280723.536
p'			I	T					Ι	+		Ι	+	+	+	+	I	+	Ι	+		Ι	+	I	+						
K_c'	20	-	22	16	24	24	24	24	12	11	19	19	25	25	25	25	2	9	2	×	19	11	12	13	12	19	25	25	25	25	20
K_a'	- 1	9	0	9	0	0	Ţ	Ţ	10	10	4	С	0	0			16	16	15	15	4	11	11	10	10	С	Ţ	0	Ţ	0	4
J'	22	9	23	21	24	24	24	24	21	21	22	22	25	25	25	25	22	22	22	22	22	22	22	22	22	22	25	25	25	25	23
d			I	Ι					Ι	+		Ι	+	+	+	+	Ι	+	Ι	+		Ι	+	Ι	+						
K_c	21	-	23	17	25	25	25	25	13	12	20	20	26	26	26	26	8	2	8	6	20	12	13	14	13	20	26	26	26	26	21
K_{a}	-3	L-	0	9	-	0	0	Ţ	10	10	4	С	0	-	0	-	16	16	15	15	4	11	11	10	10	З	Ţ	0	0	Ţ	4
ſ	23	2	24	22	25	25	25	25	22	52	23	23	26	26	26	26	23	23	23	23	23	23	23	23	23	23	26	26	26	26	24
Spec.	$E^{-13}C_{1}$	$E^{-13}C_1$	$A^{-13}C_{1}$	$A^{-13}C_2$	$E^{-13}C_1$	$E^{-13}C_1$	$E^{-13}C_1$	$E^{-13}C_1$	$A^{-13}C_1$	$A^{-13}C_{1}$	$E^{-13}C_2$	$A^{-13}C_2$	$E^{-13}C_1$	$A^{-13}C_{2}$	$A^{-13}C_2$	$A^{-13}C_{2}$	$A^{-13}C_2$	$E^{-13}C_1$	$E^{-13}C_1$	$E^{-13}C_1$	$E^{-13}C_1$	$E^{-13}C_{1}$	$E^{-13}C_2$								



Fig. 2. Selected lines of A/E-H¹³COOCH₃ $v_t = 1$ (modelled in red) and A/E-HCOO¹³CH₃ $v_t = 1$ (modelled in dark blue) toward Orion-IRc2. The continuous cyan line corresponds to all lines already modelled in our previous papers except H¹³COOCH₃ and HCOO¹³CH₃ $v_t = 1$ (see Sect. 5.1). A v_{LSR} of 7 km s⁻¹ is assumed.

has to be considered as the total intensity of the detected feature and an upper limit for the intensity of the methyl formate species in this study. Nevertheless, in Table 5 we found some values of the $T_{\rm mb}^{\rm obs}$ lower than those predicted by the model (synthetic spectrum HCOO¹³CH₃ and H¹³COOCH₃ $\nu_t = 1$); uncertainties in the removed baselines are the most probable source for these discrepancies. The uncertainty in the radial velocity was adopted from the spectral resolution of our data. Most unblended detected lines show a radial velocity of $\simeq 7 \,\mathrm{km \, s^{-1}}$, in agreement with the expected velocity for emission from the compact ridge component where organic saturated O-rich molecules have the largest abundances inside the region. In addition, most detected lines of methyl formate, its isotopologues, and its first vibrationally excited state appear at the same radial velocity (see e.g. Carvajal et al. 2009; Margulès et al. 2010; Favre et al. 2011; Tercero et al. 2012; López et al., in prep.).

5.3. Modelling the data

To model the emission of H¹³COOCH₃ and HCOO¹³CH₃ $v_t = 1$ we used an excitation and radiative transfer code: MADEX (Cernicharo 2012). This time we applied LTE conditions because we lacked collisional rates for these species. Two compact ridge components ($v_{LSR} = 7 \text{ km s}^{-1}$ and $\Delta v = 3 \text{ km s}^{-1}$) are enough to model the line profiles: one at 110 K, with diameter of 15", and a column density of (6 ± 3) × 10¹³ cm⁻² for each state (A and E) of each isotopologue, and a hotter and inner compact ridge at 250 K with a diameter of 10", and a column



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Fig. 3. Selected lines of A/E-HCOOCH₃ $v_t = 1$ (modelled in dark blue) toward Orion-IRc2. The dashed red line and the green continuous line correspond to methyl formate in the ground state and the sum of the emission of $v_t = 1$ and $v_t = 0$, respectively. The continuous cyan line corresponds to all lines already modelled in our previous papers except HCOOCH₃ $v_t = 0$, 1 (see Sect. 5.1). A v_{LSR} of 9 km s⁻¹ is assumed. The *x*-axis is the frequency in GHz.

density of $(6 \pm 3) \times 10^{14}$ cm⁻² for each state (A and E) of each isotopologue (both components are 7" offset from the pointing position; interferometric maps show this component offset $\simeq 7''$ southwest from the source I – located 0.5'' south of IRc2, Menten & Reid 1995; see e.g. Neill et al. 2013 for ALMA cycle 0 maps). The beam dilution for each line (depending on the frequency) has been also taken into account in the calculation of the emerging line intensities. The obtained synthetic spectra is shown in Fig. 2. We obtained source averaged-column densities over the source size given for each component. Uncertainties of the column density results were estimated to be between 20-30% for this line survey (see Tercero et al. 2010) taking into account errors introduced by different sources, such as the spatial overlap of the different cloud components, the modest angular resolution, or pointing errors. Nevertheless, high overlap problems add another source of uncertainty (rising to 50%) for results obtained by means of weak lines such as those of H¹³COOCH₃ and HCOO¹³CH₃ v_t = 1. All together we obtained a total column density of $(1.4 \pm 0.7) \times 10^{15} \text{ cm}^{-2}$ for each ¹³C isotopologue of vibrationally excited methyl formate.

5.3.1. Isotope ratios

To obtain the ${}^{12}\text{C}/{}^{13}\text{C}$ ratio, we included our model for A/E-HCOOCH₃ $v_t = 1$ (López et al., in prep.). Figure 3 shows selected lines of methyl formate in its first vibrationally excited state in the 2 mm window (from 130 to 178 GHz). We modelled these species assuming all components listed in Table 4 and the hotter and inner compact ridge described above. The differences between the considered components with respect to the model of the ${}^{13}\text{C}$ isotopologues arise because the much more stronger lines of the main isotopologue in all the spectral band (from 80 to 280 GHz) allow us to distinguish the contribution of several cloud components in the line profiles, although some of them, such as the plateau contribution, are very low. For

HCOOCH₃ $v_t = 1$ we obtained a total column density of $(1.9 \pm 0.6) \times 10^{16} \text{ cm}^{-2}$.

The column density ratio between the two vibrationally excited isotopologues (main and ¹³C) of methyl formate yields a ¹²C/¹³C ratio between 6–35 (considering the error bars and assuming the same partition function for both species). Taking into account the weakness of the ¹³C lines, the derived ratios are compatible with the value of ¹²C/¹³C \approx 35 obtained with the column density ratios between ¹³C-HCOOCH₃ and HCOOCH₃ in the ground state (see Carvajal et al. 2009; Margulès et al. 2010) and agrees quite well with previous results of this ratio in Orion KL (Johansson et al. 1984; Blake et al. 1987; Demyk et al. 2007; Tercero et al. 2010; Daly et al. 2013; Esplugues et al. 2013b).

5.3.2. Vibrational temperatures

We can estimate vibrational temperatures from

$$\frac{\exp\left(-\frac{E_{v_t=1}}{T_{vib}}\right)}{f_v} = \frac{N({}^{13}\text{C} - \text{HCOOCH}_3 \quad v_t = 1)}{N({}^{13}\text{C} - \text{HCOOCH}_3)},$$
(1)

where $E_{\nu,=1}$ is the energy of the vibrational state (187.6 K), $T_{\rm vib}$ is the vibrational temperature, f_{ν} is the vibrational partition function, $N({}^{13}\text{C}\text{-HCOOCH}_3 v_t = 1)$ is the column density of the vibrational state, and $N({}^{13}\text{C}\text{-HCOOCH}_3)$ is the total column density of ${}^{13}\text{C}$ methyl formate. Taking into account that $N({}^{13}\text{C}\text{-HCOOCH}_3) = N(\text{ground}) \times f_{\nu}$ and assuming the same partition function for these species in the ground- and the first vibrationally excited states, we only need the energy of the vibrational state and the column densities of ${}^{13}\text{C}\text{-HCOOCH}_3$ $v_t = 1$ and ${}^{13}\text{C}\text{-HCOOCH}_3 v_t = 0$ to derive the vibrational temperatures.

In Carvajal et al. (2009) we did not consider the hotter component of the compact ridge for column density calculations. After several works on methyl formate (Carvajal et al. 2009; Margulès et al. 2010; Tercero et al. 2012; Coudert et al. 2013; López et al., in prep.) we realize that this hotter component plays an important role in the global analysis of this molecule. In Margulès et al. (2010) we introduced a hot compact ridge to model the lines of main isotopologue of the methyl formate to properly reproduce the line profiles. After that, the study of the vibrationally excited states (this work and A. López et al., in prep.) confirms that a hot compact ridge is needed to understand the emission of this molecule. Therefore, quite strong lines such as those of ¹³C-HCOOCH₃ require the hot compact ridge component to obtain the best fit. Consequently, here we again modeled the emission from H¹³COOCH₃ and HCOO¹³CH₃ taking into account a compact ridge at a temperature of 250 K and a source size of 10". As in Carvajal et al. (2009), all cloud components of Table 4 where taken into account to reproduce the line profiles. Using MADEX and assuming LTE conditions, the obtained column densities for each ¹³C isotopologue and each state (A and E) are $(2.0 \pm 0.6) \times 10^{15}$ cm⁻² and $(2.0 \pm 0.6) \times 10^{14}$ cm⁻² for the T = 250 K compact ridge and the T = 110 K compact ridge, respectively, and $(1.0 \pm 0.3) \times 10^{13} \text{ cm}^{-2}$ for the plateau, the extended ridge, and the hot core. These values yield a total column density of a factor 3 higher than obtained in Carvajal et al. (2009). This difference is mostly due to the reduced source size when we introduced the hotter component of the compact ridge.

At this point, we were ready to derive the vibrational temperatures. We assumed that both gases (ground-state and vibrationally excited) are spatially coincident, so the calculated vibrational temperatures have to be considered as lower limits. A $T_{\rm vib} = 156 \,\mathrm{K}$ was obtained for H¹³COOCH₃ and HCOO¹³CH₃ $v_t = 1$ in each compact ridge component. This value is similar to the averaged kinetic temperature we adopted in this model (180 K). This result indicates that collisions are probably the main mechanism to populate the vibrationally excited levels of methyl formate.

6. Conclusion

We measured the spectra of HCOO¹³CH₃ from 75 to 940 GHz. We analyzed 2881 lines from the first excited torsional states, 5728 new lines corresponding to the ground-vibrational states were also measured and added to the previous ones from Carvajal et al. (2009). The global fit of the 9455 lines, taking into account the internal rotation motion, was made using the rho-axis-method and the BELGI code. Fifty-two parameters could be determined, the global fit of both states permitted us to obtain uncorrelated values of the torsional parameters: F, V_3 , and V_6 . The value of barrier height ($V_3 = 369.93168(395) \text{ cm}^{-1}$) has been significantly improved.

Owing to this work and Carvajal et al. (2009), accurate predictions of line positions and intensities were performed. In the line survey of Orion-KL with the IRAM 30 m telescope, this permits detecting 135 spectral features in the range 80–280 GHz; there are no missing transitions in this range. This is the first detection of an excited vibrational state of a methyl formate isotopologue.

At the beginning of the line identification process of this line survey (in 2005), nearly 8000 spectral features were U lines. In 2007 we began a close collaboration between astronomers and spectroscopists to reduce the uncertainty due to unidentified lines in spectral surveys. New isotopologues of ethyl cyanide and methyl formate (Demyk et al. 2007; Margulès et al. 2009, 2010; Carvajal et al. 2009; Tercero et al. 2012; Coudert et al. 2013) and new vibrationally excited states of formamide (Motiyenko et al. 2012), ethyl cyanide (Daly et al. 2013), and vinyl cyanide (López et al. 2014) have been studied in spectroscopy laboratories and detected in Orion KL for the first time space. Altogether, we reduced 3000 of unidentified lines through these studies. At this point, we were ready to search for new molecular species, detecting methyl acetate and gauche-ethyl formate (Tercero et al. 2013) and *qauche*-ethyl mercaptan (Kolesniková et al. 2014) for the first time in space, and providing the tentative detection of phenol (Kolesniková et al. 2013). These new species together with the work of López et al. (in prep., about vibrationally excited CH₃OCOH and CH₃COOH) account for $\simeq 1000$ lines. Up to date, we have reduced 4000 of U lines in the survey of Orion KL. We expect that a large number of the still unidentified lines arise from vibrationally modes of abundant species. However, low-lying vibrationally excited states of abundant molecules such as ethyl cyanide (for example the $v_{13} = 2/v_{21} = 2$ state at ~420 cm⁻¹) or methanol are still only poorly characterized in the laboratory, and certainly contribute to the remaining U-lines. The main goal of our studies is to provide the community with a fully analysed line survey of Orion that will constitute a template for ALMA observations of warm clouds.

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Appendix A: Part of the supplementary tables available at the CDS

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~~	\mathbf{K}_{c}'	þ,	v''_i	J.,,	K_a''	K_c''	<i>"</i> ,"	Obs. Freq. $(Unc.)^{p}$	Calc. Freq (Unc.) ^c	$ObsCalc^{d}$	$S(i \to f)^e$	Lower energy ¹	Ref^g
	50	+		49	0	49	+	519745.410(30)	519745.377(5)	0.033	137.102	560.0328	NEW B
	51	+	1	50	0	50	+	530 000.398(30)	530000.382(5)	0.016	22.972	577.3696	NEW B
	52	+	1	51	0	51	+	540 252.626(30)	540 252.652(5)	-0.026	153.599	595.0485	NEW B
	53	+	1	52	0	52	+	550 502.111(30)	550502.133(6)	-0.022	154.358	613.0694	NEW B
	54	+	1	53	0	53	+	560748.773(30)	560748.773(6)	0.000	137.761	631.4322	NEW B
	55	+	1	54	0	54	+	570 992.521(30)	570 992.518(7)	0.003	148.615	650.1368	NEW B
	56	+	1	55	0	55	+	581 233.298(30)	581 233.318(7)	-0.020	35.289	669.1830	NEW B
	57	+	1	56	0	56	+	591 471.082(30)	591471.118(8)	-0.036	181.673	688.5709	NEW B
	58	+	1	57	0	57	+	601 705.819(30)	$601\ 705.866(9)$	-0.047	184.679	708.3002	NEW B
	59	+	1	58	0	58	+	611 937.436(30)	611937.510(10)	-0.074	187.823	728.3710	NEW B
	60	+	1	59	0	59	+	622 165.944(30)	622 165.996(11)	-0.052	190.689	748.7830	NEW B
	50		1	49	0	49		519732.280(30)	519732.271(5)	0.009	159.484	559.5257	NEW B
	51		1	50	0	50		529 985.311(30)	529 985.287(6)	0.024	161.679	576.8621	NEW B
	52		1	51	0	51		540 235.593(30)	540 235.566(6)	0.027	122.669	594.5405	NEW B
	53		1	52	0	52		550483.081(30)	550483.060(6)	0.021	114.755	612.5608	NEW B
	54		1	53	0	53		560727.716(30)	560727.709(6)	0.007	172.372	630.9230	NEW B
	55		1	54	0	54		570 969.450(30)	570 969.463(7)	-0.013	175.587	649.6268	NEW B
	56		1	55	0	55		581 208.238(30)	$581\ 208.268(7)$	-0.030	178.795	668.6723	NEW B
	57		1	56	0	56		591 444.057(30)	591444.072(8)	-0.015	182.011	688.0593	NEW B
	58		1	57	0	57		601 676.811(30)	$601\ 676.819(9)$	-0.008	185.223	707.7878	NEW B
	59		1	58	0	58		611 906.456(30)	611906.456(9)	0.000	188.431	727.8576	NEW B
	60		1	59	0	59		622 132.891(30)	622 132.935(10)	-0.044	191.656	748.2686	NEW B

Notes. ^(a) Upper and lower state quantum numbers are indicated by ' and " respectively. Torsion-rotation levels of A species have a "parity" label; levels of E species have a signed K_a value (Herbst et al. 1984). ^(b) Observed $v_t = 0$ and 1 microwave transitions in MHz, with estimated uncertainties in parentheses (in kHz). ^(c) Calculated line frequency in MHz with calculated uncertainty in kHz. ^(d) Differences among the experimental and computed frequencies. ^(e) Calculated line strengths in D**2 (for details of the calculation procedure, see Sect. 4). ^(f) Lower state energy (zn^{-1}) referred to the $J = K_a = 0$ A-species energy level taken as the zero of the energy (zero-point torsional energy: 71.0704 cm⁻¹). ^(g) Sources of data are explained in footnote b of Table 1. Blended lines are indicated with a capital letter B.

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